

## United States Patent and Trademark Office

United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS P.O. Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov UNITED STATES DEPARTMENT OF COMMERCE

APPLICATION NUMBER FILING OR 371(C) DATE FIRST NAMED APPLICANT ATTY. DOCKET NO./TITLE 13/943,395 07/16/2013 Stephen Harrison 102663.0002US5/99

> **CONFIRMATION NO. 2954** POA ACCEPTANCE LETTER

23370 KILPATRICK TOWNSEND & STOCKTON LLP Mailstop: IP Docketing - 22 1100 PEACHTREE STREET **SUITE 2800** ATLANTA, GA 30309



Date Mailed: 12/07/2017

## NOTICE OF ACCEPTANCE OF POWER OF ATTORNEY

This is in response to the Power of Attorney filed 12/04/2017.

The Power of Attorney in this application is accepted. Correspondence in this application will be mailed to the above address as provided by 37 CFR 1.33.

> Questions about the contents of this notice and the requirements it sets forth should be directed to the Office of Data Management, Application Assistance Unit, at (571) 272-4000 or (571) 272-4200 or 1-888-786-0101.

/dgela/	



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FILING OR 371(C) DATE FIRST NAMED APPLICANT ATTY. DOCKET NO./TITLE APPLICATION NUMBER 13/943,395 07/16/2013 Stephen Harrison 102663.0002US5/99

24392 FISH IP LAW, LLP 2603 Main Street Suite 1000 Irvine, CA 92614

**CONFIRMATION NO. 2954** 

**POWER OF ATTORNEY NOTICE** 

Date Mailed: 12/07/2017

## NOTICE REGARDING CHANGE OF POWER OF ATTORNEY

This is in response to the Power of Attorney filed 12/04/2017.

 The Power of Attorney to you in this application has been revoked by the assignee who has intervened as provided by 37 CFR 3.71. Future correspondence will be mailed to the new address of record(37 CFR 1.33).

> Questions about the contents of this notice and the requirements it sets forth should be directed to the Office of Data Management, Application Assistance Unit, at (571) 272-4000 or (571) 272-4200 or 1-888-786-0101.

/dgela/	
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Doc Code: PA..

Document Description: Power of Attorney

PTO/AIA/82A (07-13) Approved for use through 11/30/2014. OMB 0651-0051

Approved for use through 11/30/2014. OMB 0651-0051 U.S. Patent and Trademark Office; U.S. DEPARTMENT OF COMMERCE

Under the Paperwork Reduction Act of 1995, no persons are required to respond to a collection of information unless it displays a valid OMB control number.

# TRANSMITTAL FOR POWER OF ATTORNEY TO ONE OR MORE REGISTERED PRACTITIONERS

NOTE: This form is to be submitted with the Power of Attorney by Applicant form (PTO/AIA/82B) to identify the application to which the Power of Attorney is directed, in accordance with 37 CFR 1.5, unless the application number and filing date are identified in the Power of Attorney by Applicant form. If neither form PTO/AIA/82A nor form PTO/AIA/82B identifies the application to which the Power of Attorney is directed, the Power of Attorney will not be recognized in the application. Application Number 13/943,395 07-16-2013 Filing Date Stephen Harrison First Named Inventor Title Selective Recovery of Manganese, Lead and Zinc 1756 Art Unit PHASGE, ARUN S Examiner Name 102555-1054659 Attorney Docket Number SIGNATURE of Applicant or Patent Practitioner Signature /Chrystal A. Tomblyn/ Date (Optional) 12/1/17 Registration 76,327 Name Chrystal A. Tomblyn Number Title (if Applicant is a Attorney for Applicant juristic entity) Alger Alternative Energy, LLC Applicant Name (if Applicant is a juristic entity) NOTE: This form must be signed in accordance with 37 CFR 1.33. See 37 CFR 1.4(d) for signature requirements and certifications. If more than one applicant, use multiple forms. \*Total of \_\_\_\_\_ forms are submitted.

This collection of information is required by 37 CFR 1.131, 1.32, and 1.33. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.11 and 1.14. This collection is estimated to take 3 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. **SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.** 

Doc Code: PA.. Document Description: Power of Attorney

PTO/AIA/82B (07-13)

Approved for use through 01/31/2018. OMB 0651-0035

U.S. Patent and Trademark Office; U.S. DEPARTMENT OF COMMERCE Under the Paperwork Reduction Act of 1995, no persons are required to respond to a collection of information unless it displays a valid OMB control number

## POWER OF ATTORNEY BY APPLICANT

	y revoke all pre es below.	vious powers of attorney given	in the applicati	on identified in <u>either</u>	the attached transmittal letter or
30000000000000000000000000000000000000	A	pplication Number	000000000000000000000000000000000000000	Filing Date	
9000000000	(Note:	The boxes above may be left blar	nk if information is	s provided on form PTO/	'AIA/82A.)
$ \mathbf{V} $					as my/our attorney(s) or agent(s), and with for the application referenced in
		insmittal letter (form PTO/AIA/82A)		ve: [	with for the application referenced in
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					attorney(s) or agent(s), and to transact
		he United States Patent and Trade nittal letter (form PTO/AIA/82A) or			patent application referenced in the PTO/AIA/82C.)
	recognize or or the boxes a		address for th	ne application identif	ied in the attached transmittal
		sociated with the above-mentioned	d Customer Num	ber	
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I am the	Applicant (if the	Applicant is a juristic entity, list the	e Applicant name	in the box):	
Alg	er Alternative	Energy, LLC			
	Inventor or Join	nt Inventor (title not required below)	 )		
	Legal Represer	ntative of a Deceased or Legally In	capacitated Inver	ntor (title not required be	low)
$\square$	Assignee or Pe	rson to Whom the Inventor is Unde	er an Obligation t	o Assign (provide signer	's title if applicant is a juristic entity)
		herwise Shows Sufficient Propriets concurrently being filed with this o			
		SIGNAT	URE of Applicat	nt for Patent	
************	~~~~~~~~~~	se title is supplied below) is authoriz	ed to act on behal	If of the applicant (e.g., wh	nere the applicant is a juristic entity).
Sign		Stopphen bourrison	*****************************	Date (Optional)	July 13, 2016
Nam	8	Stephen Harrison			
Title		Chief Technology Officer, Alge			
		is form must be signed by the applic ore than one applicant, use multiple i		with 37 CFR 1.33. See 3	7 CFR 1.4 for signature requirements
Tota	l of	forms are submitted.			

This collection of information is required by 37 CFR 1.131, 1.32, and 1.33. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.11 and 1.14. This collection is estimated to take 3 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. **SEND TO: Commissioner** for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

U.S. Patent and Trademark Office; U.S. DEPARTMENT OF COMMERCE

Under the Paperwork Reduction Act of 1995, no persons are required to respond to a collection of information unless it contains a valid OMB control number.

Application Data Sheet 37 CFR 1		476	Attorney Docket Number		076165.4	900099	102555-1054659				
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Title of	Invention	Selec	tive Recovery of	Manga	nese, Lead a	and Zin	Ç				
The appl	ication data sh	l neet is pa	rt of the provisions	al or non	provisional app	lication	for which it is	being subm	itted. The	following form contains t	he
This doc	ument may be	comple	format specified b ted electronically cluded in a paper	and sub	mitted to the					CFR 1.76. onic Filing System (EFS	) or the
	•		CFR 5.2								
Por	tions or all o	of the ap	plication associ	ated wi	th this Applic	cation	Data Shee	t may fall	under a	Secrecy Order purs l electronically.)	uant to
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	Stephen				•			Harrison			
Resid	ence Infor	nation	(Select One)	⊕ US	Residency	0	Non US Re	sidency	○ Activ	ve US Military Service	
City	Benicia			State	/Province	CA	Count	ry of Resi	idence	US	
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Invent	or 3								F	Remove	
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Resid	ence Infor	nation	(Select One)	<ul><li>US</li></ul>	Residency	$\overline{}$	Non US Re	sidency	O Activ	ve US Military Service	

Application Data S	heet 37 CFR 1	76 Attorney Docket Number		<del>076165.000009</del> 102333-1034639						
A possocious a cela		Application Number 13/943,395				,395			*******	
Title of Invention Sele	ective Recovery of M	langanese	e, Lead a	and Zinc			~~~~~~~	~~~~~~		
City Milpitas	s	tate/Pro	vince	CA	Count	ry of Resi	dencė	us		~~~
Mailing Address of Inve	ntor:						·····		<del>;</del>	
Address 1	Address 1 1126 Ridgemont Drive									
Address 2										
City Milpitas				(	State/Pro	vince	CA			
Postal Code	95035			Coun	tryi	US			***************************************	
Inventor 4							R	emove		
Legal Name										
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Elizabeth						Geler				
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City San Rafael	s	tate/Pro	vince	CA	Count	ry of Resi	dence	US		
Mailing Address of Inve	ntor:									
Address 1	8 Seville Drive			*************						
Address 2							- <b>-</b>			
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All Inventors Must Be generated within this for				ormatio	n blocks	may be		Add		
Correspondence	Information	s 						<b>77-7-1</b>		
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Title of the Invention	Selective Reco	very of Ma	anganes	e, Lead	and Zinc					
Attorney Docket Numl	9 10255	5-1054	1659	Small En	tity Statu	s Claime	ed ⊠			
Application Type	Nonprovisional		***************************************						***************************************	
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Total Number of Draw	ing Sheets (if any	r) 17		T	Suggest	ted Figure	for Put	lication	(if any)	
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Annlication Da	ta Sheet 37 CFR 1.76	Attorney Docket Number	<del>076165.000000</del> <u>102555-1054659</u>						
Application Da	ta offeet of OFN 1.70	Application Number	13/943,395						
Title of Invention	Selective Recovery of Manga	nese, Lead and Zinc							
Publication I	Publication Information:								
Request Early	Publication (Fee required a	t time of Request 37 CFR 1.2	219)						
35 U.S.C. 122 subject of an a	Request Not to Publish. I hereby request that the attached application not be published under  35 U.S.C. 122(b) and certify that the invention disclosed in the attached application has not and will not be the subject of an application filed in another country, or under a multilateral international agreement, that requires publication at eighteen months after filing.								
Representativ	/e Information:								

this information in the Appli Either enter Customer Nun	epresentative information should be provided for all practitioners having a power of attorney in the application. Providing his information in the Application Data Sheet does not constitute a power of attorney in the application (see 37 CFR 1.32). Ither enter Customer Number or complete the Representative Name section below. If both sections are completed the customer number will be used for the Representative Information during processing.				
Please Select One:	Customer Number	US Patent Practitioner	Limited Recognition (37 CFR 11.9)		
Customer Number	<del>35979</del> <u>23370</u>				

## Domestic Benefit/National Stage Information:

This section allows for the applicant to either claim benefit under 35 U.S.C. 119(e), 120, 121, or 365(c) or indicate National Stage entry from a PCT application. Providing this information in the application data sheet constitutes the specific reference required by 35 U.S.C. 119(e) or 120, and 37 CFR 1.78.

Prior Application Status	Pending		Remove
Application Number	Continuity Type	Prior Application Number	Filing Date (YYYY-MM-DD)
·	Continuation of	13/539106	2012-06-29
Prior Application Status	Expired		Remove
Application Number	Continuity Type	Prior Application Number	Filing Date (YYYY-MM-DD)
3/539106	non provisional of	61/502736	2011-06-29

Additional Domestic Benefit/National Stage Data may be generated within this form by selecting the **Add** button.

## Foreign Priority Information:

This section allows for the applicant to claim priority to a foreign application. Providing this information in the application data sheet constitutes the claim for priority as required by 35 U.S.C. 119(b) and 37 CFR 1.55(d). When priority is claimed to a foreign application that is eligible for retrieval under the priority document exchange program (PDX)<sup>1</sup> the information will be used by the Office to automatically attempt retrieval pursuant to 37 CFR 1.55(h)(1) and (2). Under the PDX program, applicant bears the ultimate responsibility for ensuring that a copy of the foreign application is received by the Office from the participating foreign intellectual property office, or a certified copy of the foreign priority application is filed, within the time period specified in 37 CFR 1.55(g)(1).

Application Data Sheet 37 CFR 1.76			Attorney Docket Number	<del>076165.000000</del> <u>102555-1054659</u>		
			Application Number	13/943,39:	5	
Title of Invention	Selectiv	e Recovery of Mangal	nese, Lead and Zinc			
Application Number Coun		Country	Filing Date (YYYY	-MM-DD)	Remove Access Code <sup>i</sup> (if applicable)	
Additional Foreign  Add button.	Priority	Data may be gener	ated within this form by sel	ecting the		
·			**************************************		***************************************	

# Statement under 37 CFR 1.55 or 1.78 for AIA (First Inventor to File) Transition Applications

This application (1) claims priority to or the benefit of an application filed before March 16, 2013 and (2) also
contains, or contained at any time, a claim to a claimed invention that has an effective filing date on or after March
16, 2013.
NOTE: By providing this statement under 37 CFR 1.55 or 1.78, this application, with a filing date on or after March
16, 2013, will be examined under the first inventor to file provisions of the AIA.

## **Authorization to Permit Access:**

 Authorization to Permit Access to the Instant Application by the Participating Offices
If checked, the undersigned hereby grants the USPTO authority to provide the European Patent Office (EPO), the Japan Patent Office (JPO), the Korean Intellectual Property Office (KIPO), the World Intellectual Property Office (WIPO), and any other intellectual property offices in which a foreign application claiming priority to the instant patent application is filed access to the instant patent application. See 37 CFR 1.14(c) and (h). This box should not be checked if the applicant does not wish the EPO, JPO, KIPO, WIPO, or other intellectual property office in which a foreign application claiming priority to the instant patent application is filed to have access to the instant patent application.
In accordance with 37 CFR 1.14(h)(3), access will be provided to a copy of the instant patent application with respect to: 1) the instant patent application-as-filed; 2) any foreign application to which the instant patent application claims priority under 35 U.S.C. 119(a)-(d) if a copy of the foreign application that satisfies the certified copy requirement of 37 CFR 1.55 has been filed in the instant patent application; and 3) any U.S. application-as-filed from which benefit is sought in the instant patent application.
In accordance with 37 CFR 1.14(c), access may be provided to information concerning the date of filing this Authorization.

## **Applicant Information:**

Providing assignment information in this section does not substitute for compliance with any requirement of part 3 of Title 37 of CFR to have an assignment recorded by the Office.

Annliastian N	مغم گهمه	427 CED 476	Attorney Doc	ket Number	076165.00	0099 102333-1034039	
Application Data Sheet 37 CFR 1.76		Application N	umber	13/943,395			
Title of Invention Selective Recovery of Manganese, Lead and Zinc							
Applicant 1		· ·					
The information to be 1.43; or the name an who otherwise shows applicant under 37 C	provided in d address o sufficient p FR 1.46 (as ogether with	n this section is the na f the assignee, person proprietary interest in to signee, person to who	ime and address in to whom the in the matter who is om the inventor i	of the legal rep ventor is under the applicant us s obligated to a	resentative v an obligation Inder 37 CFF ssign, or pen	s section should not be completed who is the applicant under 37 CF in to assign the invention, or person 1.46. If the applicant is an isson who otherwise shows sufficient on are also the applicant should include the applicant should be applied to the applied	
Assignee	<del>a, janarana ana ana ana ana ana ana ana ana </del>	◯ Legal Re	epresentative un	der 35 U.S.C.	117	Joint Inventor	
Person to whom	he inventor	is obligated to assign.		O Person	who shows	sufficient proprietary interest	
If applicant is the le	gal repres	entative, indicate th	e authority to f	ile the patent a	application,	the inventor is:	
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Name of the Dece	ased or Le	gally Incapacitated	Inventor :				
If the Applicant is	an Organi	zation check here.					
Organization Nan	T		Iternative En	iergy, LLC	······································		
Mailing Address	Informati	on For Applicant:		·			
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Address 2							
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Assignee Inf	ormatio	on including	Non-Appli	cant Assi	gnee in	formation:	
Providing assignmen have an assignment			not subsitute for	compliance wit	h any require	ement of part 3 of Title 37 of CFF	
Assignee 1							
application publication	n . An assig llicant. For a	nee-applicant identifi	ed in the "Applica	ant Information"	section will	ired to be included on the patent appear on the patent application s an assignee is also desired on	
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Application Data Sheet 37 CFR 1.76		Attorney Docket Numbe	r <del>076165.000099</del> _	102555-1054659	
Application Da	ata Sheet 37 Crr 1.70	Application Number	13/943,395		
Title of Invention	Selective Recovery of Manga	nese, Lead and Zinc			
Prefix	Given Name	Middle Name	Family Name	Suffix	
Mailing Address	Information For Non-Applic	cant Assignee:			
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City	}	State/Pro	vince		
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Phone Number		Fax Numl	per		
Email Address					
Additional Assigne	e Data may be generated wit	hin this form by selecting t	he Add button.		

## Signature:

Signature	/Chrystal A. Tomblyn/	Date (YYYY-MM-DD)	2017-12-01
First Name	Chrystal A. Last Name Tomblyn	Registration Number	76,327

This collection of information is required by 37 CFR 1.76. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.14. This collection is estimated to take 23 minutes to complete, including gathering, preparing, and submitting the completed application data sheet form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

Electronic Acknowledgement Receipt					
EFS ID:	31119683				
Application Number:	13943395				
International Application Number:					
Confirmation Number:	2954				
Title of Invention:	Selective Recovery of Manganese, Lead and Zinc				
First Named Inventor/Applicant Name:	Stephen Harrison				
Customer Number:	24392				
Filer:	Chrystal A. Tomblyn/Myrla Hernandez				
Filer Authorized By:	Chrystal A. Tomblyn				
Attorney Docket Number:	102663.0002US5/99				
Receipt Date:	04-DEC-2017				
Filing Date:	16-JUL-2013				
Time Stamp:	15:45:41				
Application Type:	Utility under 35 USC 111(a)				

## **Payment information:**

Submitted with Payment	no
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## File Listing:

Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)
1	Assignee showing of ownership per 37 CFR 3.73	13943395_373c_Statement.pdf	85763 f0b82414ca3fcfcaa95bbead6ec4a6cae074 d02b	no	3
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Warnings: ESM / PETITIONER EX 1014\_11

Information:					
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2	Power of Attorney	13943395_POA.pdf	d06ea1a4f26957d09b4cecfe66f76d11880d 87d9	no	2
Warnings:					
Information:					
			810276		
3	Application Data Sheet	13943395_ADS.pdf	424f560be1e270f08e3848946f9ee45d6148 92e9	no	6
Warnings:					
Information:					
This is not an U	SPTO supplied ADS fillable form				
		Total Files Size (in bytes)	10	82777	

This Acknowledgement Receipt evidences receipt on the noted date by the USPTO of the indicated documents, characterized by the applicant, and including page counts, where applicable. It serves as evidence of receipt similar to a Post Card, as described in MPEP 503.

## New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

## New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

STATEMENT UNDER 37 CFR 3.73(c)
Applicant/Patent Owner: Alger Alternative Energy, LLC
Application No./Patent No.: 13/943,395 / 9,238,851 Filed/Issue Date: 07-16-2013 / 01-19-2016
Titled: Selective Recovery of Manganese, Lead and Zinc
Alger Alternative Energy, LLC , a Limited Liability Company
(Name of Assignee) (Type of Assignee, e.g., corporation, partnership, university, government agency, etc.)
states that, for the patent application/patent identified above, it is (choose one of options 1, 2, 3 or 4 below):
1. The assignee of the entire right, title, and interest.
2. An assignee of less than the entire right, title, and interest (check applicable box):
The extent (by percentage) of its ownership interest is%. Additional Statement(s) by the owners holding the balance of the interest <u>must be submitted</u> to account for 100% of the ownership interest.
There are unspecified percentages of ownership. The other parties, including inventors, who together own the entire right, title and interest are:
Additional Statement(s) by the owner(s) holding the balance of the interest <u>must be submitted</u> to account for the entire right, title, and interest.
3. The assignee of an undivided interest in the entirety (a complete assignment from one of the joint inventors was made). The other parties, including inventors, who together own the entire right, title, and interest are:
Additional Statement(s) by the owner(s) holding the balance of the interest <u>must be submitted</u> to account for the entire right, title, and interest.
4. The recipient, via a court proceeding or the like (e.g., bankruptcy, probate), of an undivided interest in the entirety (a complete transfer of ownership interest was made). The certified document(s) showing the transfer is attached.
The interest identified in option 1, 2 or 3 above (not option 4) is evidenced by either (choose one of options A or B below):
A. An assignment from the inventor(s) of the patent application/patent identified above. The assignment was recorded in the United States Patent and Trademark Office at Reel, Frame, or for which a copy thereof is attached.
B. A chain of title from the inventor(s), of the patent application/patent identified above, to the current assignee as follows:
1. From: Harrison et al. To: Simbol Inc.
The document was recorded in the United States Patent and Trademark Office at
Reel 031428 Frame 0328 or for which a copy thereof is attached.
2. From: Simbol Inc. To: Alger Alternative Energy, LLC
The document was recorded in the United States Patent and Trademark Office at Reel 043117 , Frame 0923 , or for which a copy thereof is attached.

[Page 1 of 2]
This collection of information is required by 37 CFR 3.73(b). The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.11 and 1.14. This collection is estimated to take 12 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

		STATEME	NT UNDER 37 CFR 3.73(	<u>C)</u>
3. From:			To:	
	The docume	nt was recorded in the	United States Patent and Trader	mark Office at
	Reel	, Frame	, or for which a copy the	ereof is attached.
4. From:			To:	
	The docume	nt was recorded in the	United States Patent and Trader	mark Office at
	Reel	, Frame	, or for which a copy the	ereof is attached.
5. From:			To:	
	The docume	nt was recorded in the	United States Patent and Trader	mark Office at
	Reel	, Frame	, or for which a copy the	ereof is attached.
6. From:			To:	
	The docume	nt was recorded in the	United States Patent and Trader	mark Office at
	Reel	, Frame	, or for which a copy the	ereof is attached.
Ac	dditional document	s in the chain of title are	e listed on a supplemental sheet	(s).
			mentary evidence of the chain o tted for recordation pursuant to 3	f title from the original owner to the 37 CFR 3.11.
				t(s)) must be submitted to Assignment cords of the USPTO. See MPEP 302.08]
The undersi	gned (whose title is	supplied below) is aut	horized to act on behalf of the a	ssignee.
/Chrystal	A. Tomblyn/			12/1/17
Signature			<del></del>	Date
Chrysta	al A. Tombly	n		76,327
Printed or Ty	yped Name			Title or Registration Number

[Page 2 of 2]

## Privacy Act Statement

The **Privacy Act of 1974 (P.L. 93-579)** requires that you be given certain information in connection with your submission of the attached form related to a patent application or patent. Accordingly, pursuant to the requirements of the Act, please be advised that: (1) the general authority for the collection of this information is 35 U.S.C. 2(b)(2); (2) furnishing of the information solicited is voluntary; and (3) the principal purpose for which the information is used by the U.S. Patent and Trademark Office is to process and/or examine your submission related to a patent application or patent. If you do not furnish the requested information, the U.S. Patent and Trademark Office may not be able to process and/or examine your submission, which may result in termination of proceedings or abandonment of the application or expiration of the patent.

The information provided by you in this form will be subject to the following routine uses:

- The information on this form will be treated confidentially to the extent allowed under the Freedom of Information Act (5 U.S.C. 552) and the Privacy Act (5 U.S.C 552a). Records from this system of records may be disclosed to the Department of Justice to determine whether disclosure of these records is required by the Freedom of Information Act.
- 2. A record from this system of records may be disclosed, as a routine use, in the course of presenting evidence to a court, magistrate, or administrative tribunal, including disclosures to opposing counsel in the course of settlement negotiations.
- 3. A record in this system of records may be disclosed, as a routine use, to a Member of Congress submitting a request involving an individual, to whom the record pertains, when the individual has requested assistance from the Member with respect to the subject matter of the record.
- 4. A record in this system of records may be disclosed, as a routine use, to a contractor of the Agency having need for the information in order to perform a contract. Recipients of information shall be required to comply with the requirements of the Privacy Act of 1974, as amended, pursuant to 5 U.S.C. 552a(m).
- 5. A record related to an International Application filed under the Patent Cooperation Treaty in this system of records may be disclosed, as a routine use, to the International Bureau of the World Intellectual Property Organization, pursuant to the Patent Cooperation Treaty.
- 6. A record in this system of records may be disclosed, as a routine use, to another federal agency for purposes of National Security review (35 U.S.C. 181) and for review pursuant to the Atomic Energy Act (42 U.S.C. 218(c)).
- 7. A record from this system of records may be disclosed, as a routine use, to the Administrator, General Services, or his/her designee, during an inspection of records conducted by GSA as part of that agency's responsibility to recommend improvements in records management practices and programs, under authority of 44 U.S.C. 2904 and 2906. Such disclosure shall be made in accordance with the GSA regulations governing inspection of records for this purpose, and any other relevant (i.e., GSA or Commerce) directive. Such disclosure shall not be used to make determinations about individuals.
- 8. A record from this system of records may be disclosed, as a routine use, to the public after either publication of the application pursuant to 35 U.S.C. 122(b) or issuance of a patent pursuant to 35 U.S.C. 151. Further, a record may be disclosed, subject to the limitations of 37 CFR 1.14, as a routine use, to the public if the record was filed in an application which became abandoned or in which the proceedings were terminated and which application is referenced by either a published application, an application open to public inspection or an issued patent.
- 9. A record from this system of records may be disclosed, as a routine use, to a Federal, State, or local law enforcement agency, if the USPTO becomes aware of a violation or potential violation of law or regulation.



## United States Patent and Trademark Office

UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS

P.O. Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov

ISSUE DATE CONFIRMATION NO. APPLICATION NO. PATENT NO. ATTORNEY DOCKET NO. 13/943,395 01/19/2016 9238851

102663.0002US5/99

2954

24392

7590

12/29/2015

FISH & TSANG LLP ROBERT D. FISH 2603 Main Street **Suite 1000** Irvine, CA 92614-6232

## **ISSUE NOTIFICATION**

The projected patent number and issue date are specified above.

## **Determination of Patent Term Adjustment under 35 U.S.C. 154 (b)**

(application filed on or after May 29, 2000)

The Patent Term Adjustment is 147 day(s). Any patent to issue from the above-identified application will include an indication of the adjustment on the front page.

If a Continued Prosecution Application (CPA) was filed in the above-identified application, the filing date that determines Patent Term Adjustment is the filing date of the most recent CPA.

Applicant will be able to obtain more detailed information by accessing the Patent Application Information Retrieval (PAIR) WEB site (http://pair.uspto.gov).

Any questions regarding the Patent Term Extension or Adjustment determination should be directed to the Office of Patent Legal Administration at (571)-272-7702. Questions relating to issue and publication fee payments should be directed to the Application Assistance Unit (AAU) of the Office of Data Management (ODM) at (571)-272-4200.

APPLICANT(s) (Please see PAIR WEB site http://pair.uspto.gov for additional applicants):

SIMBOL Inc., Pleasanton, CA; Stephen Harrison, Benicia, CA; Samaresh Mohanta, Pleasanton, CA; C.V. Krishnamohan Sharma, Milpitas, CA; Elizabeth Geler, San Rafael, CA;

The United States represents the largest, most dynamic marketplace in the world and is an unparalleled location for business investment, innovation, and commercialization of new technologies. The USA offers tremendous resources and advantages for those who invest and manufacture goods here. Through SelectUSA, our nation works to encourage and facilitate business investment. To learn more about why the USA is the best country in the world to develop technology, manufacture products, and grow your business, visit <u>SelectUSA.gov</u>.

### PART B - FEE(S) TRANSMITTAL

### Complete and send this form, together with applicable fee(s), to: Mail Mail Stop ISSUE FEE

09/08/2015

Commissioner for Patents P.O. Box 1450 Alexandria, Virginia 22313-1450

(571)-273-2885 or <u>Fax</u>

INSTRUCTIONS: This form should be used for transmitting the ISSUE FEE and PUBLICATION FEE (if required). Blocks 1 through 5 should be completed where appropriate. All further correspondence including the Patent, advance orders and notification of maintenance fees will be mailed to the current correspondence address as indicated unless corrected below or directed otherwise in Block 1, by (a) specifying a new correspondence address; and/or (b) indicating a separate "FEE ADDRESS" for maintenance fee notifications.

CURRENT CORRESPONDENCE ADDRESS (Note: Use Block 1 for any change of address)

Note: A certificate of mailing can only be used for domestic mailings of the Fee(s) Transmittal. This certificate cannot be used for any other accompanying papers. Each additional paper, such as an assignment or formal drawing, must have its own certificate of mailing or transmission.

24392 7590 FISH & TSANG LLP ROBERT D. FISH 2603 Main Street **Suite 1000** 

Certificate of Mailing or Transmission

I hereby certify that this Fee(s) Transmittal is being deposited with the United States Postal Service with sufficient postage for first class mail in an envelope addressed to the Mail Stop ISSUE FEE address above, or being facsimile transmitted to the USPTO (571) 273-2885, on the date indicated below.

Marutzzella Castro	(Depositor's name
/Marutzzella Castro/	(Signature
December 8, 2015	(Date

Irvine, CA 9261	4-6232		1 /	Marutzzena	Castro	1	(Signature)
			Ī	December 8, 1	2015		(Date)
APPLICATION NO.	FILING DATE		FIRST NAMED INVENTO	OR ATTORNEY DO		NEY DOCKET NO.	CONFIRMATION NO.
13/943,395 07/16/2013			Stephen Harrison		1026	63.0002US5/99	2954
TITLE OF INVENTION	N: Selective Recovery of	Manganese, Lead and Zir	nc				
APPLN. TYPE	ENTITY STATUS	ISSUE FEE DUE	PUBLICATION FEE DUE	PREV. PAID ISSU	JE FEE	TOTAL FEE(S) DUE	DATE DUE
nonprovisional	SMALL	\$480	\$0	\$0		\$480	12/08/2015
EXAM	MINER	ART UNIT	CLASS-SUBCLASS	7			
	, ARUN S	1756	205-560000	_			
	ence address or indication		2. For printing on the	natant front nago. I	int		
CFR 1.363).		,	(1) The names of up	to 3 registered pate		ys 1 <u>Fish 8</u>	k Tsang, LLP
☐ Change of corresp Address form PTO/S	oondence address (or Cha B/122) attached.	inge of Correspondence	or agents OR, alternat		a mambai	2	
☐ "Fee Address" indication (or "Fee Address" Indication form PTO/SB/47; Rev 03-02 or more recent) attached. Use of a Customer			(2) The name of a single firm (having as a member a registered attorney or agent) and the names of up to 2 registered patent attorneys or agents. If no name is listed, no name will be printed.				
Number is required.		A TO BE PRINTED ON	THE PATENT (print or ty	1			
					nee is ide	ntified below, the do	ocument has been filed for
		pletion of this form is NO					
(A) NAME OF ASSI			(B) RESIDENCE: (CIT	Y and STATE OR	COUNTR	(1)	
Simbol, In	ic.		Pleasan	ton, CA			
Please check the appropr	riate assignee category or	categories (will not be p	rinted on the patent):	Individual 🗚	orporation	n or other private gro	oup entity 🔲 Government
4a. The following fee(s)	are submitted:	4	b. Payment of Fee(s): (Ple		my previo	ously paid issue fee	shown above)
XX ssue Fee	No small entity discount [	: <del>***</del> -4\	A check is enclosed.  Payment by credit ca		0 :441-		
	of Copies						iciency, or credits any
			overpayment, to Dep	osit Account Numb	për 5003	(énclose a	iciency, or credits any nextra copy of this form).
5. Change in Entity Sta	tus (from status indicate	d above)					
	ng micro entity status. Se		NOTE: Absent a valid c fee payment in the micro	ertification of Micro entity amount wil	o Entity S I not be ac	Status (see forms PTC eccepted at the risk of	D/SB/15A and 15B), issue application abandonment.
	g small entity status. See	NOTE: If the application was previously under micro entity status, checking this box will be taken to be a notification of loss of entitlement to micro entity status.					
	ng to regular undiscounte		entity status, as applicab	ole.			tlement to small or micro
NOTE: This form must b	oe signed in accordance v	with 37 CFR 1.31 and 1.3	3. See 37 CFR 1.4 for sign	nature requirements	and certi	fications.	
Authorized Signature	/Martin Fesse	nmaier/		DateI	Decem	ber 8, 2015	
Typed or printed nam	e Martin Fess	enmaier		Registration 1	No	466	97

#### Page 2 of 3 ESM / PETITIONER EX 1014 17

Electronic Patent Application Fee Transmittal							
Application Number:	139	943395					
Filing Date:	16-	16-Jul-2013					
Title of Invention:	Selective Recovery of Manganese, Lead and Zinc						
First Named Inventor/Applicant Name:	Stephen Harrison						
Filer:	Martin Fessenmaier/Marutzzella Castro						
Attorney Docket Number:	102	2663.0002US5/99					
Filed as Large Entity							
Filing Fees for Utility under 35 USC 111(a)							
Description		Fee Code	Quantity	Amount	Sub-Total in USD(\$)		
Basic Filing:							
Pages:							
Claims:							
Miscellaneous-Filing:							
Petition:							
Patent-Appeals-and-Interference:							
Post-Allowance-and-Post-Issuance:							

Description	Fee Code	Quantity	Amount	Sub-Total in USD(\$)		
Utility Appl Issue Fee	1501	1	960	960		
Publ. Fee- Early, Voluntary, or Normal	1504	1	0	0		
Extension-of-Time:						
Miscellaneous:						
	Tot	al in USD	(\$)	960		

Electronic Acknowledgement Receipt			
EFS ID:	24301588		
Application Number:	13943395		
International Application Number:			
Confirmation Number:	2954		
Title of Invention:	Selective Recovery of Manganese, Lead and Zinc		
First Named Inventor/Applicant Name:	Stephen Harrison		
Customer Number:	24392		
Filer:	Martin Fessenmaier/Marutzzella Castro		
Filer Authorized By:	Martin Fessenmaier		
Attorney Docket Number:	102663.0002US5/99		
Receipt Date:	08-DEC-2015		
Filing Date:	16-JUL-2013		
Time Stamp:	19:08:40		
Application Type:	Utility under 35 USC 111(a)		

## **Payment information:**

Submitted with Payment	yes
Payment Type	Deposit Account
Payment was successfully received in RAM	\$960
RAM confirmation Number	17717
Deposit Account	500341
Authorized User	CASTRO, MARUTZZELLA

The Director of the USPTO is hereby authorized to charge indicated fees and credit any overpayment as follows:

Charge any Additional Fees required under 37 C.F.R. Section 1.16 (National application filing, search, and examination fees)

Charge any Additional Fees required under 37 C.F.R. Section 1.17 (Patent application and recognition propersing feet) 20

Charge any Additional Fees required under 37 C.F.R. Section 1.19 (Document supply fees)

Charge any Additional Fees required under 37 C.F.R. Section 1.20 (Post Issuance fees)

Charge any Additional Fees required under 37 C.F.R. Section 1.21 (Miscellaneous fees and charges)

## File Listing:

Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)
1	Issue Fee Payment (PTO-85B)	PTOL-85.pdf	101163	no	1
'	issue ree rayment (rro 656)	1 10E 03.pai	90012680125fec4e71440dcd269d80d80d6 48f8b	110	'
Warnings:					
Information:					
2	Fee Worksheet (SB06)	fee-info.pdf	32109	no	2
	ree worksheet (5500)	rec imo.pui	79fd02a8c60d394c3b1c2abd4b57a64d65b 1e57c	110	-
Warnings:					
Information:					
		Total Files Size (in bytes)	13	33272	

This Acknowledgement Receipt evidences receipt on the noted date by the USPTO of the indicated documents, characterized by the applicant, and including page counts, where applicable. It serves as evidence of receipt similar to a Post Card, as described in MPEP 503.

### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

UNITED STATES PATENT AND TRADEMARK OFFICE COMMISSIONER FOR PATENTS P.O.BOX 1450 ALEXANDRIA VA 22313-1451 PRESORTED FIRST-CLASS MAIL U.S. POSTAGE PAID POSTEDIGITAL NNNNN

FISH & TSANG LLP ROBERT D. FISH 2603 Main Street Suite 1000 Irvine, CA 92614-6232



## Courtesy Reminder for Application Serial No: 13/943,395

Attorney Docket No: 102663.0002US5/99

Customer Number: 24392

Date of Electronic Notification: 09/08/2015

This is a courtesy reminder that new correspondence is available for this application. If you have not done so already, please review the correspondence. The official date of notification of the outgoing correspondence will be indicated on the form PTOL-90 accompanying the correspondence.

An email notification regarding the correspondence was sent to the following email address(es) associated with your customer number:

rfish@fishiplaw.com patents@fishiplaw.com

To view your correspondence online or update your email addresses, please visit us anytime at https://sportal.uspto.gov/secure/myportal/privatepair. If you have any questions, please email the Electronic Business Center (EBC) at EBC@uspto.gov or call 1-866-217-9197.

UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS P.O. Box 1450 Alexandria, Virginia 22313-1450

## NOTICE OF ALLOWANCE AND FEE(S) DUE

FISH & TSANG LLP ROBERT D. FISH 2603 Main Street Suite 1000 Irvine, CA 92614-6232 09/08/2015

EXAMINER

PHASGE, ARUN S

ART UNIT PAPER NUMBER

1756

DATE MAILED: 09/08/2015

APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
13/943,395	07/16/2013	Stephen Harrison	102663.0002US5/99	2954

TITLE OF INVENTION: Selective Recovery of Manganese, Lead and Zinc

APPLN. TYPE	ENTITY STATUS	ISSUE FEE DUE	PUBLICATION FEE DUE	PREV. PAID ISSUE FEE	TOTAL FEE(S) DUE	DATE DUE
nonprovisional	SMALL	\$480	\$0	\$0	\$480	12/08/2015

THE APPLICATION IDENTIFIED ABOVE HAS BEEN EXAMINED AND IS ALLOWED FOR ISSUANCE AS A PATENT. PROSECUTION ON THE MERITS IS CLOSED. THIS NOTICE OF ALLOWANCE IS NOT A GRANT OF PATENT RIGHTS. THIS APPLICATION IS SUBJECT TO WITHDRAWAL FROM ISSUE AT THE INITIATIVE OF THE OFFICE OR UPON PETITION BY THE APPLICANT. SEE 37 CFR 1.313 AND MPEP 1308.

THE ISSUE FEE AND PUBLICATION FEE (IF REQUIRED) MUST BE PAID WITHIN <u>THREE MONTHS</u> FROM THE MAILING DATE OF THIS NOTICE OR THIS APPLICATION SHALL BE REGARDED AS ABANDONED. <u>THIS STATUTORY PERIOD CANNOT BE EXTENDED.</u> SEE 35 U.S.C. 151. THE ISSUE FEE DUE INDICATED ABOVE DOES NOT REFLECT A CREDIT FOR ANY PREVIOUSLY PAID ISSUE FEE IN THIS APPLICATION. IF AN ISSUE FEE HAS PREVIOUSLY BEEN PAID IN THIS APPLICATION (AS SHOWN ABOVE), THE RETURN OF PART B OF THIS FORM WILL BE CONSIDERED A REQUEST TO REAPPLY THE PREVIOUSLY PAID ISSUE FEE TOWARD THE ISSUE FEE NOW DUE.

#### HOW TO REPLY TO THIS NOTICE:

I. Review the ENTITY STATUS shown above. If the ENTITY STATUS is shown as SMALL or MICRO, verify whether entitlement to that entity status still applies.

If the ENTITY STATUS is the same as shown above, pay the TOTAL FEE(S) DUE shown above.

If the ENTITY STATUS is changed from that shown above, on PART B - FEE(S) TRANSMITTAL, complete section number 5 titled "Change in Entity Status (from status indicated above)".

For purposes of this notice, small entity fees are 1/2 the amount of undiscounted fees, and micro entity fees are 1/2 the amount of small entity fees

II. PART B - FEE(S) TRANSMITTAL, or its equivalent, must be completed and returned to the United States Patent and Trademark Office (USPTO) with your ISSUE FEE and PUBLICATION FEE (if required). If you are charging the fee(s) to your deposit account, section "4b" of Part B - Fee(s) Transmittal should be completed and an extra copy of the form should be submitted. If an equivalent of Part B is filed, a request to reapply a previously paid issue fee must be clearly made, and delays in processing may occur due to the difficulty in recognizing the paper as an equivalent of Part B.

III. All communications regarding this application must give the application number. Please direct all communications prior to issuance to Mail Stop ISSUE FEE unless advised to the contrary.

IMPORTANT REMINDER: Utility patents issuing on applications filed on or after Dec. 12, 1980 may require payment of maintenance fees. It is patentee's responsibility to ensure timely payment of maintenance fees when due.

#### PART B - FEE(S) TRANSMITTAL

## Complete and send this form, together with applicable fee(s), to: Mail Mail Stop ISSUE FEE

Commissioner for Patents P.O. Box 1450

Alexandria, Virginia 22313-1450 (571)-273-2885 or <u>Fax</u>

INSTRUCTIONS: This form should be used for transmitting the ISSUE FEE and PUBLICATION FEE (if required). Blocks 1 through 5 should be completed where appropriate. All further correspondence including the Patent, advance orders and notification of maintenance fees will be mailed to the current correspondence address as indicated unless corrected below or directed otherwise in Block 1, by (a) specifying a new correspondence address; and/or (b) indicating a separate "FEE ADDRESS" for maintenance fee sufficient in the current correspondence address; and/or (b) indicating a separate "FEE ADDRESS" for

Note: A certificate of mailing can only be used for domestic mailings of the

CURRENT CORRESPOND	ENCE ADDRESS (Note: Use B	lock 1 for any change of addre	ss) pap	ers. Each additional pa		or any other accompanying nt or formal drawing, must
FISH & TSAN ROBERT D. FI	IG LLP SH	3/2015	I he Stat add tran	ereby certify that this E	ate of Mailing or Trans ee(s) Transmittal is being sufficient postage for firs op ISSUE FEE address (571) 273-2885, on the da	mission g deposited with the United st class mail in an envelope above, or being facsimile ate indicated below.
2603 Main Stree Suite 1000	ા					(Depositor's name)
Irvine, CA 9261	4-6232					(Signature)
						(Date)
APPLICATION NO.	FILING DATE		FIRST NAMED INVENTOR	AT AT	TORNEY DOCKET NO.	CONFIRMATION NO.
13/943,395	07/16/2013		Stephen Harrison	•	102663.0002US5/99	2954
TITLE OF INVENTION	7: Selective Recovery of	Manganese, Lead and 2	•		_ <b>_</b>	
APPLN. TYPE	ENTITY STATUS	ISSUE FEE DUE	PUBLICATION FEE DUE	PREV. PAID ISSUE FE	E TOTAL FEE(S) DUE	DATE DUE
nonprovisional	SMALL	\$480	\$0	\$0	\$480	12/08/2015
1. Change of correspond CFR 1.363).  Change of correspond CFR 1.363).  Change of correspond CFR 1.363.  The Address form PTO/S PTO/SB/47; Rev 03-6	, ARUN S ence address or indication on dence address (or Chab/122) attached. lication (or "Fee Address 12 or more recent) attached. IND RESIDENCE DATA less an assignee is ident thin 37 CFR 3.11. Com	"Indication form ed. Use of a Customer A TO BE PRINTED O	(1) The names of up to or agents OR, alternati (2) The name of a sing registered attorney or	o 3 registered patent att vely, de firm (having as a me agent) and the names o orneys or agents. If no n printed. pe) patent. If an assignee is assignment.	mber a f up to ame is 3	ocument has been filed for
4a. The following fee(s)  Issue Fee	are submitted:  No small entity discount p	<u> </u>	e printed on the patent):  4b. Payment of Fee(s): (Ple:  A check is enclosed.  Payment by credit can  The director is hereby overpayment, to Depo	ase first reapply any p rd. Form PTO-2038 is a	reviously paid issue fee ttached. te required fee(s), any de	·
	tus (from status indicate ng micro entity status. Se g small entity status. See	ee 37 CFR 1.29	fee payment in the micro	entity amount will not was previously under i	be accepted at the risk of nicro entity status, check	O/SB/15A and 15B), issue application abandonment.
Applicant changing	ng to regular undiscounte	d fee status.			•	tlement to small or micro

Authorized Signature Date

NOTE: This form must be signed in accordance with 37 CFR 1.31 and 1.33. See 37 CFR 1.4 for signature requirements and certifications.

Typed or printed name Registration No.

Page 2 of 3

entity status, as applicable

ESM / PETITIONER EX 1014 24

PTOL-85 Part B (10-13) Approved for use through 10/31/2013.

OMB 0651-0033

U.S. Patent and Trademark Office; U.S. DEPARTMENT OF COMMERCE



## UNITED STATES PATENT AND TRADEMARK OFFICE

UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS

P.O. Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov

APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
13/943,395	07/16/2013	Stephen Harrison	102663.0002US5/99	2954
24392 75	90 09/08/2015		EXAM	INER
FISH & TSANG		PHASGE, ARUN S		
ROBERT D. FISH 2603 Main Street			ART UNIT	PAPER NUMBER
Suite 1000			1756	
Irvine, CA 92614-6	5232		DATE MAILED: 09/08/201	5

## Determination of Patent Term Adjustment under 35 U.S.C. 154 (b)

(Applications filed on or after May 29, 2000)

The Office has discontinued providing a Patent Term Adjustment (PTA) calculation with the Notice of Allowance.

Section 1(h)(2) of the AIA Technical Corrections Act amended 35 U.S.C. 154(b)(3)(B)(i) to eliminate the requirement that the Office provide a patent term adjustment determination with the notice of allowance. See Revisions to Patent Term Adjustment, 78 Fed. Reg. 19416, 19417 (Apr. 1, 2013). Therefore, the Office is no longer providing an initial patent term adjustment determination with the notice of allowance. The Office will continue to provide a patent term adjustment determination with the Issue Notification Letter that is mailed to applicant approximately three weeks prior to the issue date of the patent, and will include the patent term adjustment on the patent. Any request for reconsideration of the patent term adjustment determination (or reinstatement of patent term adjustment) should follow the process outlined in 37 CFR 1.705.

Any questions regarding the Patent Term Extension or Adjustment determination should be directed to the Office of Patent Legal Administration at (571)-272-7702. Questions relating to issue and publication fee payments should be directed to the Customer Service Center of the Office of Patent Publication at 1-(888)-786-0101 or (571)-272-4200.

### OMB Clearance and PRA Burden Statement for PTOL-85 Part B

The Paperwork Reduction Act (PRA) of 1995 requires Federal agencies to obtain Office of Management and Budget approval before requesting most types of information from the public. When OMB approves an agency request to collect information from the public, OMB (i) provides a valid OMB Control Number and expiration date for the agency to display on the instrument that will be used to collect the information and (ii) requires the agency to inform the public about the OMB Control Number's legal significance in accordance with 5 CFR 1320.5(b).

The information collected by PTOL-85 Part B is required by 37 CFR 1.311. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.14. This collection is estimated to take 12 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, Virginia 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, Virginia 22313-1450. Under the Paperwork Reduction Act of 1995, no persons are required to respond to a collection of information unless it displays a valid OMB control number.

## **Privacy Act Statement**

The Privacy Act of 1974 (P.L. 93-579) requires that you be given certain information in connection with your submission of the attached form related to a patent application or patent. Accordingly, pursuant to the requirements of the Act, please be advised that: (1) the general authority for the collection of this information is 35 U.S.C. 2(b)(2); (2) furnishing of the information solicited is voluntary; and (3) the principal purpose for which the information is used by the U.S. Patent and Trademark Office is to process and/or examine your submission related to a patent application or patent. If you do not furnish the requested information, the U.S. Patent and Trademark Office may not be able to process and/or examine your submission, which may result in termination of proceedings or abandonment of the application or expiration of the patent.

The information provided by you in this form will be subject to the following routine uses:

- 1. The information on this form will be treated confidentially to the extent allowed under the Freedom of Information Act (5 U.S.C. 552) and the Privacy Act (5 U.S.C 552a). Records from this system of records may be disclosed to the Department of Justice to determine whether disclosure of these records is required by the Freedom of Information Act.
- 2. A record from this system of records may be disclosed, as a routine use, in the course of presenting evidence to a court, magistrate, or administrative tribunal, including disclosures to opposing counsel in the course of settlement negotiations.
- 3. A record in this system of records may be disclosed, as a routine use, to a Member of Congress submitting a request involving an individual, to whom the record pertains, when the individual has requested assistance from the Member with respect to the subject matter of the record.
- 4. A record in this system of records may be disclosed, as a routine use, to a contractor of the Agency having need for the information in order to perform a contract. Recipients of information shall be required to comply with the requirements of the Privacy Act of 1974, as amended, pursuant to 5 U.S.C. 552a(m).
- 5. A record related to an International Application filed under the Patent Cooperation Treaty in this system of records may be disclosed, as a routine use, to the International Bureau of the World Intellectual Property Organization, pursuant to the Patent Cooperation Treaty.
- 6. A record in this system of records may be disclosed, as a routine use, to another federal agency for purposes of National Security review (35 U.S.C. 181) and for review pursuant to the Atomic Energy Act (42 U.S.C.
- 7. A record from this system of records may be disclosed, as a routine use, to the Administrator, General Services, or his/her designee, during an inspection of records conducted by GSA as part of that agency's responsibility to recommend improvements in records management practices and programs, under authority of 44 U.S.C. 2904 and 2906. Such disclosure shall be made in accordance with the GSA regulations governing inspection of records for this purpose, and any other relevant (i.e., GSA or Commerce) directive. Such disclosure shall not be used to make determinations about individuals.
- 8. A record from this system of records may be disclosed, as a routine use, to the public after either publication of the application pursuant to 35 U.S.C. 122(b) or issuance of a patent pursuant to 35 U.S.C. 151. Further, a record may be disclosed, subject to the limitations of 37 CFR 1.14, as a routine use, to the public if the record was filed in an application which became abandoned or in which the proceedings were terminated and which application is referenced by either a published application, an application open to public inspection or an issued patent.
- 9. A record from this system of records may be disclosed, as a routine use, to a Federal, State, or local law enforcement agency, if the USPTO becomes aware of a violation or potential violation of law or regulation. ESM/PETITIONER EX 1014\_26

	Application No. 13/943,395	Applicant(s) HARRISON E						
Notice of Allowability	Examiner	Art Unit	AIA (First Inventor to					
	ARUN S. PHASGE	1756	File) Status No					
The MAILING DATE of this communication appears on the cover sheet with the correspondence address All claims being allowable, PROSECUTION ON THE MERITS IS (OR REMAINS) CLOSED in this application. If not included herewith (or previously mailed), a Notice of Allowance (PTOL-85) or other appropriate communication will be mailed in due course. THIS  NOTICE OF ALLOWABILITY IS NOT A GRANT OF PATENT RIGHTS. This application is subject to withdrawal from issue at the initiative  of the Office or upon petition by the applicant. See 37 CFR 1.313 and MPEP 1308.								
1. X This communication is responsive to <u>terminal disclaimer filed</u>	<u>ł 4/14/15</u> .							
	A declaration(s)/affidavit(s) under <b>37 CFR 1.130(b)</b> was/were filed on  An election was made by the applicant in response to a restriction requirement set forth during the interview on; the restriction							
<ol> <li>The allowed claim(s) is/are <u>32-47</u>. As a result of the allowed Highway program at a participating intellectual property offic <u>http://www.uspto.gov/patents/init_events/pph/index.jsp</u> or ser</li> </ol>	claim(s), you may be eligible to bendered	For more inforr						
4. Acknowledgment is made of a claim for foreign priority under	35 U.S.C. § 119(a)-(d) or (f).							
Certified copies:  a) ☐ All b) ☐ Some *c) ☐ None of the:  1. ☐ Certified copies of the priority documents have 2. ☐ Certified copies of the priority documents have 3. ☐ Copies of the certified copies of the priority doc  International Bureau (PCT Rule 17.2(a)).  * Certified copies not received:	been received in Application No		pplication from the					
Applicant has THREE MONTHS FROM THE "MAILING DATE" of noted below. Failure to timely comply will result in ABANDONMI THIS THREE-MONTH PERIOD IS NOT EXTENDABLE.		complying with	the requirements					
5. CORRECTED DRAWINGS ( as "replacement sheets") must	be submitted.							
including changes required by the attached Examiner's Paper No./Mail Date	Amendment / Comment or in the Or	ffice action of						
Identifying indicia such as the application number (see 37 CFR 1.6 each sheet. Replacement sheet(s) should be labeled as such in the			not the back) of					
<ol> <li>DEPOSIT OF and/or INFORMATION about the deposit of BI attached Examiner's comment regarding REQUIREMENT FO</li> </ol>	OLOGICAL MATERIAL must be sub	omitted. Note th	ne					
Attachment(s)  1. Notice of References Cited (PTO-892)  2. Information Disclosure Statements (PTO/SB/08), Paper No./Mail Date  3. Examiner's Comment Regarding Requirement for Deposit of Biological Material  4. Interview Summary (PTO-413), Paper No./Mail Date	5. ☐ Examiner's Amendn 6. ☑ Examiner's Stateme 7. ☐ Other							
/ARUN S PHASGE/ Primary Examiner, Art Unit 1756								

U.S. Patent and Trademark Office PTOL-37 (Rev. 08-13)

Notice of Allowability

Part of Paper No./Mail Date 20150810

Application/Control Number: 13/943,395 Page 2

Art Unit: 1756

## **REASONS FOR ALLOWANCE**

The following is an examiner's statement of reasons for allowance: none of the prior art of record fairly disclose or render obvious the claimed method for recovering zinc, or manganese or both from geothermal brine with the steps recited in the claims.

Any comments considered necessary by applicant must be submitted no later than the payment of the issue fee and, to avoid processing delays, should preferably accompany the issue fee. Such submissions should be clearly labeled "Comments on Statement of Reasons for Allowance."

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Arun S. Phasge whose telephone number is (571) 272-1345. The examiner can normally be reached on MONDAY-THURSDAY, 7:30-6:00.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Keith D. Hendricks can be reached on (571) 272-1401. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9306.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR.

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Art Unit: 1756

Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <a href="http://pair-direct.uspto.gov">http://pair-direct.uspto.gov</a>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

/ARUN S PHASGE/ Primary Examiner, Art Unit 1724

## Search Notes

Application/Control No.		Applicant(s)/Patent Under Reexamination
	13943395	HARRISON ET AL.
	Examiner	Art Unit
	ARUN S PHASGE	1756

CPC- SEARCHED		
Symbol	Date	Examiner
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Symbol	Date	Examiner		

	US CLASSIFICATION SEARCHE	:D	
Class	Subclass	Date	Examiner
205	539, 540, 573, 602-608	12/22/14	asp

SEARCH NOTE	:S	
Search Notes	Date	Examiner
inventor name searched in eDAN	12/22/14	asp
WEST his attached	12/24/14	asp
WESt his attached	8/10/15	asp

	INTERFERENCE SEARCH		
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## WEST Search History for Application 13943395

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	(Geothermal.clm. and zinc.clm.) and silica.clm.	PGPB	OR	YES		08- 10- 2015
	(Geothermal.cim.) and manganese.cim.	PGP8	OR	YES		08- 10- 2015
	(Geothermal.clm. and manganese.clm.) and silica.clm.	PGPB	OR	YES		08- 10- 2015
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	c22b47/00.cpc. or c22b19/00.cpc. or c25c1/10.cpc. or c25c1/16.cpc.	PGPB, USPT, USOC, EPAB, JPAB, DWPI, TDBD, FPRS	OR	YES		12- 23- 2014
	(c22b47/00.cpc. or c22b19/00.cpc. or c25c1/10.cpc. or c25c1/16.cpc. ) and (manganese and zinc)	PGPB, USPT, USOC, EPAB, JPAB, DWPI, TDBO, FPRS	OR	YES		12- 23- 2014
	(c22b47/00.cpc, or c22b19/00.cpc, or c25c1/10.cpc, or c25c1/16.cpc, ) and (manganese)	PGPB, USPT, USOC, EPAB, JPAB, DWPI, TDBD, FPRS	OR	YES		12- 23- 2014
	(c22b47/00.cpc, or c22b19/00.cpc, or c25c1/10.cpc, or c25c1/16.cpc, and (manganese) ) and zinc	PGPB, USPT, USOC, EPAB, JPAB, DWPI, TDBD, FPRS	OR	YES		12- 23- 2014
	(c22b47/00.cpc, or c22b19/00.cpc, or c25c1/10.cpc, or c25c1/16.cpc, and (manganese) and zinc) and silica	PGPB, USPT, USOC, EPAB, JPAB, DWPI, TDBO, FPRS	OR	YES		12- 23- 2014
	manganese or	PGPB, USPT, USOC, EPAB, JPAB, DWPI, TDBD, FPRS	OR	YES		12- 23- 2014
	manganese or mn	PGPB, USPT, USOC, EPAB, JPAB, DWPI, TDBD, FPAS	OR	YES		12- 23- 2014
	(manganese or mn ) and (zinc or zn)	PGPB, USPT, USOC, EPAB, JPAB, DWPI, TDBD, FPRS	OR	YES		12- 23- 2014
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	(manganese or mn and (zinc or zn) ) and silica	EPAB, JPAB, DWPI, TDBD, FPRS	OR	YES	23- 2014
	(manganese or mn and (zinc or zn) and silica) and brine	TDBD, FPRS	OR	YES	12- 23- 2014
	(manganese or mn and (zinc or zn) and	PGPB, USPT, USOC, EPAB, JPAB, DWPI, TDBO, FPRS	OR	YES	12- 23- 2014
		PGPB, USPT, USOC, EPAB, JPAB, DWPI, TDBD, FPRS	OR	YES	12- 23- 2014

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# Index of Claims 13943395 Examiner Application/Control No. Applicant(s)/Patent Under Reexamination HARRISON ET AL. Art Unit 1756

<b>✓</b>	Rejected	-	Cancelled	N	Non-Elected	A	Appeal
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	Application/Control No.	Applicant(s)/Patent Under Reexamination
Index of Claims	13943395	HARRISON ET AL.
	Examiner	Art Unit
	ARUN S PHASGE	1756

<b>✓</b>	Rejected	-	Cancelled	N	Non-Elected	Α	Appeal
=	Allowed	÷	Restricted	I	Interference	0	Objected

Claims	renumbered	in the same	order as prese	ented by a	ppiicant		☐ CPA	□ Т.0	D. 🗆	R.1.47
CL	AIM					DATE				
Final	Original	12/22/2014	08/10/2015							
	37	<b>√</b>	=							
	38	<b>√</b>	=							
	39	✓	=							
	40	✓	=							
	41	✓	=							
	42	✓	=							
	43	✓	=							
	44	✓	=							
	45	✓	=							
	46	✓	=							
	47	<b>√</b>	=							

## Issue Classification



13943395

HARRISON ET AL.

Applicant(s)/Patent Under Reexamination

Examiner

ARUN S PHASGE

Art Unit

1756

CPC				
Symbol			Туре	Version
C22B	47	/ 00	F	2013-01-01
C22B	19	/ 00	I	2013-01-01

CPC Combination Sets										
Symbol	Туре	Set	Ranking	Version						

NONE	Total Claims Allowed:				
(Assistant Examiner)	(Date)	16			
/ARUN S PHASGE/ Primary Examiner.Art Unit 1756	8/10/15	O.G. Print Claim(s)	O.G. Print Figure		
(Primary Examiner)	(Date)	1, 10	1, 2		

## Issue Classification



Application/Control No.	Applicant(s)/Patent Under Reexamination						
13943395	HARRISON ET AL.						
Examiner	Art Unit						
ARUN S PHASGE	1756						

US ORIGINAL CLASSIFICATION						INTERNATIONAL CLASSIFICATION									
CLASS SUBCLASS					CLAIMED						NON-CLAIMED				
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CROSS REFERENCE(S)															
CLASS	ASS SUBCLASS (ONE SUBCLASS PER BLOCK)			CK)											

NONE	Total Claims Allowed:				
(Assistant Examiner)	(Date)	16			
/ARUN S PHASGE/ Primary Examiner.Art Unit 1756	8/10/15	O.G. Print Claim(s)	O.G. Print Figure		
(Primary Examiner)	(Date)	1, 10	1, 2		

# Issue Classification



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Application/Control No.		Applicant(s)/Patent Under Reexamination
	13943395	HARRISON ET AL.
	Examiner	Art Unit
	ARUN S PHASGE	1756

	Claims re	numbere	d in the sa	ame orde	r as prese	ented by a	applicant		СР	A [2	T.D.		R.1.	47	
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	13		29	14	45										
	14		30	15	46										
	15		31	16	47										
	16	1	32												

NONE		Total Claims Allowed:		
(Assistant Examiner)	(Date)	1	6	
/ARUN S PHASGE/ Primary Examiner.Art Unit 1756	8/10/15	O.G. Print Claim(s)	O.G. Print Figure	
(Primary Examiner)	(Date)	1, 10	1, 2	



#### United States Patent and Trademark Office

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www.uspto.gov UNITED STATES DEPARTMENT OF COMMERCE

APPLICATION NUMBER FILING OR 371(C) DATE FIRST NAMED APPLICANT ATTY. DOCKET NO./TITLE 13/943,395 07/16/2013 Stephen Harrison 10001108.0099

24392 FISH & TSANG LLP ROBERT D. FISH 2603 Main Street Suite 1000 Irvine, CA 92614-6232

**CONFIRMATION NO. 2954** POA ACCEPTANCE LETTER



Date Mailed: 04/20/2015

#### NOTICE OF ACCEPTANCE OF POWER OF ATTORNEY

This is in response to the Power of Attorney filed 04/14/2015.

The Power of Attorney in this application is accepted. Correspondence in this application will be mailed to the above address as provided by 37 CFR 1.33.

/ytdemisse/			

Office of Data Management, Application Assistance Unit (571) 272-4000, or (571) 272-4200, or 1-888-786-0101



## United States Patent and Trademark Office

UNITED STATES DEPARTMENT OF COMMI United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS PO. Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov UNITED STATES DEPARTMENT OF COMMERCE

APPLICATION NUMBER FILING OR 371(C) DATE FIRST NAMED APPLICANT ATTY. DOCKET NO./TITLE 13/943,395 07/16/2013 Stephen Harrison 10001108.0099

126930 **DENTONS US LLP** P.O. Box 061080 Chicago, IL 60606-7020

**CONFIRMATION NO. 2954 POWER OF ATTORNEY NOTICE** 



Date Mailed: 04/20/2015

#### NOTICE REGARDING CHANGE OF POWER OF ATTORNEY

This is in response to the Power of Attorney filed 04/14/2015.

• The Power of Attorney to you in this application has been revoked by the applicant. Future correspondence will be mailed to the new address of record(37 CFR 1.33).

	/ytdemisse	:/				

Office of Data Management, Application Assistance Unit (571) 272-4000, or (571) 272-4200, or 1-888-786-0101

Application Number	Application/Co	R	Applicant(s)/Patent under Reexamination HARRISON ET AL.			
Document Code - DISQ	•	Internal Do	cument – DO	NOT MAIL		
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TERMINAL DISCLAIMER	☑ APPROVI	ED	☐ DISAPP	ROVED		
Date Filed : 4/14/15	to a Te	t is subject erminal aimer				
Approved/Disapproved by:						
icia D. Roberts						
18,232						

U.S. Patent and Trademark Office

### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE P.O. BOX 1450 ALEXANDRIA, VA 22313-1450

Appl No.: 13/943,395

Inventor: Stephen Harrison

Filing Date: **07/16/2013** 

Art Unit: 1756

Examiner: PHASGE, ARUN S
Attorney Docket No.: 102663.0002US5

Title Selective Recovery of Manganese and Zinc From

**Geothermal Brines** 

Mail Stop Amendment Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

#### RESPONSE TO OFFICE ACTION

Sir:

This paper responds to the Office Action dated January 16, 2015. Please enter the amendments shown herein.

- Claim amendments begin on page 2.
- Remarks begin on page 4.

#### AMENDMENTS TO THE CLAIMS

This listing of claims replaces all prior versions, and listings, of claims in the application: 1-31. (cancelled).

32. (previously presented) A method for recovering zinc from a geothermal brine, the method comprising the steps of:

providing a geothermal brine, said geothermal brine comprising zinc; selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine; and

removing the zinc from the substantially silica free brine.

- 33. (previously presented) The method of claim 32, wherein the silica and iron are selectively removed from the geothermal brine to produce a substantially silica and iron free brine.
- 34. (previously presented) The method of claim 32, wherein the zinc is removed from the substantially silica free brine by solvent extraction.
- 35. (previously presented) The method of claim 33, wherein the zinc is removed from the substantially silica free brine using an organic solvent.
- 36. (previously presented) The method of claim 33, wherein the zinc is removed from the substantially silica free brine using a phosphorous-based solvent.
- 37. (previously presented) The method of claim 33, wherein the zinc is removed from the substantially silica free brine using a solvent selected from the group consisting of phosphines, phosphoric acids, and phosphinic acids.
- 38. (previously presented) The method of claim 32, wherein the zinc is removed from the substantially silica free brine by ion exchange.
- 39. (previously presented) The method of claim 32, further comprising the step of extracting manganese from the substantially silica free brine by solvent extraction.

- 40. (previously presented) The method of claim 32, further comprising the step of extracting manganese from the substantially silica free brine by ion exchange.
- 41. (previously presented) A method for recovering manganese from a geothermal brine, the method comprising the steps of:

providing a geothermal brine, said geothermal brine comprising zinc; selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine; and

- extracting the manganese from the substantially silica free brine.
- 42. (previously presented) The method of claim 41, wherein the manganese is extracted from the substantially silica free brine by precipitation.
- 43. (previously presented) The method of claim 41, wherein the manganese is extracted from the substantially silica free brine by selective precipitation using an amine or ammonia.
- 44. (previously presented) A method for recovering zinc and manganese from a geothermal brine, the method comprising the steps of:

providing a geothermal brine, said geothermal brine comprising manganese and zinc; selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine;

selectively removing lithium chloride from the substantially silica free brine; removing the zinc from the substantially silica free brine; and extracting manganese from the substantially silica free brine.

