IN THE UNITED STATES DISTRICT COURT FOR THE DISTRICT OF DELAWARE

MEDA PHARMACEUTICALS, INC. and CIPLA LTD.)))
Plaintiff,)))
v.)
APOTEX, INC. and APOTEX CORP.)

Civil Action No. 1:14-cv-01453-LPS

Defendant.

EXPERT REPORT OF RAMPRAKASH GOVINDARAJAN, PH.D.



I. INTRODUCTION

1. My name is Ramprakash Govindarajan, Ph.D. As detailed in my current *curriculum vitae*, which is attached to this declaration as Appendix A, I became Director of Preformulation at the University of Iowa Pharmaceuticals in 2015. I am also a Clinical Assistant Professor at the College of Pharmacy, University of Iowa.

I obtained my Ph.D in Pharmaceutics from the University of Mumbai in 2002. After that,
 I was a Research Associate in Pharmaceutics at the University of Minnesota until 2006. From
 2006 to 2015, I worked in various product development roles at GlaxoSmithKline.

3. Throughout my career, I have had experience conducting experiments similar to the one described in this report, including making pharmaceutical formulations.

4. I have never testified as an expert witness or prepared an expert report in connection with litigation. A list of all my publications from the last 18 years is included in my *curriculum vitae*.

5. I have rendered all services in this case as an independent consultant. Apotex has compensated me for these services at in the amount of \$8,500. If additional testimony is required of me, I will be compensated at a rate of \$250/hour. My compensation does not depend on the the opinions formed in this Report or the outcome of this litigation.

6. If called to testify, I am prepared to testify regarding the results of my experiment as reported in this report and the attached documents. I am also prepared to testify about the methods I used and the research I conducted in obtaining those results.

II. SUMMARY OF WORK

7. Apotex retained me to formulate the composition in "Example III" of EP 0780127 ("Cramer") (attached as Appendix B). To make the formulation, I was directed to follow any instructions or guidance included in the document. To the extent the Cramer reference did not

expressly provide information I needed to formulate, I was directed to proceed as an ordinarily skilled pharmaceutical formulator would have proceeded in June 2002 after reading Cramer. I was directed not to take any steps or use any equipment that was not known and readily available to an ordinarily skilled pharmaceutical formulator in 2002. Cramer was the only document given to me by Apotex's counsel.

8. I was not told what outcome(s) to try to obtain, or what outcome would be helpful to Apotex or their opponent in this litigation. I have not reviewed any patent asserted in this litigation.

9. If I was able to make a pharmaceutical formulation as described in Example III in Cramer, I was asked to make the following observations:

- a. Observe whether Example III forms a suspension, solution, or something else.
 - i. If Example 3 does form a suspension, observe whether that suspension settles.
 - 1. If it settles, observe whether it is difficult to re-suspend.
 - 2. If it settles, opine on whether the settling is unacceptable for a pharmaceutical product.
- b. Measure the osmolality in mOsm/kg.
- c. Put the resulting composition in a nasal pump with a suitable actuator. Observe if the composition comes out as a spray, a jet, or something else.

10. I was asked to record my work and the above observations/experimental results.

III. RESULTS AND OBSERVATIONS

11. The laboratory notebook entries reflecting the work I did are attached as Exhibit C.

12. I prepared three batches of the formulation in Example III during the formulation process. I conducted my final testing and made my final observations using Batch 1345-011. My laboratory notebook records the steps I took to make each batch, as well as the ingredient and equipment I used. I am prepared to discuss each of those steps if asked to do so, and I incorporate the contents of my lab notebook to this report as if set forth herein.

13. I concluded that Cramer Example III is a suspension that would be acceptable as a pharmaceutical product. There was some settling, but no settling or sedimentation in the product that would make it pharmaceutically unacceptable.

14. I further concluded that the product could be delivered as a fine spray using a nasal spray pump. I took two videos of the mist produced by Batch 1345-011 when actuated in a nasal pump with a suitable actuator. Those videos are attached as Appendix D and Appendix E.

15. I also took osmolality measurements as requested. I determined that Batch 1345-011 had an osmolality of 529 mOsm/kg.

IV. DECLARATION

I declare under penalty of perjury under the laws of the United States that the foregoing is true and correct to the best of my knowledge.

Date: 29 - June - 2016

Signed: R. Ka-prokel

Dr. Ramprakash Govindarajan

EXHIBIT A

Ramprakash (Ram) Govindarajan, PhD Iowa City, Iowa, USA Work: +1-319-335-6339 E-mail: ramprakash-govindarajan@uiowa.edu

PROFESSIONAL APPOINTMENTS

UNIVERSITY OF IOWA, COLLEGE OF PHARMACY (2015 – PRESENT)

- Director, Preformulation, University of Iowa Pharmaceuticals
- Clinical Assistant Professor, College of Pharmacy, University of Iowa

GLAXOSMITHKLINE, RESEARCH TRIANGLE PARK (2006 - 2015)

• Manager (2013 - 2015); Investigator (2009 – 2013); Principal Scientist (2006 – 2009) Served in roles of increasing responsibility in various drug product development areas including early developability assessment, clinical phase pharmaceutical product development, manufacturing process development, technical transfer and scale-up

UNIVERSITY OF NORTH CAROLINA, CHAPEL HILL

Clinical Assistant Professor (Adjunct appointment, 2007-2015)

- Taught topics included in course MOPH-862 to graduate students in Pharmaceutics
- Served on PhD dissertation committee

ACADEMIC BACKGROUND

- Research Associate, Pharmaceutics, University of Minnesota, 2002-2006
- Ph. D. (Pharmaceutics), University of Mumbai, 1998-2002
- M. Pharm. Sci. (Pharmaceutics), University of Mumbai, 1995-1997
- B. Pharm. Sci., University of Mumbai, 1991-1995

AFFILIATIONS

- Member of American Association of Pharmaceutical Scientists (AAPS)
- Scientific Reviewer for Journals 'Pharmaceutical Research', 'Journal of Pharmaceutical Sciences' and 'The Journal of Pharmacy and Pharmaceutical Sciences'

PUBLICATIONS /PRESENTATIONS

Journal Articles

- Govindarajan, R.; Landis, M.; Hancock, B.; Gatlin, L.; Suryanarayanan, R.; Shalaev, E.; (2014), Surface Acidity and Solid-State Compatibility of Excipients with an Acid-Sensitive API: Case Study of Atorvastatin Calcium, AAPS PharmSciTech., 1-10.
- 2. Chakravarty, P.; Suryanarayanan, R.; Govindarajan, R.; (2012), Phase transformation in thiamine hydrochloride tablets: influence on tablet microstructure, physical properties, and performance, J. Pharm. Sci. 101 (4), 1410-1422.
- Chakravarty, P.; Govindarajan, R.; Suryanarayanan, R.; (2010), Investigation of solution and vapor phase mediated phase transformation in thiamine hydrochloride, J. Pharm. Sci. 99(9), 3941-3952.
- Chakravarty, P.; Berendt, R.; Munson, E.; Young, V; Govindarajan, R.; Suryanarayanan, R.; (2009), Insights into the dehydration behavior of thiamine hydrochloride (vitamin B1) hydrates: Part II, J. Pharm. Sci. 99(4), 1882 - 1895.
- Chakravarty, P.; Berendt, R.; Munson, E.; Young, V; Govindarajan, R.; Suryanarayanan, R.; (2009), Insights into the dehydration behavior of thiamine hydrochloride (vitamin B1) hydrates: Part I, J. Pharm. Sci. 99(2), 816-827.
- Chatterjee, K.; Shalaev, E.Y.; Suryanarayanan, R.; Govindarajan, R. (2008), Correlation between chemical reactivity and the Hammett acidity function in amorphous solids using inversion of sucrose as a model reaction, J. Pharm. Sci. 97(1), 274-286.
- Govindarajan, R.; Zinchuk, A.V.; Hancock, B.C.; Shalaev, E.Y.; Suryanarayanan, R., (2006), Ionization states in the microenvironment of solid dosage forms: Effect of formulation variables and processing, Pharm. Res., 23(10), 2454-2468.
- 8. Rani, M; Govindarajan, R.; Surana, R.; Suryanarayanan, R. (2006), Structure in dehydrated trehalose dihydrate evaluation of the concept of partial crystallinity, Pharm. Res., 23(10), 2356-2367.
- Govindarajan, R.; Chatterjee, K.; Gatlin, L.; Suryanarayanan, R.; Shalaev, E.Y. (2006), Impact of freeze-drying on ionization of sulfonephthalein probe molecules in trehalose-citrate systems, J. Pharm. Sci. 95(7), 1498–1510.
- Zinchuk, A.V.; Hancock, B.C.; Shalaev, E.Y.; Reddy, R.D.; Govindarajan, R.; Novak, E. (2005), The influence of measurement conditions on the Hammett acidity function of solid pharmaceutical excipients. Eur. J. Pharm. Biopharm., 61(3), 158-170.
- Govindarajan, R; Nagarsenker M.S. (2005), Formulation studies and in vivo evaluation of a flurbiprofen – hydroxypropyl beta-cyclodextrin system, Pharm. Dev. Technol., 10(1), 113-122.
- Govindarajan, R; Nagarsenker M.S. (2004), Basic drug enterosoluble polymer coevaporates: development of oral controlled release systems, Drug Dev. Ind. Pharm., 30(8), 847–857.
- Govindarajan, R; Nagarsenker M.S. (2004), Influence of preparation methodology on solid-state properties of an acidic drug–cyclodextrin system, J. Pharm. Pharmacol., 56, 725–733.

