

PUBLICATION SCHEDULE FOR 1982

BBA is published according to a volume-numbering scheme that embraces all sections of the journal. In addition, each regular section has its own sequential volume-numbering system, running parallel to the overall BBA scheme set out below.

	BIO- ENERGETICS	BIO- MEMBRANES	GENE STRUCTURE		LIPIDS AND LIPID	GENERAL SUBJECTS	MOLECULAR
	ENERIOS		AND EXPRESSION		METABOLISM		RESEARCH
Jan.	679/1 = B58/1	684/1 = M95/1 684/2 = M95/2	696/1 = N1/1	700/1 = P1/1 700/2 = P1/2	710/1 = L58/1	714/1 = G96/1	
Feb.	679/2 = B58/2	685/1 = M96/1 685/2 = M96/2	696/2 = N1/2	701/1 = P2/1 701/2 = P2/2	710/2 = L58/2	714/2 = G96/2	720/1 = C1/1
March	679/3 = B58/3	685/3 = M96/3 686/1 = M97/1	696/3 = N1/3	701/3 = P2/3 702/1 = P3/1	710/3 = L58/3	714/3 = G96/3 715/1 = G97/1	720/2 = C1/2
April	680/1 = B59/1	686/2 = M97/2 687/1 = M98/1	697/1 = N2/1	702/2 = P3/2 703/1 = P4/1	711/1 = L59/1	715/2 = G97/2	
May	680/2 = 859/2	687/2 = M98/2 688/1 = M99/1	697/2 = N2/2	703/2 = P4/2 704/1 = P5/1	711/2 = L59/2	716/1 = G98/1	720/3 = C1/3
June	680/3 = B59/3	688/2 = M99/2 688/3 = M99/3	697/3 = N2/3	704/2 = P5/2 704/3 = P5/3	711/3 = L59/3	716/2 = G98/2 716/3 = G98/3	720/4 = C1/4
July	681/1 = B60/1	689/1 = M100/1 689/2 = M100/2	698/1 = N3/1	705/1 = P6/1 705/2 = P6/2	712/1 = L60/1	717/1 = G99/1	
Aug.	681/2 = B60/2	689/3 = M100/3 690/1 = M101/1	698/2 = N3/2	705/3 = P6/3	712/2 = L60/2	717/2 = G99/2 717/3 = G99/3	721/1 = C2/1
Sept.	681/3 = B60/3	690/2 = M101/2 691/1 = M102/1	698/3 = N3/3	706/2 = P7/2 707/1 = P8/1	712/3 = L60/3	718/1 = G100/1	721/2 = C2/2
Oct.	682/1 = B61/1	691/2 = M102/2 692/1 = M103/1	699/1 = N4/1	707/2 = P8/2 708/1 = P9/1	713/1 = L61/1	718/2 = G100/2	
Nov.	682/2 = B61/2	692/2 = M103/2 692/3 = M103/3	699/2 = N4/2	708/2 = P9/2	713/2 = L61/2	719/1 = G101/1 719/2 = G101/2	721/3 = C2/3
Dec.	682/3 = B61/3	693/1 = M104/1 693/2 = M104/2	699/3 = N4/3	709/1 = P10/1 709/2 = P10/2	713/3 = L61/3	719/3 = G101/3	721/4 = C2/4

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BBA 31265

PURIFICATION AND CHARACTERIZATION OF HUMAN CĪ-ESTERASE INHIBITOR

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Key words: CI-esterase; Disulfide; Esterase inhibitor; Complement; (Human)

A new purification method for $C\overline{1}$ -esterase inhibitor is described, which is essentially a three-step procedure: precipitation with poly(ethylene glycol), chromatography on DEAE-cellulose and hydrophobic interaction chromatography on hexyl-Sepharose. The final product is a single-chain glycoprotein with a molecular weight of about 100 000 and NH_2 -terminal asparagine. The molecule is fully active as judged by complex formation with $C\overline{1}$ s. Two of its three disulphide bridges can be easily reduced and S-carboxymethylated under non-denaturing conditions without loss of activity. However, at high dithioerythritol concentration the third disulphide bridge is also cleaved and accompanied by loss of the activity, indicating that this disulphide bridge is involved in maintaining the conformation around the reactive site in the inhibitor.