- 45. (previously presented) The method of claim 44, wherein the lithium chloride is selectively removed from the substantially silica free brine by use of lithium aluminum intercalates.
- 46. (previously presented) The method of claim 44, wherein the lithium chloride is selectively removed from the substantially silica free brine by ion exchange.
- 47. (previously presented) The method of claim 44, wherein the lithium chloride is selectively removed from the substantially silica free brine by electrochemical means.

REMARKS/ARGUMENTS

**Double Patenting** 

The Office rejected claims 32-47 on the ground of nonstatutory double patenting as being

unpatentable over claims 1-20 of U.S. Patent No. 8,518,232. The applicant respectfully submits a

terminal disclaimer over the '232 patent, and the rejection should thus be overcome.

The Office provisionally rejected claims 32-47 on the ground of nonstatutory double

patenting as being unpatentable over claims 21-41 of copending Application No. 13/874,004.

Applicant will submit a terminal disclaimer in the '004 application as appropriate once all

remaining issues have been resolved in that application. The rejection should thus be overcome.

**Request For Allowance** 

Claims 32-47 are pending in this application. The applicant requests allowance of all

pending claims.

Respectfully submitted, Fish & Tsang, LLP

Date: April 14, 2015

By\_\_/Martin Fessenmaier/\_

Martin Fessenmaier

Reg. No. 46697

Fish & Tsang, LLP

2603 Main Street, Suite 1000 Irvine, CA 92614-4271

Telephone (949) 943-8300 Fax (949) 943-8358

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	<u>s</u>	STATEMENT UNDE	R 37 CFR 3.7	<u>3(b)</u>
Applicant	/Patent Owner: Stephen Harrison /	Simbol, Inc.		
			Filed/Issue [	Date: July 16, 2013
Titled:	Selective Recovery Of Manganes	e, Lead and Zinc		
Simbol, I	nc.	, a corpora	ation	
(Name of As	ssignee)	(Type o	of Assignee, e.g., cor	poration, partnership, university, government agency, etc.
states tha	at it is:			
1. 🗶	the assignee of the entire right, title,	and interest in;		
2.	an assignee of less than the entire r (The extent (by percentage) of its ov			or
3.	the assignee of an undivided interes	t in the entirety of (a c	complete assign	ment from one of the joint inventors was made)
the paten	t application/patent identified above, b	y virtue of either:		
A. 🔀	An assignment from the inventor(s) the United States Patent and Trade copy therefore is attached.	of the patent applicati mark Office at Reel	on/patent identif 031428	ried above. The assignment was recorded in, Frame 0328, or for which a
OR	A chain of title from the inventor(a)	of the notant annliastic	on/notontidontif	ind above to the current assigned as follows:
В			•	ied above, to the current assignee as follows:
				on large of Office of
	The document was record			or for which a copy thereof is attached.
		,		
	2. From:			
	The document was record			
	Reel	, Frame		or for which a copy thereof is attached.
	3. From:		To:	
	The document was record	led in the United State	es Patent and Tr	ademark Office at
	Reel	, Frame		or for which a copy thereof is attached.
	Additional documents in the chain of	of title are listed on a s	supplemental sh	eet(s).
	required by 37 CFR 3.73(b)(1)(i), the concurrently is being, submitted for re-			of title from the original owner to the assignee was,
	OTE: A separate copy (i.e., a true cop cordance with 37 CFR Part 3, to recon			nt(s)) must be submitted to Assignment Division in USPTO. <u>See</u> MPEP 302.08]
The unde	ersigned (whose title is supplied below)	is authorized to act o	n behalf of the a	assignee.
/Martin F	essenmaier/			2015-04-14
S	ignature			Date
Martin Fe	essenmaier			Agent for Applicant Simbol In
Р	rinted or Typed Name			Title

This collection of information is required by 37 CFR 3.73(b). The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.11 and 1.14. This collection is estimated to take 12 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. **SEND TO: Commissioner** for Patents, P.O. Box 1450, Alexandria, VA 22313-1450. ESM / PETITIONER EX 1014 45

#### Privacy Act Statement

The **Privacy Act of 1974 (P.L. 93-579)** requires that you be given certain information in connection with your submission of the attached form related to a patent application or patent. Accordingly, pursuant to the requirements of the Act, please be advised that: (1) the general authority for the collection of this information is 35 U.S.C. 2(b)(2); (2) furnishing of the information solicited is voluntary; and (3) the principal purpose for which the information is used by the U.S. Patent and Trademark Office is to process and/or examine your submission related to a patent application or patent. If you do not furnish the requested information, the U.S. Patent and Trademark Office may not be able to process and/or examine your submission, which may result in termination of proceedings or abandonment of the application or expiration of the patent.

The information provided by you in this form will be subject to the following routine uses:

- 1. The information on this form will be treated confidentially to the extent allowed under the Freedom of Information Act (5 U.S.C. 552) and the Privacy Act (5 U.S.C 552a). Records from this system of records may be disclosed to the Department of Justice to determine whether disclosure of these records is required by the Freedom of Information Act.
- 2. A record from this system of records may be disclosed, as a routine use, in the course of presenting evidence to a court, magistrate, or administrative tribunal, including disclosures to opposing counsel in the course of settlement negotiations.
- 3. A record in this system of records may be disclosed, as a routine use, to a Member of Congress submitting a request involving an individual, to whom the record pertains, when the individual has requested assistance from the Member with respect to the subject matter of the record
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- 6. A record in this system of records may be disclosed, as a routine use, to another federal agency for purposes of National Security review (35 U.S.C. 181) and for review pursuant to the Atomic Energy Act (42 U.S.C. 218(c)).
- 7. A record from this system of records may be disclosed, as a routine use, to the Administrator, General Services, or his/her designee, during an inspection of records conducted by GSA as part of that agency's responsibility to recommend improvements in records management practices and programs, under authority of 44 U.S.C. 2904 and 2906. Such disclosure shall be made in accordance with the GSA regulations governing inspection of records for this purpose, and any other relevant (*i.e.*, GSA or Commerce) directive. Such disclosure shall not be used to make determinations about individuals.
- 8. A record from this system of records may be disclosed, as a routine use, to the public after either publication of the application pursuant to 35 U.S.C. 122(b) or issuance of a patent pursuant to 35 U.S.C. 151. Further, a record may be disclosed, subject to the limitations of 37 CFR 1.14, as a routine use, to the public if the record was filed in an application which became abandoned or in which the proceedings were terminated and which application is referenced by either a published application, an application open to public inspection or an issued patent.
- 9. A record from this system of records may be disclosed, as a routine use, to a Federal, State, or local law enforcement agency, if the USPTO becomes aware of a violation or potential violation of law or regulation.

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# TERMINAL DISCLAIMER TO OBVIATE A DOUBLE PATENTING REJECTION OVER A "PRIOR" PATENT

Docket Number (Optional) 102663.0002US5

In re Application of: Simbol, Inc.
Application No.: 13/943395
Filed: July 16, 2013
For: Selective Recovery Of Manganese, Lead and Zinc
The owner*, <u>Simbol, Inc.</u> , of <u>100</u> percent interest in the instant application hereby disclaims, except as provided below, the terminal part of the statutory term of any patent granted on the instant application which would extend beyond the expiration date of the full statutory term of <b>prior patent</b> No. <u>8518232</u> as the term of said <b>prior patent</b> is presently shortened by any terminal disclaimer. The owner hereby agrees that any patent so granted on the instant application shall be enforceable only for and during such period that it and the <b>prior patent</b> are commonly owned. This agreement runs with any patent granted on the instant application and is binding upon the grantee, its successors or assigns.
In making the above disclaimer, the owner does not disclaim the terminal part of the term of any patent granted on the instant application that would extend to the expiration date of the full statutory term of the <b>prior patent</b> , "as the term of said <b>prior patent</b> is presently shortened by any terminal disclaimer," in the event that said <b>prior patent</b> later:  expires for failure to pay a maintenance fee; is held unenforceable; is found invalid by a court of competent jurisdiction; is statutorily disclaimed in whole or terminally disclaimed under 37 CFR 1.321; has all claims canceled by a reexamination certificate; is reissued; or is in any manner terminated prior to the expiration of its full statutory term as presently shortened by any terminal disclaimer.
Check either box 1 or 2 below, if appropriate.
1. For submissions on behalf of a business/organization (e.g., corporation, partnership, university, government agency, etc.), the undersigned is empowered to act on behalf of the business/organization.
I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.
2. The undersigned is an attorney or agent of record. Reg. No. 46697
/Martin Fessenmaier/ 2015-04-14
Signature Date
Martin Fessenmaier
Typed or printed name
949-943-8300
Telephone Number
Terminal disclaimer fee under 37 CFR 1.20(d) included.
WARNING: Information on this form may become public. Credit card information should not be included on this form. Provide credit card information and authorization on PTO-2038.
*Statement_under 37 CFR 3.73(b) is required if terminal disclaimer is signed by the assignee (owner). Form PTO/SB/96 may be used for making this certification. See MPEP § 324.

This collection of information is required by 37 CFR 1.321. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.11 and 1.14. This collection is estimated to take 12 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this bu rden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

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- A record in this system of records may be disclosed, as a routine use, to a Member of Congress submitting a request involving an individual, to whom the record pertains, when the individual has requested assistance from the Member with respect to the subject matter of the record
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- 6. A record in this system of records may be disclosed, as a routine use, to another federal agency for purposes of National Security review (35 U.S.C. 181) and for review pursuant to the Atomic Energy Act (42 U.S.C. 218(c)).
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- A record from this system of records may be disclosed, as a routine use, to a Federal, State, or local law enforcement agency, if the USPTO becomes aware of a violation or potential violation of law or regulation.

Electronic Patent Application Fee Transmittal						
Application Number: 13943395						
Filing Date:	16-	Jul-2013				
Title of Invention:	Selective Recovery of Manganese, Lead and Zinc					
First Named Inventor/Applicant Name:	Stephen Harrison					
Filer:	Martin Fessenmaier/Marutzzella Castro					
Attorney Docket Number:	100	001108.0099				
Filed as Large Entity						
Filing Fees for Utility under 35 USC 111(a)						
Description		Fee Code	Quantity	Amount	Sub-Total in USD(\$)	
Basic Filing:						
Pages:						
Claims:						
Miscellaneous-Filing:						
Petition:						
Patent-Appeals-and-Interference:						
Post-Allowance-and-Post-Issuance:						
Extension-of-Time:						

Description	Fee Code	Fee Code Quantity		Sub-Total in USD(\$)	
Miscellaneous:					
Statutory or Terminal Disclaimer	1814	1	160	160	
	Tot	al in USD	(\$)	160	

Electronic Acknowledgement Receipt				
EFS ID:	22062813			
Application Number:	13943395			
International Application Number:				
Confirmation Number:	2954			
Title of Invention:	Selective Recovery of Manganese, Lead and Zinc			
First Named Inventor/Applicant Name:	Stephen Harrison			
Customer Number:	126930			
Filer:	Martin Fessenmaier/Marutzzella Castro			
Filer Authorized By:	Martin Fessenmaier			
Attorney Docket Number:	10001108.0099			
Receipt Date:	14-APR-2015			
Filing Date:	16-JUL-2013			
Time Stamp:	20:54:44			
Application Type:	Utility under 35 USC 111(a)			

# **Payment information:**

Submitted with Payment	yes
Payment Type	Deposit Account
Payment was successfully received in RAM	\$160
RAM confirmation Number	6828
Deposit Account	500341
Authorized User	

The Director of the USPTO is hereby authorized to charge indicated fees and credit any overpayment as follows:

Charge any Additional Fees required under 37 C.F.R. Section 1.16 (National application filing, search, and examination fees)

Charge any Additional Fees required under 37 C.F.R. Section 1.17 (Patent applications in dreekanting tion processing fees) 5

Charge any Additional Fees required under 37 C.F.R. Section 1.19 (Document supply fees)

Charge any Additional Fees required under 37 C.F.R. Section 1.20 (Post Issuance fees)

Charge any Additional Fees required under 37 C.F.R. Section 1.21 (Miscellaneous fees and charges)

## File Listing:

Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)
1		2015-04-14_Response1apdf	26886	yes	4
'		2013 04 14_Nesponsera_,par	004ea2f7826e1cebd8d2c4e2b7bfc594209 e6237	yes	7
	Multip	oart Description/PDF files in .	zip description		
	Document Des	Start	Eı	nd	
	Amendment/Req. Reconsiderati	1		1	
	Claims	2		3	
	Applicant Arguments/Remarks	4		4	
Warnings:					
Information:					
2	Assignee showing of ownership per 37	StmtOwnershippdf -	422942	no	2
2	CFR 3.73	Stifftownershippdi	37ee5d25091dcb83dea1b849c3d534c55b 1c642a	110	2
Warnings:					
Information:					
3	Power of Attorney	POA_signed.pdf	1383149	no	3
3	1 ower of Attorney	r on_signed.pur	3173468dd4e29a21c92424e43807b1bfc56 9c547	110	<u> </u>
Warnings:					
Information:					
4	Terminal Disclaimer Filed	TD_2US2.pdf	374011	no	2
7	reminal Disclaimer Flied	1 <i>b</i> _2032.pdi	b5892d8f7a72a5d7f608ce973664a604ded 93ab3	110	2
Warnings:					
Information:					
5	Fee Worksheet (SB06)	fee-info.pdf	30214	no 2	
,	i ee wordlieet (3000)	ree inio.pui	c67987661f72b3a5e4a951aa216e69ad9c0 b3de6		
Warnings:			<u> </u>	<u> </u>	
Information:					
		Total Files Size (in bytes)	22:	37202	

This Acknowledgement Receipt evidences receipt on the noted date by the USPTO of the indicated documents, characterized by the applicant, and including page counts, where applicable. It serves as evidence of receipt similar to a Post Card, as described in MPEP 503.

#### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

#### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

#### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

## POWER OF ATTORNEY TO PROSECUTE APPLICATIONS BEFORE THE USPTO

I hereby revoke all previous powers of attorney given in the application identified in the attached statement under 37 CFR 3.73(c).								
	reby app							
	Pac OR	tioners associated with Custo	omer Number:	24392				
	Practitioner(s) named below (if more than ten patent practitioners are to be named, then a customer number must be used):							
	jananan. mmann	N9456	Regi	stration umber		Registration Number		
	in in the same of			,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,				
	janana.							
any a	and all pate		to the undersig			llice (USPTO) in connection with rds or assignments documents		
Plear	se change i	aanbba enabnoqeemoo ed	for the applicati	on identified in the	attached statement under 3	7 CFR 3.73(e) to:		
	The a	ddress associated with Custr	mer Number:	24392				
OR	Firm or		,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	<u> </u>	***************************************	d		
l	Individual	Name	****************		,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,			
	Address					·		
	City			State	***************************************	Zp		
	Country		****					
	Telephon	6			Emed			
Assignee Name and Address: Simbol, Inc. 6920 Koll Center Parkway, Suite 123, Pleasanton, CA 94566								
A copy of this form, together with a statement under 37 CFR 3.73(c) (Form PTO/AIA/96 or equivalent) is required to be Filed in each application in which this form is used. The statement under 37 CFR 3.73(c) may be completed by one of The practitioners appointed in this form, and must identify the application in which this Power of Attorney is to be filed.								
	The	individual whose signatur		RE of Assignee supplied below is		·-		
Sign	ature	4 Harri	<i>M</i>		1	3-2015		
Nam	183	Stephen Harrison			Telephone 925	-226-7418		
Titie	Title Chief Technology Officer and Vice President							

This obligation of information is required by \$7 CFR 1.31, 1.22 and 1.33. The information is required to obtain or return a benefit by the public which is to file (and by the USFTO to process) as application. Confidentiality is governed by \$5 U.S.D. 122 and \$7 CFR 1.11 and 1.14. This collection is estimated to take 3 minuses to complete, including garriering, preparing, and submitting the completed application form to the USFTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief information Officer, U.S. Platent and Pademark Office, U.S. Department of Commence, P.O. Box 1450, Alexandria, VA 22013-1450, DO NOT SENO FIEE OR COMPLETED FORMS TO THIS ACCRESS, SEND TO: Germinationer for Patents, P.O. Box 1450, Alexandria, VA 22013-1450.

If you need assistance in completing the form, call 1-800-PTO-9199 and select option 2.

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  evidence to a court, magistrate, or administrative tribunal, including disclosures to opposing counsel in
  the course of settlement negotiations.
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UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS P.O. Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov

APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
13/943,395	07/16/2013	Stephen Harrison	10001108.0099	2954
126930 DENTONS US	7590 01/16/201 LLP	5	EXAM	INER
P.O. Box 06108 Chicago, IL 606	80		PHASGE,	ARUN S
			ART UNIT	PAPER NUMBER
			1756	
			NOTIFICATION DATE	DELIVERY MODE
			01/16/2015	ELECTRONIC

## Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Notice of the Office communication was sent electronically on above-indicated "Notification Date" to the following e-mail address(es):

patents.us@dentons.com

	Application No. 13/943,395	Applicant(s) HARRISON ET AL.						
Office Action Summary	Examiner ARUN S. PHASGE	Art Unit 1756	AIA (First Inventor to File) Status No					
The MAILING DATE of this communication appears on the cover sheet with the correspondence address Period for Reply								
A SHORTENED STATUTORY PERIOD FOR REPLY THIS COMMUNICATION.  - Extensions of time may be available under the provisions of 37 CFR 1.13 after SIX (6) MONTHS from the mailing date of this communication.  - If NO period for reply is specified above, the maximum statutory period w - Failure to reply within the set or extended period for reply will, by statute, Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	ill apply and will expire SIX (6) MONTHS from to cause the application to become ABANDONEI	ely filed the mailing date of 0 (35 U.S.C. § 133)	this communication.					
Status								
1) Responsive to communication(s) filed on								
A declaration(s)/affidavit(s) under 37 CFR 1.1								
2a) ☐ This action is <b>FINAL</b> . 2b) ☑ This								
3) An election was made by the applicant in responsible to the restriction requirement and election	•		g the interview on					
<ul> <li>the restriction requirement and election</li> <li>Since this application is in condition for allowant</li> </ul>	•		n the merite is					
closed in accordance with the practice under <i>E</i>	·							
Disposition of Claims*	, ,							
5) Claim(s) <u>32-47</u> is/are pending in the application	1.							
5a) Of the above claim(s) is/are withdraw								
6) Claim(s) is/are allowed.								
7)⊠ Claim(s) <u>32-47</u> is/are rejected.								
8) Claim(s) is/are objected to.								
9) Claim(s) are subject to restriction and/or	·							
* If any claims have been determined <u>allowable</u> , you may be eli			way program at a					
participating intellectual property office for the corresponding ap	•							
http://www.uspto.gov/patents/init_events/pph/index.jsp or send	an inquiry to <u>PPHieedback@uspto.d</u>	<u>ov</u> .						
Application Papers								
10) The specification is objected to by the Examiner		- - -						
11) The drawing(s) filed on is/are: a) accessory applicant may not request that any objection to the company.			2)					
Replacement drawing sheet(s) including the correcti			· ·					
		00.000 10. 000 0	77 OTT 1.121(d).					
Priority under 35 U.S.C. § 119 12) ☐ Acknowledgment is made of a claim for foreign	priority under 25 LLC C 8 110(a)	(d) or (f)						
Certified copies:	priority under 35 0.5.C. § 119(a)	-(u) or (i).						
a) ☐ All b) ☐ Some** c) ☐ None of the:								
1. Certified copies of the priority document	s have been received.							
2. Certified copies of the priority document		on No						
3. Copies of the certified copies of the prio	rity documents have been receive	ed in this Nati	onal Stage					
application from the International Bureau	(PCT Rule 17.2(a)).							
** See the attached detailed Office action for a list of the certifie	d copies not received.							
Attachment(s)								
1) Notice of References Cited (PTO-892)	3) Interview Summary							
2) Information Disclosure Statement(s) (PTO/SB/08a and/or PTO/S Paper No(s)/Mail Date 8/15/13, 9/4/13.	Paper No(s)/Mail Da 4) Other:	te						

Application/Control Number: 13/943,395 Page 2

Art Unit: 1756

The present application is being examined under the pre-AIA first to invent

provisions.

**DETAILED ACTION** 

Double Patenting

The nonstatutory double patenting rejection is based on a judicially created

doctrine grounded in public policy (a policy reflected in the statute) so as to prevent the

unjustified or improper timewise extension of the "right to exclude" granted by a patent

and to prevent possible harassment by multiple assignees. A nonstatutory double

patenting rejection is appropriate where the claims at issue are not identical, but at least

one examined application claim is not patentably distinct from the reference claim(s)

because the examined application claim is either anticipated by, or would have been

obvious over, the reference claim(s). See, e.g., In re Berg, 140 F.3d 1428, 46 USPQ2d

1226 (Fed. Cir. 1998); In re Goodman, 11 F.3d 1046, 29 USPQ2d 2010 (Fed. Cir.

1993); In re Longi, 759 F.2d 887, 225 USPQ 645 (Fed. Cir. 1985); In re Van Ornum,

686 F.2d 937, 214 USPQ 761 (CCPA 1982); In re Vogel, 422 F.2d 438, 164 USPQ 619

(CCPA 1970); and *In re Thorington*, 418 F.2d 528, 163 USPQ 644 (CCPA 1969).

A timely filed terminal disclaimer in compliance with 37 CFR 1.321(c) or 1.321(d)

may be used to overcome an actual or provisional rejection based on a nonstatutory

double patenting ground provided the reference application or patent either is shown to

Application/Control Number: 13/943,395 Page 3

Art Unit: 1756

be commonly owned with this application, or claims an invention made as a result of

activities undertaken within the scope of a joint research agreement. A terminal

disclaimer must be signed in compliance with 37 CFR 1.321(b).

The USPTO internet Web site contains terminal disclaimer forms which may be

used. Please visit http://www.uspto.gov/forms/. The filing date of the application will

determine what form should be used. A web-based eTerminal Disclaimer may be filled

out completely online using web-screens. An eTerminal Disclaimer that meets all

requirements is auto-processed and approved immediately upon submission. For more

information about eTerminal Disclaimers, refer to

http://www.uspto.gov/patents/process/file/efs/guidance/eTD-info-I.jsp.

Claims 32-47 are rejected on the ground of nonstatutory double patenting as

being unpatentable over claims 1-20 of U.S. Patent No. 8,518,232. Although the claims

at issue are not identical, they are not patentably distinct from each other because the

claims when read in light of the specification are clearly encompassed and rendered

obvious by the claims of the prior patent.

Claims 32-47 are rejected on the ground of nonstatutory double patenting as

being unpatentable over claims 1-20 of U.S. Patent No. 8,518,232. Although the claims

at issue are not identical, they are not patentably distinct from each other because the

Application/Control Number: 13/943,395

Art Unit: 1756

claims when read in light of the specification are clearly encompassed and rendered

obvious by the claims of the prior patent.

Claims 32-47 are provisionally rejected on the ground of nonstatutory double

patenting as being unpatentable over claims 21-41 of copending Application No.

13/874,004. Although the claims at issue are not identical, they are not patentably

distinct from each other because the claims when read in light of the specification are

clearly encompassed and rendered obvious by the claims of the copending application.

This is a provisional nonstatutory double patenting rejection because the

patentably indistinct claims have not in fact been patented.

Conclusion

Any inquiry concerning this communication or earlier communications from the

examiner should be directed to ARUN S. PHASGE whose telephone number is

(571)272-1345. The examiner can normally be reached on MONDAY-THURSDAY,

7:30-6:00.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's

supervisor, Keith D. Hendricks can be reached on (571) 272-1401. The fax phone

number for the organization where this application or proceeding is assigned is 571-

273-8300.

Page 4

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Art Unit: 1756

Information regarding the status of an application may be obtained from the

Patent Application Information Retrieval (PAIR) system. Status information for

published applications may be obtained from either Private PAIR or Public PAIR.

Status information for unpublished applications is available through Private PAIR only.

For more information about the PAIR system, see http://pair-direct.uspto.gov. Should

you have questions on access to the Private PAIR system, contact the Electronic

Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a

USPTO Customer Service Representative or access to the automated information

system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/ARUN S PHASGE/ Primary Examiner, Art Unit 1756

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Application/Control No.   13/943,395     Applicant(s)/Patent Under Reexamination   HARRISON ET AL.	on
13/943,395	on
Examiner   Art Unit   1756   Page 1 of	on
Name   Classificat	on
*         Document Number Country Code-Number-Kind Code         Date MM-YYYY         Name         Classificat           *         A         US-8,454,816         06-2013         Harrison et al.         205/540           *         B         US-8,518,232         08-2013         Harrison et al.         205/539           C         US-         D         US-         US-           E         US-         US-         US-         US-	
*         Country Code-Number-Kind Code         MM-YYYY         Name         Classificat           *         A         US-8,454,816         06-2013         Harrison et al.         205/540           *         B         US-8,518,232         08-2013         Harrison et al.         205/539           C         US-         D         US-         US-           E         US-         US-         US-	
*       A       US-8,454,816       06-2013       Harrison et al.       205/540         *       B       US-8,518,232       08-2013       Harrison et al.       205/539         C       US-       US-       E       US-       US-       US-	
*         B         US-8,518,232         08-2013         Harrison et al.         205/538           C         US-         US-         E         US-         US-	
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FOREIGN PATENT DOCUMENTS	
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NON-PATENT DOCUMENTS	
* Include as applicable: Author, Title Date, Publisher, Edition or Volume, Pertinent Pages)	
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\*A copy of this reference is not being furnished with this Office action. (See MPEP § 707.05(a).) Dates in MM-YYYY format are publication dates. Classifications may be US or foreign.

U.S. Patent and Trademark Office PTO-892 (Rev. 01-2001)

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# Search Notes Application/Control No. Applicant(s)/Patent Under Reexamination HARRISON ET AL. Examiner ARUN S PHASGE 1756

CPC- SEARCHE	ĒD	
Symbol	Date	Examiner
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SEARCH NOTE	ES	
Search Notes	Date	Examiner
inventor name searched in eDAN	12/22/14	asp
WEST his attached	12/24/14	asp

Subclass

539, 540, 573, 602-608

	INTERFERENCE SEARCH		
US Class/ CPC Symbol	US Subclass / CPC Group	Date	Examiner
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12/22/14

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PTO/SB/08a (01-10)
Approved for use through 07/31/2012. OMB 0651-0031
U.S. Patent and Trademark Office; U.S. DEPARTMENT OF COMMERCE
Under the Paperwork Reduction Act of 1995, no persons are required to respond to a collection of information unless it contains a valid OMB control number.

	Application Number		13943395	
INTERPRETATION PROGRAMME	Filing Date		2013-07-16	
INFORMATION DISCLOSURE	First Named Inventor Stephe		ohen Harrison	
( Not for submission under 37 CFR 1.99)	Art Unit			
(Not for submission under 57 of K 1.55)	Examiner Name	Unkno	own	
	Attorney Docket Number	er	076165.000099	

	U.S.PATENTS					
Examiner Initial*	Cite No	Patent Number	Kind Code <sup>1</sup>	Issue Date	Name of Patentee or Applicant of cited Document	Pages, Columns, Lines where Relevant Passages or Relevant Figures Appear
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( Not for submission under 37 CFR 1.99)

Application Number		13943395	
Filing Date		2013-07-16	
First Named Inventor	Stepl	hen Harrison	
Art Unit	•	·	
Examiner Name Unki		nown .	
Attorney Docket Number		076165 000099	

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If you wish	to add	additional U.S. Pater	it citatio	n information p	lease click the Add button.	
			U.S.P	ATENT APPLI	CATION PUBLICATIONS	Y
Examiner Initial*	Cite No	Publication Number	Kind Code <sup>1</sup>	Publication Date	Name of Patentee or Applicant of cited Document	Pages,Columns,Lines where Relevant Passages or Relevant Figures Appear
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( Not for submission under 37 CFR 1.99)

Application Number		13943395
Filing Date		2013-07-16
First Named Inventor	Steph	nen Harrison
Art Unit		
Examiner Name	Unkno	own
Attorney Docket Numb	er	076165.000099

No	Foreign Document Number <sup>3</sup>	Country Code <sup>2</sup> İ	Kind Code <sup>4</sup>	Publication Date	Name of Patentee or Applicant of cited Document	Pages,Columns,Lines where Relevant Passages or Relevant Figures Appear	T5	
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Application Number		13943395			
Filing Date		2013-07-16			
First Named Inventor	Steph	en Harrison			
Art Unit					
Examiner Name Unkn		own			
Attorney Docket Numb	er	076165.000099			

	EXAMINER SIGNATURE										
Examiner Signature /Arun Phasge/ Date Considered 12/22/2014											
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Standard ST.3). 3 For Japanese	patent documents, the indication of the year of priate symbols as indicated on the document un	901.04. <sup>2</sup> Enter office that issued the document, the reign of the Emperor must precede the serial der WIPO Standard ST.16 if possible. <sup>5</sup> Applicar	number of the patent documen								

( Not for submission under 37 CFR 1.99)

Application Number		13943395				
Filing Date		2013-07-16				
First Named Inventor	Step	hen Harrison				
Art Unit						
Examiner Name	Unkr	nown				
Attorney Docket Numb	er	076165.000099	***************************************			

	CERTIFICATION STATEMENT										
Plea	Please see 37 CFR 1.97 and 1.98 to make the appropriate selection(s):										
	That each item of information contained in the information disclosure statement was first cited in any communication from a foreign patent office in a counterpart foreign application not more than three months prior to the filing of the information disclosure statement. See 37 CFR 1.97(e)(1).										
OR	OR										
	That no item of information contained in the information disclosure statement was cited in a communication from a foreign patent office in a counterpart foreign application, and, to the knowledge of the person signing the certification after making reasonable inquiry, no item of information contained in the information disclosure statement was known to any individual designated in 37 CFR 1.56(c) more than three months prior to the filing of the information disclosure statement. See 37 CFR 1.97(e)(2).										
	See attached ce	rtification statement.									
	The fee set forth	in 37 CFR 1.17 (p) has been submitted her	ewith.								
$\boxtimes$	A certification st	atement is not submitted herewith.									
	SIGNATURE  A signature of the applicant or representative is required in accordance with CFR 1.33, 10.18. Please see CFR 1.4(d) for the form of the signature.										
Sigi	nature	Kaethika	Date (YYYY-MM-DD)	2013-08-15							
Nar	ne/Print	Karthika Perumal	Registration Number	69296							
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This collection of information is required by 37 CFR 1.97 and 1.98. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.14. This collection is estimated to take 1 hour to complete, including gathering, preparing and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. **SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.** 



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## **BIB DATA SHEET**

#### **CONFIRMATION NO. 2954**

SERIAL NUMBER   FILING O			7 371(c)	371(c) CLASS GRO			ROUP ART UNIT ATTORNEY DOCKET NO.				
13/943,39	5	07/16/2			205		1756		10	0001108.0099	
		RULI	E								
	APPLICANTS SIMBOL Inc., Pleasanton, CA, Assignee (with 37 CFR 1.172 Interest);										
INVENTORS Stephen Harrison, Benicia, CA; Samaresh Mohanta, Pleasanton, CA; C.V. Krishnamohan Sharma, Milpitas, CA; Elizabeth Geler, San Rafael, CA;											
This appl wh	** <b>CONTINUING DATA</b> ***********************************										
** FOREIGN A	PPLICA	TIONS *****	******	*****	·						
	** IF REQUIRED, FOREIGN FILING LICENSE GRANTED ** ** SMALL ENTITY ** 07/31/2013										
Foreign Priority claime		Yes No	☐ Met af	tor	STATE OR		IEETS	TOT		INDEPENDENT	
	ditions met ARUN S P Examiner's	HASGE/	Allowa	ince	COUNTRY CA	DRA	WINGS 17	CLAII 16		CLAIMS 3	
ADDRESS											
P.O. Box Chicago,	DENTONS US LLP P.O. Box 061080 Chicago, IL 60606-7020 UNITED STATES										
TITLE											
Selective	Recove	ery of Mangai	nese, Lead	d and 2	Zinc						
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## WEST Search History for Application 13943395

Delete Selected Add More Searches Close eRed Folder Select All Plurals | Thesaurus | Query Database Operator | Date All Int. All P/A Prior Art PGPB, USPT, USOC, 12~ c22b47/00.cpc, or c22b19/00.cpc, or ...: EPAB, JPAB, DWPI, OR YES 23c25c1/10.cpc, or c25c1/16.cpc. TDBD, FPRS 2014 PGPB, USPT, USOC. (c22b47/00.cpc, or c22b19/00.cpc, or 12-.... EPAB, JPAB, DWPL OR YES c25c1/10.epc, or c25c1/16.cpc, ) and 23-TOBD, FPRS (manganese and zinc) 2014 (c22b47/00.cpc, or c22b19/00.cpc, or PGPB, USPT, USOC, 12-...: c25c1/10.cpc, or c25c1/16.cpc.) and EPAB, JPAB, DWPI, OR YES 23~ TDBD. FPRS 2014 (manganese) (c22b47/00.cpc, or c22b19/00.cpc, or PGPB, USPT, USOC, 12-OR YES c25c1/10.cpc, or c25c1/16.cpc, and EPAB, JPAB, DWPI, 23-(manganese) ) and zinc TDBD, FPRS 2014 PGPB, USPT, USOC, 12-(c22b47/00.cpc, or c22b19/00.cpc, or .... c25c1/10.cpc, or c25c1/16.cpc, and EPAB, JPAB, DWPL OR YES 23-TDBD, FPRS 2014 (manganese) and zinc ) and silica PGPB, USPT, USOC. 12-...: EPAB, JPAB, DWPI, OR YES 23manganese or TOBO, FPRS 2014 PGPB, USPT, USOC, 12-EPAB, JPAB, DWPI, OB YES .... manganese or mn 23-TDBD, FPRS 2014 PGPB, USPT, USOC, 12~ .... EPAB, JPAB, DWPI. OR YES 23-(manganese or mn ) and (zinc or zn) TDBD, FPRS 2014 PGPB, USPT, USOC, 12-(manganese or mn and (zinc or zn) ) and ...: YES EPAB, JPAB, DWPL, OR 23~ silica TOBO, FPRS 2014 PGPB, USPT, USOC, 12-(manganese or mn and (zinc or zn) and .... EPAB, JPAB, DWPI, OB YES 23silica) and brine TDBD, FPRS 2014 (manganese or mn and (zinc or zn) and PGPB, USPT, USOC, 12~ ...: EPAB, JPAB, DWPL, OH. YES silica and brine ) and (manganese same 23-TD8D, FPRS 2014 (manganese or min and (zinc or zn) and PGPB, USPT, USOC, 12-...: silica and brine and (manganese same zinc) OR YES 23-EPAB, JPAB, DWPI, ) and iron TOBD, FPRS 2014