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- 14. Nagarsenker, M.S; Meshram, R.N.; Ramprakash G. (2000). "Solid dispersion of hydroxypropyl beta-cyclodextrin and ketorolac: enhancement of in-vitro dissolution rates, improvement in anti-inflammatory activity and reduction in ulcerogenicity in rats", J. Pharm. Pharmacol., 52 (8), 949-956.
- 15. Nagarsenker, M.S.; Garad, S.D; Ramprakash G. (2000) "Design, optimization and evaluation of domperidone coevaporates", J. Control. Rel., 62, 31-38.
- 16. Nagarsenker, M.S.; Govindarajan, R., (1998) Solid dispersion for extended release of Verapamil HCl, Pharm. Pharmacol. Commun., 4 (7), 331-334.

Book Chapters

- 1. Govindarajan, R; Kinetic processes in Pharmaceutics, In: Pharmaceutics: Basic principles and applications to pharmacy practice, Ed. Dash, A; Singh, S; Tolman, J; (2013)
- Govindarajan, R; Suryanarayanan, R. (2006), Processing-induced phase transformations and their implications on pharmaceutical product quality, In: Hilfiker, Rolf (Ed.) Polymorphism: In the pharmaceutical industry, Wiley-VCH Verlag GmbH & Co., KGaA, Weinheim, Germany, p 333-364.

Invited Talks

- 1. Govindarajan, R; "Effect of water and buffer on the Hammett acidity function in amorphous lyophiles, Amorph-2014 Symposium, Cambridge, UK, Jul 2014
- 2. Govindarajan, R "Partial crystallinity in dehydrated trehalose dihydrate: the one-state crystallinity", Research Triangle Institute, Research Triangle Park, North Carolina, Jun 2009.
- 3. Suryanarayanan, R and Govindarajan R., "Phase transformations during pharmaceutical processing-potential implications on product performance", Dr. Reddy's Laboratories Ltd., Nov 24, 2005, Hyderabad, India
- 4. Govindarajan, R., "Evaluation of 'acidity' of Pharmaceutical solids using indicator probes", AAPS Pharmaceutics and Drug Delivery (PDD) conference, June 2004, Philadelphia, PA.
- 5. Govindarajan, R., "Solid-state acidity of amorphous freeze-dried systems and surface acidity of solid dosage forms", Pfizer Global R&D, April 2003, Groton, CT.

Conference presentations

- Campbell D, Patel R, Whigham T, Rhodes C, Palmer J, Bianco J, Everhart K, Farrier C, Clowes R, Roberts J and Govindarajan, R. Development of Fixed Dose Combinations for Extended Release of Metformin and Glimepiride. Poster presentation, at the American Association of Pharmaceutical Scientists (AAPS) Annual Meeting 2014, San Diego, CA.
- 2. Chakravarty, P.; Govindarajan, R.; Suryanarayanan, Effect of in situ Phase Transformation on Thiamine Hydrochloride Tablets, Poster presentation, American Association of Pharmaceutical Scientists (AAPS) Annual Meeting 2009, Los Angeles, CA.
- 3. Chakravarty, P.; Govindarajan, R.; Suryanarayanan, R. Stability and behavior of thiamine hydrochloride phases: role of thermodynamics, kinetics and crystal structure, Poster presentation, American Association of Pharmaceutical Scientists (AAPS) Annual Meeting 2008, Atlanta, GA.

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- 4. Chakravarty, P.; Govindarajan, R.; Suryanarayanan, R. Hydrate Forms of Thiamine Hydrochloride: Dehydration Behavior and Phase Transformation, Poster presentation, American Association of Pharmaceutical Scientists (AAPS) Annual Meeting 2007, San Diego CA.
- 5. Govindarajan, R.; Rani, M.; Surana, R.; Suryanarayanan, R. Structure in dehydrated trehalose dihydrate evaluation of the concept of partial crystallinity, Poster presentation, American Association of Pharmaceutical Scientists (AAPS) Annual Meeting 2006, San Antonio, TX.
- 6. Chakravarty, P.; Govindarajan, R.; Suryanarayanan, R. Phase transformations in Thiamine HCl during simulated pharmaceutical processing, Poster presentation, American Association of Pharmaceutical Scientists (AAPS) Annual Meeting 2006, San Antonio, TX.
- Liao, X.; Govindarajan, R.; Suryanarayanan, R., Use of Atomic Force Microscopy to Detect Surface Crystallization, Poster presentation, American Association of Pharmaceutical Scientists (AAPS) Annual Meeting 2005, Nashville, TN.
- Govindarajan, R.; Chatterjee, K.; Gatlin, L. A.; Shalaev, E. Y.; Suryanarayanan, R., Evaluation of buffered amorphous lyophiles using indicator probes: use of the Hammett acidity function, Poster presentation, American Association of Pharmaceutical Scientists (AAPS) Annual Meeting 2005, Nashville, TN.
- 9. Govindarajan, R.; Gatlin, L.; Shalaev, E.Y.; Suryanarayanan, R., Effect of water content on the acidity and ionization states in buffered amorphous sugar lyophiles., Poster presentation, American Association of Pharmaceutical Scientists (AAPS) Annual Meeting 2004, Baltimore, MD.
- 10. Govindarajan, R.; Zinchuk, A.V.; Hancock, B.C.; Shalaev, E.Y.; Suryanarayanan, R., Surface acidity of pharmaceutical excipients: Effect of formulation variables, Poster presentation, American Association of Pharmaceutical Scientists (AAPS) Annual Meeting 2004, Baltimore, MD.
- 11. Rani, M; Surana, R.; Govindarajan, R.; Pyne, A.; Suryanarayanan, R., Study of enthalpic relaxation in ternary frozen aqueous solutions using differential scanning calorimetry, Poster presentation, American Association of Pharmaceutical Scientists (AAPS) Annual Meeting 2004, Baltimore, MD.
- Govindarajan, R.; Shalaev, E.Y.; Gatlin, L.; Suryanarayanan, R., Effect of citrate buffer concentration on the degree of ionization of an indicator probe in an amorphous sugar lyophile, Poster presentation, July 28-31, 2004: NSF Conference on Freeze-Drying of Pharmaceuticals and Biologicals, Breckenridge, CO.
- 13. Govindarajan, R.; Chatterjee, K.; Gatlin, L.; Shalaev, E.Y.; Suryanarayanan, R., Study of the solidstate acid/base relationships in amorphous freeze-dried sugar-citrate systems using colyophilized pH indicators and diffuse reflectance spectroscopy, Poster presentation, American Association of Pharmaceutical Scientists (AAPS) Annual Meeting 2003, Salt Lake City, UT.
- 14. Chatterjee, K.; Govindarajan, R.; Gatlin, L.; Suryanarayanan, R.; Shalaev, E.Y., Effect of buffers on the acid/base properties of amorphous freeze-dried systems, Poster presentation, American Association of Pharmaceutical Scientists (AAPS) Annual Meeting 2003, Salt Lake City, UT.
- 15. Govindarajan, R.; Chatterjee, K.; Hancock, B.C.; Shalaev, E.Y.; Suryanarayanan, R., Evaluation of the surface acidity of pharmaceutical excipients and formulations using pH indicators as probe molecules and diffuse reflectance spectroscopy, Poster presentation, American Association of Pharmaceutical Scientists (AAPS) Annual Meeting 2003, Salt Lake City, UT.
- 16. Rani, M; Govindarajan, R.; Surana, R.; Suryanarayanan, R., Preparation, characterization and physical stability evaluation of the alpha-polymorph of anhydrous trehalose, Poster presentation, American Association of Pharmaceutical Scientists (AAPS) Annual Meeting 2003, Salt Lake City, UT.
- 17. Rani, M; Govindarajan, R.; Surana, R.; Suryanarayanan, R., Effect Of Glass Fragility On The Enthalpic Relaxation Measured By Differential Scanning Calorimetry, Poster presentation, American Association of Pharmaceutical Scientists (AAPS) Annual Meeting 2003, Salt Lake City, UT.
- 18. Govindarajan, R.; Chatterjee, K.; Hancock, B.C.; Shalaev, E.Y. Suryanarayanan, R., Surface acidity of pharmaceutical excipients and formulations, Poster presentation, AAPS Northeast Regional Discussion group (NERDG) meeting 2003, Rocky Hill, CT.
- Govindarajan, R; Nagarsenker, MS., Effect of succinic acid on release of verapamil HCl from hydrophilic matrix, Poster presentation, 53rd Indian Pharmaceutical Congress, December 2001, New Delhi, India.