Introduction

Human plasma contains several protease inhibitors of well-known functions. Among these, the C\overline{\text{T}}-esterase inhibitor is important in the regulation of the complement system [1] and possibly also in the kininogen system [2] and the intrinsic fibrinolysis [3]. Several purification methods for the C\overline{\text{T}}-esterase inhibitor have been described [1,4-8], although few seem to give fully satisfactory preparations suitable for functional studies. In order to perform such studies we have therefore developed a simple method for purification of C\overline{\text{T}}-esterase inhibitor which can be operated on a large scale and results in a high yield of pure product, with properties as close to the native molecule as possible.

Materials and Methods

Human plasma. Freshly frozen citrated human plasma was obtained from the Blood Centre of

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Umeå University Hospital (courtesy of Dr. B. Cedergren).

CĪs. This was prepared essentially according to the method of Chapuis et al. [9]. The final product was a homogeneous protein as established by SDS-polyacrylamide gel electrophoresis, all of which could form a stable complex with CĪesterase inhibitor, if this was added in excess.

Reagents. Hexyl-Sepharose was obtained by coupling 100 ml 1 M hexylamine (Kebo AB, Stockholm, Sweden), adjusted to pH 9 with hydrochloric acid, to 1000 ml CNBr-activated Sepharose 4B [10]. Sepharose 4B® and DEAE-Sephacel® were from Pharmacia Fine Chemicals (Uppsala, Sweden). Polybrene was from Aldrich (Beerse, Belgium) and the chromogenic tri-peptide-p-nitroanilide substrate S-2400 was a gift from Kabi peptide Research (Mölndal, Sweden, courtesy of Dr. Lars Mellstam). All reagents and chemicals used were of analytical grade.

 NH_2 -terminal amino acid determination. This was performed by manual Edman degradation [11], and the phenylthiohydantoins were identified by thin-layer chromatography as described previously [12].

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Amino acid analysis. The samples were hydrolyzed in constant boiling HCl in vacuo for 24, 48 and 72 h. Analysis was performed on an LKB amino acid analyzer according to the procedures given by the manufacturers.

Immunochemical analysis. Rabbit antiserum to the Cī-esterase inhibitor was raised by injection of the purified protein mixed with Freund's complete adjuvant. Booster doses were given together with Freund's incomplete adjuvant when the specific antibody titers were decreasing.

Electroimmunoassay was performed according to the method of Laurell [13] and crossed immunoelectrophoresis essentially as described by Clarke and Freeman [14].

Determination of $C\overline{l}s$ and $C\overline{l}$ -esterase inhibitor concentrations. The $C\overline{l}s$ concentration was determined from its absorbance at 280 nm, using an $A_{280\text{nm}}^{1\%}$ of 9.5 [15].

The $\overline{\text{C1s}}$ activity was measured using the chromogenic substrate S-2400: a 140 nM solution of $\overline{\text{C1s}}$ gave a ΔA_{410} of 0.043/min at a substrate concentration of 0.63 mM in 0.1 M sodium phosphate buffer, pH 7.3, at 25°C.

CĪ-esterase inhibitor concentrations were determined spectrophotometrically using an $A_{280\text{nm}}^{1\%}$ of 5.0 [1], or by titration against purified CĪs in the following way: 25 μ l of a CĪ-esterase inhibitor sample was incubated for 10 min with 25 μ l CĪs (final concentration 140 nM) in a cuvette. Then 0.7 ml 0.1 M phosphate buffer, pH 7.3, and 200 μ l 3 mM S-2400 were added and residual CĪs activity was measured from the $\Delta A_{410\text{nm}}$, as described above. The CĪ-esterase inhibitor could then be calculated from the decrease in $\Delta A_{410\text{nm}}$.

Polyacrylamide gel electrophoresis. SDS-poly-acrylamide gel electrophoresis according to the method of Weber and Osborn [16] was performed on non-reduced and reduced samples, as previously described [17].