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# Index of Claims 13943395 Examiner Application/Control No. Applicant(s)/Patent Under Reexamination HARRISON ET AL. Art Unit 1756

<b>✓</b>	Rejected	-	Cancelled	N	Non-Elected	A	Appeal
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	Application/Control No.	Applicant(s)/Patent Under Reexamination
Index of Claims	13943395	HARRISON ET AL.
	Examiner	Art Unit
	ARUN S PHASGE	1756

<b>✓</b>	✓ Rejected		- Cancelled			N Non-Elected				Α	Appeal		eal	
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U.S. Patent and Trademark Office Part of Paper No.: 20141222

Doc code: IDS Doc description: Information Disclosure Statement (IDS) Filed PTO/SB/08a (01-10)
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INFORMATION DISCLOSURE STATEMENT BY APPLICANT (Not for submission under 37 CFR 1.99)  Application Number 13943395  Filing Date 2013-07-16  First Named Inventor Harrison, Stephen, et al.  Art Unit Examiner Name Unknown  Attorney Docket Number 076165.000099	2013-07-16		
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Examiner Initial*	Cite No	Patent Number	Kind Code <sup>1</sup>	Issue Date	Name of Patentee or Applicant of cited Document	Pages,Columns,Lines where Relevant Passages or Relevant Figures Appear
/A.P./	1	5229003	B1	1993-07-20	Duyvesteyn, Willem	
/A.P./	2	5244491	B1	1993-09-14	Brown, et al.	
/A.P./	3	5246684	B1	1993-09-21	Brown, et al.	
/A.P./	4	5358700	B1	1994-10-25	Brown, et al.	
/A.P./	5	5656172	B1	1997-08-12	Kitz, et al.	
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Application Number		13943395			
Filing Date		2013-07-16			
First Named Inventor Harris		on, Stephen, et al.			
Art Unit					
Examiner Name	Unkno	own			
Attorney Docket Number		076165.000099			

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/A.P./	1		DOMQUIST, R. Gordon, Economic Benefits of Mineral Extraction from Geothermal Brines, The Minerals, Metals & erials Society, 2006, pp. 553-558, Vol. 6, US										
/A.P./	2		GALLUP, D. L., et al., Heavy Metal Sulfide Scaling in a Production Well at the Salton Sea Geothermal Field, Geothermal Resources Council Transactions, 1990, pp. 1583-1590, Vol. 14, Part II, Davis, CA, US										
/A.P./	3		s, Journal of the			om geothermal brines to co of Geochemistry and Cosm							
/A.P./	4	MANCEAU, Alain, et al American Mineralogist,				silicate domains in geother	mal brine precipitates,						
/A.P./	5	PASCUA, C. S., et al., Geothermics, 2007, pp.				ieval of silica from spent ge	othermal brine,						
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/A.P./	7		CHULTZE, L. E., et al., Operation of a Mineral Recovery Unit on Brine from the Salton Sea Known Geothermal esource Area, Report of Investigations 8680, Bureau of Mines, 1982, pp. 1-12, Vol. 505, US										

# INFORMATION DISCLOSURE STATEMENT BY APPLICANT (Not for submission under 37 CFR 1.99) Application Number 13943395 Filing Date 2013-07-16 First Named Inventor Harrison, Stephen, et al. Art Unit Examiner Name Unknown Attorney Docket Number 076165.000099

/A.P./	SCHULTZE, L.E., Techniques for Recovering Metal Values From Postflash Geothermal Brines, Transactions, 1984, pp. 159-162, Vol. 8, Davis, CA, US									
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Standard S <sup>-</sup> <sup>4</sup> Kind of do	T.3). <sup>3</sup> F cument	of USPTO Patent Documents at <a href="https://www.uspto.gov">www.uspto.gov</a> or MPEP 901.04. 2 Enter office of Japanese patent documents, the indication of the year of the reign of the Employ the appropriate symbols as indicated on the document under WIPO Standard anslation is attached.	peror must precede the ser	ial number of the patent doc	ument.					

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Application Number		13943395
Filing Date		2013-07-16
First Named Inventor	Harris	son, Stephen, et al.
Art Unit		
Examiner Name Unknown		own
Attorney Docket Numb	er	076165.000099

		CERTIFICATION	STATEMENT						
Plea	ise see 37 CFR 1	.97 and 1.98 to make the appropriate selecti	on(s):						
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- 3. A record in this system of records may be disclosed, as a routine use, to a Member of Congress submitting a request involving an individual, to whom the record pertains, when the individual has requested assistance from the Member with respect to the subject matter of the record.
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APPLICATION NUMBER GROUP ART UNIT FILE WRAPPER LOCATION PATENT NUMBER 13/943,395 1756 1756



#### Correspondence Address/Fee Address Change

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126930 **DENTONS US LLP** P.O. Box 061080 Chicago, IL 60606-7020



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APPLICATION NUMBER FILING OR 371(C) DATE FIRST NAMED APPLICANT ATTY. DOCKET NO./TITLE 13/943,395 07/16/2013 Stephen Harrison 076165.000099

26263 **DENTONS US LLP** P.O. BOX 061080 CHICAGO, IL 60606-1080

**CONFIRMATION NO. 2954** POA ACCEPTANCE LETTER



Date Mailed: 07/03/2014

#### NOTICE OF ACCEPTANCE OF POWER OF ATTORNEY

This is in response to the Power of Attorney filed 06/30/2014.

The Power of Attorney in this application is accepted. Correspondence in this application will be mailed to the above address as provided by 37 CFR 1.33.

/ltaba/					
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APPLICATION NUMBER

FILING OR 371(C) DATE

FIRST NAMED APPLICANT

ATTY. DOCKET NO./TITLE 076165.000099

13/943,395

Bracewell & Giuliani LLP

Houston, TX 77208-1389

P.O. Box 61389

07/16/2013

Stephen Harrison

**CONFIRMATION NO. 2954** 

**POWER OF ATTORNEY NOTICE** 

Date Mailed: 07/03/2014

#### NOTICE REGARDING CHANGE OF POWER OF ATTORNEY

This is in response to the Power of Attorney filed 06/30/2014.

• The Power of Attorney to you in this application has been revoked by the applicant. Future correspondence will be mailed to the new address of record(37 CFR 1.33).

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I hereby revoke all previous powers of attorney given in the application identified in the attached statement under 37 CFR 3.73(c).								
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					ignee of Re			
	The	Individual whose sig					alf of the assignee	
Signa	ature	Z. No.	visor	\		Date May 30th 2014		4
Nam	3	Stephen Harris	son			Telephone 925.	226.7418	
Title Chief Technology Officer & Vice President								

This collection of information is required by 37 CFR 1.31, 1.32 and 1.33. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.11 and 1.14. This collection is estimated to take 3 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

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Applicant/Patent Owner		July 16, 2012
Application No./Patent I		Filed/Issue Date: July 16, 2013
Titled: Selective Rec	overy of Manganese, Lead an	corporation
(Name of Assignee)	•	ype of Assignee, e.g., corporation, partnership, university, government agency, etc.)
		ove, it is (choose one of options 1, 2, 3 or 4 below):
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- 2. A record from this system of records may be disclosed, as a routine use, in the course of presenting evidence to a court, magistrate, or administrative tribunal, including disclosures to opposing counsel in the course of settlement negotiations.
- 3. A record in this system of records may be disclosed, as a routine use, to a Member of **Congress** submitting a request involving an individual, to whom the record pertains, when the individual **has** requested assistance from the Member with respect to the subject matter of the **record**.
- 4. A record in this system of records may be disclosed, as a routine use, to a contractor of the Agency having need for the information in order to perform a contract. Recipients of information shall be required to comply with the requirements of the Privacy Act of 1974, as amended, pursuant to 5 U.S.C. 552a(m).
- 5. A record related to an International Application filed under the Patent Cooperation Treaty in this system of records may be disclosed, as a routine use, to the International Bureau of the World Intellectual Property Organization, pursuant to the Patent Cooperation Treaty.
- 6. A record in this system of records may be disclosed, as a routine use, to another federal agency for purposes of National Security review (35 U.S.C. 181) and for review pursuant to the Atomic Energy Act (42 U.S.C. 218(c)).
- 7. A record from this system of records may be disclosed, as a routine use, to the Administrator, General Services, or his/her designee, during an inspection of records conducted by GSA as part of that agency's responsibility to recommend improvements in records management practices and programs, under authority of 44 U.S.C. 2904 and 2906. Such disclosure shall be made in accordance with the GSA regulations governing inspection of records for this purpose, and any other relevant (i.e., GSA or Commerce) directive. Such disclosure shall not be used to make determinations about individuals.
- 8. A record from this system of records may be disclosed, as a routine use, to the public after either publication of the application pursuant to 35 U.S.C. 122(b) or issuance of a patent pursuant to 35 U.S.C. 151. Further, a record may be disclosed, subject to the limitations of 37 CFR 1.14, as a routine use, to the public if the record was filed in an application which became abandoned or in which the proceedings were terminated and which application is referenced by either a published application, an application open to public inspection or an issued patent.
- 9. A record from this system of records may be disclosed, as a routine use, to a Federal, State, or local law enforcement agency, if the USPTO becomes aware of a violation or potential violation of law or regulation.

Electronic Acknowledgement Receipt			
EFS ID:	19456037		
Application Number:	13943395		
International Application Number:			
Confirmation Number:	2954		
Title of Invention:	Selective Recovery of Manganese, Lead and Zinc		
First Named Inventor/Applicant Name:	Stephen Harrison		
Customer Number:	35979		
Filer:	Heather Megan Khassian/Debra Odum		
Filer Authorized By:	Heather Megan Khassian		
Attorney Docket Number:	076165.000099		
Receipt Date:	30-JUN-2014		
Filing Date:	16-JUL-2013		
Time Stamp:	16:39:18		
Application Type:	Utility under 35 USC 111(a)		

## **Payment information:**

Submitted with Payment	no
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## File Listing:

Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)
1	Power of Attorney	20140630Power of Attorney.pdf	31601 7b547bd5ba65234f50c03113a6d52444f18 65ac1	no	1

#### Warnings:

Information:	ESM / PETITIONER EX 1014 87	

2	Assignee showing of ownership per 37	20140630 Assignee Statement U	83890	no	3
	CFR 3.73.	nder 37 CFR 373. pdf	21fb023c9e3fb80c8e86f5d9c725d4ee7503 81e1		3
Warnings:					
Information:					
		Total Files Size (in bytes):	1	15491	

This Acknowledgement Receipt evidences receipt on the noted date by the USPTO of the indicated documents, characterized by the applicant, and including page counts, where applicable. It serves as evidence of receipt similar to a Post Card, as described in MPEP 503.

#### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

#### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

#### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.



#### United States Patent and Trademark Office

UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS PO Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov

APPLICATION	FILING or	GRP ART				
NUMBER	371(c) DATE	UNIT	FIL FEE REC'D	ATTY.DOCKET.NO	TOT CLAIMS	IND CLAIMS
13/943 395	07/16/2013	1756	1660	076165 000099	16	3

35979 Bracewell & Giuliani LLP P.O. Box 61389 Houston, TX 77208-1389 CONFIRMATION NO. 2954 REPLACEMENT FILING RECEIPT



Date Mailed: 12/02/2013

Receipt is acknowledged of this non-provisional patent application. The application will be taken up for examination in due course. Applicant will be notified as to the results of the examination. Any correspondence concerning the application must include the following identification information: the U.S. APPLICATION NUMBER, FILING DATE, NAME OF APPLICANT, and TITLE OF INVENTION. Fees transmitted by check or draft are subject to collection. Please verify the accuracy of the data presented on this receipt. If an error is noted on this Filing Receipt, please submit a written request for a Filing Receipt Correction. Please provide a copy of this Filing Receipt with the changes noted thereon. If you received a "Notice to File Missing Parts" for this application, please submit any corrections to this Filing Receipt with your reply to the Notice. When the USPTO processes the reply to the Notice, the USPTO will generate another Filing Receipt incorporating the requested corrections

#### Inventor(s)

Stephen Harrison, Benicia, CA; Samaresh Mohanta, Pleasanton, CA; C.V. Krishnamohan Sharma, Milpitas, CA; Elizabeth Geler, San Rafael, CA;

Applicant(s)

SIMBOL Inc., Pleasanton, CA

**Assignment For Published Patent Application** 

SIMBOL Inc., Pleasanton, CA

Power of Attorney: The patent practitioners associated with Customer Number 35979

Domestic Priority data as claimed by applicant

This application is a CON of 13/539,106 06/29/2012 PAT 8518232 which claims benefit of 61/502,736 06/29/2011

**Foreign Applications** for which priority is claimed (You may be eligible to benefit from the **Patent Prosecution Highway** program at the USPTO. Please see <a href="http://www.uspto.gov">http://www.uspto.gov</a> for more information.) - None. Foreign application information must be provided in an Application Data Sheet in order to constitute a claim to foreign priority. See 37 CFR 1.55 and 1.76.

If Required, Foreign Filing License Granted: 07/31/2013

The country code and number of your priority application, to be used for filing abroad under the Paris Convention,

is **US 13/943,395** 

Projected Publication Date: Request for Non-Publication Acknowledged

Non-Publication Request: Yes

page 1 of 3

Early Publication Request: No \*\* SMALL ENTITY \*\*
Title

Selective Recovery of Manganese, Lead and Zinc

**Preliminary Class** 

205

Statement under 37 CFR 1.55 or 1.78 for AIA (First Inventor to File) Transition Applications: No

#### PROTECTING YOUR INVENTION OUTSIDE THE UNITED STATES

Since the rights granted by a U.S. patent extend only throughout the territory of the United States and have no effect in a foreign country, an inventor who wishes patent protection in another country must apply for a patent in a specific country or in regional patent offices. Applicants may wish to consider the filing of an international application under the Patent Cooperation Treaty (PCT). An international (PCT) application generally has the same effect as a regular national patent application in each PCT-member country. The PCT process **simplifies** the filing of patent applications on the same invention in member countries, but **does not result** in a grant of "an international patent" and does not eliminate the need of applicants to file additional documents and fees in countries where patent protection is desired.

Almost every country has its own patent law, and a person desiring a patent in a particular country must make an application for patent in that country in accordance with its particular laws. Since the laws of many countries differ in various respects from the patent law of the United States, applicants are advised to seek guidance from specific foreign countries to ensure that patent rights are not lost prematurely.

Applicants also are advised that in the case of inventions made in the United States, the Director of the USPTO must issue a license before applicants can apply for a patent in a foreign country. The filing of a U.S. patent application serves as a request for a foreign filing license. The application's filing receipt contains further information and guidance as to the status of applicant's license for foreign filing.

Applicants may wish to consult the USPTO booklet, "General Information Concerning Patents" (specifically, the section entitled "Treaties and Foreign Patents") for more information on timeframes and deadlines for filing foreign patent applications. The guide is available either by contacting the USPTO Contact Center at 800-786-9199, or it can be viewed on the USPTO website at http://www.uspto.gov/web/offices/pac/doc/general/index.html.

For information on preventing theft of your intellectual property (patents, trademarks and copyrights), you may wish to consult the U.S. Government website, http://www.stopfakes.gov. Part of a Department of Commerce initiative, this website includes self-help "toolkits" giving innovators guidance on how to protect intellectual property in specific countries such as China, Korea and Mexico. For questions regarding patent enforcement issues, applicants may call the U.S. Government hotline at 1-866-999-HALT (1-866-999-4258).

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#### Title 35, United States Code, Section 184

#### Title 37, Code of Federal Regulations, 5.11 & 5.15

#### **GRANTED**

The applicant has been granted a license under 35 U.S.C. 184, if the phrase "IF REQUIRED, FOREIGN FILING LICENSE GRANTED" followed by a date appears on this form. Such licenses are issued in all applications where the conditions for issuance of a license have been met, regardless of whether or not a license may be required as set forth in 37 CFR 5.15. The scope and limitations of this license are set forth in 37 CFR 5.15(a) unless an earlier license has been issued under 37 CFR 5.15(b). The license is subject to revocation upon written notification. The date indicated is the effective date of the license, unless an earlier license of similar scope has been granted under 37 CFR 5.13 or 5.14.

This license is to be retained by the licensee and may be used at any time on or after the effective date thereof unless it is revoked. This license is automatically transferred to any related applications(s) filed under 37 CFR 1.53(d). This license is not retroactive.

The grant of a license does not in any way lessen the responsibility of a licensee for the security of the subject matter as imposed by any Government contract or the provisions of existing laws relating to espionage and the national security or the export of technical data. Licensees should apprise themselves of current regulations especially with respect to certain countries, of other agencies, particularly the Office of Defense Trade Controls, Department of State (with respect to Arms, Munitions and Implements of War (22 CFR 121-128)); the Bureau of Industry and Security, Department of Commerce (15 CFR parts 730-774); the Office of Foreign AssetsControl, Department of Treasury (31 CFR Parts 500+) and the Department of Energy.

#### **NOT GRANTED**

No license under 35 U.S.C. 184 has been granted at this time, if the phrase "IF REQUIRED, FOREIGN FILING LICENSE GRANTED" DOES NOT appear on this form. Applicant may still petition for a license under 37 CFR 5.12, if a license is desired before the expiration of 6 months from the filing date of the application. If 6 months has lapsed from the filing date of this application and the licensee has not received any indication of a secrecy order under 35 U.S.C. 181, the licensee may foreign file the application pursuant to 37 CFR 5.15(b).

#### SelectUSA

The United States represents the largest, most dynamic marketplace in the world and is an unparalleled location for business investment, innovation, and commercialization of new technologies. The U.S. offers tremendous resources and advantages for those who invest and manufacture goods here. Through SelectUSA, our nation works to promote and facilitate business investment. SelectUSA provides information assistance to the international investor community; serves as an ombudsman for existing and potential investors; advocates on behalf of U.S. cities, states, and regions competing for global investment; and counsels U.S. economic development organizations on investment attraction best practices. To learn more about why the United States is the best country in the world to develop technology, manufacture products, deliver services, and grow your business, visit <a href="http://www.SelectUSA.gov">http://www.SelectUSA.gov</a> or call +1-202-482-6800.



#### United States Patent and Trademark Office

UNITED STATES DEPARTMENT OF COMMI United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS PO. Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov UNITED STATES DEPARTMENT OF COMMERCE

APPLICATION NUMBER FILING OR 371(C) DATE FIRST NAMED APPLICANT ATTY. DOCKET NO./TITLE 13/943,395 07/16/2013 Stephen Harrison 076165.000099

35979 Bracewell & Giuliani LLP P.O. Box 61389 Houston, TX 77208-1389

**CONFIRMATION NO. 2954** POA ACCEPTANCE LETTER



Date Mailed: 12/02/2013

#### NOTICE OF ACCEPTANCE OF POWER OF ATTORNEY

This is in response to the Power of Attorney filed 11/25/2013.

The Power of Attorney in this application is accepted. Correspondence in this application will be mailed to the above address as provided by 37 CFR 1.33.

	/mbeyene/				
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Office of Data Management, Application Assistance Unit (571) 272-4000, or (571) 272-4200, or 1-888-786-0101

#### POWER OF ATTORNEY BY APPLICANT

This Power of Attorney by Applicant is directed to the application titled <b>SELECTIVE</b>
RECOVERY OF MANGANESE, LEAD AND ZINC,
the application of which is attached hereto;
X the US Application No. 13/943,395 filed on July 16, 2013; or
the PCT international application number filed on
I hereby revoke all previous powers of attorney given in the above-identified application.
I hereby appoint the practitioners associated with Bracewell & Giuliani LLP's U.S. Patent
and Trademark Office Customer No. 35979 as my/our attorney(s) or agent(s), and to transact all
business in the U.S. Patent and Trademark Office connected therewith for the application
referenced above.
Please direct all correspondence to the address associated with:
U.S. Patent and Trademark Office Customer No. 35979
In the event of questions, the attorney in charge can be reached at michaels@bgllp.com
or (713) 221.1112.
I am the Applicant's Assignee or Person to Whom the Inventor is Under an Obligation to
Assign.  SIGNATURE: DATE: 9 25 2013
NAME: Stephen Harrison
TITLE: Chief Technology Officer & Vice-President
COMPANY: Simbol Inc.

Electronic Acknowledgement Receipt			
EFS ID:	17498007		
Application Number:	13943395		
International Application Number:			
Confirmation Number:	2954		
Title of Invention:	Selective Recovery of Manganese, Lead and Zinc		
First Named Inventor/Applicant Name:	Stephen Harrison		
Customer Number:	35979		
Filer:	Karthika Perumal/Debra Odum		
Filer Authorized By:	Karthika Perumal		
Attorney Docket Number:	076165.000099		
Receipt Date:	25-NOV-2013		
Filing Date:	16-JUL-2013		
Time Stamp:	16:55:32		
Application Type:	Utility under 35 USC 111(a)		

## **Payment information:**

Submitted with Payment	no
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## File Listing:

Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)
1	Application Data Sheet	20131125MarkedUPApplicatio	144317	no	6
1	Application Data Sheet	nDataSheet.PDF	a2bcb047f680508a992c6a5f4d21b523a01a f292		

#### **Warnings:**

Information: ESM / PETITIONER EX 1014\_94

This is not an l	JSPTO supplied ADS fillable form				
2	Assignee showing of ownership per 37	20131125StatementUnder373.	36450	no	2
_	CFR 3.73.	pdf	1cb372f63b206346dc72153b2c11fa583e5 1391a		
Warnings:					
Information	:				
3	Power of Attorney	20131125PowerofAttorney.PDF	32305	no	1
_	, 5,7,4,5,1,5,		d4f4afe550f8981cce604010d02d72db009d 0ca3		
Warnings:					
Information	:				
		213072			

This Acknowledgement Receipt evidences receipt on the noted date by the USPTO of the indicated documents, characterized by the applicant, and including page counts, where applicable. It serves as evidence of receipt similar to a Post Card, as described in MPEP 503.

#### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

#### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

#### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

Application Data Sheet 37 CFR 1.76				Attorney Docket Number			076165.000099				
				Application Number							
Title of	tle of Invention Selective Recovery of Manganese, Lead and Zinc										
										ollowing form contains	the
This doc	bibliographic data arranged in a format specified by the United States Patent and Trademark Office as outlined in 37 CFR 1.76.  This document may be completed electronically and submitted to the Office in electronic format using the Electronic Filing System (EFS) or the document may be printed and included in a paper filed application.										
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	Steph	en						Harrison		-	
Resid	ence l	nformation	(Select One)	<ul><li>US</li></ul>	Residency	0	Non US Re	esidency	O Activ	e US Military Servic	е
City	Benic	а		State/	Province	CA	Count	ry of Resi	dence	us	
Mailing	Addre	ss of Invent	or:								. "
Addres	ss 1		134 Panoram	a Drive							
Addres	ss 2										
City		Benicia					State/Pro	vince	CA		
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Addres	ss 1		3278 Curtis C	irde				<del></del> .	:	·	
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Application Data She			eet 37 CFR 1.76			t Number	076165	.000099					
Application Data Sheet 37 CFK 1.76 Application Nu					mber								
Title of	Title of Invention Selective Recovery of Manganese, Lead and Zinc												
City	Milpit	tas			State/	Province	CA	Count	ry of Res	sidencė	us		
Mailing	Mailing Address of Inventor:												
Addres	ss 1			1126 Ridgemo	ont Drive	<del></del>							
Addres	ss 2												
City		Milpitas						State/Pro	vince	CA			
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An	Addı	ess is be	eing	provided for	the co	rresponde	nce li	nformation	of this a	pplicatio	n.		
Custo	mer N	umber		35979					·	· ·			
Email Address docketing@bgllp.com							Add	Email .	Remove	Email			
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Total N	Numb	er of Dra	wing	Sheets (if a	ny)	17		Sugges	ted Figui	re for Pul	olication (	if any)	

Limited Recognition (37 CFR 11.9)

Under the Paperwork Reduction Act of 1995, no persons are required to respond to a collection of information unless it contains a valid OMB control number,

Annlication Da	ata Sheet 37 CFR 1.76	Attorney Docket Number	076165.000099						
Application Da	ita Sileet 37 CFK 1,70	Application Number							
Title of Invention Selective Recovery of Manganese, Lead and Zinc									
Publication Information:									
Request Early	/ Publication (Fee required a	t time of Request 37 CFR 1.2	219)						
Request Not to Publish. I hereby request that the attached application not be published under  35 U.S.C. 122(b) and certify that the invention disclosed in the attached application has not and will not be the subject of an application filed in another country, or under a multilateral international agreement, that requires publication at eighteen months after filing.									
D									
Representative Information:									
Representative information should be provided for all practitioners having a power of attorney in the application. Providing this information in the Application Data Sheet does not constitute a power of attorney in the application (see 37 CFR 1.32). Either enter Customer Number or complete the Representative Name section below. If both sections are completed the customer Number will be used for the Representative Information during processing.									

### **Domestic Benefit/National Stage Information:**

35979

Customer Number

This section allows for the applicant to either claim benefit under 35 U.S.C. 119(e), 120, 121, or 365(c) or indicate National Stage entry from a PCT application. Providing this information in the application data sheet constitutes the specific reference required by 35 U.S.C. 119(e) or 120, and 37 CFR 1.78.

US Patent Practitioner

Prior Application Status	Pending	Remove						
Application Number	Continuity Type	Prior Application Number	Filing Date (YYYY-MM-DD)					
	Continuation of	13/539106	2012-06-29					
Prior Application Status	Expired		Remove					
Application Number	Continuity Type	Prior Application Number	Filing Date (YYYY-MM-DD)					
13/539106	non provisional of	61/502736	2011-06-29					
Additional Domestic Renefit/National Stage Data may be generated within this form								

Additional Domestic Benefit/National Stage Data may be generated within this form by selecting the **Add** button.

### Foreign Priority Information:

Please Select One:

**Customer Number** 

This section allows for the applicant to claim priority to a foreign application. Providing this information in the application data sheet constitutes the claim for priority as required by 35 U.S.C. 119(b) and 37 CFR 1.55(d). When priority is claimed to a foreign application that is eligible for retrieval under the priority document exchange program (PDX)<sup>1</sup> the information will be used by the Office to automatically attempt retrieval pursuant to 37 CFR 1.55(h)(1) and (2). Under the PDX program, applicant bears the ultimate responsibility for ensuring that a copy of the foreign application is received by the Office from the participating foreign intellectual property office, or a certified copy of the foreign priority application is filed, within the time period specified in 37 CFR 1.55(g)(1).

Application Data Sheet 37 CFR 1.76		Attorney Docket Number	076165.000	099					
Application Data Sheet 37 CFK 1.76			Application Number						
Title of Invention	tle of Invention Selective Recovery of Manganese, Lead and Zinc								
		·			Remove				
Application Nu	mber	Country <sup>i</sup>	Filing Date (YYYY	-MM-DD)	Access Code <sup>i</sup> (if applicable)				
Additional Foreign Priority Data may be generated within this form by selecting the									
Add button.									

## Statement under 37 CFR 1.55 or 1.78 for AIA (First Inventor to File) Transition Applications

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#### **Authorization to Permit Access:**

A the effection to Dermit Access to the Instant Application by the Portionating Offices
Authorization to Permit Access to the Instant Application by the Participating Offices
If checked, the undersigned hereby grants the USPTO authority to provide the European Patent Office (EPO), the Japan Patent Office (JPO), the Korean Intellectual Property Office (KIPO), the World Intellectual Property Office (WIPO), and any other intellectual property offices in which a foreign application claiming priority to the instant patent application is filed access to the instant patent application. See 37 CFR 1.14(c) and (h). This box should not be checked if the applicant does not wish the EPO, JPO, KIPO, WIPO, or other intellectual property office in which a foreign application claiming priority to the instant patent application is filed to have access to the instant patent application.
In accordance with 37 CFR 1.14(h)(3), access will be provided to a copy of the instant patent application with respect to: 1) the instant patent application-as-filed; 2) any foreign application to which the instant patent application claims priority under 35 U.S.C. 119(a)-(d) if a copy of the foreign application that satisfies the certified copy requirement of 37 CFR 1.55 has been filed in the instant patent application; and 3) any U.S. application-as-filed from which benefit is sought in the instant patent application.
In accordance with 37 CFR 1.14(c), access may be provided to information concerning the date o f filing this Authorization.

## **Applicant Information:**

Providing assignment information in this section does not substitute for compliance with any requirement of part 3 of Title 37 of CFR to have an assignment recorded by the Office.

Application Data Sheet 37 CFR 1.76			Attorney Docke	et Number	076165.000099		
			Application Nu	mber			
Title of Invention	Selecti	ve Recovery of Mangar	nese, Lead and Zir	nc			
Applicant 1							
The information to be 1.43; or the name and who otherwise shows applicant under 37 CF	provided address sufficient R 1.46 (a gether w	in this section is the na of the assignee, person proprietary interest in tassignee, person to who	ame and address on to whom the inve the matter who is to om the inventor is	of the legal rep entor is under he applicant u obligated to as	R 1.45), this section should not be completed. resentative who is the applicant under 37 CFR an obligation to assign the invention, or person under 37 CFR 1.46. If the applicant is an ssign, or person who otherwise shows sufficient inventors who are also the applicant should be		
<ul><li>Assignee</li></ul>		◯ Legal Re	epresentative unde	er 35 U.S.C. 1	Joint Inventor		
Person to whom th	ne invento	or is obligated to assign.		Person	who shows sufficient proprietary interest		
If applicant is the leg	gal repre	esentative, indicate th	ne authority to file	e the patent a	application, the inventor is:		
Name of the Doos	and ar l	egally Incapacitated	Inventor:				
		nization check here.			·		
Organization Nam	e <u>sı</u>	MBOL Inc.	· · · · · · · · · · · · · · · · · · ·				
Mailing Address	Informa	tion For Applicant:					
Address 1		6920 Koll Center Par	kway				
Address 2							
City		<u>Pleasanton</u>		State/Provin	ice <u>CA</u>		
Country US		•		Postal Code	94566		
Phone Number				Fax Number			
Email Address							
Additional Applicant	Data m	ay be generated with	nin this form by s	electing the	Add button.		
Assignee Info	ormat	ion including	Non-Applic	ant Assi	gnee Information:		
Providing assignment have an assignment r			not subsitute for c	ompliance witl	h any requirement of part 3 of Title 37 of CFR to		
Assignee 1							
application publication	i . An ass icant. Fo	ignee-applicant identifi	ed in the "Applicar	nt Information"	ation, is desired to be included on the patent section will appear on the patent application entification as an assignee is also desired on the		
If the Assignee is a	an Orgai	nization check here.					

Application Data Sheet 37 CFR 1.76			Attorney Docket Number		07616	076165.000099	
			Application N	lumber			
Title of Invention	Selective	e Recovery of Manganese, Lead and Zinc					
Prefix <b>Giv</b>		ven Name	Middle Name		Family	Name	Suffix
Mailing Address	Informati	on For Non-Applic	ant Assignee	:			
Address 1							
Address 2							
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Country i				Postal Co	de		
Phone Number				Fax Number			
Email Address							
Additional Assignee Data may be generated within this form by selecting the Add button.							
Signature:							
NOTE. This form reget he signed in accordance with 27 CED 1.22. See 27 CED 1.4 for signeture requirements and							

Signature	nature kaethika			Date (YYYY-MM-DD)	2013-11-12
First Name	Karthika	Last Name	Perumal	Registration Number	69296

This collection of information is required by 37 CFR 1.76. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.14. This collection is estimated to take 23 minutes to complete, including gathering, preparing, and submitting the completed application data sheet form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. **SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.** 

Approved for use through 01/31/2013. OMB 0651-0031

U.S. Patent and Trademark Office; U.S. DEPARTMENT OF COMMERCE

Under the Paperwork Reduction Act of 1995, no persons are required to respond to a collection of information unless it displays a valid OMB control number.