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- 20. Govindarajan, R; Nagarsenker, MS., Cyclodextrin based oral delivery of flurbiprofen: effect of method of preparation on drug release and studies on in-vivo performance, Poster presentation, 28th Controlled Release Society Annual Symposium, June 2001, San Diego, California, USA.
- 21. Govindarajan, R; Nagarsenker, MS., pH independent oral controlled release of verapamil HCl: solubilisation in the microenvironment using enterosoluble polymer or organic acid, Poster presentation, 28th Controlled Release Society Annual Symposium, June 2001, San Diego, California, USA.
- 22. Govindarajan, R; Nagarsenker, MS, Development of oral controlled release systems for pH independent release of a weak base, Poster presentation, 3rd International Symposium on Advances in Technology and business potential of new drug delivery systems, October 2000, Ootacamund, India.
- 23. Nagarsenker, MS.; Meshram, R.; Govindarajan, R, Solid dispersion of hydroxypropyl betacyclodextrin and ketorolac: enhancement of dissolution rates, improvement in anti-inflammatory activity and reduction in ulcerogenicity, Poster presentation, 10th International Cyclodextrin Symposium, May 2000, Ann Arbor, Michigan.
- 24. Govindarajan, R; Nagarsenker, MS, Role of hydroxypropyl beta-cyclodextrin in improving dissolution and reducing ulcerogenicity of flurbiprofen: effect of l-tyrosine on bioavailability from coevaporate, Poster presentation, 10th International Cyclodextrin Symposium, May 2000, Ann Arbor, Michigan.
- 25. Govindarajan, R; Nagarsenker, MS, Solid dispersion of Flurbiprofen with Hydroxypropyl betacyclodextrin, Poster presentation, 4th International symposium on innovations in pharmaceutical sciences and technology, February 2000, Ahmedabad, India.
- 26. Govindarajan, R; Nagarsenker, MS, Verapamil HCl coevaporates using enterosoluble polymer: Development of oral controlled release systems, Podium presentation, 50th Indian Pharmaceutical Congress and the 17th Asian Congress of Pharmaceutical Sciences, December 1998, Mumbai, India.

EXHIBIT B



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1 item in my patents list Displaying selected publications

Publication	Title	Page
EP0780127 (A1)	A nasal spray containing a steroid an	2

(19)	Europäisches Patentamt European Patent Office Office européen des brevets EUROPEAN PAT	(11) EP 0 780 127 A1 ENT APPLICATION
(21)	Date of publication: 25.06.1997 Bulletin 1997/26 Application number: 96308852.1 Date of filing: 05.12.1996	 (51) Int Cl.⁶ A61K 31/57, A61K 31/58, A61K 31/56, A61K 31/495, A61K 31/445, A61K 31/495, (A61K31/56, 31:495), (A61K31/56, 31:495), (A61K31/57, 31:495), (A61K31/57, 31:495), (A61K31/57, 31:55), (A61K31/58, 31:495), (A61K31/58, 31:495),
	Designated Contracting States. AT BE CH DE DK ES FI FR GB GR IE IT LI LU NL PT SE Priority: 19.12.1995 US 574791	 (72) Inventor: Cramer, Ronald Dean Cincinnati, Ohio 45215 (US) (74) Representative: Woof, Victoria Procter & Gamble Pharmaceuticals Ltd.,
(71)	Applicant: THE PROCTER & GAMBLE COMPANY Cincinnati, Ohio 45202 (US)	Patent Department, Lovett House, Lovett Road Staines, Middlesex TW18 3AZ (GB)

spray compositions comprising a safe and effective

(57) The present invention relates to novel nasal amount of a glucocorticosteroid and an antihistamine possessing leukotriene inhibiting properties.

EP 0 780 127 A1

Printed by Jauve 75001 PARIS (FR)

Description

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TECHNICAL FIELD

The present invention relates to novel nasal spray compositions comprising a safe and effective amount of a glucocorticosteroid and an antihistamine.

BACKGROUND OF THE INVENTION

Allergic disorders remain a leading cause of both acute and chronic illnesses the world over These illnesses are often times present in the form of acute or chronic rhinoconjunctivitis. The symptoms of allergic rhinoconjunctivitis are reddening of the eyes, ocular secretions, nasal congestion, ocular and palatial irritation, sneezing and hypersecretion. These symptoms occur following exposure to allergens. The most common allergens are grass and/or tree pollens, hence, allergic rhinoconjunctivitis is most common during the spring and summer months.

The symptoms of allergic rhinoconjunctivitis are believed to be due primarily to the stimulation of H-1 receptors by histamine followed by reflexive activation of parasympathetic nerves causing increases in nasal secretion and obstruction. Histamine is initially released from the tissue mast cells upon sensitization of the mast cells. This sensitization results when airborne allergens combine with specific IgE antibodies attached to mast cell membranes.

Antihistamines and/or decongestants have traditionally been the drugs of choice in treating allergic rhinoconjunctivitis. Other forms of therapy include the use of cromolyn sodium, hypertonic sall solutions or immunotherapy.

In addition, Hagen et al., <u>U.S. Patent 4,767,612</u>, discloses nasal corticosteroid therapy as an effective means of treating allergic rhinoconjunctivitis; and is herein incorporated by reference in its entirety. Notwithstanding the many disclosures in the area of allergic rhinoconjunctivitis, there is still a need for additional formulations which provide improved symptomatic relief with increased user acceptance and compliance.

The present inventor has found that by combining a nasal corticosteroid with a leukotriene inhibiting antihistamine, improved intranasal compositions result, providing improved relief of symptoms generally associated with either seasonal or perennial allergic rhinoconjunctivitis.

It is, therefore, an object of the present invention to provide pharmaceutical compositions having improved effectiveness in the treatment of symptoms generally associated with either seasonal or perennial allergic rhinoconjunctivitis.

A further object of the present invention is to provide a safe and effective method for treating the symptoms of seasonal or perennial allergic rhinoconjunctivitis.

These objects and other objects will become more apparent from the detailed description that follows.

SUMMARY OF THE INVENTION

The present invention relates to pharmaceutical compositions for nasal administration comprising:

 a) a sale and effective amount of a glucocorticoid selected from the group consisting of beclomethasone, flunisolide, triamcinolone, fluticasone, mometasone, budesonide, pharmaceutically acceptable salts thereof and mixtures thereof;

b) a safe and effective amount of a leukotriene inhibiting antihistamine selected from the group consisting of celirizine, loratadine, azelastine, pharmaceutically acceptable salts thereof, optically active racemates thereof and mixtures thereof; and

c.) an intranasal carrier.

The intranasal carrier of the present invention is preferably aqueous.

The present invention also relates to a method for the treatment of symptoms associated with seasonal or perennial allergic rhinoconjunctivitis comprising the administration of a safe and effective amount of the intranasal pharmaceutical compositions of the present invention. By "symptoms of seasonal or perennial allergic rhinoconjunctivitis" or "symptoms associated with seasonal or perennial allergic rhinoconjunctivitis" or "symptoms associated with seasonal or perennial allergic rhinoconjunctivitis", ocular secretions, reddening of the eyes, sneezing, mucoid hypersecretion, nasal congestion and itching.