Reduction and S-carboxymethylation. Reduction of the C\overline{1}-esterase inhibitor was performed using a protein concentration of about 2 mg/ml. Dithioerythritol was added to final concentrations of 0.03-200 mM. For reductions at non-denaturating conditions, a 0.1 M sodium phosphate buffer, pH 7.3, was used. For reductions at denaturing conditions, the same buffer, containing 6 M guanidinium chloride, was used. After incubation for 30 min at

25°C in nitrogen athmosphere, [3H]ICH2COONa was added to a final concentration of 80 mM, and incubation was continued in the dark for another 5 min. To the 200 mM dithioerythritol sample, 450 mM ICH2COONa was used. The labelled protein was isolated by gel filtration on Sephadex G-50 columns (0.8 cm² \times 9 cm) equilibrated with 0.1 M sodium phosphate buffer, pH 7.3. Control samples were similarly treated, except that no dithioerythritol was added. The separated protein samples were analyzed by radioactivity measurement (Packard Tri-carb liquid scintillator) and protein determination by A280 or amino acid analysis after acid hydrolysis. As a reference protein, α_2 -antiplasmin (10 mg/ml) was treated in the same way in order to obtain the specific radioactivity of the iodoacetic acid, since it is known that this protein contains three disulphide bridges and no free SH group [18,19].

Results

Purification of CI-esterase inhibitor

All procedure were carried out at +5°C. About 1 litre of plasma depleted in plasminogen by treatment with lysine-Sepharose [20] was thawed, made 20 mM in EDTA, and polybrene was added to a final concentration of 0.6 g/l. The plasma was subsequently diluted with an equal volume of PEG 6000 (12%, w/v). After stirring for 30 min, the precipitate was removed by centrifugation, and to the clear supernatant were added about 300 ml of DEAE-Sephacel®, equilibrated with 0.04 M phosphate buffer, pH 7.0.

After stirring for 30 min, the gel was washed on a Büchner funnel with about 15 litres of the same phosphate buffer. The gel was then degassed and packed in a column (5×15 cm), and subsequently eluted employing a linear gradient of NaCl from 0 to 0.3 M (total volume 1800 ml) in the same phosphate buffer (Fig. 1).

The C1-esterase inhibitor concentration in the eluted fractions was determined by electroimmunoassay. Generally it was eluted in the 0.15-0.18 M NaCl range. The fractions containing antigen were pooled, concentrated to about 30 ml and made 0.4 M in (NH₄)₂SO₄ by addition of an appropriate amount of the solid substance.

The concentrated DEAE-Sephacel® pool was

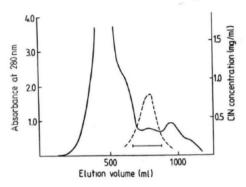


Fig. 1. Elution of the DEAE-Sephacel column by a linear gradient from 0.04 M sodium phosphate buffer, pH 7.0, to the same buffer containing 0.3 NaCl. A₂₈₀ (———); CĪ-esterase inhibitor (CIN) concentration according to Laurell electrophoresis (-----).

then applied to a column of hexyl-Sepharose (5×20 cm), equilibrated with 0.4 M (NH₄)₂SO₄ in 0.04 M phosphate buffer, pH 7.0. The CI-esterase inhibitor passed through the column unadsorbed, whereas the contaminating proteins were either adsorbed or delayed (Fig. 2).

The first peak containing the C\(\bar{l}\)-esterase inhibitor was pooled as indicated, and subsequently dialyzed against 0.1 M sodium phosphate buffer, pH 7.3, in the cold. The solution was thereafter concentrated to give a final concentration of about 3 mg protein/ml and stored frozen at -80°C. Typically about 80-100 mg of the purified C\(\bar{l}\)-

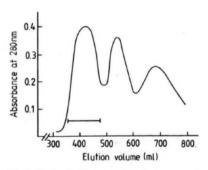


Fig. 2. Hydrophobic interaction chromatography on hexyl-Sepharose. The pooled fraction from the DEAE-Sephacel column was applied to the hexyl-Sepharose column equilibrated with 0.04 M sodium phosphate buffer containing 0.4 M (NH₄)₂SO₄. The first peak consists of essentially pure C\overline{1}-esterase inhibitor.

esterase inhibitor were obtained from 1 litre plasma.