STATEMENT UNDER 37 CFR 3.73(c)	
Applicant/Patent Owner: Simbol Inc.	
Application No./Patent No.: 13/943,395 Filed/Issue Date: July 16, 2013	
Titled: Selective Recovery of Manganese, Lead and Zinc	
Simbol Inc. Corporation	
(Name of Assignee) (Type of Assignee, e.g., corporation, partnership, university, government ag	gency, etc.)
states that, for the patent application/patent identified above, it is (choose one of options 1, 2, 3 or 4 below):	
1.  The assignee of the entire right, title, and interest.	
2. An assignee of less than the entire right, title, and interest (check applicable box):	
The extent (by percentage) of its ownership interest is	e owners
There are unspecified percentages of ownership. The other parties, including inventors, who together right, title and interest are:	own the entire
Additional Statement(s) by the owner(s) holding the balance of the interest <u>must be submitted</u> to accouright, title, and interest.	unt for the entire
3. The assignee of an undivided interest in the entirety (a complete assignment from one of the joint inventor. The other parties, including inventors, who together own the entire right, title, and interest are:	rs was made).
Additional Statement(s) by the owner(s) holding the balance of the interest <u>must be submitted</u> to account right, title, and interest.	nt for the entire
4. The recipient, via a court proceeding or the like (e.g., bankruptcy, probate), of an undivided interest in the complete transfer of ownership interest was made). The certified document(s) showing the transfer is attached.	entirety (a
The interest identified in option 1, 2 or 3 above (not option 4) is evidenced by either (choose one of options A or B	3 below):
A. An assignment from the inventor(s) of the patent application/patent identified above. The assignment was the United States Patent and Trademark Office at Reel 031428, Frame 0328, or for which thereof is attached.	
B. A chain of title from the inventor(s), of the patent application/patent identified above, to the current assign	ee as follows:
1. From: To:	
The document was recorded in the United States Patent and Trademark Office at	
Reel, Frame, or for which a copy thereof is attached.	
2. From: To:	A A A A A A A A A A A A A A A A A A A
The document was recorded in the United States Patent and Trademark Office at	
Reel, Frame, or for which a copy thereof is attached.	

[Page 1 of 2]
This collection of information is required by 37 CFR 3.73(b). The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.11 and 1.14. This collection is estimated to take 12 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form analytic varyed to suppose the complete of complete the complete of complete and the complete of complet Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. **SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450**.

If you need assistance in completing the form, call 1-800-PTO-9199 and select option 2.

		<u>STATEME</u>	NT UNDER 37 CFR 3.73	<u>(c)</u>	
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☐ Add	ditional document	s in the chain of title are	e listed on a supplemental shee	et(s).	
			nentary evidence of the chain o ted for recordation pursuant to	of title from the original owner to the 37 CFR 3.11.	
				nt(s)) must be submitted to Assignment ecords of the USPTO. See MPEP 302.08]	
The undersig	ned (whose title i	s supplied below) is autl	horized to act on behalf of the a	assignee.	
Ka	ethika			November 25, 2013	
Signature				Date	
Karthika	a Perumal			69296	
Printed or Ty	ped Name		·	Title or Registration Number	

[Page 2 of 2]



#### UNITED STATES PATENT AND TRADEMARK OFFICE

UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS P.O. Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov

APPLICATION NUMBER FILING OR 371(C) DATE FIRST NAMED APPLICANT ATTY. DOCKET NO./TITLE

13/943,395 07/16/2013 Stephen Harrison

076165.000099 CONFIRMATION NO. 2954

35979 Bracewell & Giuliani LLP P.O. Box 61389 Houston, TX 77208-1389 IMPROPER CPOA LETTER



Date Mailed: 11/20/2013

#### NOTICE REGARDING POWER OF ATTORNEY

This is in response to the power of attorney filed 11/12/2013. The power of attorney in this application is not accepted for the reason(s) listed below:

• The power of attorney has not been accepted because the party who is giving power has not been identified. Power of attorney may only be signed by the applicant for patent (37 CFR 1.42) or the patent owner. A party who is not the applicant must become the applicant in accordance with 37 CFR 1.46(c) and appoint any power of attorney in compliance with 37 CFR 3.71 and 3.73. For a reissue application, reexamination proceeding, or supplemental examination proceeding, a patent owner who was not the applicant under 37 CFR 1.46 must appoint any power of attorney in compliance with 37 CFR 3.71 and 3.73. See 37 CFR 1.32(b)(4).

/hchristian/		
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Office of Data Management, Application Assistance Unit (571) 272-4000, or (571) 272-4200, or 1-888-786-0101



#### United States Patent and Trademark Office

UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS PC. Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov

APPLICATION NUMBER FILING OR 371(C) DATE FIRST NAMED APPLICANT ATTY. DOCKET NO./TITLE

13/943,395 07/16/2013 Stephen Harrison

076165.000099

35979 Bracewell & Giuliani LLP P.O. Box 61389 Houston, TX 77208-1389 CONFIRMATION NO. 2954 IMPROPER CFR REQUEST



Date Mailed: 11/20/2013

#### RESPONSE TO REQUEST FOR CORRECTED FILING RECEIPT

Power of Attorney, Claims, Fees, System Limitations, and Miscellaneous

In response to your request for a corrected Filing Receipt, the Office is unable to comply with your request because:

Any request to correct or update the name of the applicant must include an application data sheet (ADS) in compliance with 37 CFR 1.76 specifying the correct or updated name of the applicant in the applicant information section. Any request to change the applicant after an original applicant has been specified under 37 CFR 1.46(b) must include a new ADS in compliance with 37 CFR 1.76 specifying the applicant in the applicant information section and comply with 37 CFR 3.71 and 3.73. See 37 CFR 1.46(c).

/hchristian/			
Office of Data Management	Application Assistance Unit (571) 272-4	1000 or (571) 272-420(	) or 1-888-786-0101

#### POWER OF ATTORNEY BY APPLICANT

This Power of Attorney by Applicant is directed to the application titled <b>SELECTIVE</b>
RECOVERY OF MANGANESE, LEAD AND ZINC,
the application of which is attached hereto;
X the US Application No. 13/943,395 filed on July 16, 2013; or
the PCT international application number filed on
I hereby revoke all previous powers of attorney given in the above-identified application.
I hereby appoint the practitioners associated with Bracewell & Giuliani LLP's U.S. Patent
and Trademark Office Customer No. 35979 as my/our attorney(s) or agent(s), and to transact all
business in the U.S. Patent and Trademark Office connected therewith for the application
referenced above.
Please direct all correspondence to the address associated with:
U.S. Patent and Trademark Office Customer No. 35979
In the event of questions, the attorney in charge can be reached at michaels@bgllp.com
or (713) 221.1112.
I am the Applicant's Assignee or Person to Whom the Inventor is Under an Obligation to
Assign.
SIGNATURE: 4 DATE: 9 25 (2013
SIGNATURE: DATE: 1 / 20 ( AC )
NAME: Stephen Harrison
TITLE: Chief Technology Officer & Vice-President
COMPANY: Simbol Inc.

Electronic Acknowledgement Receipt				
EFS ID:	17378323			
Application Number:	13943395			
International Application Number:				
Confirmation Number:	2954			
Title of Invention:	Selective Recovery of Manganese, Lead and Zinc			
First Named Inventor/Applicant Name:	Stephen Harrison			
Customer Number:	35979			
Filer:	Karthika Perumal/Debra Odum			
Filer Authorized By:	Karthika Perumal			
Attorney Docket Number:	076165.000099			
Receipt Date:	12-NOV-2013			
Filing Date:	16-JUL-2013			
Time Stamp:	15:14:11			
Application Type:	Utility under 35 USC 111(a)			

## **Payment information:**

Submitted with Payment	no
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## File Listing:

Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)
1	Application Data Sheet	20131112ApplicationDataShee t.pdf	136008	no	6
1			6a58e9da73f5188ccd75f1a9d989f83785be 9051		

#### **Warnings:**

Information: ESM / PETITIONER EX 1014\_107

This is not an USPTO supplied ADS fillable form							
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Warnings:							
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This Acknowledgement Receipt evidences receipt on the noted date by the USPTO of the indicated documents, characterized by the applicant, and including page counts, where applicable. It serves as evidence of receipt similar to a Post Card, as described in MPEP 503.

#### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

#### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

#### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

Application Da	oto Choot 27 CED 4 76	Attorney Docket Number	076165.000099			
Application Da	ta Sheet 37 CFR 1.76	Application Number	·			
Title of Invention	Selective Recovery of Manganese, Lead and Zinc					
bibliographic data arrar This document may be	nged in a format specified by the Un	ited States Patent and Trademark C mitted to the Office in electronic fo	being submitted. The following form contains the Office as outlined in 37 CFR 1.76.  Firmat using the Electronic Filing System (EFS) or the			

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-	C.V. Krishnamohan							Sharma			

Residence Information (Select One) 

US Residency

Active US Military Service

Non US Residency

Application Data Sheet 37 CFR 1.76					t Number	076165	.000099						
Application Nul					mber								
Title of	Title of Invention Selective Recovery of Manganese, Lead and Zinc												
City	Milpit	tas			State/	Province	CA	Count	ry of Res	sidencė	us		
Mailing	Addr	ess of In	vento	or:	·						·		
Addres	ss 1			1126 Ridgemo	ont Drive	<del></del>							
Addres	ss 2												
City		Milpitas						State/Pro	vince	CA			
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Invent	or 4	4								R	emove		
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Custo	mer N	umber		35979					·				
Email Address docketing@bgllp			gllp.con	n				Add	Email .	Remove	Email		
Appli	Application Information:												
Title o	f the I	nvention	1	Selective Re	covery	of Manganes	se, Lea	d and Zinc					
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Subjec	ct Mat	ter		Utility									
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Limited Recognition (37 CFR 11.9)

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Application Da	ta Sheet 37 CFR 1.76	Attorney Docket Number	076165.000099						
Application Da	ta Sileet 37 CFK 1,76	Application Number							
Title of Invention	Selective Recovery of Manga	nese, Lead and Zinc							
Publication Information:									
Request Early	Publication (Fee required a	time of Request 37 CFR 1.2	219)						
Request Not to Publish. I hereby request that the attached application not be published under  35 U.S.C. 122(b) and certify that the invention disclosed in the attached application has not and will not be the subject of an application filed in another country, or under a multilateral international agreement, that requires publication at eighteen months after filing.									
Representative Information:									
Representative information should be provided for all practitioners having a power of attorney in the application. Providing									

## **Domestic Benefit/National Stage Information:**

35979

Number will be used for the Representative Information during processing.

Customer Number

This section allows for the applicant to either claim benefit under 35 U.S.C. 119(e), 120, 121, or 365(c) or indicate National Stage entry from a PCT application. Providing this information in the application data sheet constitutes the specific reference required by 35 U.S.C. 119(e) or 120, and 37 CFR 1.78.

this information in the Application Data Sheet does not constitute a power of attorney in the application (see 37 CFR 1.32).

Either enter Customer Number or complete the Representative Name section below. If both sections are completed the customer

O US Patent Practitioner

Prior Application Status	Pending		Remove				
Application Number	Continuity Type	Prior Application Number	Filing Date (YYYY-MM-DD)				
	Continuation of	13/539106	2012-06-29				
Prior Application Status	Expired		Remove				
Application Number	Continuity Type	Prior Application Number	Filing Date (YYYY-MM-DD)				
13/539106	non provisional of	61/502736	2011-06-29				
Additional Domestic Benefit/National Stage Data may be generated within this form							

Additional Domestic Benefit/National Stage Data may be generated within this form by selecting the **Add** button.

## Foreign Priority Information:

Please Select One:

**Customer Number** 

This section allows for the applicant to claim priority to a foreign application. Providing this information in the application data sheet constitutes the claim for priority as required by 35 U.S.C. 119(b) and 37 CFR 1.55(d). When priority is claimed to a foreign application that is eligible for retrieval under the priority document exchange program (PDX)<sup>1</sup> the information will be used by the Office to automatically attempt retrieval pursuant to 37 CFR 1.55(h)(1) and (2). Under the PDX program, applicant bears the ultimate responsibility for ensuring that a copy of the foreign application is received by the Office from the participating foreign intellectual property office, or a certified copy of the foreign priority application is filed, within the time period specified in 37 CFR 1.55(g)(1).

Application Data Short 27 CED 4 76			Attorney Docket Number	076165.000099					
Application Da	Application Data Sheet 37 CFR 1.76			·					
Title of Invention	f Invention Selective Recovery of Manganese, Lead and Zinc								
Remove									
Application Number Country <sup>i</sup>		Filing Date (YYY)	/-MM-DD)	Access Code <sup>i</sup> (if applicable)					
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Additional Foreign Priority Data may be generated within this form by selecting the Add button.									

# Statement under 37 CFR 1.55 or 1.78 for AIA (First Inventor to File) Transition Applications

This application (1) claims priority to or the benefit of an application filed before March 16, 2013 and (2) also contains, or contained at any time, a claim to a claimed invention that has an effective filing date on or after March 16, 2013.
NOTE: By providing this statement under 37 CFR 1.55 or 1.78, this application, with a filing date on or after March 16, 2013, will be examined under the first inventor to file provisions of the AIA.

## **Authorization to Permit Access:**

Authorization to Permit Access to the Instant Application by the Participating Offices
If checked, the undersigned hereby grants the USPTO authority to provide the European Patent Office (EPO), the Japan Patent Office (JPO), the Korean Intellectual Property Office (KIPO), the World Intellectual Property Office (WIPO), and any other intellectual property offices in which a foreign application claiming priority to the instant patent application is filed access to the instant patent application. See 37 CFR 1.14(c) and (h). This box should not be checked if the applicant does not wish the EPO, JPO, KIPO, WIPO, or other intellectual property office in which a foreign application claiming priority to the instant patent application is filed to have access to the instant patent application.
In accordance with 37 CFR 1.14(h)(3), access will be provided to a copy of the instant patent application with respect to: 1) the instant patent application-as-filed; 2) any foreign application to which the instant patent application claims priority under 35 U.S.C. 119(a)-(d) if a copy of the foreign application that satisfies the certified copy requirement of 37 CFR 1.55 has been filed in the instant patent application; and 3) any U.S. application-as-filed from which benefit is sought in the instant patent application.
In accordance with 37 CFR 1.14(c), access may be provided to information concerning the date o f filing this Authorization.

# **Applicant Information:**

Providing assignment information in this section does not substitute for compliance with any requirement of part 3 of Title 37 of CFR to have an assignment recorded by the Office.

Application Data Sheet 37 CFR 1.76		Attorney Docket Number		076165.000099					
Аррисацоп Ба	Application Data Sneet 37 CFR 1.76 Application								
Title of Invention Selective Recovery of Manganese, Lead and Zinc									
Applicant 1									
The information to be   1.43; or the name and who otherwise shows applicant under 37 CF	orovided address sufficient R 1.46 (a gether wi	in this section is the na of the assignee, person proprietary interest in tassignee, person to who	ame and address on to whom the inve the matter who is to om the inventor is	of the legal rep entor is under the applicant u obligated to as	R 1.45), this section should not be completed. resentative who is the applicant under 37 CFR an obligation to assign the invention, or person under 37 CFR 1.46. If the applicant is an ssign, or person who otherwise shows sufficient inventors who are also the applicant should be				
<ul><li>Assignee</li></ul>		◯ Legal Re	epresentative und	er 35 U.S.C. 1	Joint Inventor				
Person to whom th	e invento	or is obligated to assign.	(	Person	who shows sufficient proprietary interest				
If applicant is the leg	al repre	sentative, indicate th	ne authority to file	e the patent a	application, the inventor is:				
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Organization Name	e si	MBOL Inc.	1. 1						
Mailing Address I	nforma	tion For Applicant:		,					
Address 1		6920 Koll Center Par	kway						
Address 2									
City		Pleasanton		State/Provin	nce CA				
Country US				Postal Code	94566				
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Additional Applicant	Data m	ay be generated with	nin this form by s	electing the	Add button.				
Assignee Info	rmat	ion including	Non-Applic	ant Assi	gnee Information:				
Providing assignment have an assignment re			not subsitute for c	ompliance with	h any requirement of part 3 of Title 37 of CFR to				
Assignee 1									
Complete this section if assignee information, including non-applicant assignee information, is desired to be included on the patent application publication. An assignee-applicant identified in the "Applicant Information" section will appear on the patent application publication as an applicant. For an assignee-applicant, complete this section only if identification as an assignee is also desired on the patent application publication.									
If the Assignee is a	ın Orgai	nization check here.							

Application Data Sheet 37 CFR 1.76			Attorney Docket Number		076165.	000099		
			Application Number					
Title of Invention	Selective	ve Recovery of Manganese, Lead and Zinc						
Prefix Giv		en Name	Middle Name		Family Na	ame	Suffix	
Mailing Address I	nformatio	on For Non-Applic	ant Assignee	:				
Address 1								
Address 2								
City			State/Province					
Country i				Postal Cod	de			
Phone Number			Fax Number				·	
Email Address								
Additional Assignee Data may be generated within this form by selecting the Add button.								
Signature:								
NOTE: This form	must be si	gned in accordanc	e with 37 CFR	1.33. See 3	37 CFR 1.4	for signatur	e requirements and	

First Name Karthika Last Name Perumal Regist

Date (YYYY-MM-DD) 2013-11-12

Registration Number 69296

Additional Signature may be generated within this form by selecting the Add button.

This collection of information is required by 37 CFR 1.76. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.14. This collection is estimated to take 23 minutes to complete, including gathering, preparing, and submitting the completed application data sheet form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. **SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.** 

certifications.
Signature

UNITED STATES PATENT AND TRADEMARK OFFICE COMMISSIONER FOR PATENTS P.O.BOX 1450 ALEXANDRIA VA 22313-1451 PRESORTED
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Bracewell & Giuliani LLP P.O. Box 61389 Houston, TX 77208-1389

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# **Courtesy Reminder for Application Serial No: 13/943,395**

Attorney Docket No: 076165.000099

Customer Number: 35979

Date of Electronic Notification: 10/28/2013

This is a courtesy reminder that new correspondence is available for this application. If you have not done so already, please review the correspondence. The official date of notification of the outgoing correspondence will be indicated on the form PTOL-90 accompanying the correspondence.

An email notification regarding the correspondence was sent to the following email address(es) associated with your customer number: docketing@bgllp.com

To view your correspondence online or update your email addresses, please visit us anytime at https://sportal.uspto.gov/secure/myportal/privatepair. If you have any questions, please email the Electronic Business Center (EBC) at EBC@uspto.gov or call 1-866-217-9197.



### United States Patent and Trademark Office

INITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office
Address: COMMISSIONER FOR PATENTS
PO. Box 1450
Alexandria, Virginia 22313-1450
www.uspto.gov

FILING OR 371(C) DATE FIRST NAMED APPLICANT ATTY. DOCKET NO./TITLE APPLICATION NUMBER 13/943,395 07/16/2013

Stephen Harrison

35979 Bracewell & Giuliani LLP P.O. Box 61389 Houston, TX 77208-1389

**CONFIRMATION NO. 2954 IMPROPER CPOA LETTER** 



Date Mailed: 10/28/2013

076165.000099

#### NOTICE REGARDING POWER OF ATTORNEY

This is in response to the power of attorney filed 10/17/2013. The power of attorney in this application is not accepted for the reason(s) listed below:

• The power of attorney has not been accepted because the party who is giving power of attorney has not been identified. Power of attorney may only be signed by the applicant for patent (37 CFR 1.42) or the patent owner. A patent owner who was not the applicant must appoint any power of attorney in compliance with 37 CFR 3.71 and 3.73. See 37 CFR 1.32(b)(4).

/sleutchit/			

Office of Data Management, Application Assistance Unit (571) 272-4000, or (571) 272-4200, or 1-888-786-0101



### UNITED STATES PATENT AND TRADEMARK OFFICE

UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS PC. Box 1450 Alexandria, Vignnia 22313-1450 www.uspto.gov

APPLICATION NUMBER FILING OR 371(C) DATE FIRST NAMED APPLICANT ATTY. DOCKET NO./TITLE

13/943,395 07/16/2013 Stephen Harrison

076165.000099

35979 Bracewell & Giuliani LLP P.O. Box 61389 Houston, TX 77208-1389 CONFIRMATION NO. 2954 IMPROPER CFR REQUEST



Date Mailed: 10/28/2013

#### RESPONSE TO REQUEST FOR CORRECTED FILING RECEIPT

Power of Attorney, Claims, Fees, System Limitations, and Miscellaneous

In response to your request for a corrected Filing Receipt, the Office is unable to comply with your request because:

Any request to correct or update the name of the applicant must include an application data sheet (ADS) in compliance with 37 CFR 1.76 specifying the correct or updated name of the applicant in the applicant information section. Any request to change the applicant after an original applicant has been specified under 37 CFR 1.46(b) must include a new ADS in compliance with 37 CFR 1.76 specifying the applicant in the applicant information section and comply with 37 CFR 3.71 and 3.73. See 37 CFR 1.46(c).

/sleutchit/	
	_
Office of Data Management, Application Assistance Unit (571	) 272-4000, or (571) 272-4200, or 1-888-786-0101

## POWER OF ATTORNEY BY APPLICANT

This Power of Attorney by Applicant is directed to the application titled SELECTIVE
RECOVERY OF MANGANESE, LEAD AND ZINC,
the application of which is attached hereto;
X the US Application No. 13/943,395 filed on July 16, 2013; or
the PCT international application number filed on
I hereby revoke all previous powers of attorney given in the above-identified application.
I hereby appoint the practitioners associated with Bracewell & Giuliani LLP's U.S. Patent
and Trademark Office Customer No. 35979 as my/our attorney(s) or agent(s), and to transact all
business in the U.S. Patent and Trademark Office connected therewith for the application
referenced above.
Please direct all correspondence to the address associated with:
U.S. Patent and Trademark Office Customer No. 35979
In the event of questions, the attorney in charge can be reached at michaels@bgllp.com
or (713) 221.1112.
I am the Applicant's Assignee or Person to Whom the Inventor is Under an Obligation to
Assign.  SIGNATURE: DATE: 9 25 2013
NAME: Stephen Harrison
TITLE: Chief Technology Officer & Vice-President
COMPANY: Simbol Inc.

Electronic Ac	knowledgement Receipt
EFS ID:	17150323
Application Number:	13943395
International Application Number:	
Confirmation Number:	2954
Title of Invention:	Selective Recovery of Manganese, Lead and Zinc
First Named Inventor/Applicant Name:	Stephen Harrison
Customer Number:	35979
Filer:	Karthika Perumal/Debra Odum
Filer Authorized By:	Karthika Perumal
Attorney Docket Number:	076165.000099
Receipt Date:	17-OCT-2013
Filing Date:	16-JUL-2013
Time Stamp:	10:51:57
Application Type:	Utility under 35 USC 111(a)
Application Type:  Payment information:	Utility under 35 USC 111(a)

# **Payment information:**

Submitted with Payment	no
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# File Listing:

Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)
1	Power of Attorney	20131017Power of Attorney.pdf	<b>29111</b> 3fc6deeb2655c77aa04c0c8fbc7f7fa1e6d3c	no	1
			ced		

#### **Warnings:**

Information:	ESM / PETITIONER EX 1014 119

This Acknowledgement Receipt evidences receipt on the noted date by the USPTO of the indicated documents, characterized by the applicant, and including page counts, where applicable. It serves as evidence of receipt similar to a Post Card, as described in MPEP 503.

#### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

#### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

#### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

#### DECLARATION FOR PATENT APPLICATION

As a below named inventor, I hereby declare that:

This Declaration is directed to the application titled SELECTIVE RECOVERY OF MANGANESE, LEAD AND ZINC,

	the application of which is attached hereto;	
X	the US Application No. 13/943,395 filed o	n July 16, 2013; or
	the PCT international application number	filed on

My residence, post office address and citizenship are as stated below my name.

The above-identified application was made or authorized by me.

I believe that we are the original, first and sole inventors of the subject matter which is claimed and for which a patent is sought on the invention entitled:

I acknowledge the duty to review and understand the contents of the above-identified application, including the claims.

I acknowledge the duty to disclose information that is material to the examination of this application in accordance with Title 37, Code of Federal Regulations, Ssec. 1.56(a).

I hereby claim the benefit under Title 35, United States Code, §119(e) of any United States provisional application(s) listed below:

<u>NUMBER</u>	<u>DATE FILED</u>
61/502,736	June 29, 2011

I hereby state that I have reviewed and understand the contents of the above-identified specification, including the claims, as amended by any amendment referred to above.

I acknowledge the duty to disclose information which is material to patentability as defined in 37 CFR §1.56, including for continuation-in-part applications, material information which became available between the filing date of the prior application and the national or PCT international filing date of the continuation-in-part application.

I hereby claim foreign priority benefits under Title 35, United States Code, Sec. 119 of any foreign application(s) for patent or inventor's certificate listed below and have also identified below any foreign application for patent or inventor's certificate having a filing date before that of the application on which priority is claimed.

# PRIOR FOREIGN APPLICATION(S)

NUMBER	COUNTRY	DATE FILED	PRIORITY C	LAIMED
		Name of the Association of the State of the	YES	_NO
States application application is not first paragraph o	claim the benefit under n listed below and, inso t disclosed in any prior Uf Title 37, Code of Federathe prior application as	far as the subject United States applieral Regulations, S	matter of each of the cation in the manner Sec. 1.56(a), which of	e claims of this provided by the courred between
	nt Application No: 13/5 nt Application No:		Filing Date: June 29, Filing Date:	2012
	acknowledge that any Title 18 of United State ) years, or both.			
Please dir	ect all correspondence to	o the address assoc	iated with:	
.U.	S. Patent and Trademarl	c Office <b>Custome</b> i	Number 35979.	
In the eve or (713) 221-111	ent of questions, the atto 2.	orney in charge ca	n be reached at mich	aels@bgllp.com
INVENTOR'S N	AME: Stephen Harris	on		
SIGNATURE:	19 hours	2		
DATE:	9/25/201	5		
RESIDENCE: 1	34 Panorama Drive, Ben	icia, CA 94510		

POST OFFICE ADDRESS: SAME

INVENTOR'S NAME: Samaresh Mohanta

SIGNATURE:

DATE: 09/25/2013

RESIDENCE: 2410, Brandini Drive, Dublin, CA, 94568

POST OFFICE ADDRESS: SAME

INVENTOR'S NAME: C.V. Krishnamohan Sharma

SIGNATURE:

TE: Sept. 25 2018

RESIDENCE: 1126 Ridgemont Dr., Milpitas, CA 95035

POST OFFICE ADDRESS: Same

#4349215.1

#### DOCKET NO. 076165.000099

INVENTOR'S NAME: Placeth Geler

SIGNATURE:

DATE: 10/17/13

RESIDENCE: 8 Seville Dr., San Rafael, CA 94903

POST OFFICE ADDRESS: Same

Electronic Ack	knowledgement Receipt
EFS ID:	17158548
Application Number:	13943395
International Application Number:	
Confirmation Number:	2954
Title of Invention:	Selective Recovery of Manganese, Lead and Zinc
First Named Inventor/Applicant Name:	Stephen Harrison
Customer Number:	35979
Filer:	Karthika Perumal/Debra Odum
Filer Authorized By:	Karthika Perumal
Attorney Docket Number:	076165.000099
Receipt Date:	17-OCT-2013
Filing Date:	16-JUL-2013
Time Stamp:	17:16:02
Application Type:	Utility under 35 USC 111(a)

# **Payment information:**

Submitted with Payment	no
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# File Listing:

Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)
1	Oath or Declaration filed	2013 1017 Declaration.pdf	61435	no	5
•	Sam S. Designation med	20.0.000	29416811af4fd330be20125e331dae93d07 a9def	***	

## **Warnings:**

Information: ESM / PETITIONER EX 1014\_126

This Acknowledgement Receipt evidences receipt on the noted date by the USPTO of the indicated documents, characterized by the applicant, and including page counts, where applicable. It serves as evidence of receipt similar to a Post Card, as described in MPEP 503.

#### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

#### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

#### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.



### United States Patent and Trademark Office

FILING OR 371(C) DATE

07/16/2013

UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS PC. Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov

www.uspto.gov

PPLICANT ATTY. DOCKET NO./TITLE

FIRST NAMED APPLICANT

NOTICE

Stephen Harrison

076165.000099

**CONFIRMATION NO. 2954** 

35979 Bracewell & Giuliani LLP P.O. Box 61389 Houston, TX 77208-1389

APPLICATION NUMBER 13/943,395

Date Mailed: 09/13/2013

#### INFORMATIONAL NOTICE TO APPLICANT

Applicant is notified that the above-identified application contains the deficiencies noted below. No period for reply is set forth in this notice for correction of these deficiencies. However, if a deficiency relates to the inventor's oath or declaration, the applicant must file an oath or declaration in compliance with 37 CFR 1.63, or a substitute statement in compliance with 37 CFR 1.64, executed by or with respect to each actual inventor no later than the expiration of the time period set in the "Notice of Allowability" to avoid abandonment. See 37 CFR 1.53(f).

The item(s) indicated below are also required and should be submitted with any reply to this notice to avoid further processing delays.

A properly executed inventor's oath or declaration has not been received for the following inventor(s):

Stephen Harrison

Samaresh Mohanta

C.V. Krishnamohan Sharma

Elizabeth Geler

Applicant may submit the inventor's oath or declaration at any time before the Notice of Allowance and Fee(s) Due, PTOL-85, is mailed.

	PATEN	IT APPLI		ON FEE DE titute for Form	TERMINAT PTO-875	TON REC	ORD		Applica 13/94	tion or Docket Num 3,395	nber
	APPLIC	CATION AS			olumn 2)	SMA	ALL E	NTITY	OR	OTHEF SMALL	R THAN ENTITY
	FOR	NUMBE			ER EXTRA	RATE(\$	)	FEE(\$)	1	RATE(\$)	FEE(\$)
	IC FEE FR 1.16(a), (b), or (c))	N.	/A		N/A	N/A	十	70	1	N/A	
SEA	RCH FEE FR 1.16(k), (i), or (m))	N.	/A		N/A	N/A	$\top$	300	1	N/A	
XΑ	MINATION FEE FR 1.16(o), (p), or (q))	N,	/A		N/A	N/A	$\dashv$	360	1	N/A	
ОТ	AL CLAIMS FR 1.16(i))	16	minus	20= *		× 40	=	0.00	OR		
NDE	EPENDENT CLAIMS FR 1.16(h))	3	minus	3 = *		× 210	=	0.00	1		
PF	PLICATION SIZE	sheets of p \$310 (\$155 50 sheets of	aper, the for smale for fraction	and drawings of application sall entity) for each thereof. See CFR 1.16(s).	size fee due is ach additional			0.00			
/IUL	TIPLE DEPENDENT	CLAIM PRES	SENT (3	7 CFR 1.16(j))				0.00	1		
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		(Column 1)		(Column 2)	(Column 3)	SMA	ALL E	NTITY	OR <b>1</b>	OTHEF SMALL	R THAN ENTITY
	F	(Column 1)  CLAIMS  REMAINING  AFTER  MENDMENT	Minus	(Column 2) HIGHEST NUMBER PREVIOUSLY PAID FOR	PRESENT EXTRA	RATE(\$)		NTITY  ADDITIONAL FEE(\$)		SMALL RATE(\$)	ENTITY
	Total * (37 CFR 1.16(i))	CLAIMS REMAINING AFTER	Minus	HIGHEST NUMBER PREVIOUSLY PAID FOR	PRESENT EXTRA			ADDITIONAL	OR	SMALL  RATE(\$)  x =	ENTITY ADDITION
	Total (37 CFR 1.16(i))  Independent (37 CFR 1.16(h))	CLAIMS REMAINING AFTER MENDMENT	Minus Minus	HIGHEST NUMBER PREVIOUSLY PAID FOR	PRESENT EXTRA	RATE(\$)		ADDITIONAL		SMALL RATE(\$)	ENTITY ADDITION.
	Total (37 CFR 1.16(h))  Independent (37 CFR 1.16(h))  Application Size Fee (3	CLAIMS REMAINING AFTER MENDMENT	Minus	HIGHEST NUMBER PREVIOUSLY PAID FOR	PRESENT EXTRA	RATE(\$)	=	ADDITIONAL	OR OR	SMALL  RATE(\$)  x =	ENTITY ADDITION.
	Total (37 CFR 1.16(i))  Independent (37 CFR 1.16(h))	CLAIMS REMAINING AFTER MENDMENT	Minus	HIGHEST NUMBER PREVIOUSLY PAID FOR	PRESENT EXTRA	RATE(\$)	= =	ADDITIONAL	OR	SMALL  RATE(\$)  x =  x =	ENTITY ADDITIONA
	Total (37 CFR 1.16(h))  Independent (37 CFR 1.16(h))  Application Size Fee (3	CLAIMS REMAINING AFTER MENDMENT	Minus	HIGHEST NUMBER PREVIOUSLY PAID FOR	PRESENT EXTRA	RATE(\$)	= =	ADDITIONAL	OR OR	SMALL  RATE(\$)  x =	ENTITY ADDITIONA
AIVICINDIVICIAL A	Total (37 CFR 1.16(i))  Independent (37 CFR 1.16(h))  Application Size Fee (3	CLAIMS REMAINING AFTER MENDMENT  OF CFR 1.16(s)) N OF MULTIPL  (Column 1)	Minus	HIGHEST NUMBER PREVIOUSLY PAID FOR ***  DENT CLAIM (37	PRESENT EXTRA	RATE(\$)  x  x  TOTAL	= =	ADDITIONAL	OR OR OR	SMALL  RATE(\$)  x =  x =	ENTITY ADDITIONA
	Total (37 CFR 1.16(i))  Independent (37 CFR 1.16(h))  Application Size Fee (3  FIRST PRESENTATIO	CLAIMS REMAINING AFTER MENDMENT  7 CFR 1.16(s)) N OF MULTIPL	Minus	HIGHEST NUMBER PREVIOUSLY PAID FOR	PRESENT EXTRA	RATE(\$)  x  x  TOTAL	= = =	ADDITIONAL	OR OR OR	SMALL  RATE(\$)  x =  x =	ENTITY  ADDITION, FEE(\$)
	Total (37 CFR 1.16(i))  Independent (37 CFR 1.16(h))  Application Size Fee (3  FIRST PRESENTATIO	CLAIMS REMAINING AFTER MENDMENT  FOR THE CONTROL OF	Minus	HIGHEST NUMBER PREVIOUSLY PAID FOR	PRESENT EXTRA  = = CFR 1.16(j))  (Column 3)  PRESENT	x x TOTAL ADD'L FE	= = =	ADDITIONAL FEE(\$)	OR OR OR	SMALL  RATE(\$)  X =  X =  TOTAL  ADD'L FEE	ENTITY  ADDITION, FEE(\$)
	Total (37 CFR 1.16(i)) Independent (37 CFR 1.16(h)) Application Size Fee (3 FIRST PRESENTATIO	CLAIMS REMAINING AFTER MENDMENT  FOR THE CONTROL OF	Minus E DEPEN	HIGHEST NUMBER PREVIOUSLY PAID FOR ***  ***  COolumn 2)  HIGHEST NUMBER PREVIOUSLY PAID FOR	PRESENT EXTRA  =  CFR 1.16(j))  (Column 3)  PRESENT EXTRA	RATE(\$)  X  X  TOTAL ADD'L FE	= = =	ADDITIONAL FEE(\$)	OR OR OR	SMALL  RATE(\$)  x =  x =  TOTAL ADD'L FEE  RATE(\$)	ENTITY  ADDITION, FEE(\$)
	Total (37 CFR 1.16(ii))	CLAIMS REMAINING AFTER MENDMENT  FOR 1.16(s)  N OF MULTIPL  (Column 1)  CLAIMS REMAINING AFTER MENDMENT	Minus  E DEPEN  Minus	HIGHEST NUMBER PREVIOUSLY PAID FOR  ***  COolumn 2)  HIGHEST NUMBER PREVIOUSLY PAID FOR  **  ***  ***  ***  ***  ***  ***  *	PRESENT EXTRA  =  =  CFR 1.16(j))  (Column 3)  PRESENT EXTRA	RATE(\$)  X  X  TOTAL ADD'L FE  RATE(\$)	= = = :E	ADDITIONAL FEE(\$)	OR OR OR OR	SMALL  RATE(\$)	ADDITION/ FEE(\$)
	Total (37 CFR 1.16(i))  Independent (37 OFR 1.16(h))  Application Size Fee (3  FIRST PRESENTATIO	CLAIMS REMAINING AFTER MENDMENT  67 CFR 1.16(s)) N OF MULTIPL  (Column 1) CLAIMS REMAINING AFTER MENDMENT	Minus  E DEPEN  Minus  Minus	HIGHEST NUMBER PREVIOUSLY PAID FOR  (Column 2) HIGHEST NUMBER PREVIOUSLY PAID FOR  **	PRESENT EXTRA  =  =  CFR 1.16(j))  (Column 3)  PRESENT EXTRA  =  =	RATE(\$)  X  X  TOTAL ADD'L FE  RATE(\$)	= = = :E	ADDITIONAL FEE(\$)	OR OR OR	SMALL  RATE(\$)	ADDITION/ FEE(\$)

<sup>\*\*</sup> If the "Highest Number Previously Paid For" IN THIS SPACE is less than 3, enter "3".