By "safe and effective amount," as used herein, is an amount that is effective to mitigate and/or treat the symptoms for which the active ingredient is indicated in a human without undue adverse side effects commensurate with a reasonable risk/benefit ratio.

By "leukotriene inhibiting antihistamine, "as used herein, is meant an antihistamine effective in inhibiting or reducing In vivo the biosynthesis of and/or cellular release of leukotrienes or otherwise modulating mammalian leukotriene levels.

The pH of the compositions is preferably from about 4.5 to about 9, more preferably from about 6 to about 7. All percentages and ratios herein are by weight unless otherwise specified. Additionally, all measurements are

made at 25°C unless otherwise specified.

DETAILED DESCRIPTION OF THE INVENTION

The compositions of the present invention contain the essential components as well as various optional components as indicated below.

More specifically, the compositions of the instant invention are for nasal administration and contain a therapeutically effective amount of the herein described pharmaceutical agents. They are preferably provided as isotonic aqueous solutions, suspensions or viscous compositions which may be buffered to a selected pH.

Essential Ingredients

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Glucocorticoid Agents

Agents within this class have potent glucocorticoid activity and weak mineralocorticoid activity. Glucocorticoid agents most useful to the present invention include those selected from the group consisting of beclomethasone, flunisolide, triamcinolone, fluticasone, mometasone, budesonide, pharmaceutically acceptable salts thereol and mixtures thereol.

When used in the compositions of the present invention, the glucocorticoid component is preferably present at a concentration of from about 0.001% to about 0.2%, more preferably from about 0.01% to about 0.1%.

Leukotriene Inhibiting, Antihistaminic Agents

Antihistamines useful to the present invention are histamine H-1 receptor antagonists which also reduce mammalian leukotriene levels. Such H-1 receptor antihistamines may be selected from among the following groups of antihistamines: piperazines, phenothiazines, piperidines.

Examples of useful leukotriene inhibiting antihistamines include cetirizine, loratadine, azelastine and the like, optically active racemates thereof, pharmaceutically acceptable salts thereof and mixtures thereof. When used in the compositions of the present invention, the antihistamine component is preferably present at a concentration of from about 0.01% to about 4.0%, more preferably from about 0.01% to about 1%.

Pharmaceutically-Acceptable Aqueous Nasal Carrier.

One other essential component of the present invention is a pharmaceutically-acceptable intranasal carrier. Preferred for use herein are aqueous saline solution carriers. These solutions which generally contain sodium chloride as the salt are fully described in Remington's Pharmaceutical Sciences, 17th edition (1985) p. 835, which is herein incorporated by reference. The salt is present in the solution at a level of about 0.01% to about 2%, preferably from about 0.5% to about 1.0%.

The combination of any of the above described antihistamines and glucocorticoids can be conveniently administered nasally to warm-blooded animals to elicit the desired therapeutic response by formulating it into a nasal dosage form, together with a nontoxic pharmaceutically-acceptable nasal carrier. Suitable nontoxic pharmaceutically-acceptable nasal carriers are known to those skilled in the art and are also fully disclosed in Remington's Pharmaceutical Sciences, 17th edition, 1985. Obviously, the choice of suitable carrier forms will depend on the exact nature of the particular nasal dosage form required, e.g., whether the drug(s) is to be formulated into a nasal solution (for use as drops or as a spray), a nasal suspension, a nasal ointment, a nasal gel or another nasal form. Preferred nasal dosage forms are solutions, suspensions and gels, which normally contain sodium chloride in a major amount of water (preferably purified water) in addition to the antihistamine and glucocorticoid. Minor amounts of other ingredients such as pH adjusters (e.g., an acid such as HCI), emulsifiers or dispersing agents, buffering agents, preservatives, wetting agents and jelling agents (e.g., methylcellulose) may also be present. Most preferably, the nasal composition is isotonic,

i.e., it has the same osmotic pressure as blood and lacrimal fluid.

Preferably the composition is applied to the nasal mucosa via topical application of a safe and effective amount of the composition to treat nasal symptoms. The amount of the antihistamine and glucocorticoid combination and frequency of topical application to the nasal mucosa may vary, depending upon personal or medical needs, but it is suggested, as an example, that topical application range from about once per day to about four times daily, preferably twice daily, most preferably once daily. As a practical matter the selected therapeutic compositions will normally be prepared in unit dosage forms or actuations to contain therapeutically effective amounts of the selected antihistamine and glucocorticoid combination. In specific instances fractions of these dosage units or multiple dosage units will be employed. Typically, dosage units may be prepared to deliver from about 0.5 mcg to about 100 mcg of the glucocorticoid

agent and from about 5 mcg to about 1000 mcg of the antihistaminic agent per spray actuation (e.g., 50 mg to about 200 mg of the spray composition). A typical dose contains one to four sprays per nostril.

Optional Ingredients

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Optional ingredients useful in the present invention include decongestants. Decongestants useful to the present invention may be selected from among the class of sympathomimetic agents; examples of which include pseudoephedrine, desoxyephedrine, propylhexedrine, phenylpropanolamine, xylometazoline, phenylephrine, tetrahydrozoline, naphazoline, oxymetazoline, tramazoline and pharmaceutically acceptable salts thereof. Also useful as decongestants are the 5-(2-imidazolinylamino)benzimedazole compounds. Mixtures of these decongestants can also be used.

When used in the compositions of the present invention, the sympathomimetic agents may be incorporated at concentrations, preferably, of from about 0.01% to about 0.5%, more preferably from about 0.05% to about 0.1%.

The compositions of the present invention may also contain antiallergics. Suitable antiallergics include, but are not limited to, cromolyn, ketotifen, N-allyl-(dichloro-3, 4-benzyl)-2-methylamino-2-propanol-1, AP-582 (Pharmaprojects No. 3055-under investigation by Ariad Pharmaceuticals), Andolast, oxatamide and pharmaceutically-acceptable salts thereof. Mixtures of these antiallergics may also be used.

Similarly, mucolytics such as acetylcysteine and anticholinergics such as ipratropium bromide may also be used in the compositions of the present invention.

Also of optional use in the compositions of the present invention are nonopiate analgesics such as oxaprozin. The intranasal use of oxaprozin is described in Namiki et al., <u>Studies on improvement of pharmaceutical preparations prescribed in hospitals. VI. oxaprozin nasal spray</u>, Drug Design and Delivery 1968;2:pp. 311-321, herein incorporated by reference. Further examples of preferred nonopiate analgesics include, but are not limited to, acetaminophen, ace-tylsalicylic acid, ibuprofen, etodolac, fenbuprofen, fenoprofen, flurbiprofen, indomethacin, ketoprofen, naproxen, pharmaceutically-acceptable salts thereof, optically active racemates thereof and mixtures thereof. Still further examples of such drugs are disclosed in U.S. Patent No. 4,522,828, to Sunshine et al., issued June 11, 1985; this patent being incorporated herein by reference in its entirety.

Synthetic opiate analgesics such as butorphanol may also be incorporated into the compositions of the present invention. The intranasal use of butorphanol is described in Baumel, <u>Migraine: A pharmacologic review with newer options and delivery modalities</u>, Neurology 1994;44(supp):pp. s13-s17, herein incorporated by reference. Further examples of preferred synthetic opioid analgesics include alfentanil, buprenorphine, fentanyl, meperidine, methadone, nalbuphine, natrexone, propoxyphene, pentazocine, sufenanil, pharmaceutically-acceptable salts thereof and mixtures thereof.

Compounds commonly known as lipoxygenase inhibitors and receptor antagonists are also optionally useful in the compositions of the present invention. Suitable lipoxygenase inhibitors are described in U.S. Patent 4,873,259, to Summers et al., issued October 10 1989 and European Patent Application 318093, both of which are herein incorporated by reference. Lipoxygenase antagonists suitable for use in the present invention include Zafirlukast (Accolate, Zeneca).