Characterization of the purified $C\overline{I}$ -esterase inhibitor

Homogeneity. The product obtained after the hexyl-Sepharose step is typically a single-chain protein which is more than 95% pure as shown by SDS-polyacrylamide gel electrophoresis on reduced and non-reduced samples (Fig. 3). On comparison with reference proteins its molecular weight was estimated to about 100000. On crossed immunoelectrophoresis, using a rabbit antiserum raised against the purified protein, only one precipitation arc is observed for the purified CIesterase inhibitor, as well as for human pooled plasma (Fig. 4). Within experimental error the electrophoretic mobility of the pure product is identical or very similar to the native C1-esterase inhibitor in plasma. No signs of this inhibitor in complex with proteases such as Cls or Clr have been detected in normal plasma samples.

The amino acid composition (Table I) is in excellent agreement with that reported by Haupt et al. [4] for most amino acids. The only exception is tyrosine, where our figure is considerably lower. The amino acid composition reported by Harpel [1] is, however, quite different. On NH₂-terminal amino acid analysis, about 1 mol asparagine per mol protein was found. No other amino acid could be detected.

Activity. Titration of the purified $C\overline{l}$ -esterase inhibitor with $C\overline{l}$ s resulted in an inverse linear relationship between the amount of $C\overline{l}$ -esterase inhibitor added and remaining $C\overline{l}$ s activity. A 10% molar excess of the inhibitor was sufficient to abolish the $C\overline{l}$ s activity.

SDS-polyacrylamide gel electrophoretic analysis of the reaction mixture obtained using Cls in



Fig. 3. SDS-polyacrylamide gel electrophoretograms of the purified CĪ-esterase inhibitor: (a), reduced sample; (b), non-reduced sample.

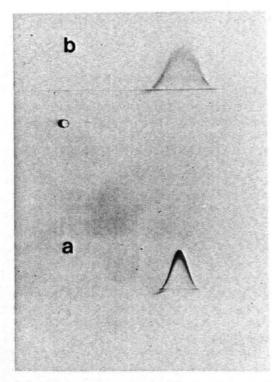


Fig. 4. Crossed immunoelectrophoresis of (a) purified $C\bar{l}$ -esterase inhibitor, (b) normal human plasma. The second dimension was run into a gel containing 1% (v/v) rabbit anti- $C\bar{l}$ -esterase inhibitor.

excess revealed that all of the C\overline{\text{I}}-esterase inhibitor reacted to form a complex with the C\overline{\text{I}}s light chain. When the C\overline{\text{I}}-esterase inhibitor was in excess all of the C\overline{\text{I}}s was converted to the enzyme-inhibitor complex, indicating that both the inhibitor and the C\overline{\text{I}}s preparations fully retain their activities.

Stability. A fresh preparation of $\overline{C1}$ -esterase inhibitor, dialyzed against 0.1 M phosphate buffer, pH 7.3, was divided into aliquots and kept at $+5^{\circ}$ C, -20° C or -80° C. Sodium azide (0.04%) was added as preservant in the $+5^{\circ}$ C aliquots. At different times, ranging between 0-50 days, samples were checked for inhibitory activity against $\overline{C1}$ s. At -80° C or at $+5^{\circ}$ C no decline in activity could be detected. However, at -20° C a small but steady decline in activity was observed, amounting

TABLE I

AMINO ACID COMPOSITION OF THE PURIFIED PREPARATION OF CĪ-ESTERASE INHIBITOR

The figures represent the mean value of three determinations.

A	μmol	Amino acid	μmol 100 mg	
Amino acid	100 mg	Amino acid		
Aspartic acid	56.5	Methionine	14.5	
Threonine a	59.9	Isoleucine	19.8	
Serine a	52.5	Leucine	62.7	
Glutamic acid	61.0	Tyrosine	8.3	
Proline	42.6	Phenylalanine	29.2	
Glycine	17.0	Histidine	9.0	
Alanine	38.0	Lysine	36.9	
Cysteine a	5.5	Arginine	16.0	

a Extrapolated values.