The "Highest Number Previously Paid For" (Total or Independent) is the highest found in the appropriate box in column 1.



#### United States Patent and Trademark Office

UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS P.O. Box 1450 Alexandria, Virginia 22313-1450 WWW.18910.gov

APPLICATION	FILING or	GRP ART				
NUMBER	371(c) DATE	UNIT	FIL FEE REC'D	ATTY.DOCKET.NO	TOT CLAIMS	IND CLAIMS
13/943 395	07/16/2013	1778	1660	076165 000099	16	3

35979 Bracewell & Giuliani LLP P.O. Box 61389 Houston, TX 77208-1389 CONFIRMATION NO. 2954
UPDATED FILING RECEIPT



Date Mailed: 09/13/2013

Receipt is acknowledged of this non-provisional patent application. The application will be taken up for examination in due course. Applicant will be notified as to the results of the examination. Any correspondence concerning the application must include the following identification information: the U.S. APPLICATION NUMBER, FILING DATE, NAME OF APPLICANT, and TITLE OF INVENTION. Fees transmitted by check or draft are subject to collection. Please verify the accuracy of the data presented on this receipt. If an error is noted on this Filing Receipt, please submit a written request for a Filing Receipt Correction. Please provide a copy of this Filing Receipt with the changes noted thereon. If you received a "Notice to File Missing Parts" for this application, please submit any corrections to this Filing Receipt with your reply to the Notice. When the USPTO processes the reply to the Notice, the USPTO will generate another Filing Receipt incorporating the requested corrections

#### Inventor(s)

Stephen Harrison, Benicia, CA; Samaresh Mohanta, Pleasanton, CA; C.V. Krishnamohan Sharma, Milpitas, CA; Elizabeth Geler, San Rafael, CA;

#### Applicant(s)

Stephen Harrison, Benicia, CA; Samaresh Mohanta, Pleasanton, CA; C.V. Krishnamohan Sharma, Milpitas, CA; Elizabeth Geler, San Rafael, CA;

#### **Assignment For Published Patent Application**

SIMBOL Inc., Pleasanton, CA

Power of Attorney: None

#### Domestic Priority data as claimed by applicant

This application is a CON of 13/539,106 06/29/2012 PAT 8518232 which claims benefit of 61/502,736 06/29/2011

**Foreign Applications** for which priority is claimed (You may be eligible to benefit from the **Patent Prosecution Highway** program at the USPTO. Please see <a href="http://www.uspto.gov">http://www.uspto.gov</a> for more information.) - None. Foreign application information must be provided in an Application Data Sheet in order to constitute a claim to foreign priority. See 37 CFR 1.55 and 1.76.

If Required, Foreign Filing License Granted: 07/31/2013

page 1 of 3

The country code and number of your priority application, to be used for filing abroad under the Paris Convention, is **US 13/943,395** 

Projected Publication Date: Request for Non-Publication Acknowledged

Non-Publication Request: Yes

Early Publication Request: No

\*\* SMALL ENTITY \*\*

Title

Selective Recovery of Manganese, Lead and Zinc

**Preliminary Class** 

210

Statement under 37 CFR 1.55 or 1.78 for AIA (First Inventor to File) Transition Applications: No

#### PROTECTING YOUR INVENTION OUTSIDE THE UNITED STATES

Since the rights granted by a U.S. patent extend only throughout the territory of the United States and have no effect in a foreign country, an inventor who wishes patent protection in another country must apply for a patent in a specific country or in regional patent offices. Applicants may wish to consider the filing of an international application under the Patent Cooperation Treaty (PCT). An international (PCT) application generally has the same effect as a regular national patent application in each PCT-member country. The PCT process **simplifies** the filing of patent applications on the same invention in member countries, but **does not result** in a grant of "an international patent" and does not eliminate the need of applicants to file additional documents and fees in countries where patent protection is desired.

Almost every country has its own patent law, and a person desiring a patent in a particular country must make an application for patent in that country in accordance with its particular laws. Since the laws of many countries differ in various respects from the patent law of the United States, applicants are advised to seek guidance from specific foreign countries to ensure that patent rights are not lost prematurely.

Applicants also are advised that in the case of inventions made in the United States, the Director of the USPTO must issue a license before applicants can apply for a patent in a foreign country. The filing of a U.S. patent application serves as a request for a foreign filing license. The application's filing receipt contains further information and guidance as to the status of applicant's license for foreign filing.

Applicants may wish to consult the USPTO booklet, "General Information Concerning Patents" (specifically, the section entitled "Treaties and Foreign Patents") for more information on timeframes and deadlines for filing foreign patent applications. The guide is available either by contacting the USPTO Contact Center at 800-786-9199, or it can be viewed on the USPTO website at http://www.uspto.gov/web/offices/pac/doc/general/index.html.

For information on preventing theft of your intellectual property (patents, trademarks and copyrights), you may wish to consult the U.S. Government website, http://www.stopfakes.gov. Part of a Department of Commerce initiative, this website includes self-help "toolkits" giving innovators guidance on how to protect intellectual property in specific countries such as China, Korea and Mexico. For questions regarding patent enforcement issues, applicants may call the U.S. Government hotline at 1-866-999-HALT (1-866-999-4258).

#### LICENSE FOR FOREIGN FILING UNDER

#### Title 35, United States Code, Section 184

#### Title 37, Code of Federal Regulations, 5.11 & 5.15

#### **GRANTED**

The applicant has been granted a license under 35 U.S.C. 184, if the phrase "IF REQUIRED, FOREIGN FILING LICENSE GRANTED" followed by a date appears on this form. Such licenses are issued in all applications where the conditions for issuance of a license have been met, regardless of whether or not a license may be required as set forth in 37 CFR 5.15. The scope and limitations of this license are set forth in 37 CFR 5.15(a) unless an earlier license has been issued under 37 CFR 5.15(b). The license is subject to revocation upon written notification. The date indicated is the effective date of the license, unless an earlier license of similar scope has been granted under 37 CFR 5.13 or 5.14.

This license is to be retained by the licensee and may be used at any time on or after the effective date thereof unless it is revoked. This license is automatically transferred to any related applications(s) filed under 37 CFR 1.53(d). This license is not retroactive.

The grant of a license does not in any way lessen the responsibility of a licensee for the security of the subject matter as imposed by any Government contract or the provisions of existing laws relating to espionage and the national security or the export of technical data. Licensees should apprise themselves of current regulations especially with respect to certain countries, of other agencies, particularly the Office of Defense Trade Controls, Department of State (with respect to Arms, Munitions and Implements of War (22 CFR 121-128)); the Bureau of Industry and Security, Department of Commerce (15 CFR parts 730-774); the Office of Foreign AssetsControl, Department of Treasury (31 CFR Parts 500+) and the Department of Energy.

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No license under 35 U.S.C. 184 has been granted at this time, if the phrase "IF REQUIRED, FOREIGN FILING LICENSE GRANTED" DOES NOT appear on this form. Applicant may still petition for a license under 37 CFR 5.12, if a license is desired before the expiration of 6 months from the filing date of the application. If 6 months has lapsed from the filing date of this application and the licensee has not received any indication of a secrecy order under 35 U.S.C. 181, the licensee may foreign file the application pursuant to 37 CFR 5.15(b).

#### SelectUSA

The United States represents the largest, most dynamic marketplace in the world and is an unparalleled location for business investment, innovation, and commercialization of new technologies. The U.S. offers tremendous resources and advantages for those who invest and manufacture goods here. Through SelectUSA, our nation works to promote and facilitate business investment. SelectUSA provides information assistance to the international investor community; serves as an ombudsman for existing and potential investors; advocates on behalf of U.S. cities, states, and regions competing for global investment; and counsels U.S. economic development organizations on investment attraction best practices. To learn more about why the United States is the best country in the world to develop technology, manufacture products, deliver services, and grow your business, visit <a href="http://www.SelectUSA.gov">http://www.SelectUSA.gov</a> or call +1-202-482-6800.

# Amendments to the Drawings

Please replace Figure 14 in the application with the Replacement Figure 14 that is being submitted along with this Response.

<u>Remarks</u>

Applicant has provided Replacement Figure 14 in response to the Notice to File

Corrected Papers. Replacement Figure 14 is a clearer version of the previously submitted Figure

14. Additionally, several elements have been renumbered to correctly identify items in

accordance with the specification as filed.

Should the Commissioner deem any fees as being due, including fees for extensions of

time, the Commissioner is authorized to debit said fees from, or credit overpayments to,

Bracewell & Giuliani LLP, USPTO Deposit Account Number 50-0259, Reference No.

076165.000099.

The Examiner is requested to contact the undersigned via telephone if a telephone

conference would expedite or aid the prosecution and examination of this application.

Respectfully submitted,

Date: September 4, 2013

/Karthika Perumal/

Karthika Perumal

Reg. No. 69,296

BRACEWELL & GIULIANI LLP

P.O. Box 61389

Houston, Texas 77208-1389

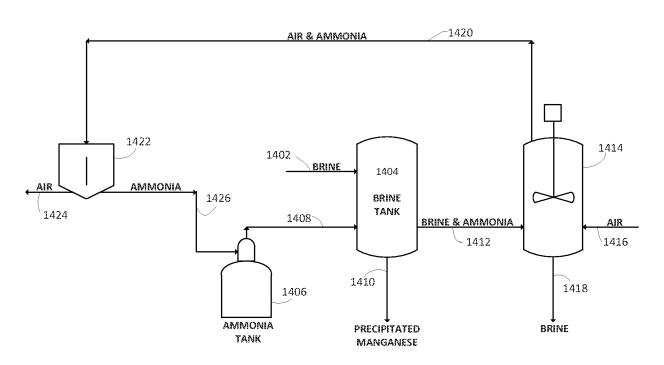
Telephone: (713) 221-1342

Facsimile: (800) 404-3970

-3-

# Replacement Figure 14

#### <u>1400</u>



Electronic Acl	Electronic Acknowledgement Receipt						
EFS ID:	16755140						
Application Number:	13943395						
International Application Number:							
Confirmation Number:	2954						
Title of Invention:	Selective Recovery of Manganese, Lead and Zinc						
First Named Inventor/Applicant Name:	Stephen Harrison						
Customer Number:	35979						
Filer:	Karthika Perumal/Debra Odum						
Filer Authorized By:	Karthika Perumal						
Attorney Docket Number:	076165.000099						
Receipt Date:	04-SEP-2013						
Filing Date:	16-JUL-2013						
Time Stamp:	08:12:20						
Application Type:	Utility under 35 USC 111(a)						

# **Payment information:**

Submitted with Payment	no

# File Listing:

Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)
1		20130904Responsetofilecorrect	41850	yes	4
		ed application papers.pdf	8a6fe305eacd72c4a66a5f557487c2d4c372 f88e	,	

	Multipart Description/PDF files in .zip description							
	Document Description	Start	End					
	Miscellaneous Incoming Letter	1	1					
	Applicant Response to Pre-Exam Formalities Notice	2	2					
	Applicant Arguments/Remarks Made in an Amendment	3	3					
	Drawings-only black and white line drawings	4	4					
Varnings:		1	1					

#### ٧

#### Information:

Total Files Size (in bytes):	41850

This Acknowledgement Receipt evidences receipt on the noted date by the USPTO of the indicated documents, characterized by the applicant, and including page counts, where applicable. It serves as evidence of receipt similar to a Post Card, as described in MPEP 503.

#### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

#### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

#### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

#### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Patent Application of:

Stephen HARRISON, et al.

Examiner: Not Assigned

Serial No.: 13/943,395

Art Unit: Not Assigned

Filed: July 16, 2013

Confirmation No.: 2954

For: SELECTIVE RECOVERY OF

MANGANESE, LEAD AND ZINC

Attorney Docket No. 076165.000099

## RESPONSE TO NOTICE TO FILE CORRECTED PAPERS

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Dear Sir,

Applicant is responding to the Notice to file Corrected Papers dated August 22, 2013, by submitting a replacement Figure 14.

Amendment to the drawings is on page 2 of this Response.

Remarks begin on page 3 of this Response.

Doc code: IDS Doc description: Information Disclosure Statement (IDS) Filed PTO/SB/08a (01-10)
Approved for use through 07/31/2012. OMB 0651-0031
U.S. Patent and Trademark Office; U.S. DEPARTMENT OF COMMERCE

Under the Paperwork Reduction Act of 1995, no persons are required to respond to a collection of information unless it contains a valid OMB control number.

	Application Number		13943395
	Filing Date		2013-07-16
INFORMATION DISCLOSURE	First Named Inventor	Harris	on, Stephen, et al.
STATEMENT BY APPLICANT (Not for submission under 37 CFR 1.99)	Art Unit		
( too for our house and or or of the five)	Examiner Name	Unkno	own
	Attorney Docket Number		076165.000099

***************************************	U.S.PATENTS									
Examiner Initial*	Cite No	Patent Number	Kind Code <sup>1</sup>	Issue Date	Name of Patentee or Applicant of cited Document	Pages,Columns,Lines where Relevant Passages or Relevant Figures Appear				
	1	5229003	B1	1993-07-20	Duyvesteyn, Willem					
	2	5244491	B1	1993-09-14	Brown, et al.					
	3	5246684	B1	1993-09-21	Brown, et al.					
	4	5358700	B1	1994-10-25	Brown, et al.					
	5	5656172	B1	1997-08-12	Kitz, et al.					
If you wisl	n to add	additional U.S. Paten	t citatio	n information pl	ease click the Add button.	<u> </u>				
			U.S.P	ATENT APPLI	CATION PUBLICATIONS					
Examiner Initial*	Cite No	Publication Number	Kind Code <sup>1</sup>	Publication Date	Name of Patentee or Applicant of cited Document	Pages,Columns,Lines where Relevant Passages or Relevant Figures Appear				
	1									
If you wisl	n to add	additional U.S. Publis	hed Ap	plication citation	n information please click the Add	d button.				
	FOREIGN PATENT DOCUMENTS									

# INFORMATION DISCLOSURE STATEMENT BY APPLICANT

( Not for submission under 37 CFR 1.99)

Application Number		13943395
Filing Date		2013-07-16
First Named Inventor Harris		con, Stephen, et al.
Art Unit	danaanaanaan	
Examiner Name Unknown		n
Attorney Docket Number		076165.000099

Examiner Initial*	Cite No	Foreign Document Number <sup>3</sup>	Country Code <sup>2</sup> i	Kind Code <sup>4</sup>	Publication Date	Name of Patentee or Applicant of cited Document	Pages,Columns,Lines where Relevant Passages or Relevant Figures Appear	775
	1							
If you wisl	h to ac	dd additional Foreign P	atent Document	citation	information p	lease click the Add butto	n	
			NON-PATE	NT LITE	RATURE DO	CUMENTS		
Examiner Initials*	Cite No		nal, serial, symp	oosium,	catalog, etc),	the article (when approp date, pages(s), volume-i		<b>T</b> 5
	1	BLOOMQUIST, R. Gord Materials Society, 2006,			Mineral Extraction	on from Geothermal Brines	, The Minerals, Metals &	
	2	GALLUP, D. L., et al., Heavy Metal Sulfide Scaling in a Production Well at the Salton Sea Geothermal Field, Geothermal Resources Council Transactions, 1990, pp. 1583-1590, Vol. 14, Part II, Davis, CA, US						
	3	GALLUP, D. L., et al., Laboratory investigation of silica removal from geothermal brines to control silica scaling and produce usable silicates, Journal of the International Association of Geochemistry and Cosmochemistry, 2003, pp. 1597-1612, Vol. 18, Elsevier, UK						
	4	MANCEAU, Alain, et al., Nanometer0sized, divalent-Mn, hydrous silicate domains in geothermal brine precipitates, American Mineralogist, 2005, pp. 371-831, Vol. 90, US						
	5	PASCUA, C. S., et al., Uptake of dissolved arsenic during the retrieval of silica from spent geothermal brine, Geothermics, 2007, pp. 230-242, Vol. 36, Elsevier, UK						
	6	POTAPOV, V. V., Silica Precipitation From Hydrothermal Solution, Journal of Mining Science, 2004, pp. 101-112, Vol. 40, No. 1, Plenum Publishing Corporation						
	7	SCHULTZE, L. E., et al., Operation of a Mineral Recovery Unit on Brine from the Salton Sea Known Geothermal Resource Area, Report of Investigations 8680, Bureau of Mines, 1982, pp. 1-12, Vol. 505, US						

### Application Number 13943395 Filing Date 2013-07-16 INFORMATION DISCLOSURE First Named Inventor Harrison, Stephen, et al. STATEMENT BY APPLICANT Art Unit ( Not for submission under 37 CFR 1.99) Unknown **Examiner Name** Attorney Docket Number 076165.000099 SCHULTZE, L.E., Techniques for Recovering Metal Values From Postflash Geothermal Brines, Transactions, 1984, 8 pp. 159-162, Vol. 8, Davis, CA, US If you wish to add additional non-patent literature document citation information please click the Add button **EXAMINER SIGNATURE Examiner Signature Date Considered** \*EXAMINER: Initial if reference considered, whether or not citation is in conformance with MPEP 609. Draw line through a citation if not in conformance and not considered. Include copy of this form with next communication to applicant. 1 See Kind Codes of USPTO Patent Documents at www.USPTO.GOV or MPEP 901.04. 2 Enter office that issued the document, by the two-letter code (WIPO Standard ST.3). 3 For Japanese patent documents, the indication of the year of the reign of the Emperor must precede the serial number of the patent document.

4 Kind of document by the appropriate symbols as indicated on the document under WIPO Standard ST.16 if possible. 5 Applicant is to place a check mark here if

English language translation is attached.

# INFORMATION DISCLOSURE STATEMENT BY APPLICANT

(Not for submission under 37 CFR 1.99)

Application Number		13943395
Filing Date		2013-07-16
First Named Inventor Harris		son, Stephen, et al.
Art Unit		
Examiner Name Unkn		own
Attorney Docket Number		076165.000099

		CERTIFICATION	STATEMENT					
Plea	ase see 37 CFR 1	.97 and 1.98 to make the appropriate selection	on(s):					
	That each item of information contained in the information disclosure statement was first cited in any communication from a foreign patent office in a counterpart foreign application not more than three months prior to the filing of the information disclosure statement. See 37 CFR 1.97(e)(1).							
OR								
	That no item of information contained in the information disclosure statement was cited in a communication from a foreign patent office in a counterpart foreign application, and, to the knowledge of the person signing the certification after making reasonable inquiry, no item of information contained in the information disclosure statement was known to any individual designated in 37 CFR 1.56(c) more than three months prior to the filing of the information disclosure statement. See 37 CFR 1.97(e)(2).							
	See attached ce	rtification statement.						
	The fee set forth	in 37 CFR 1.17 (p) has been submitted here	with.					
$\boxtimes$	A certification sta	atement is not submitted herewith.						
	ignature of the ap 1 of the signature.	SIGNAT plicant or representative is required in accord		8. Please see CFR 1.4(d) for the				
Sign	nature	/Karthika Perumal/	Date (YYYY-MM-DD)	2013-09-04				
Name/Print Karthika Perumal Registration Number			Registration Number	69296				
pub 1.14 app requ	lic which is to file ( 1. This collection in lication form to the uire to complete the	rmation is required by 37 CFR 1.97 and 1.98. (and by the USPTO to process) an applicatio is estimated to take 1 hour to complete, include USPTO. Time will vary depending upon the lis form and/or suggestions for reducing this to the list of Commerce, P.O.	<ul> <li>n. Confidentiality is gover ding gathering, preparing a individual case. Any cor purden, should be sent to</li> </ul>	rned by 35 U.S.C. 122 and 37 CFR and submitting the completed mments on the amount of time you the Chief Information Officer, U.S.				

FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria,

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The Privacy Act of 1974 (P.L. 93-579) requires that you be given certain information in connection with your submission of the attached form related to a patent application or patent. Accordingly, pursuant to the requirements of the Act, please be advised that: (1) the general authority for the collection of this information is 35 U.S.C. 2(b)(2); (2) furnishing of the information solicited is voluntary; and (3) the principal purpose for which the information is used by the U.S. Patent and Trademark Office is to process and/or examine your submission related to a patent application or patent. If you do not furnish the requested information, the U.S. Patent and Trademark Office may not be able to process and/or examine your submission, which may result in termination of proceedings or abandonment of the application or expiration of the patent.

The information provided by you in this form will be subject to the following routine uses:

- 1. The information on this form will be treated confidentially to the extent allowed under the Freedom of Information Act (5 U.S.C. 552) and the Privacy Act (5 U.S.C. 552a). Records from this system of records may be disclosed to the Department of Justice to determine whether the Freedom of Information Act requires disclosure of these record s.
- A record from this system of records may be disclosed, as a routine use, in the course of presenting evidence to a court, magistrate, or administrative tribunal, including disclosures to opposing counsel in the course of settlement negotiations.
- 3. A record in this system of records may be disclosed, as a routine use, to a Member of Congress submitting a request involving an individual, to whom the record pertains, when the individual has requested assistance from the Member with respect to the subject matter of the record.
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- 5. A record related to an International Application filed under the Patent Cooperation Treaty in this system of records may be disclosed, as a routine use, to the International Bureau of the World Intellectual Property Organization, pursuant to the Patent Cooperation Treaty.
- 6. A record in this system of records may be disclosed, as a routine use, to another federal agency for purposes of National Security review (35 U.S.C. 181) and for review pursuant to the Atomic Energy Act (42 U.S.C. 218(c)).
- 7. A record from this system of records may be disclosed, as a routine use, to the Administrator, General Services, or his/her designee, during an inspection of records conducted by GSA as part of that agency's responsibility to recommend improvements in records management practices and programs, under authority of 44 U.S.C. 2904 and 2906. Such disclosure shall be made in accordance with the GSA regulations governing inspection of records for this purpose, and any other relevant (i.e., GSA or Commerce) directive. Such disclosure shall not be used to make determinations about individuals.
- 8. A record from this system of records may be disclosed, as a routine use, to the public after either publication of the application pursuant to 35 U.S.C. 122(b) or issuance of a patent pursuant to 35 U.S.C. 151. Further, a record may be disclosed, subject to the limitations of 37 CFR 1.14, as a routine use, to the public if the record was filed in an application which became abandoned or in which the proceedings were terminated and which application is referenced by either a published application, an application open to public inspections or an issued patent.
- 9. A record from this system of records may be disclosed, as a routine use, to a Federal, State, or local law enforcement agency, if the USPTO becomes aware of a violation or potential violation of law or regulation.

Electronic Acknowledgement Receipt				
EFS ID:	16758969			
Application Number:	13943395			
International Application Number:				
Confirmation Number:	2954			
Title of Invention:	Selective Recovery of Manganese, Lead and Zinc			
First Named Inventor/Applicant Name:	Stephen Harrison			
Customer Number:	35979			
Filer:	Karthika Perumal/Debra Odum			
Filer Authorized By:	Karthika Perumal			
Attorney Docket Number:	076165.000099			
Receipt Date:	04-SEP-2013			
Filing Date:	16-JUL-2013			
Time Stamp:	13:46:23			
Application Type:	Utility under 35 USC 111(a)			

# **Payment information:**

Submitted with Payment	no
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# File Listing:

Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)
1	Information Disclosure Statement (IDS)	20130904In formation Disclosur	69781	no	5
Form (SB08)	eStatement.pdf	e1ff89c02d196d81d1f4a7dbb235924f6d25 429e			

## **Warnings:**

Information: ESM / PETITIONER EX 1014\_144

This is not an U	SPTO supplied IDS fillable form				
2	Non Patent Literature	Gallup 1990.pdf	9144381 120927b2f2fbbcbd3201a01479790445d89 0e4c2	no	10
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Warnings:					
Information:					
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This Acknowledgement Receipt evidences receipt on the noted date by the USPTO of the indicated documents, characterized by the applicant, and including page counts, where applicable. It serves as evidence of receipt similar to a Post Card, as described in MPEP 503.

#### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

#### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

#### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.



35979

#### United States Patent and Trademark Office

INITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS P.O. Sox 1450 Alexandria, Virginia 22313-1450 www.uspto.gov

APPLICATION NUMBER

FILING OR 371(C) DATE

FIRST NAMED APPLICANT

ATTY. DOCKET NO./TITLE 076165.000099

13/943,395

Bracewell & Giuliani LLP

Houston, TX 77208-1389

P.O. Box 61389

07/16/2013

Stephen Harrison

**CONFIRMATION NO. 2954** 

WITHDRAWAL NOTICE

\*OC00000063332882\*

Date Mailed: 08/22/2013

#### Letter Regarding a New Notice and/or the Status of the Application

If a new notice or Filing Receipt is enclosed, applicant may disregard the previous notice mailed on 08/05/2013. The time period for reply runs from the mail date of the new notice. Within the time period for reply, applicant is required to file a reply in compliance with the requirements set forth in the new notice to avoid abandonment of the application.

Registered users of EFS-Web may alternatively submit their reply to this notice via EFS-Web. https://sportal.uspto.gov/authenticate/AuthenticateUserLocalEPF.html

For more information about EFS-Web please call the USPTO Electronic Business Center at 1-866-217-9197 or visit our website at <a href="http://www.uspto.gov/ebc.">http://www.uspto.gov/ebc.</a>

If the reply is not filed electronically via EFS-Web, the reply must be accompanied by a copy of the new notice.

If the Office previously granted a petition to withdraw the holding of abandonment or a petition to revive under 37 CFR 1.137, the status of the application has been returned to pending status.

Office of Data Management, Application Assistance Unit (571) 272-4000, or (571) 272-4200, or 1-888-786-0101



#### UNITED STATES PATENT AND TRADEMARK OFFICE

UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS P.O. Box 1450 Alexandria, Virginia 22313-1450 WWW.18910.gov

APPLICATION	FILING or	GRP ART				
NUMBER	371(c) DATE	UNIT	FIL FEE REC'D	ATTY.DOCKET.NO	TOT CLAIMS	IND CLAIMS
13/943 395	07/16/2013	1778	1660	076165 000099	16	3

**CONFIRMATION NO. 2954** 

FILING RECEIPT

\*CC00000063332219\*

Date Mailed: 08/22/2013

35979 Bracewell & Giuliani LLP P.O. Box 61389 Houston, TX 77208-1389

Receipt is acknowledged of this non-provisional patent application. The application will be taken up for examination in due course. Applicant will be notified as to the results of the examination. Any correspondence concerning the application must include the following identification information: the U.S. APPLICATION NUMBER, FILING DATE, NAME OF APPLICANT, and TITLE OF INVENTION. Fees transmitted by check or draft are subject to collection. Please verify the accuracy of the data presented on this receipt. If an error is noted on this Filing Receipt, please submit a written request for a Filing Receipt Correction. Please provide a copy of this Filing Receipt with the changes noted thereon. If you received a "Notice to File Missing Parts" for this application, please submit any corrections to this Filing Receipt with your reply to the Notice. When the USPTO processes the reply to the Notice, the USPTO will generate another Filing Receipt incorporating the requested corrections

#### Inventor(s)

Stephen Harrison, Benicia, CA; Samaresh Mohanta, Pleasanton, CA; C.V. Krishnamohan Sharma, Milpitas, CA; Elizabeth Geler, San Rafael, CA;

#### Applicant(s)

Stephen Harrison, Benicia, CA; Samaresh Mohanta, Pleasanton, CA; C.V. Krishnamohan Sharma, Milpitas, CA; Elizabeth Geler, San Rafael, CA;

#### **Assignment For Published Patent Application**

SIMBOL Inc., Pleasanton, CA

Power of Attorney: None

#### Domestic Priority data as claimed by applicant

This application is a CON of 13/539,106 06/29/2012 PAT 8518232 which claims benefit of 61/502,736 06/29/2011

**Foreign Applications** for which priority is claimed (You may be eligible to benefit from the **Patent Prosecution Highway** program at the USPTO. Please see <a href="http://www.uspto.gov">http://www.uspto.gov</a> for more information.) - None. Foreign application information must be provided in an Application Data Sheet in order to constitute a claim to foreign priority. See 37 CFR 1.55 and 1.76.

If Required, Foreign Filing License Granted: 07/31/2013

page 1 of 3

The country code and number of your priority application, to be used for filing abroad under the Paris Convention, is **US 13/943,395** 

Projected Publication Date: Request for Non-Publication Acknowledged

Non-Publication Request: Yes

Early Publication Request: No

\*\* SMALL ENTITY \*\*

Title

Selective Recovery of Manganese, Lead and Zinc

**Preliminary Class** 

210

Statement under 37 CFR 1.55 or 1.78 for AIA (First Inventor to File) Transition Applications: No

#### PROTECTING YOUR INVENTION OUTSIDE THE UNITED STATES

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Almost every country has its own patent law, and a person desiring a patent in a particular country must make an application for patent in that country in accordance with its particular laws. Since the laws of many countries differ in various respects from the patent law of the United States, applicants are advised to seek guidance from specific foreign countries to ensure that patent rights are not lost prematurely.

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For information on preventing theft of your intellectual property (patents, trademarks and copyrights), you may wish to consult the U.S. Government website, http://www.stopfakes.gov. Part of a Department of Commerce initiative, this website includes self-help "toolkits" giving innovators guidance on how to protect intellectual property in specific countries such as China, Korea and Mexico. For questions regarding patent enforcement issues, applicants may call the U.S. Government hotline at 1-866-999-HALT (1-866-999-4258).