Leukotriene receptor antagonists may also be incorporated into the compositions of the present invention. Suitable examples include, but are not limited to, experimental agents such as LY171883, Wy-45,911, LY163443, ONO-RS-411 and ONO-RS-347 and ICI 198,615. A more detailed discussion of leukotriene receptor antagonists is found in Fleisch, J. H., <u>Development of Cysteinyl Leukotriene Receptor Antagonists</u>, Vol. 12 Advances in Inflammation Research 173-189 (A. Lewis et al. ed, 1988), herein incorporated by reference in its entirety.

Various aromatic components (e.g., aldehydes and esters) may also be used. These aromatics include, for example, menthol, camphor, eucalyptol, benzaldehyde (cherry, almond); citral (lemon, lime); neral; decanal (orange, lemon); aldehyde C-8, aldehyde C-9 and aldehyde C-12 (citrus fruits); tolyl aldehyde (cherry, almond); 2,6-dimethyl-octanal (green fruit); and 2-dodecenal (citrus, mandarin). Additional aromatic components suitable for use in the present invention include those described in <u>U.S. Patent 4,136,163</u> to Watson et al., <u>U.S. Patent 4,459,425</u> to Amano et al., and <u>U.S. Patent 4,230,668</u> to Rowsell et al.; all of which are herein incorporated by reference. Mixtures of these aromatics can also be used.

The desired isotonicity of the compositions of this invention may be accomplished using, for example, the sodium chloride already present, or other pharmaceutically-acceptable agents such as dextrose, boric acid, citric acid, sodium tartrate, sodium phosphate, potassium phosphate, propylene glycol or other inorganic or organic solutes or mixtures thereof. Sodium chloride is preferred particularly for buffers containing sodium ions. Further examples of sodium chloride equivalents are disclosed in <u>Remington's Pharmaceutical Sciences</u> pp. 1491-1497 (Altonso Gennaro 18th ed 1990).

Viscosity of the compositions may be maintained at the selected level using a pharmaceutically-acceptable thickening agent. Methyl cellulose is preferred because it is readily and economically available and is easy to work with. Other suitable thickening agents include, for example, xanthan gum, microcrystalline cellulose, carboxymethyl cellu-

lose, hydroxypropyl cellulose, hydroxypropyl methyl cellulose, hydroxymethyl cellulose, hydroxypethyl cellulose, carboxyvinyl polymer, carbomer, and the like or pharmaceutical salts thereof. Mixtures of such thickening agents may also be used. The preferred concentration of the thickener will depend upon the agent selected. The important point is to use an amount which will achieve the selected viscosity. Viscous compositions are normally prepared from solutions by the addition of such thickening agents.

Preferred compositions within the scope of this invention will contain from about 0.01% to about 5% of a humectant to inhibit drying of the mucous membrane and to prevent irritation. Any of a variety of pharmaceutically-acceptable humectants can be employed including, for example sorbitol, propylene glycol, polyethylene glycol, glycerol or mixtures thereof. As with the thickeners, the concentration will vary with the selected agent, although the presence or absence of these agents, or their concentration is not an essential feature of the invention.

Enhanced absorption across the nasal membrane can be accomplished employing a therapeutically acceptable surfactant. Typical useful surfactants for these therapeutic compositions include polyoxyethylene derivatives of fatty acid partial esters of sorbitol anhydrides such as Polysorbate 80. Polyoxyl 40 Stearate, Polyoxylethylene 50 Stearate and Octoxynol, as well as Oxyethylated tertiary octyl phenol formaldehyde polymer (available from Sterling Organics as tyloxapol) or mixtures thereof. The usual concentration is from 0.5% to 10% based on the total weight.

A pharmaceutically-acceptable preservative is generally employed to increase the shelf life of the compositions of the present invention. Benzyl alcohol is suitable, although a variety of preservatives including, for example, parabens, phenylethyl alcohol, thirmerosal, chlorobutanol, phenylmecuric acetate or benzalkonium chloride may also be employed. The most preferred preservative system for use herein comprises a combination of benzalkonium chloride, chlorhexidine gluconate and disodium EDTA. A suitable concentration of the preservative will be from 0.001% to 2% based on the total weight, although there may be appreciable variation depending upon the agent selected Mixtures of these preservatives may also be used.

Other Optional Components. A variety of additional ingredients may be added to the emulsion compositions of the present invention. These additional ingredients include various polymers for aiding the film-forming properties and substantivity of the formulation, antioxidants, and agents suitable for aesthetic purposes such as fragrances, pigments, and colorings.

The compositions can also contain low levels of insoluble ingredients added, for example for visual effect purposes, e.g. thermochromic liquid crystalline materials such as the microencapsulated cholesteryl esters and chiral nematic (nonsterol) based chemicals such as the (2-methylbutyl) phenyl 4-alkyl(oxy)benzoates available from Hallcrest, Glenview, Illinois 60025, U.S.A. Mixtures of these ingredients may also be used.

EXAMPLES

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The following examples further describe and demonstrate embodiments within the scope of the present invention. The examples are given solely for the purpose of illustration and are not to be construed as limitations of the present invention, as many variations thereof are possible without departing from the spirit and scope of the invention.

Example |

The intranasally administered pharmaceutical composition of the present invention is prepared by combining the lollowing components utilizing conventional mixing techniques similar to that described below.

Component	Wgt %
beclomethasone diproprionate, monohydrate	0.042
loratadine	0.200
avicel RC - 5911	1.200
dextrose	5.100
polysorbate 80	0.025
benzalkonium chloride	0.040
phenylethyl alcohol	0.250
distilled water	q.s. to vol

¹microcrystalline cellulose and sodium carboxymethyl cellulose, supplied by FMC corporation.

In an appropriately sized vessel, the dextrose, polysorbate 80 and benzalkonium chloride are added one at a time to water with mixing, allowing each to dissolve or completely disperse before adding the next. To this is added, with mixing, a premixed slurry of the avicel and water. Upon forming a uniform solution, the becomethasone, loratadine

and phenylethyl alcohol are added. After all the ingredients are added, purified water is used to bring the batch to the appropriate weight.

Administration of approximately 0.4 grams of the composition is used for topical nasal application to provide relief from allergy or allergy-like symptoms.

Example II

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The intranasally administered pharmaceutical composition of the present invention is prepared by combining the following components utilizing conventional mixing techniques similar to that described in Example I.

Component	Wgt %
flunisolide	0.025
cetirizine	0.200
propylene glycol	2.000
polyethylene glycol	1.000
sodium chloride	0.900
ethylenediamine tetraacetic acid	0.050
benzalkonium chloride	0.010
distilled water	q.s. to vol.

Administration of approximately 0.4 grams of the composition is used for topical nasal application to provide relief from allergy or allergy-like symptoms.

Example III

The intranasally administered pharmaceutical composition of the present invention is prepared by combining the following components utilizing conventional mixing techniques similar to that described in Example).

Component	Wgt %
triamcinolone acetonide	0.050
azelastine HCI	0.070
polysorbate 80	0.050
glycerin	2.000
hydroxypropyl methyl cellulose	1.000
sodium chloride	0.900
ethylenediamine tetraacetic acid	0.050
benzalkonium chloride	0.020
distilled water	q.s. to vol

Administration of approximately 0.4 grams of the composition is used for topical nasal application to provide relief from allergy or allergy-like symptoms. Additionally, substantially similar results are also obtained using, in whole or in part, equivalent amounts of other glucocorticoid agents such as fluticasone, mometasone, budesonide, pharmaceulically acceptable salts thereof and mixtures thereof. Furthermore, the above described compositions may also contain a decongestant such as pseudoephedrine, phenylpropanolamine, phenylephrine, tetrahydrozoline, naphazoline, oxymetazoline, tramazoline, 5-(2-imidazolinylamino)benzimedazoles, optically active racemates thereof, pharmaceutically acceptable salts thereof and mixtures thereof. Those skilled in the art will quickly realize other suitable ingredients, diluents and dosage forms (or readily ascertain such using routine experimentation) which may further be incorporated into the above compositions without departing from the scope and spirit of the present invention.

Claims

1. A pharmaceutical composition comprising.

a) a safe and effective amount of a glucocorticoid selected from the group consisting of beclomethasone,

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flunisolide, triamcinolone, fluticasone, mometasone, budesonide, pharmaceutically acceptable salts thereof and mixtures thereof;

 b) a safe and effective amount of a leukotriene inhibiting antihistamine selected from the group consisting of cetirizine, loratadine, azelastine, pharmaceutically acceptable salts thereof, optically active racemates thereof and mixtures thereof; and

c.) an intranasal carrier.