Valine

to about 50% during the observation period.

37.1

The disulphide bridges in $C\overline{I}$ -esterase inhibitor. Amino acid analysis has indicated six cysteines in the molecule (Table I), thus forming three disulphide bridges. This has been investigated further by reduction of the $C\overline{I}$ -esterase inhibitor with dithioerythritol in the presence of 6 M guanidinium chloride, followed by S-carboxymethylation with $[^3H]CH_2COONa$. Using α_2 -antiplasmin as a standard, known to possess three disulphide bridges [19], we found 5.9 cysteines in the $C\overline{I}$ -esterase inhibitor. The small radioactivity incorporated nonspecifically prior to reduction, corresponding to about 0.8 residues, is then already subtracted. Conclusively, the $C\overline{I}$ -esterase inhibitor contains three disulphide bridges.

Partial reduction of the C1-esterase inhibitor under non-denaturing conditions using different dithioerythritol concentrations followed by S-carboxymethylation with [3H]ICH2COONa is shown in Fig. 5. Two of the three disulphide bridges seem to be easily cleaved even at 3 mM dithioerythritol under non-denaturing conditions. However, to cleave the third disulphide bridge the dithioerythritol concentration has to be increased to 200 mM. Included in Fig. 5 are also the results from activity measurements of the reduced C1-esterase inhibitor samples. Reduction and S-carboxymethylation of the two easily reduced disulphide bridges are performed without loss of

activity. However, cleavage of the third disulphide bridge results in almost complete loss of activity (Fig. 5).

Discussion

Many of the previously published purification procedures for CI-esterase inhibitor are tedious and result in low yields and varying amounts of degraded forms of the inhibitor. The method recently published by Reboul et al. [5] is simple, but in our hands it only gave a 50%-pure material with several contaminants at low concentrations. Therefore the present method seems to be superior in its simplicity, high yield and quality of the final product. A yield of 80–100 mg/litre plasma represents about 45–55%, assuming a concentration of 180 mg/l in normal plasma [1]. The present method can, furthermore, easily be operated on a large scale.

The purified protein is a single-chain glycoprotein with a molecular weight of about 100000, and asparagine is found to be its NH_2 -terminal amino acid. The purified preparation has electrophoretic mobility in the α_2 region identical to that of the $C\overline{1}$ -esterase inhibitor in plasma, as judged by crossed immunoelectrophoresis. It is also fully active, as judged by titration against $C\overline{1}$ s or analysis of a mixture of $C\overline{1}$ -esterase inhibitor and $C\overline{1}$ s in molar excess by SDS-polyacrylamide gel electro-

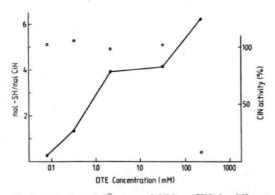


Fig. 5. Reduction of CĪ-esterase inhibitor (CIN) by different concentrations of dithioerythritol (DTE) followed by S-carboxymethylation with [3H]ICH₂COONa. The number of S-carboxymethylcysteines were calculated from the specific radioactivity (••••••). Included in the figure are also the results from activity measurements (O).

phoresis. If kept frozen at -80° C the material is stable for months, but not if frozen at -20° C. Lyophilization always results in loss of appreciable amounts of activity.

Recently we reported [19,21] that two of the three disulphide bridges in α_2 -antiplasmin could be reduced and S-carboxymethylated under nondenaturating conditions without affecting the inhibitory activity. Cleavage of the third disulphide bridge could only be performed in the presence of denaturing agents and was accompanied by complete loss of activity [19,21]. In the present study, similar results are obtained with CI-esterase inhibitor, although the third disulphide bridge in this case can be cleaved if a high concentration of dithioerythritol is used. This still leads to inactivation of the molecule, indicating that the 'stable' disulphide bridge is indeed important in maintaining the crucial conformation around the reactive site of the inhibitor. The similar behaviour of the disulphide bridges in α_2 -antiplasmin and CI-esterase inhibitor may indicate a structural relationship between these molecules. This will be further elucidated by amino acid sequence analyses, which are already in progress.

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