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#### Title 35, United States Code, Section 184

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	PATEN	IT APPLI		ON FEE DE titute for Form	TERMINAT PTO-875	TON REC	ORD		Applica 13/94		nber
	APPLIC	CATION AS			olumn 2)	SMA	ALL E	NTITY	OR		
	FOR	NUMBE			ER EXTRA	RATE(\$	)	FEE(\$)	1	RATE(\$)	FEE(\$)
	IC FEE FR 1.16(a), (b), or (c))	N.	/A		N/A	N/A	十	70	1	N/A	
SEA	RCH FEE FR 1.16(k), (i), or (m))	N.	/A		N/A	N/A	$\top$	300	1	N/A	
XΑ	MINATION FEE FR 1.16(o), (p), or (q))	N,	/A		N/A	N/A	$\dashv$	360	1	N/A	
ОТ	AL CLAIMS FR 1.16(i))	16	minus	20= *		× 40	=	0.00	OR		
NDE	EPENDENT CLAIMS FR 1.16(h))	3	minus	3 = *		× 210	=	0.00	1		
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		(Column 1)		(Column 2)	(Column 3)	SMA	ALL E	NTITY	OR <b>1</b>		
	F	(Column 1)  CLAIMS  REMAINING  AFTER  MENDMENT	Minus	(Column 2) HIGHEST NUMBER PREVIOUSLY PAID FOR	PRESENT EXTRA	RATE(\$)		NTITY  ADDITIONAL FEE(\$)		SMALL RATE(\$)	ENTITY
	Total * (37 CFR 1.16(i))	CLAIMS REMAINING AFTER	Minus	HIGHEST NUMBER PREVIOUSLY PAID FOR	PRESENT EXTRA			ADDITIONAL	OR	SMALL  RATE(\$)  x =	ENTITY ADDITION
	Total (37 CFR 1.16(i))  Independent (37 CFR 1.16(h))	CLAIMS REMAINING AFTER MENDMENT	Minus Minus	HIGHEST NUMBER PREVIOUSLY PAID FOR	PRESENT EXTRA	RATE(\$)		ADDITIONAL		SMALL RATE(\$)	ENTITY ADDITION.
	Total (37 CFR 1.16(h))  Independent (37 CFR 1.16(h))  Application Size Fee (3	CLAIMS REMAINING AFTER MENDMENT	Minus	HIGHEST NUMBER PREVIOUSLY PAID FOR	PRESENT EXTRA	RATE(\$)	=	ADDITIONAL	OR OR	SMALL  RATE(\$)  x =	ENTITY ADDITION.
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AIVICINDIVICIAL A	Total (37 CFR 1.16(i))  Independent (37 CFR 1.16(h))  Application Size Fee (3	CLAIMS REMAINING AFTER MENDMENT  OF CFR 1.16(s)) N OF MULTIPL  (Column 1)	Minus	HIGHEST NUMBER PREVIOUSLY PAID FOR ***  DENT CLAIM (37	PRESENT EXTRA	RATE(\$)  x  x  TOTAL	= =	ADDITIONAL	OR OR OR	SMALL  RATE(\$)  x =  x =	ENTITY ADDITIONA
	Total (37 CFR 1.16(i))  Independent (37 CFR 1.16(h))  Application Size Fee (3  FIRST PRESENTATIO	CLAIMS REMAINING AFTER MENDMENT  7 CFR 1.16(s))	Minus	HIGHEST NUMBER PREVIOUSLY PAID FOR	PRESENT EXTRA	RATE(\$)  x  x  TOTAL	= = =	ADDITIONAL	OR OR OR	SMALL  RATE(\$)  x =  x =	ENTITY  ADDITION, FEE(\$)
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	Total (37 CFR 1.16(ii))	CLAIMS REMAINING AFTER MENDMENT  FOR 1.16(s)  N OF MULTIPL  (Column 1)  CLAIMS REMAINING AFTER MENDMENT	Minus  E DEPEN  Minus	HIGHEST NUMBER PREVIOUSLY PAID FOR  ***  COolumn 2)  HIGHEST NUMBER PREVIOUSLY PAID FOR  **  ***  ***  ***  ***  ***  ***  *	PRESENT EXTRA  =  =  CFR 1.16(j))  (Column 3)  PRESENT EXTRA	RATE(\$)  X  X  TOTAL ADD'L FE  RATE(\$)	= = = :E	ADDITIONAL FEE(\$)	OR OR OR OR	SMALL  RATE(\$)	ADDITION/ FEE(\$)
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The "Highest Number Previously Paid For" IN THIS SPACE is less than 3, enter "3".
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APPLICATION NUMBER FILING OR 371(C) DATE FIRST NAMED APPLICANT ATTY. DOCKET NO./TITLE

13/943,395 07/16/2013 Stephen Harrison

076165.000099 CONFIRMATION NO. 2954

35979 Bracewell & Giuliani LLP P.O. Box 61389 Houston, TX 77208-1389 FORMALITIES LETTER



Date Mailed: 08/22/2013

#### NOTICE TO FILE CORRECTED APPLICATION PAPERS

#### Filing Date Granted

An application number and filing date have been accorded to this application. The application is informal since it does not comply with the regulations for the reason(s) indicated below. Applicant is given TWO MONTHS from the date of this Notice within which to correct the informalities indicated below. Extensions of time may be obtained by filing a petition accompanied by the extension fee under the provisions of 37 CFR 1.136(a).

The required item(s) identified below must be timely submitted to avoid abandonment:

- Replacement drawings in compliance with 37 CFR 1.84 and 37 CFR 1.121(d) are required. The drawings submitted are not acceptable because:
  - The drawings have a line quality that is too light to be reproduced (weight of all lines and letters must be heavy enough to permit adequate reproduction) or text that is illegible (reference characters, sheet numbers, and view numbers must be plain and legible) see 37 CFR 1.84(I) and (p)(1)); See Figure(s) 14.

Applicant is cautioned that correction of the above items may cause the specification and drawings page count to exceed 100 pages. If the specification and drawings exceed 100 pages, applicant will need to submit the required application size fee.

#### **Items Required To Avoid Processing Delays:**

Applicant is notified that the above-identified application contains the deficiencies noted below. No period for reply is set forth in this notice for correction of these deficiencies. However, if a deficiency relates to the inventor's oath or declaration, the applicant must file an oath or declaration in compliance with 37 CFR 1.63, or a substitute statement in compliance with 37 CFR 1.64, executed by or with respect to each actual inventor no later than the expiration of the time period set in the "Notice of Allowability" to avoid abandonment. See 37 CFR 1.53(f).

• A properly executed inventor's oath or declaration has not been received for the following inventor(s):

Stephen Harrison

Samaresh Mohanta

C.V. Krishnamohan Sharma

Elizabeth Geler

Applicant may submit the inventor's oath or declaration at any time before the Notice of Allowance and Fee(s) Due, PTOL-85, is mailed.

Replies must be received in the USPTO within the set time period or must include a proper Certificate of Mailing or Transmission under 37 CFR 1.8 with a mailing or transmission date within the set time period. For more information and a suggested format, see Form PTO/SB/92 and MPEP 512.

Replies should be mailed to:

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If you are not using EFS-Web to submit your reply, you must include a copy of this notice.

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Office of Data N	Management,	Application	Assistance	Unit (571)	272-4000.	or (571)	272-4200.	or 1-888	-786-0101

Doc description: Information Disclosure Statement (IDS) Filed

Approved for use through 07/31/2012. OMB 0651-0031
U.S. Patent and Trademark Office; U.S. DEPARTMENT OF COMMERCE

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#### **Application Number** 13943395 Filing Date 2013-07-16 **INFORMATION DISCLOSURE** First Named Inventor Stephen Harrison STATEMENT BY APPLICANT Art Unit ( Not for submission under 37 CFR 1.99) **Examiner Name** Unknown Attorney Docket Number 076165.000099

				U.S.	PATENTS		
Examiner Initial*	Cite No					Name of Patentee or Applicant of cited Document	Pages,Columns,Lines where Relevant Passages or Relevant Figures Appear
	1	4016075		1977-04-05	Wilkins		
	2	5254225		1993-10-19	Gallup		
	3	6682644		2004-01-27	Featherstone et al.		
	4	5135652		1992-08-04	Boateng		
	5	4116856		1978-09-26	Lee et al.		
	6	4159311		1979-06-26	Lee et al.		
	7	4221767		1980-09-09	Lee et al.		
	8	4291001		1981-09-22	Repsher et al.		

( Not for submission under 37 CFR 1.99)

Application Number		13943395	
Filing Date		2013-07-16	
First Named Inventor Steph		nen Harrison	
Art Unit	•		
Examiner Name Unknown		own	
Attorney Docket Numb	er	076165 000099	

9	4347327	1982-08-31	Lee et al.	
10	4348295	1982-09-07	Burba, III	
11	4348296	1982-09-07	Bauman, et al.	
12	4348297	1982-09-07	Bauman, et al.	
13	4376100	1983-03-08	Lee et al.	
14	4430311	1984-02-07	Lee et al.	
15	4461714	1984-07-24	Burba, III	
16	4472362	1984-09-18	Burba, III	
17	4540509	1985-09-10	Burba, III	
18	4727167	1988-02-23	Burba, III, et al.	
19	5389349	1995-02-14	Bauman et al.	

# INFORMATION DISCLOSURE STATEMENT BY APPLICANT (Not for submission under 37 CFR 1.99) Application Number 13943395 Filing Date 2013-07-16 First Named Inventor Stephen Harrison Art Unit Examiner Name Unknown Attorney Docket Number 076165.000099

	20	5599516		1997-02-04	Bauman et al.
	21	6017500	V dev de version de la constante de la constan	2000-01-25	Mehta
	22	6280693		2001-08-28	Bauman et al.
	23	6555078		2003-04-29	Mehta
	24	6458184		2002-10-01	Featherstone
	25	8454816		2013-06-04	Harrison, et al.
If you wis	h to add	l additional U.S. Patent	citatio	n information pl	ease click the Add button.

#### **U.S.PATENT APPLICATION PUBLICATIONS**

Examiner Initial*	Cite No	Publication Number	Kind Code <sup>1</sup>	Publication Date	Name of Patentee or Applicant of cited Document	Pages,Columns,Lines where Relevant Passages or Relevant Figures Appear
	1	20010000597	A1	2001-05-03	Featherstone	
-	2	20030226761	A1	2003-12-11	Featherstone et al.	

If you wish to add additional U.S. Published Application citation information please click the Add button.

#### **FOREIGN PATENT DOCUMENTS**

( Not for submission under 37 CFR 1.99)

Application Number		13943395
Filing Date		2013-07-16
First Named Inventor	Steph	en Harrison
Art Unit		
Examiner Name	Unkno	own
Attorney Docket Number		076165.000099

Examiner Initial*	Cite No	Foreign Document Number³	Country Code <sup>2</sup> İ	Kind Code4	Publication Date	Name of Patentee or Applicant of cited Document	Pages,Columns,Lines where Relevant Passages or Relevant Figures Appear	<b>T</b> 5			
	1										
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Examiner Initials*  Cite No  Include name of the author (in CAPITAL LETTERS), title of the article (when appropriate), title of the item (book, magazine, journal, serial, symposium, catalog, etc), date, pages(s), volume-issue number(s), publisher, city and/or country where published.											
	1	COLE ET AL., "Zinc Sol (2003), pp. 91-137	COLE ET AL., "Zinc Solvent Extraction in the Process Industries", 24(2) Mineral Proc. & Extractive Metallurgy Rev (2003), pp. 91-137								
	2	DREISINGER ET AL., "New Developments in the Boleo Copper-Cobalt-Zinc-Manganese Project", available at http://bajamining.com/_resources/Reports/alta_paper_2006_boleo_final.pdf									
	3					cions. Simulation of Counter essing (2004), pp. 113-120	r-Current Extraction				
	4	KAWAI ET AL., "Solvent extraction of zinc(II) and manganese(II) with 5,10,15,20-tetraphenyl-21H, 23H-porphine(TPP) through the metal exchange reaction of lead(II)-TPP", 7 Solvent Extr. Res. Dev. Japan (2000), pp. 36-43  LEE ET AL., "Solvent Extraction of Zinc from Strong Hydrochloric Acid Solution with Alamine 336", 30(7) Bull. Korean Chem. Soc. (2009), pp. 1526-1530									
	5										
	6										
If you wis	h to a	dd additional non-pater	it literature docu	ıment cit	tation informat	tion please click the Add	button				

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Application Number		13943395
Filing Date		2013-07-16
First Named Inventor	Steph	en Harrison
Art Unit		
Examiner Name	ame Unknown	
Attorney Docket Number		076165.000099

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EXAMINER SIGNATURE				
Examiner Signature	Date Considered			
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Standard ST.3). <sup>3</sup> For Japanese patent documents, the indication	GOV or MPEP 901.04. <sup>2</sup> Enter office that issued the document, by the two-letter code (WIPO of the year of the reign of the Emperor must precede the serial number of the patent document. e document under WIPO Standard ST.16 if possible. <sup>5</sup> Applicant is to place a check mark here if			

( Not for submission under 37 CFR 1.99)

Application Number		13943395
Filing Date		2013-07-16
First Named Inventor	Steph	en Harrison
Art Unit		
Examiner Name	Unkn	own
Attorney Docket Numb	er	076165.000099

	CERTIFICATION STATEMENT			
Plea	Please see 37 CFR 1.97 and 1.98 to make the appropriate selection(s):			
	That each item of information contained in the information disclosure statement was first cited in any communication from a foreign patent office in a counterpart foreign application not more than three months prior to the filing of the information disclosure statement. See 37 CFR 1.97(e)(1).			
OR				
	foreign patent of after making rea any individual de	information contained in the information diffice in a counterpart foreign application, an sonable inquiry, no item of information containsignated in 37 CFR 1.56(c) more than the 37 CFR 1.97(e)(2).	id, to the knowledge of th ained in the information dis	e person signing the certification sclosure statement was known to
	See attached ce	rtification statement.		
	The fee set forth	in 37 CFR 1.17 (p) has been submitted here	ewith.	
$\boxtimes$	A certification sta	atement is not submitted herewith.		
	SIGNATURE  A signature of the applicant or representative is required in accordance with CFR 1.33, 10.18. Please see CFR 1.4(d) for the form of the signature.			
Sigr	nature	kaethika	Date (YYYY-MM-DD)	2013-08-15
Nan	ne/Print	Karthika Perumal	Registration Number	69296

This collection of information is required by 37 CFR 1.97 and 1.98. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.14. This collection is estimated to take 1 hour to complete, including gathering, preparing and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. **SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.** 

Electronic Acknowledgement Receipt		
EFS ID:	16603296	
Application Number:	13943395	
International Application Number:		
Confirmation Number:	2954	
Title of Invention:	Selective Recovery of Manganese, Lead and Zinc	
First Named Inventor/Applicant Name:	Stephen Harrison	
Customer Number:	35979	
Filer:	Karthika Perumal/Debra Odum	
Filer Authorized By:	Karthika Perumal	
Attorney Docket Number:	076165.000099	
Receipt Date:	15-AUG-2013	
Filing Date:	16-JUL-2013	
Time Stamp:	16:36:20	
Application Type: Utility under 35 USC 111(a)		

## **Payment information:**

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## File Listing:

Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /₊zip	Pages (if appl.)
1	Information Disclosure Statement (IDS)	2013 0815 Information Disclosur	93457	no	6
'	Form (SB08)	eStatement.pdf	28824488c0331494235b90e3948995445e7 b2c5f		

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			ec2d63398d93b498c2f47a3bc17cb714f750 49b0	110	
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5	Non Patent Literature	Kawai.PDF	347015	no	8
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6	Non Patent Literature	Lee.PDF	350404	no	5
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		Total Files Size (in bytes)	203	31001	

This Acknowledgement Receipt evidences receipt on the noted date by the USPTO of the indicated documents, characterized by the applicant, and including page counts, where applicable. It serves as evidence of receipt similar to a Post Card, as described in MPEP 503.

#### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

#### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

#### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

#### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Patent Application of: Stephen HARRISON, et al. Examiner: Not Assigned 13/943,395 Serial No.: Art Unit: Not Assigned Filed: Confirmation No.: July 16, 2013 2954 For: SELECTIVE RECOVERY OF Attorney Docket No. 076165.000099 MANGANESE, LEAD AND ZINC

## SECOND PRELIMINARY AMENDMENT AND RESPONSE TO NOTICE TO FILE CORRECTED PAPERS

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Dear Sir,

Applicant is responding to the Notice to file Corrected Papers dated August 5, 2013, by submitting a replacement Figure 14 along with this Second Preliminary Amendment.

By this Second Preliminary Amendment, please amend the above-identified application as set forth below.

Amendments to the Specification begin on page 2 of this Second Preliminary Amendment.

Amendments to the Drawings begin on page 8 of this Second Preliminary Amendment.

Amendments to Claims begin on page 9 of this Second Preliminary Amendment.

Remarks begin on page 12 of this Second Preliminary Amendment.

#### Amendment to the Specification

Please replace paragraphs [0084], [0087], [0094], [0096], [0098], [0099], [00100], [00101], [00107], [00110], [00113], [00115], [00116], and [00128] of the specification with the corresponding replacement paragraphs [0084], [0087], [0094], [0096], [0098], [0099], [00100], [00101], [00107], [00110], [00113], [00115], [00116], and [00128] described below.

#### Replacement paragraph [0084]:

**[0084]** In another embodiment, as provided in Figure 7, process 700 for the recovery of manganese or zinc from a geothermal brine is provided. As previously discussed, a first step <u>710</u> of the process includes the removal of iron and silica from the brine solution. In certain embodiments, the iron is oxidized and base is added to control the pH of the solution to about 5 and 6. In certain embodiments, iron is oxidized with air and the base is calcium oxide or calcium hydroxide.

#### Replacement paragraph [0087]:

[0087] In another embodiment, as provided in Figure 8, process 800 for the recovery of manganese or zinc from a geothermal brine is provided. As previously discussed, a first step 810 of the process includes the removal of iron and silica from the brine solution. In certain embodiments, the iron is oxidized and base is added to control the pH of the solution to between about 4.5 and 6, preferably between about 4.75 and 5.5. In certain embodiments, iron is oxidized with air and the base is calcium oxide or calcium hydroxide.

#### Replacement paragraph [0094]:

[0094] Referring now to Figure 10, in another aspect, a **tenth** process <u>1000</u> for the recovery of manganese and/or zinc from a geothermal brine is provided. First step 1010 of the process includes the removal of iron and silica from the brine solution, as previously described herein. In certain embodiments, the iron is oxidized and base is added to control the pH of the solution to

between about 4.5 and 6, preferably between about 4.75 and 5.5. In certain embodiments, iron is oxidized with air and the base is calcium oxide or calcium hydroxide.

#### Replacement paragraph [0096]:

[0096] The acid solution is filtered in step 1040 to produce a manganese <u>and zinc</u> containing solution and <u>zine precipitates remove remaining solids</u>. <u>Zine precipitates are electrochemically converted to zine metal in step 1050</u>. The solution is passed to zine solvent extraction step 1060 to recovery remaining zine. The solution is passed to zine extraction step 1050 to selectively remove zinc. The zinc containing solution may be optionally purified using electrochemical means 1060. The manganese containing solution from the filtration step is provided to a reduction step 1070 wherein the manganese containing solution is contacted with a reducing agent, such as SO<sub>2</sub>. In step 1080, the reduced manganese can be recovered from the solution by electrolytically depositing manganese dioxide, such as by electrowinning.

#### Replacement paragraph [0098]:

[0098] Referring now to Figure 11, in another aspect, a **tenth** process <u>1100</u> for the recovery of manganese and/or zinc from a geothermal brine is provided. First step 1110 of the process includes the removal of iron and silica from the brine solution, as previously described herein. In certain embodiments, the iron is oxidized and base is added to control the pH of the solution to between about 4.5 and 6, preferably between about 4.75 and 5.5. In certain embodiments, iron is oxidized with air and the base is calcium oxide or calcium hydroxide

#### Replacement paragraph [0099]:

[0099] Following the removal of the iron and silica, in precipitation step 1120, additional base is added to cause the precipitation of manganese and zinc. The manganese and zinc precipitates are collected by known means and, in step 1130, dissolved in an ammonium sulfate solution. Optionally, the ammonium sulfate solution that includes the manganese and zinc can be purified. In certain embodiments, the remaining brine solution from extraction step 1020 1120, from

which the manganese and zinc have been removed, can then be reinjected into the geothermal well from which the brine was originally removed.

#### Replacement paragraph [00100]:

manganese <u>and zinc</u> containing solution and <u>zine precipitates remove remaining solids</u>. Zine precipitates are electrochemically converted to zinc metal in step 1050. The solution is passed to zinc solvent extraction step 1060 to recovery remaining zinc. The solution is passed to zinc extraction step 1150 to selectively remove zinc. The zinc containing solution may be optionally purified using electrochemical means 1160. The manganese containing solution from the filtration step is provided to a reduction step 1070 1170 wherein the manganese containing solution is contacted with a reducing agent, such as SO<sub>2</sub>. In step 1080 1180, the reduced manganese can be recovered from the solution by electrolytically depositing manganese dioxide, such as by electrowinning.

#### Replacement paragraph [00101]:

[00101] Optionally, as shown by the dashed line, following recovery of zinc in electrochemical recovery step 1060 1160, a sulfuric acid-rich solution can be recycled to zinc extraction step 1050 1150. Similarly, as shown by the dashed line, following the electrochemical recovery of manganese in step 1080 1180, the remaining brine solution can be recycled to either precipitation step 1020 1120 or dissolution step 1030 1130.

#### Replacement paragraph [00107]:

[00107] Referring now to Figure 12, one embodiment of the present invention is provided. Process 100 1200 for the selective removal of manganese from a manganese containing solution, such as a geothermal brine, is provided. Brine 1202 is provided via line 1204 to manganese reaction tank 1206. Ammonia is supplied via line 1208 to manganese reaction tank 1206, where it contacts the manganese containing solution to selectively precipitate manganese having a purity of greater than 95%, alternatively greater than about 97%, alternatively greater than about

99%. The manganese reaction tank 1206 is maintained at a pH of at least about 6.8, alternatively at least about 8.2, alternatively at least about 8.4, to limit co-precipitation of other metal ions. A manganese-oxide/hydroxide precipitate can be collected from reaction tank 1206 via line 1212. In certain embodiments, it is believed that the manganese-oxide/hydroxide may include a high percentage of Mn<sub>3</sub>O<sub>4</sub>. The brine, having a reduced concentration of manganese, also referred to as a manganese depleted brine solution, can optionally be supplied via line 1210 to a holding tank 1214. In certain embodiments, the brine can be supplied via line 1210 into an injection well (not shown). Air is supplied via line 1216 to produce a reduced pH brine solution having a pH of less than about 7, alternatively less than about 6, alternatively between about 5 and 6. The reduced pH solution can be supplied from holding tank 1214 via line 1217 to zinc reaction tank 1218, which can also be supplied with lime supplied via line 1220, to increase the pH to greater than about 7, alternatively be a pH of between about 7.2 and 7.7, alternatively about 7.5, thereby causing the zinc to precipitate. The zinc precipitate can be collected via line 1222, and the remaining brine solution, having a reduced concentration of both manganese and zinc, also referred to as a manganese and zinc depleted brine solution, can be removed via line 1220 1224. Brine removed via line 1224 can be supplied to an alternate process for the recovery of additional metal ions, or alternatively can be supplied to an injection well (not shown).

#### Replacement paragraph [00110]:

Referring now to Figure 14, one embodiment of the present invention is provided. Brine is supplied from a holding tank or directly from the source, such as a geothermal well, via line 1402 to brine tank 1404. To brine tank 1404, ammonia can be supplied via line 1408 from ammonia tank 1406. Ammonia is supplied in known amounts to selectively precipitate manganese present in the brine, which other ions remain in solution. The solid manganese precipitate can be collected from brine tank 1404 via line 1410. The solution in brine tank 1404, which includes brine (having a lower manganese concentration than originally supplied) and ammonia are supplied via line 1412 to tank 1414, which can include stirring means, such as a mechanical stirrer, and can be supplied with air via line 1416. Air can be added via line 1416 to reduce the pH of the solution selectively to less than 7, preferably between 5 and 6. Ammonia can be removed as a gas from tank 1414 via line 1420 and ammonia-free brine can be removed

from the tank via line 1418. Ammonia removed via line 1420 can be supplied to a separation tank 1422 wherein air is separated via line 1424 and ammonia is separated via line 1426, and can be recycled back to ammonia tank 1406. In certain embodiments, ammonia tank 1406 can be supplied with fresh make up ammonia, as needed or desired.

#### Replacement paragraph [00113]:

[00113]As shown in Figure 16, a process 1600 for the extraction of various metals from brine, according to one aspect of the present invention, is provided. Brine is supplied via line 1602 and is combined with air supplied via line 1604, lime (20% solution by volume) supplied via line 1606, and flocculant (0.025% by volume in water) supplied via line 1608 in silica removal reactor 1610. Exemplary flow rates are as follows: brine (6 gal/min), flocculant (0.01 gal/min), air (100 cfm), and lime (0.5 lb/min). In silica removal reactor, a solid silica precipitate is formed, which is then removed via line 1612. Air and water vapor can be removed from silica removal reactor 1610 vai line 1614. Under the exemplary flow conditions, for a brine solution having a silica concentration of about 10 ppm, production of the wet silica cake is about 15 lb/hr. A silica lean brine solution can then be supplied via line 1616 to a lithium extraction reactor 1618. Lithium extraction reactor 1618 can include a lithium aluminum intercalate or other extraction medium that has been prepared for the purposes of extracting lithium. Lithium extraction reactor 1618 can include a water inlet 1620, a lithium salt extraction line 1622 for removal of lithium, typically as a chloride salt, after the salt has extracted from the silica lean brine. Water vapor can be removed from reactor 1618 via line 1624.

#### Replacement paragraph [00115]:

[00115] Following the lead extraction, the remaining solution is supplied <u>via line 1640</u> to manganese extraction reactor 1642, which can include any of the several different examples of manganese extraction that have been described herein. A remaining brine solution, having reduced concentration of silica, lithium, zinc, lead and manganese, can be collected via line 1644 and either injected into a geothermal well, or supplied to further extraction or other processes. A solid manganese oxide and/or manganese hydroxide precipitate (which can include MnO<sub>4</sub>,

MnO<sub>2</sub>, and/or Mn(OH)<sub>2</sub>) can be collected via line 1646. Preferably, air is excluded during the manganese precipitation.

Replacement paragraph [00116]:

[00116]Referring now to Figure 17, a process 1700 for the extraction of various metals from brine, according to one aspect of the present invention, is provided. Brine is supplied via line 1702 and is combined with air supplied via line 1704, lime (20% solution by volume) supplied via line 1706, and flocculant (0.025% by volume in water) supplied via line 1708 in silica removal reactor 1710. Exemplary flow rates are as follows: brine (6 gal/min), flocculant (0.01 gal/min), air (100 cfm), and lime (0.5 lb/min). In silica removal reactor 1710, a solid silica precipitate is formed, which is then removed via line 1712. Air and water vapor can be removed from silica removal reactor 1710 via line 1714. Under the exemplary flow conditions, for a brine solution having a silica concentration of about 10 ppm, production of the wet silica cake is about 15 lb/hr. A silica lean brine solution can then be supplied via line 1716 to a zinc extraction reactor 1718, which in certain embodiments can be an ion exchange resin that is designed to extract zinc ions, while allowing other ions to pass through the membrane. Zinc chloride can be collected from the extraction process via line 1720. The remaining brine solution, having had silica and zinc extracted therefrom, can be supplied via line 1722 to lead extraction reactor 1724. In certain embodiments, the brine solution supplied via line 1722 can have a pH of between about 5 and 6, and a temperature of between about 75°C and 105°C In lead extraction reactor 1724, the brine is contacted with a sulfide compound, such as sodium sulfide, hydrogen sulfide, sodium hydrogen sulfide, calcium sulfide, and the like, which can be supplied to the reactor via line 1726, to form a lead sulfide precipitate. The lead sulfide precipitate can be removed from reactor 1724 via line 1728.

Replacement paragraph [00128]:

[**00128**] Example **5** <u>6</u>

#### **Amendments to the Drawings**

Please replace Figure 14 in the application with the replacement Figure 14 that is being submitted along with this Second Preliminary Amendment.

#### Amendment to the Claims

Claims 1-31 have been cancelled. Claims 32-47 are new.

This listing of claims will replace all prior versions, and listing of claims in the application.

- 1-31. (Cancelled).
- 32. (New) A method for recovering zinc from a geothermal brine, the method comprising the steps of:

providing a geothermal brine, said geothermal brine comprising zinc;

selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine; and

removing the zinc from the substantially silica free brine.

- 33. (New) The method of claim 32, wherein the silica and iron are selectively removed from the geothermal brine to produce a substantially silica and iron free brine.
- 34. (New) The method of claim 32, wherein the zinc is removed from the substantially silica free brine by solvent extraction.
- 35. (New) The method of claim 33, wherein the zinc is removed from the substantially silica free brine using an organic solvent.
- 36. (New) The method of claim 33, wherein the zinc is removed from the substantially silica free brine using a phosphorous-based solvent.
- 37. (New) The method of claim 33, wherein the zinc is removed from the substantially silica free brine using a solvent selected from the group consisting of phosphines, phosphoric acids, and phosphinic acids.
- 38. (New) The method of claim 32, wherein the zinc is removed from the substantially silica free brine by ion exchange.

- 39. (New) The method of claim 32, further comprising the step of extracting manganese from the substantially silica free brine by solvent extraction.
- 40. (New) The method of claim 32, further comprising the step of extracting manganese from the substantially silica free brine by ion exchange.
- 41. (New) A method for recovering manganese from a geothermal brine, the method comprising the steps of:

providing a geothermal brine, said geothermal brine comprising zinc;

selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine; and

extracting the manganese from the substantially silica free brine.

- 42. (New) The method of claim 41, wherein the manganese is extracted from the substantially silica free brine by precipitation.
- 43. (New) The method of claim 41, wherein the manganese is extracted from the substantially silica free brine by selective precipitation using an amine or ammonia.
- 44. (New) A method for recovering zinc and manganese from a geothermal brine, the method comprising the steps of:

providing a geothermal brine, said geothermal brine comprising manganese and zinc; selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine;

selectively removing lithium chloride from the substantially silica free brine; removing the zinc from the substantially silica free brine; and extracting manganese from the substantially silica free brine.

- 45. (New) The method of claim 44, wherein the lithium chloride is selectively removed from the substantially silica free brine by use of lithium aluminum intercalates.
- 46. (New) The method of claim 44, wherein the lithium chloride is selectively removed from the substantially silica free brine by ion exchange.
- 47. (New) The method of claim 44, wherein the lithium chloride is selectively removed from the substantially silica free brine by electrochemical means.

Remarks

Paragraphs [0084], [0087], [0094], [0096], [0098], [0099], [00100], [00101], [00107],

[00110], [00113], [00115], [00116], and [00128] of the specification have been amended to

correct typographical errors and to properly reflect the numbers in the drawings. No new matter

has been added by these amendments to the specification.

Applicant has amended the claims for the purpose of filing a continuation application.

No new matter has been added by these newly added claims.

Prior to examination of this application, please cancel claims 1-31, and consider claims

32-47 as presented on pages 9-11.

Applicant has provided replacement Figure 14 in response to the Notice to file Corrected

Papers. Replacement Figure 14 is a clearer version of the previously submitted Figure 14.

Additionally, several elements have been renumbered to correctly identify items in accordance

with the specification as filed.

Should the Commissioner deem any fees as being due, including fees for extensions of

time, the Commissioner is authorized to debit said fees from, or credit overpayments to,

Bracewell & Giuliani LLP, USPTO Deposit Account Number 50-0259, Reference No.

076165.000099.

The Examiner is requested to contact the undersigned via telephone if a telephone

conference would expedite or aid the prosecution and examination of this application.

Respectfully submitted,

Date: August 14, 2013

/Karthika Perumal/

Karthika Perumal

Reg. No. 69,296

BRACEWELL & GIULIANI LLP

P.O. Box 61389

Houston, Texas 77208-1389

Telephone: (713) 221-1342

Facsimile: (800) 404-3970

-12-

### **Replacement Figure 14**

1400

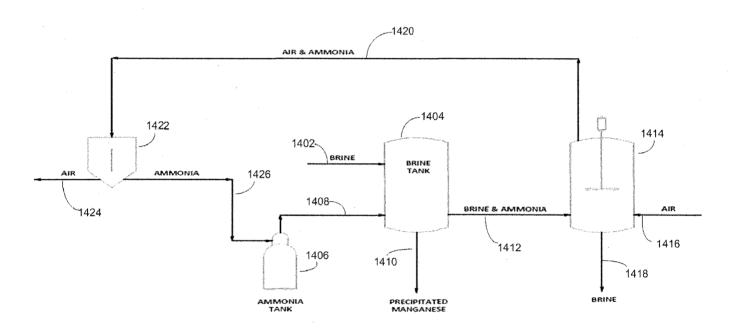


Figure 14

Electronic Acknowledgement Receipt		
EFS ID:	16590052	
Application Number:	13943395	
International Application Number:		
Confirmation Number:	2954	
Title of Invention:	Selective Recovery of Manganese, Lead and Zinc	
First Named Inventor/Applicant Name:	Stephen Harrison	
Customer Number:	35979	
Filer:	Karthika Perumal/Debra Odum	
Filer Authorized By:	Karthika Perumal	
Attorney Docket Number:	076165.000070	
Receipt Date:	14-AUG-2013	
Filing Date:	16-JUL-2013	
Time Stamp:	15:56:15	
Application Type: Utility under 35 USC 111(a)		

## **Payment information:**

Submitted with Payment	no
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## File Listing:

Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)
1		20130814SecondPreliminaryA	129399	ves	13
'		mendment.pdf	38a287ac57454924e1bfc4d5815a9c79255 6275b	, l	13

	Multipart Description/PDF files in .zip description				
	Document Description	Start	End		
	Miscellaneous Incoming Letter	1	1		
	Specification	2	2		
	Claims	3	11		
	Applicant Arguments/Remarks Made in an Amendment	12	12		
	Drawings-only black and white line drawings	13	13		
Varnings:					

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#### Information:

Total Files Size (in bytes):	129399

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#### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

#### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

#### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.