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2. A composition according to Claim 1 in the form of an isotonic aqueous solution

3 A composition according to Claim 1 or 2 wherein the glucocorticoid is selected from the group consisting of beclomethasone, budesonide, fluticasone and mixtures thereof.

4. A pharmaceutical composition according to any of Claims 1-3, which further comprises a sympathomimetic amine selected from the group consisting of pseudoephedrine, desoxyephedrine, propylhexedrine, phenyl-propanolamine, xylometazoline, phenylephrine, tetrahydrozoline, naphazoline, oxymetazoline, tramazoline, 5-(2-imidazolinylamino)benzimedazoles, pharmaceutically acceptable salts thereof, optically active race-mates thereof and mixtures thereof.

5. A pharmaceutical composition according to any of Claims 1-4, which further comprises a non-steroidal anti inflammatory agent, or optically active racemates thereof and mixtures thereof.

6. A pharmaceutical composition according to any of Claims 1-5, which further comprises a lipoxygenase inhibitor or antagonist, a leukotriene receptor antagonist, a nonopiate analgesic, a mucolytic, an antiallergic, and pharmaceutically acceptable salts thereof and mixtures thereof.

APOTEX_AZFL 0052958



EUROPEAN SEARCH REPORT

Application Number EP 96 30 8852

A	Citation of document with indication	where appropriate	Relevant	CLASSIFICATION OF THE	
Category	of relevant passages	, where appropriate.	to claim	APPLICATION (Int.Cl.6)	
X,Y ADVANCES IN THERAPY, vol. 12, no. 6, November 1995 - December 1995, USA, pages 340-349, XP000651473 DROUIN ET AL: "ADDING LORATADINE TO TOPICAL NASAL STEROID THERAPY IMPROVES MODERATELY SEVERE SEASONAL ALLERGIC RHINOCONJUNCTIVITIS" * the whole document *			1-6	A51K31/57 A61K31/58 A61K31/56 A51K31/495 A61K31/495 A61K31/445 A61K31/55 //(A61K31/56, A61K31:495), (A61K31:405)	
Х,Ү	CLINICAL AND EXPERIMENT/ vol. 22, no. 10, October pages 916-922, XP0006510 ARMITAGE ET AL : "INVES TENDENCY TO WHEEZE IN PO PATIENTS" * the whole document *	- 1992, UK, 560 STICATION OF THE	1-6	A61K31:445), (A61K31/56, A61K31/55), (A61K31:55), (A61K31:495), (A61K31:495), (A61K31:57, A61K31:445), (A61K31:57, A61K31:55),	
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	pages 225-233, XP0006514	134 JATION OF		TECHNICAL FIELDS SEARCHED (Int.Cl.6)	
	FLUTICASONE PROPIONATE / TAKEN ALONE AND IN COMB CETIRIZINE IN THE PROPH OF SEASONAL ALLERGIC RH * the whole document *	INATION WITH (LACTIC TREATMENT		A61K	
D,Y	US 4 767 612 A (HAGEN N 30 August 1988 * abstract *	CHOLAS S ET AL) 1,2,4-0			
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	The present search report has been dra	the state of the state of the state of the			
Place of search MUNICH		Date of completing of the search 1 April 1997	He	Franker Prera, S	
X: par Y: par do	CATEGORY OF CITED DOCLMENTS ticularly relevant if taken alone (cularly relevant if combined with another ument of the same category hological background	I : theory or principl F : earlier patent doc after the filing do D : document cited i I : document cited fi	e underlying th sument, hut pu ite n the application or other reason	hè invention blished en, m on is	
O . nen-written disclosure P : intermediate document		& : member of the sa document			

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EUROPEAN SEARCH REPORT

Application Number EP 96 30 8852

	DOCUMENTS CONSID		1	The second second second
ategory	Citation of document with indi of relevant passa		Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.CL6)
	EP 0 605 203 A (SENJU 1994 * the whole document 		1-6	(A61K31/58, A61K31:495). (A61K31/58, A61K31:445)
				TECHNICAL FIELDS SEARCHED (Int.Cl.6)
	The present search report has been	n drawn up for all claims		
	Place of search	Date of completion of the search		Examiner
1.1	MUNICH	1 April 1997 He	rrera. S	
Y: pai doi A: tec O: no	CATEGORY OF CITED DOCUMENT ficularly relevant if taken alone tscularly relevant if combined with anoth unent of the same category hnological background n-writhan disclosure remodiate document	K : earlier patent o after the filing or D : document citer L : document citer	locument, but put date I in the application for other reasons	n s

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EXHIBIT C

1) Azdastine Ha (0.07.1) and Trancinatone 1 Acetonide (0.057) noral & pray composition - Background in formation 2) Batch 1345-004: Trial batch preparation of triam cinplone are tonide (0.05%) and 4 azilastine HU (0.07%) nasal spray formulation 3) Batch 1345-008: Trial batch preparation of trancinolone autonide (0.05.) and 8 azelastin MCI (0.07%) nasal spray Formulation 4) Batch 1345-011 - Preparation of a SUDML batch 11 of triancinolone actionide (0.05%) and azelastine Hal (0.07%) nasal spray suspension and evaluation of the batch.

Triancinolone Actonide & Azdastine Ha nasal 1345 1 1345 - 001 Winston - Strawn 1345 - 001 Trial preparation and \$\$ \$\$p 10 Jun 2016 background Background 10 - June - 2016 The propose is to determine an appropriate prepara--tion provedure for the composition described below prepare a batch with the said compo--sition and evaluate physical properties of the final batch. Ingredient and Concentration. Triancinatione Autonide - 0.05%. NV Azolastine Hal - 0.05%. W/V Polyson bate 80 - 0.05%. W/V Glycenin - 2.0%. W/V Hub another and 0.05%. Hypromellose - 1.0% w/v Sodium chloride - 0.9% w/v Etholene diamine tetra acitic acid - 0.05 %. W/V Bentzalkonium chloride - 0.02%. W/V Distilled water - q.s. to final volume US Patent 6017909 supports a solubility for the HCL Sult of azilantine as 1%. Band on This it is expected that in the Composition Superted above, azilantine could be completely in Solution. Block and Patel (1973) Secret Subility of Trian-- cinclone Automate in water to be < 20 mg/ml with Suchurd Saukility @ higher ionic strong the. Ignsin the power ble Solubli zing effects of the Surfactant (Inteen So) and glycenin, it is likely that the concentration of Itsiam cinclone automide would excued the Solubility of the drug. The Surfactant (Polyson bate So) would help with we thing of the Suspended particles and hypromethore Solution. to decrease Sedimentation sales. 9 Kappeket 10-Jun-2016

See page 1 R. Ry 10 Jun 2016 Polyeon bat 80 and glyunin can help with initial dispersion and Surface wetting of the drug particles before their addition to the bulk lof the agheous vehicle. chloride provides the desired tonicity Lodium (osmolality) to the preparation Ethylene diamine tetrautic acid; a chilating agent may serve as a stabilizer against Odighadation. Odlida tive Benzalkonium Chloride could unction aban anti-microbial preservative. / EDIA may Contribute anti- microbial activity of well Azdantine HCl, polyson bat 80, hypromelloze, Sodium chlomide, I glycin, and benzalkonium Chlande, beige & g. 10-Istin-2016, have Sufficient aqueous dollability to be in Solution" in This Composition. Trancinolone aretonide is expected to remain pastially undissolved and sub pended is the Vehicle !! The reported Salubility of EDTA is 1 in 500 parts (from the Hand book I of Pharma central Exciptients) and hence should also be in Solution But there are other reports of much lower solubility. From Noshino et al 1 (1958) The Salubility at 200 is 1:153 mm (~ 33.6 mg/100mL) In the process of preparation therefore rapid EDIAL was not expected. disso by fiots 8 Yoshino et al (1958), Bull. Chem. Suc. Jap 31(7), 892 Black and Vatel, J Phasm Sci 62(4), 617, 1972 1973 9 fy 10 Jun 2016 Analysin by Reviewed by Date: 10 Jun 2016 Date 3 To Page No.

PTX1664-00026

3 See page 1 Rep 10 Jun 2016 Chemicals used - Hypromellose USP, Type 2910, Pharmacoat 606, Shih Etsu Lot # 1088348 - Benzalkonium Chlonde Sigma, Material B6295 Lot 开60K 2609 - Sodium chloride USP, Sigma Lot # CDBC0348V - Elycerint USP, Sigma Aldrich Lot - SHBG1128V - PolySoobat 80 Ph. Eur Sigma Lot - BCBQ 4721V - Ethylenediamine tetraactic acid (EDTA) Aldrich. (99,09951) Lat 井 MKBW 24.70V - Triamaindone Autonide USP, Signa Aldrich - Azelastine HCL, Signa Aldrich Lot # LRAA5303 - Pusified Water USP, from UIP port # Z Ryn Jun 2016, Equipment list - IKA Euroster 20 overhead Stirror, Model Euro--ST 20D 5001 - Fisher Thermix magnetic stirver and hot plate Model 11-495, Stral # 118 - IKA RO 10 magnetic stirroz - Branson Ultorbonic bath Model CPX 5800H Senial # EG0021670670B Metther Toledo XSE 105 dual Trange kalance Senal # 13617383248 - Ohans analytical balance AX124 (Sno B530216186) TKA T25 digital Ultra Turrax homogenizor Model T25 DEI #03445541 - IKA homogenizaton tool SZSN - 18G NWR Nortex mixur SN-160315017 S Rp 10 Jun 2016 S. Ka-p-hL 10 Jun 2016

PTX1664-00027

4 Trancinolone Autonide and Azdastine Hal 1345-004 1345-004 Torial Suspension Breparration S. Ly 10 Jun 2016 Purpose - Breparation of a trial bath of triamainstone outonide (0105%) and azalastine Hkl (0.07%) naral Spray. Expt. Ref 1345-004 Composition for a 100 m2 batch. (ingredient details - page 3) Ingredient, Jasget gty Actual quantity 0.059 OTramcinclone Actomide 0.050119 OAzilastre, Ha 0.070239 1 Yolyson bate 80 0.059 0.05040% B Gly win 2002 1.9949 1-000 0.99899 OHypromellose Lofum chloride 0.900 0.90129 DEdetic Acid 0.059 0.029 0.019819 Benzalkonium chloride Drumfied water gsloomL 95 100m2 Rocedure: In this attempt to prepare a norsal 2 kay with the comprission listed above ingridiate 2 , O and O will be weathed out and transfind to appreximately soo & Rp 10 Jun 2016 50ml of devonind water under string till all of them dissolve. O will be added under vignous stirring to generate a vortex. O will be shally sprinklid, into the Solution. After addition is low pluti stirring will continue till a clear dispension is obtained 3, 6, 0 and 7 will be added signen trully into a vial and the contents will be mixed well on a vostex mixor followed by Donicetion. A small postion of the purified water I will be added under Stirring to obtain a pourable dispersion. This dispersion will be added to the aqueous phase containing Q, O, O & and a partial of the water, while being mixed with a Thomogenizor. 10 Jun 2016 5

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See page 4.

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when homo genization is complete. Sufficient water will be used to vinse the container originally used to propose a mixture of O B B LO and transford to the bulk container. Homogenization will be continued till a uniform dispersion is visually confirmed. Additional water may be added to adjust the final volume to 100ml and appropriate additional mixing will be cannot appropriate additional mixing will be filled into a surfable contained for observation.

11- Jun-2016 - This providence will invorporate edetic acid as a subservin glycuin an poly--sorbate so and upon addition to the tequeous phase with homogenization is expected to dissolve to the extent of its substituty in the rehall

13-Jun -2016

Benzalkonium Chloride was transford (weight out) ina 20.1 - vial. and approximately 10ml of water was added. The Conferts were mixed manually to obtain a Solution. In a 100ml beaker, approximately 30ml of water was transformed and the weight grantity of Sodium chloride was added to it under stimp using a magnetic Str - bar. when the Sodium chloride had dissolved, the benzalkonium chloride had dissolved, the benzalkonium chloride solution was transformed to the beaker and addition was transformed to the beaker. Azilartine Hell was then weight and transformed to the agneous Solution. Under Stiming. After about a minute of string 13 Jun 2016

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S. Fy 13 Jun 2016

the beaker was placed in an ultra Sonic bath for Imin to break up the few remaining undissalued aggregato of azelantine Ha to facilitate dissolution. An additional Iminut of stirring yielded a char Solution. hypromellor was weighed out and spinked spinkled into the solution. The solution was continued to stir to hydrate and dispense the hypromellor to yield a char Solution. This aqueous solution was stirred using a mag-netic stir box while the second pre-mik of ingredients was prepared. Elycenn, Polyson batel \$0, tramcing lone a letonide and editic bid were weighed out and trans--ford into a 20ml vial. The contents were mixed on a vortex mixer for 30 seconds to of water was added to this shorn while being mixed with a magnetic stir- box. This fielded a hazy dispersion of the solids in the water + glycisin + polyses bat so vehicle. The magnetic stir - bare were removed from both containors. The beakir containing the aqueous, Solution was plaud on the homoge. nizor stand and the contents were mixed bat 2000 spm. The dispersion containing the undisvolved Solids was added to the beakers while being mixed. Additional water was added to the vial, and the since water was added to the beaker in two postions of 10mL each. Homogeni-zation was continued for 5 min at 6000 spm. The homogenization tool was raised on the beaker Covered. The contents were allowed to stand for 13 - Jun-2016

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See page 4

Ry 13 Jun 2016

the foam to subside. OThe final volume was made up to 100 m2, and homogenized at 6000 ~ pm 15 min. The preparation was stored overnight to be obtained the next day in a stoppind vial.

14 - Jun - 2016 The preparation was a cloudy dispersion. It exhibited presence of fine Settled Solid upon shaking. The Solid was easily redispersed upon gentle agitation by overturning the vhal 3-4 times. I overturning the beaks and homogenized at 1000 spin for 10 min. The final preparation was transferred back to a Stoppord 100 mL capacity Berun Vial and Stand at soom ten perature.

DON 13-Jun-2016, the final volume was made up by transforming the unitents to a 250 mL beaker marked for 100 mL volume and making up volume by adding water used to rinse out the original contained and B. Bp 14-Jun-2016.

Conclusion - A uniform Suspension product was prepared. Although Settling of the Suspended Red Lid was obserred, the foldiment was readily and uniformly resuspended with gentle agitation.

S. R. 14 Jun 2016

S. Ha-popl 14- Jun 2016

8 Triamcinolone Autonide and Azolastine HU noval 1345-008 Total Suspension proportion. S. Ly 14 Jun 2016 Purpose - Preparation of a trial batch of a normal spray composition. Containing 0.057. W/ of triamcinelone aletonide and 0.071. W/ of azelastine HCL. (14 Jun 2016) Composition. for a 100 m 2 batch. (ingredient details - pg3) Tanget gty Ingredient Actual gty 0.049818 Difficiencinatione a letonide 0050 E Azilastine HCI 0.079 0.070369 3 Yolyson bat 80 0.059 0.050399 6 Glywin Oflypromellore 2.009 2.01629 1.000 0.999292 0.908 8) Sodium chloride 0.90029 0.053 A Edetre Acid 0.04991 8 0.029 3 Benzalkonium Chloride 0.02027g gs 100mL D Runfied water 95 100 ml In this preparation procedure EDTA will be dissolved in the aqueous phase with the aid of heat and a string of glywin, polyson bati so and triamcinolone will be added to the aqueous plane containing the remaining ingredients, followed by homo--genization and making up of the volume. Beparaton. - I Benzalkonium chloride weighed but ina zomz Vid. 10mz) water added and mixed - In a 100 mL beaker, approximately 60 mL 7, water heated to Force and EDTA added to the heated water with Stirring. Stirring continued till the EDTA is dissolved. Took - Lodium chloride added and stirring continued G. Lapet 14 Jun 2016 0 To Page Wo

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Ser page 8

S. Kp 14 Jun 2016

the clear Solution was obtained.

- Benzalkonium chloride Solution added to the beaker and an additional 5m2 of water used to since the container and transford - The Solution was allowed to cool down to

below 35°C.

- Azilas the Ha was wayhed out and toans --found to the Solution under Stirring, which was continued the all the added drug dissolved.
- While stirring with a magnetic stir- ber to create a vortex, the hypomethore was slowly added in partiens and allowed to wet and get suspended. The mixing was continued filla char & bution was obtained

- This Solution was kept aside while the triam -- cinclone Sluon was prepared. - Triam cinclone actionid, glycerin and polysobat 80 was weighed and transferred into a 20mL vial The contexts were mixed well on a vertex mixer to obtain a uniform sturn. Approximately Sml of water was added to the viel and the contents mixed well to obtain a Suspension. - The agreeus mixture of @ @ @ @ @ and a

portion of the water was transferred to a 250ml Beaker. The Shing of O in a mixture of Band O with some water was added to the agreeus mixture with home genization. at 3000 spm. The two containers were sinced with approximately 15ml of water and the rinse was transfind to the main container.

-Homogenization was continued at 1000 spm for

Analyse bus S tappeter 14 Jun 2016

10 See page 8

> - The home genization tool was saised, for was allowed to subside to enable making up the volume - Sufficient water was added after de-foaming to make up the volume to 100 ml. - The final mixture was homegenized at 6000mm or 5 additional minutes. - The final product was trainsferred to a 100 mL capacity Serum bottle with Stopper and allowed to stand at scoon temperature - After about I has of standing the product was observed and compared with the batch Departed in 1345-004. I Both preparations booked Similar to the naked age. Botthe exhibited preserve of a fine Sediment and visually no difference was observed between the two. In both Cases, gently rotating the vials redispersed the file Sediment uniformly and no large due pended a hazy appearance with uniformly dispersed Solid phase Conclusion - The modified preparation method which infiliard heat to first dissolve EDTA yielded a uniform Suppression. which was similar to the batch obtained in 1345-004. Settling of the Swspended solid ocurned over time, but the Sidiment was easily and uniformly resuspendable S. Ry 14 Jun - 2016 R. Ra-prett 14- Jun - 2016

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11 Triamaino have Actomale and Azlastine Ha noval spray 1345-011 Batch puparation and analysis. S. Ry 16 Jun 2016 Purpose - Reputation of a SOOML batch of nosal Spray containing 0 of y. Triamcinolone Autonide and 0107. Azelastine Hal. 16-June-2016 Composition (Ingredient details on Page 3) Tanget gty Actual gty 0:24968g Ingredients 0.250 9 Triamcinolone Autonide 0.34999 0.350 g - Azolowhine HCL - Ethylene diamine tetraacetic acid 0.2509 0.25039 Polyson bati 80 0.25109 10.0087 8 10.0000 Glycoin 0.1000 0.10050 - Benzalkonium Chloride 4.5009 4.50229 - Lodium chloride 95 500mt 16 Jun 2016 95 500mt - Hypromellose 5.00409 5.0009 Punfied water 9.5 500 mL 9-5 500ml Beparation 1) The 500ml graduation on a 750ml beakur Was confirmed using 500ml of water. 2) Benzalkonium Chlonde was weighd out into a 20ml vial. Approximately 10ml of water was added and the contents mixed well on a Vostex mixur 3) Approximately 250 mL of Rinfied water heated to Force 4) EDTA added to water with string. After most of the orded solid had dissolved, the beaker was placed in an ultrasonic beth for 30 seconds and an additional 100ml of water was added Stirring rushfed in a chart do bution with not sky 16 Jun 2016 no undissolved particles observed 5) Sodium Chloride was transferred to the 16 Jun 2016 12

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12 See page 11 S. Lp 11 Jun 2016 Solution under stirring using a magnetic stir-box Heating was stopped and the ablution was allowed to cool that cold durafan with 6) When Salution temporation was ~ 35°C, azilantine Hel was transferred to the Solution under Stissing. The container was then placed in an Ulforesnic bath for Iminute and Istirring was continued tilla, clear Solution resulted. 7) Benzalkonium Chloside Salution was transferred to the beaker, with Stirring. An additional 10ml of water was used to since out the vid and transferred to the beaker with Stirring 8) While stirring to create a vertex, the hyper--methore was added in small partions till all the added hypnomethou wids we that 9) Stirring was untinued 10) Ina Zoml vial, polyssibati 80, glycenin and trancingtone autonicle wern 8 20 16 Jun 2016 were weighed out or transferred. 11) The contents were mixed in a vostex mixed to obtain a smooth uniform allung: An additional SML of water was added to this aling and the contents were mixed on a Vorted mixer to obtain a uniform dispersion. 12) The aqueous Solution phase from step 9 was a chair Solution It was transformed to the 750mL beaker from Step 1. While the contents were mixed using a humogenizer, the dispersion from Step 11 was added. 13) The vial from Step11 and the beaker from Step9. wore stinsed with 10 mL portions of purified 16-Jun-2016

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water 4 times and the sinses were transferred to the main container with continued homoge -nization at 3000 spm. 14) The mixture obtained was the homogenized at 7400 rpm for 30 min 15) The homogenizing tool was traised and the preparation watt (BRp 16 Jun 2016) was allowed to Istand to allow The batch to 'de-form' 16) After 45 min, de foaming was complet. Additional pumped water was coverfully added to make--lip the final volume to 500m2. 17) Settled Solid was observed while making up the volume. The final to (8 By 16 Jun 2016) Sur Kension Was homogenized & Stoo opm. for Smin and then at 10,000 opm for 30 min. The Surperion was allowed to dederate and mixed well with a stirrer & filled into a 1000mL Capacity LDPE Rottle. 18) After the a fine Sectionent was observed which was steadily and uniformly stesuspendable Sectimentation behavior The bulk container containing the final Suspension product was then mixed will manually to uniformly suspend the Solid particle. A 100mL ph from of the Suspension I was transferred to a 100mL gradua ted cylindr and allowed to stand un disturbed I on the lab bench. The instial Suspendien appeared hozy but uniform trite Sky 16-Jun-2016. Discrete Suspended particulate matter could not be visually discended. After 10 min. There was no change in the visual And A Larphi 16 - Jun - 2016 14 37

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See page 11 S. Ry 16 Jun 2016 appeariance of the druppension. After 20 minutes the Suspension continued After 130 minutes of standing, there was some visual evidence of standing, there was some visual evidence of settled at the bottom of the cylinder. Particles After 45 minutes the presence of settled Solid was cleasly evident. The Suspension was allowed to Stand Overnight along with the bulk containor for observations the next day. 17 - Jun - 2016 After overnight storage all the containing that had the suppension still appeared hoizy with no- (32, 17-Jun-2016) settled I dolid at the Rottom of the container. In each case, gently investing the container fire to six times I subulted in the Solid being readily resuspended. The 100 mL Sexum vials containing batches 1345-004 and 1345-008 had been lift un disturbed for 3 days (12 (8 & 17-Jun 2011) 14th to 17th June 2016). The Rediment in the bott & g 17 Jun 2011 Vials also Seemed to be easily rearispendable by gently indesting the vials 5-6 times. yielding Va Uniform Euspension 8. Rp 18 Jun 2016 Manager A. La-pokt 18-Jun-2016

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15 France See page 11 Rp 17 Jun 2016 Measurement of Osmolatity 17- Jun - 2016 - Fiske One - Ten Osmo meter Asset ID 498658 - Sample. Triancinolone Acetonide (0.05%) and Azelastine HCl (0.07%) nosal spray duspension Lot # NB 1345-011 Standards used: 200 mOsm \$ 20 17-Jun - 2016 290 mosm/ky and 500 m 0 sm/kg Sample measurement 1 - 529 m 0 sm/kg Sample measurement 2 - 529 m 0 sm/kg Sam ble Sray delivery On 16-Jun 2016 after the buck suspension was mixed and transformed to the graduated cylindor for observation of the declimentation behavior, a portion of the devis pension was filled into a 160 cc char glass bottle and a Screw cap with a noval spray pump (Aptar VP-7 pum Part # 5766, Lot # M6242) was used pump to chose the bottle: The delivery of the Suspension by the Spray pump was resnally observed. and the I resultant delivery pation of the Suspension was recorded using a video Cambra. It was observed that the prepared Suspension (lot 1345-011) was delivered as a Spray Using the nasal Spray pump. Conclusion - A SODINL batch of Triam cinolone Autonia (0.05%) and azelastine HU (0.07%) noval sprany Suspension was prepared. Huniform product was obtained. Settling occurred but was steadily & uniformly representable. The product Could be delivered as a fine spray using a nasal spray pump. Osmolality was measured. 18 Jun - 2016 To Page No. END

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EXHIBIT D Video Provided via FTP

EXHIBIT E Video Provided via FTP