#### United States Patent and Trademark Office

INITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS P.O. Sox 1450 Alexandria, Virginia 22313-1450 www.uspto.gov

ATTY. DOCKET NO./TITLE APPLICATION NUMBER FILING OR 371(C) DATE FIRST NAMED APPLICANT

13/943,395 07/16/2013 Stephen Harrison

076165.000070 **CONFIRMATION NO. 2954** 

35979 Bracewell & Giuliani LLP P.O. Box 61389 Houston, TX 77208-1389 **FORMALITIES LETTER** 

Date Mailed: 08/05/2013

#### NOTICE TO FILE CORRECTED APPLICATION PAPERS

#### Filing Date Granted

An application number and filing date have been accorded to this application. The application is informal since it does not comply with the regulations for the reason(s) indicated below. Applicant is given TWO MONTHS from the date of this Notice within which to correct the informalities indicated below. Extensions of time may be obtained by filing a petition accompanied by the extension fee under the provisions of 37 CFR 1.136(a).

The required item(s) identified below must be timely submitted to avoid abandonment:

- Replacement drawings in compliance with 37 CFR 1.84 and 37 CFR 1.121(d) are required. The drawings submitted are not acceptable because:
  - The drawings must be reasonably free from erasures and must be free from alterations, overwriting, interlineations, folds, and copy marks. See Figure(s) 14.
  - The drawings submitted to the Office are not electronically reproducible because portions of figures 14 are missing and/or blurry.

Applicant is cautioned that correction of the above items may cause the specification and drawings page count to exceed 100 pages. If the specification and drawings exceed 100 pages, applicant will need to submit the required application size fee.

#### **Items Required To Avoid Processing Delays:**

Applicant is notified that the above-identified application contains the deficiencies noted below. No period for reply is set forth in this notice for correction of these deficiencies. However, if a deficiency relates to the inventor's oath or declaration, the applicant must file an oath or declaration in compliance with 37 CFR 1.63, or a substitute statement in compliance with 37 CFR 1.64, executed by or with respect to each actual inventor no later than the expiration of the time period set in the "Notice of Allowability" to avoid abandonment. See 37 CFR 1.53(f).

A properly executed inventor's oath or declaration has not been received for the following inventor(s):

Stephen Harrison

Samaresh Mohanta

C.V. Krishnamohan Sharma

Elizabeth Geler

Applicant may submit the inventor's oath or declaration at any time before the Notice of Allowance and Fee(s) Due, PTOL-85, is mailed.

Replies must be received in the USPTO within the set time period or must include a proper Certificate of Mailing or Transmission under 37 CFR 1.8 with a mailing or transmission date within the set time period. For more information and a suggested format, see Form PTO/SB/92 and MPEP 512.

Replies should be mailed to:

Mail Stop Missing Parts Commissioner for Patents P.O. Box 1450 Alexandria VA 22313-1450

Registered users of EFS-Web may alternatively submit their reply to this notice via EFS-Web. <a href="https://sportal.uspto.gov/authenticate/AuthenticateUserLocalEPF.html">https://sportal.uspto.gov/authenticate/AuthenticateUserLocalEPF.html</a>

For more information about EFS-Web please call the USPTO Electronic Business Center at **1-866-217-9197** or visit our website at <a href="http://www.uspto.gov/ebc.">http://www.uspto.gov/ebc.</a>

If you are not using EFS-Web to submit your reply, you must include a copy of this notice.

	/tle/							
Office of Data N	Management,	Application As	sistance Unit	(571) 272-40	00, or (571	) 272-4200.	or 1-888	-786-0101

PATENT APPLICATION FEE DETERMINATION RECORD Substitute for Form PTO-875							Application or Docket Number 13/943,395				
	APPLICATION AS FILED - PART I (Column 1) (Column 2) SMALL ENTITY							OR	OTHER THAN SMALL ENTITY		
FOR NUMBER FILED  BASIC FEE (37 CFR 1.16(a), (b), or (c))  N/A		D NUMBE	NUMBER EXTRA N/A		RATE(\$) N/A	FEE(\$) 70	]	RATE(\$)	FEE(\$)		
		N						N/A			
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(37 CFR 1.16(k), (i), or (m))  EXAMINATION FEE (37 CFR 1.16(i), (p), or (q))  TOTAL CLAIMS (37 CFR 1.16(i))  31 minus 20 = 11		I/A	T	N/A	360	1	N/A				
		minus	20= *	= 11		x 40 =	440	OR			
INDEPENDENT CLAIMS 5			minus	3 = *	2	,	x 210 =	420	1		
APPLICATION SIZE FEE (37 CFR 1.16(s))  If the specification and drawings exceed 100 sheets of paper, the application size fee due is \$310 (\$155 for small entity) for each additional 50 sheets or fraction thereof. See 35 U.S.C. 41(a)(1)(G) and 37 CFR 1.16(s).								0.00			
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₹	Application Size Fe	Size Fee (37 CFR 1.16(s))							]		
	FIRST PRESENTATION OF MULTIPLE DEPENDENT CLAIM (37 CFR 1.16(j))						OR				
							TOTAL ADD'L FEE		OR	TOTAL ADD'L FEE	
$\neg$	Ī	(Column 1) CLAIMS REMAINING		(Column 2) HIGHEST NUMBER	(Column 3)	Γ		ADDITIONAL	1		ADDITIONAL
NT B		AFTER AMENDMENT		PREVIOUSLY PAID FOR	EXTRA		RATE(\$)	FEE(\$)		RATE(\$)	FEE(\$)
₩ 	Total (37 CFR 1.16(i))	-	Minus		=	,	=		OR	x =	
AMENDMENT	Independent (37 CFR 1.16(h))	*	Minus	**	=	,	=		OR	x =	
⋛	Application Size Fee (37 CFR 1.16(s))								]		
	FIRST PRESENTATION OF MULTIPLE DEPENDENT CLAIM (37 CFR 1.16(j))								OR		
						_	TOTAL ADD'L FEE		OR	TOTAL ADD'L FEE	
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UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS P.O. Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov

APPLICATION	FILING or	GRP ART				
NUMBER	371(c) DATE	UNIT	FIL FEE REC'D	ATTY.DOCKET.NO	TOT CLAIMS	IND CLAIMS
13/943 395	07/16/2013	1778	1660	076165 000070	31	5

**CONFIRMATION NO. 2954** 

**FILING RECEIPT** 

\*OC00000062960160\*

Date Mailed: 08/05/2013

35979 Bracewell & Giuliani LLP P.O. Box 61389 Houston, TX 77208-1389

Receipt is acknowledged of this non-provisional patent application. The application will be taken up for examination in due course. Applicant will be notified as to the results of the examination. Any correspondence concerning the application must include the following identification information: the U.S. APPLICATION NUMBER, FILING DATE, NAME OF APPLICANT, and TITLE OF INVENTION. Fees transmitted by check or draft are subject to collection. Please verify the accuracy of the data presented on this receipt. If an error is noted on this Filing Receipt, please submit a written request for a Filing Receipt Correction. Please provide a copy of this Filing Receipt with the changes noted thereon. If you received a "Notice to File Missing Parts" for this application, please submit any corrections to this Filing Receipt with your reply to the Notice. When the USPTO processes the reply to the Notice, the USPTO will generate another Filing Receipt incorporating the requested corrections

Inventor(s)

Stephen Harrison, Benicia, CA; Samaresh Mohanta, Pleasanton, CA; C.V. Krishnamohan Sharma, Milpitas, CA; Elizabeth Geler, San Rafael, CA;

Applicant(s)

Stephen Harrison, Benicia, CA; Samaresh Mohanta, Pleasanton, CA; C.V. Krishnamohan Sharma, Milpitas, CA; Elizabeth Geler, San Rafael, CA;

Non-Applicant Assignee(s)

SIMBOL Inc., Pleasanton, CA

Power of Attorney: None

Domestic Priority data as claimed by applicant

This application is a CON of 13/539,106 06/29/2012 which claims benefit of 61/502,736 06/29/2011

**Foreign Applications** for which priority is claimed (You may be eligible to benefit from the **Patent Prosecution Highway** program at the USPTO. Please see <a href="http://www.uspto.gov">http://www.uspto.gov</a> for more information.) - None. Foreign application information must be provided in an Application Data Sheet in order to constitute a claim to foreign priority. See 37 CFR 1.55 and 1.76.

If Required, Foreign Filing License Granted: 07/31/2013

page 1 of 3

The country code and number of your priority application, to be used for filing abroad under the Paris Convention, is **US 13/943,395** 

Projected Publication Date: Request for Non-Publication Acknowledged

Non-Publication Request: Yes

Early Publication Request: No

\*\* SMALL ENTITY \*\*

Title

Selective Recovery of Manganese, Lead and Zinc

**Preliminary Class** 

210

Statement under 37 CFR 1.55 or 1.78 for AIA (First Inventor to File) Transition Applications: No

#### PROTECTING YOUR INVENTION OUTSIDE THE UNITED STATES

Since the rights granted by a U.S. patent extend only throughout the territory of the United States and have no effect in a foreign country, an inventor who wishes patent protection in another country must apply for a patent in a specific country or in regional patent offices. Applicants may wish to consider the filing of an international application under the Patent Cooperation Treaty (PCT). An international (PCT) application generally has the same effect as a regular national patent application in each PCT-member country. The PCT process **simplifies** the filing of patent applications on the same invention in member countries, but **does not result** in a grant of "an international patent" and does not eliminate the need of applicants to file additional documents and fees in countries where patent protection is desired.

Almost every country has its own patent law, and a person desiring a patent in a particular country must make an application for patent in that country in accordance with its particular laws. Since the laws of many countries differ in various respects from the patent law of the United States, applicants are advised to seek guidance from specific foreign countries to ensure that patent rights are not lost prematurely.

Applicants also are advised that in the case of inventions made in the United States, the Director of the USPTO must issue a license before applicants can apply for a patent in a foreign country. The filing of a U.S. patent application serves as a request for a foreign filing license. The application's filing receipt contains further information and guidance as to the status of applicant's license for foreign filing.

Applicants may wish to consult the USPTO booklet, "General Information Concerning Patents" (specifically, the section entitled "Treaties and Foreign Patents") for more information on timeframes and deadlines for filing foreign patent applications. The guide is available either by contacting the USPTO Contact Center at 800-786-9199, or it can be viewed on the USPTO website at http://www.uspto.gov/web/offices/pac/doc/general/index.html.

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## Title 35, United States Code, Section 184

# Title 37, Code of Federal Regulations, 5.11 & 5.15

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#### **NOT GRANTED**

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## IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

	Application of:  nen <b>HARRISON</b> , et al.	Examiner: Not Assigned
Serial No.:	13/943395	Art Unit:
Filed:	July 16, 2013	Confirmation No.: 2954
	ECTIVE RECOVERY OF ) NGANESE, LEAD AND ZINC )	Attorney Docket No. 076165.000070 (Continuation)

# **PRELIMINARY AMENDMENT**

Commissioner for Patents P.O. Box 1450 Alexandria, VA 22313-1450

Dear Sir,

By this Preliminary Amendment, please amend the above-identified application as set forth below.

Amendments to the specification begin on page 2 of this Preliminary Amendment.

Amendments to Claims begin on page 3 of this Preliminary Amendment.

Remarks begin on page 4 of this Preliminary Amendment.

# **Amendment to the Specification**

Please replace paragraph [0001] of the specification with replacement paragraph [0001]; and replace paragraph [0009] of the specification with replacement paragraph [0009].

Replacement paragraph [0001]:

This application is a continuation of and claims priority to U.S. Patent Application Serial No. 13/539,106, filed on June 29, 2012, which claims priority to U.S. Provisional Patent Application Serial No. 61/502,736, filed on June 29, 2011, all of which are incorporated herein by reference in their entirety.

Replacement paragraph [0009]:

[0009] In a second embodiment, a method for recovering zinc and manganese from a geothermal brine is provided. The method includes the steps of: providing a geothermal brine that includes manganese and zinc ions; selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine that includes manganese and zinc; removing the zinc from the substantially silica free brine by means of an ion exchange or other process; extracting manganese from the substantially silica free brine; oxidizing the manganese to produce a manganese dioxide precipitate; and recovering the manganese dioxide precipitate. In certain embodiments, the process can include the recycling of various solutions to dissolve manganese and zinc precipitates.

# **Amendment to the Claims**

Please amend Claim 20 as follows:

20. (Currently Amended) A method for recovering zinc and manganese from a geothermal brine, the method comprising the steps of:

providing a geothermal brine, said geothermal brine comprising manganese and zinc; selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine that includes manganese and zinc;

removing the zinc from the substantially silica free brine; extracting manganese from the substantially silica free brine; oxidizing the manganese to produce a manganese dioxide precipitate; and recovering the magnesium manganese dioxide precipitate.

# Remarks

Applicant has amended the specification for purposes of properly claiming priority for the presently pending application and to fix a typographical error. Claim 20 has been amended to fix a typographical error.

Should the Commissioner deem any fees as being due, including fees for extensions of time, the Commissioner is authorized to debit said fees from, or credit overpayments to, Bracewell & Giuliani LLP, USPTO Deposit Account Number 50-0259, Reference No. 076165.000070 (Continuation).

The Examiner is requested to contact the undersigned via telephone if a telephone conference would expedite or aid the prosecution and examination of this application.

Respectfully submitted,

Karthika Perumal

Reg. No. 69,296

BRACEWELL & GIULIANI LLP

P.O. Box 61389

Houston, Texas 77208-1389

Telephone: (713) 221-1342

Facsimile: (800) 404-3970

Date: July 16, 2013

Electronic Acknowledgement Receipt					
EFS ID:	16336006				
Application Number:	13943395				
International Application Number:					
Confirmation Number:	2954				
Title of Invention:	Selective Recovery of Manganese, Lead and Zinc				
First Named Inventor/Applicant Name:	Stephen Harrison				
Customer Number:	35979				
Filer:	Karthika Perumal/Debra Odum				
Filer Authorized By:	Karthika Perumal				
Attorney Docket Number:					
Receipt Date:	17-JUL-2013				
Filing Date:					
Time Stamp:	07:12:05				
Application Type:	Utility under 35 USC 111(a)				
Payment information:	1				

# **Payment information:**

Submitted with Payment	no
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# File Listing:

Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)
1	Preliminary Amendment	20130716PreliminaryAmendm	32795	no	Δ
•		ent.pdf	6ede1c32ec2ffd300fb78e0cbde8c5a2a20d 3133		

# **Warnings:**

Information: ESM / PETITIONER EX 1014\_187

This Acknowledgement Receipt evidences receipt on the noted date by the USPTO of the indicated documents, characterized by the applicant, and including page counts, where applicable. It serves as evidence of receipt similar to a Post Card, as described in MPEP 503.

#### New Applications Under 35 U.S.C. 111

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## National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

#### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

Under the Paperwork Reduction Act of 1995, no persons are required to respond to a collection of information unless it contains a valid OMB control number.

A	Application Data Sheet 37 CFR			Attorney Docket Number			076165.000070					
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OLegal Representative under 35 U.S.C. 117

CA

Middle Name

US Residency

State/Province

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**Family Name** 

Sharma

Country of Residence

Non US Residency

OParty of Interest under 35 U.S.C. 118

Active US Military Service

Suffix

City

Prefix

**Applicant 3** 

Applicant Authority Inventor

C.V. Krishnamohan

**Residence Information (Select One)** 

**Given Name** 

Milpitas

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Applicant Authority • Inventor														
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Subject Matter Utility														
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Approved for use through 01/31/2014. OMB 0651-0032
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Title of Invention   Sele	ective Recovery of Manga	nese, Lead and Zinc 				
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Request Early Pub	ication (Fee required a	t time of Request 37 CFR 1	.219)			
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Application Data Sheet 37 CFR 1.76		Attorney Docket Number	076165.000070
		Application Number	
Title of Invention	Selective Recovery of Manga	nese, Lead and Zinc	

# **Assignee Information:**

	Providing this information in the application data sheet does not substitute for compliance with any requirement of part 3 of Title 37 of the CFR to have an assignment recorded in the Office.							
Assignee 1								
If the Assignee is an Organization check here.								
Organization Name	Organization Name SIMBOL Inc.							
Mailing Address Info	rmation:							
Address 1	6920 Koll Center Parkwa	6920 Koll Center Parkway						
Address 2	Suite 123	Suite 123						
City	Pleasanton	State/Province	CA					
Country US		Postal Code	94566					
Phone Number		Fax Number						
Email Address								
Additional Assignee Data may be generated within this form by selecting the Add button.								

# Signature:

A signature of the applicant or representative is required in accordance with 37 CFR 1.33 and 10.18. Please see 37 CFR 1.4(d) for the form of the signature.								
Signature	kart	hika	Date (YYYY-MM-DD)	2013-07-16				
First Name	Karthika	Last Name	Perumal	Registration Number	69296			

This collection of information is required by 37 CFR 1.76. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.14. This collection is estimated to take 23 minutes to complete, including gathering, preparing, and submitting the completed application data sheet form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. **SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.** 

#### NONPROVISIONAL PATENT APPLICATION

Inventor:

Stephen Harrison (Canada)

Samaresh Mohanta (Canada)

C. V. Krishnamohan Sharma (USA)

Elizabeth Geler (USA)

Assignee:

Simbol Inc.

Atty Docket No.: 076165.000070

#### SELECTIVE RECOVERY OF MANGANESE, LEAD AND ZINC

[0001] This application claims priority to U.S. provisional patent application ser. no. 61/502,736, filed June 29, 2011, the disclosure of which is incorporated herein by reference in its entirety.

#### BACKGROUND OF THE INVENTION

#### Technical Field of the Invention

[0002] This invention generally relates to the field of selectively removing manganese and zinc from brines. More particularly, the invention relates to methods for the selective removal and recovery of manganese and zinc geothermal brines that include zinc and manganese, preferably without the simultaneous removal of other ions from the brines.

#### Description of the Prior Art

[0003] Geothermal brines are of particular interest for a variety of reasons. First, geothermal brines provide a source of power due to the fact that hot geothermal pools are stored at high pressure underground, which when released to atmospheric pressure, can provide a flashsteam. The flash-stream can be used, for example, to run a power plant. Additionally, geothermal brines contain useful elements, which can be recovered and utilized for secondary processes. With some geothermal waters and brines, binary processes can be used to heat a second fluid to provide steam for the generation of electricity without the flashing of the geothermal brine.

[0004] It is known that geothermal brines can include various metal ions, particularly alkali and alkaline earth metals, as well as silica, iron, lead, silver, zinc and manganese, in varying concentrations, depending upon the source of the brine. Recovery of these metals is potentially important to the chemical, pharmaceutical and electronics industries. Typically, the economic recovery of desired metals from natural brines, which may vary widely in composition, depends not only on the specific concentration of the desired metal, but also upon the concentrations of interfering ions, particularly silica, calcium and magnesium, because the presence of the interfering ions will increase recovery costs as additional steps must be taken to remove the interfering ions, before the desired metals are recovered.

[0005] One problem associated with geothermal brines when utilized for the production of electricity results from scaling and deposition of solids. Silica and other solids that are dissolved within the geothermal brine precipitate out during all stages of brine processing, particularly during the cooling of a geothermal brine, and may eventually result in fouling of the injection wells or processing equipment.

[0006] Although conventional processing of ores and brines currently employed can be used to recover a portion of the manganese and zinc present in geothermal brines, there still exists a need to develop economic methods that are selective for the removal and recovery of manganese and zinc from the brines at high yields and high purity.

#### SUMMARY OF THE INVENTION

[0007] Methods for the selective removal and recovery of manganese and zinc metals and compounds from geothermal brines are provided.

[0008] In a first embodiment, a method for recovering zinc and manganese ions from a geothermal brine is provided. The method includes the steps of: providing a geothermal brine that includes manganese and zinc ions; selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine; adjusting the pH of the substantially silica free brine to a pH suitable to form precipitates of zinc and manganese as hydroxides and oxides, such that precipitates of zinc and manganese are selectively formed and other metal precipitates are not formed; and separating the zinc and manganese precipitates from the brine. The method can further include the steps of dissolving the

precipitates of zinc and manganese to produce a zinc manganese solution; oxidizing the manganese to form a manganese precipitate and a zinc solution; separating the manganese precipitate from the zinc solution; and recovering the zinc by electrochemical means. The zinc solution can be contacted with hydrochloric acid to produce zinc chloride, or alternatively can be contacted with sulfuric acid to produce zinc sulfate.

[0009] In a second embodiment, a method for recovering zinc and manganese from a geothermal brine is provided. The method includes the steps of: providing a geothermal brine that includes manganese and zinc ions; selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine that includes manganese and zinc; removing the zinc from the substantially silica free brine by means of an ion exchange or other process; extracting manganese from the substantially silica free brine; oxidizing the manganese to produce a manganese dioxide precipitate; and recovering the magnesium dioxide precipitate. In certain embodiments, the process can include the recycling of various solutions to dissolve manganese and zinc precipitates.

[0010] In a third embodiment, a method for recovering zinc and manganese from a geothermal brine is provided. The method includes the steps of: providing a geothermal brine that includes manganese and zinc; selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine that includes manganese and zinc; extracting manganese and zinc from the substantially silica free brine to produce a manganese zinc solution; electrochemically removing manganese as manganese metal or manganese dioxide from the manganese zinc solution to produce a residual solution that includes zinc; and electrochemically removing zinc from the residual solution. In certain embodiments, the manganese dioxide and zinc can be recovered in a single electrochemical cell.

[0011] In a fourth embodiment of the invention, a method for recovering zinc and manganese from a geothermal brine is provided. The method includes the steps of: providing a geothermal brine that includes manganese and zinc; selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine that includes manganese and zinc; adjusting the pH of the substantially silica free brine to a pH suitable to form precipitates of zinc and manganese, such that precipitates of zinc and manganese are selectively formed and other metal precipitates are not formed; and separating the zinc and manganese precipitates from the brine. The fourth embodiment can also include the steps of

dissolving the precipitates of zinc and manganese to produce a zinc manganese solution; extracting zinc by solvent extraction; recovering and oxidizing the manganese to form a manganese dioxide precipitate and a zinc solution; separating the manganese precipitate from the zinc solution; and recovering the zinc by electrochemical means. In certain embodiments, the oxidation of the manganese is by chemical means. In alternate embodiments, the oxidation of manganese is by electrochemical means.

[0012] In a fifth embodiment of the invention, a method for recovering zinc and manganese from a geothermal brine is provided. The method includes the steps of: providing a geothermal brine that includes manganese and zinc; selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine that includes manganese and zinc; adjusting the pH of the substantially silica free brine to a pH suitable to form precipitates of zinc and manganese, such that precipitates of zinc and manganese are selectively formed and other metal precipitates are not formed; and separating the zinc and manganese precipitates from the brine. The fifth embodiment can further include the steps of dissolving the precipitates of zinc and manganese to produce a zinc manganese solution; extracting manganese by solvent extraction and then recovering manganese; recovering and oxidizing the dissolved manganese to form a manganese dioxide precipitate and a zinc solution; and recovering the zinc by electrochemical means. In certain embodiments, the recovery of manganese is by oxidation of the manganese is by chemical means. In alternate embodiments, the oxidation of manganese is by electrochemical means. In further embodiments, manganese is recovered by electrochemical reduction.

[0013] In a sixth embodiment of the invention, a method for recovering zinc and manganese from a geothermal brine is provided. The method includes the steps of: providing a geothermal brine that includes manganese and zinc; selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine that includes manganese and zinc; adjusting the pH of the substantially silica free brine to a pH suitable to form precipitates of zinc and manganese, such that precipitates of zinc and manganese are selectively formed and other metal precipitates are not formed; and separating the zinc and manganese precipitates from the brine. The sixth embodiment can further include the steps of dissolving the precipitates of zinc and manganese to produce a zinc manganese solution; extracting by way of a double solvent extraction both zinc and manganese in two separate streams; recovering and oxidizing the dissolved manganese to form a manganese dioxide precipitate and a zinc solution; and recovering the zinc by electrochemical means. In certain

embodiments, the oxidation of the manganese is by chemical means. In alternate embodiments, the oxidation of manganese is by electrochemical means.

[0014] In a seventh embodiment of the invention, a method for recovering zinc and manganese from a geothermal brine is provided. The method includes the steps of: providing a geothermal brine that includes manganese and zinc; selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine that includes manganese and zinc; adjusting the pH of the substantially silica free brine to a pH suitable to form precipitates of zinc and manganese as hydroxides and oxides, such that precipitates of zinc and manganese are selectively formed and other metal precipitates are not formed; and separating the zinc and manganese precipitates from the brine. The seventh embodiment can also include the steps of dissolving the precipitates of zinc and manganese to produce a zinc manganese solution; extracting by way of a double solvent extraction both zinc and manganese in two separate streams; recovering and reducing the dissolved manganese to form a manganese metal electrolytically and a zinc solution; and recovering the zinc by electrochemical means.

[0015] In an eighth embodiment of the invention, a method for recovering zinc and manganese from a geothermal brine is provided. The method includes the steps of: providing a geothermal brine that includes manganese and zinc; selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine that includes manganese and zinc; adjusting the pH of the substantially silica free brine to a pH suitable to form precipitates of zinc and manganese, such that precipitates of zinc and manganese are selectively formed and other metal precipitates are not formed; and separating the zinc and manganese precipitates from the brine. The eighth embodiment can also include the steps of dissolving the precipitates of zinc and manganese to produce a zinc manganese solution; extracting by way of a double solvent extraction both zinc and manganese in two separate streams; reacting the manganese stream to produce a manganese salt; and reacting the zinc stream to produce a zinc salt. In certain embodiments, the manganese salt is selected from manganese carbonate, manganese sulfate, and a manganese halide. In certain embodiments, the zinc salt is selected from zinc carbonate, zinc sulfate, or a zinc halide.

[0016] In a ninth embodiment of the invention, a method for recovering zinc and manganese from a geothermal brine is provided. The method includes the steps of: providing a geothermal brine that includes manganese and zinc; selectively removing silica and iron from

the geothermal brine to produce a substantially silica free brine that includes manganese and zinc; recovering zinc by contacting the substantially silica free brine with an ion exchange resin, and recovering manganese from the solution by electrolytically depositing manganese dioxide from the substantially silica free brine. Optionally, following removal of the zinc and manganese, the remaining brine solution can be recycled to the step for recovering zinc by contacting with the ion exchange resin. In an alternate embodiment, the ion exchange resin is a basic anionic exchange resin.

[0017] In a tenth embodiment, a method for recovering zinc and manganese from a geothermal brine is provided. The method includes the steps of: providing a geothermal brine that includes manganese and zinc; selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine that includes manganese and zinc. The method includes adjusting the pH of the substantially silica free brine to a pH suitable to form precipitates of zinc and manganese, such that precipitates of zinc and manganese are selectively formed and other metal precipitates are not formed, and separating the manganese and zinc precipitates from the brine. The method can further include the steps of dissolving the precipitates of zinc and manganese in an acid to produce a zinc and manganese containing acid solution with an extraction solvent to produce a first liquid phase that includes the extraction solvent and zinc and a second liquid phase that includes manganese. The first and second liquid phases are separated and then zinc is electrochemically recovered from the first liquid phase. The second liquid phase is reduced to form Mn<sup>2+</sup> and the second liquid phase is supplied to an electrochemical cell and manganese is recovered by electrochemical means.

[0018] In an eleventh embodiment, a method for recovering zinc and manganese from a geothermal brine is provided. The method includes the steps of: providing a geothermal brine that includes manganese and zinc; selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine that includes manganese and zinc. The method includes adjusting the pH of the substantially silica free brine to a pH suitable to form precipitates of zinc and manganese, such that precipitates of zinc and manganese are selectively formed and other metal precipitates are not formed, and separating the manganese and zinc precipitates from the brine. The method can further include the steps of dissolving the precipitates of zinc and manganese in ammonium sulfate to produce a zinc and manganese containing ammonium sulfate solution and extracting the zinc and manganese containing ammonium sulfate solution with an extraction solvent to produce a first liquid phase that includes the extraction solvent and zinc and a second liquid phase that includes

manganese and ammonium sulfate. The first and second liquid phases are separated and then zinc is electrochemically recovered from the first liquid phase. The second liquid phase is reduced to form Mn<sup>2+</sup> and the second liquid phase is supplied to an electrochemical cell and manganese is recovered by electrochemical means.

[0019] In a twelfth embodiment, a method for recovering zinc and manganese from a geothermal brine is provided. The method includes the steps of providing a geothermal brine that includes manganese and zinc. Selectively removing silica and iron from the geothermal brine to produce a substantially silica and iron free brine that includes manganese and zinc. Selectively precipitating manganese from the brine by the addition of a basic solution containing an amine or ammonia up to about pH 8 to 9. Producing a manganese precipitate with less than 0.15% Zn and 4% Ca and recovering the precipitate. In certain embodiments, the method can include the step of dissolving the precipitate in an acidic solution to recover the manganese salt. The method can also include the step of adding a reducing agent to ensure dissolution of at least about 95% of the manganese. In certain embodiments, the precipitate has less than about 4% of the Mg present.

#### BRIEF DESCRIPTION OF THE DRAWINGS

[0020] Figure 1 illustrates a process for the recovery of manganese and zinc from a geothermal brine according to one embodiment of the invention.

[0021] Figure 2 illustrates another process for the recovery of manganese and zinc from a geothermal brine according to another embodiment of the invention.

[0022] Figure 3 illustrates another process for the recovery of manganese and zinc from a geothermal brine according to another embodiment of the invention.

[0023] Figure 4 illustrates another process for the recovery of manganese and zinc from a geothermal brine according to another embodiment of the invention.

[0024] Figure 5 illustrates another process for the recovery of manganese and zinc from a geothermal brine according to another embodiment of the invention.

[0025] Figure 6 illustrates another process for the recovery of manganese and zinc from a geothermal brine according to another embodiment of the invention.

[0026] Figure 7 illustrates another process for the recovery of manganese and zinc from a geothermal brine according to another embodiment of the invention.

[0027] Figure 8 illustrates another process for the recovery of manganese and zinc from a geothermal brine according to another embodiment of the invention.

[0028] Figure 9 illustrates another process for the recovery of manganese and zinc from a geothermal brine according to another embodiment of the invention.

[0029] Figure 10 illustrates another process for the recovery of manganese and zinc from a geothermal brine according to another embodiment of the invention.

[0030] Figure 11 illustrates another process for the recovery of manganese and zinc from a geothermal brine according to another embodiment of the invention.

[0031] Figure 12 is an exemplary reaction scheme according to one embodiment of the present invention.

[0032] Figure 13 is an exemplary reaction scheme according to one embodiment of the present invention.

[0033] Figure 14 is a process diagram according to one embodiment of the present invention.

[0034] Figure 15 is a graph showing precipitation of manganese as a function of pH.

[0035] Figure 16 is a process diagram according to another aspect of the present invention.

[0036] Figure 17 is a process diagram according to another aspect of the present invention.

#### DETAILED DESCRIPTION OF THE INVENTION

[0037] Broadly, described herein are methods for the selective removal of manganese and zinc from solution. As used herein, the selective removal of manganese and zinc generally refers to methods to facilitate the removal of manganese and zinc from solutions that include manganese and zinc, such as geothermal brines, without the removal of other ions. Generally, in certain embodiments, the methods employ chemical means for the separation of manganese and zinc from brines. In certain embodiments, the methods may include physical means, as well as chemical means, for the separation of manganese and zinc from brines.

#### [0038] Definitions

[0039] As used herein, "brine" or "brine solution" refers to any aqueous solution that contains a substantial amount of dissolved metals, such as alkali and/or alkaline earth metal salt(s) in

water, wherein the concentration of salts can vary from trace amounts up to the point of saturation. As used herein, brine refers to both geothermal brines and waste or byproduct streams from industrial processes.

[0040] The term "amines" shall refer to primary, secondary, and tertiary amines, unless otherwise specified.

[0041] Generally, brines suitable for the methods described herein are aqueous solutions that may include alkali metal or alkaline earth chlorides, bromides, sulfates, hydroxides, nitrates, and the like, as well as natural brines. In certain brines, metals may be present. Exemplary elements present in the geothermal brines can include sodium, potassium, calcium, magnesium, lithium, strontium, barium, iron, boron, silica, manganese, chlorine, zinc, aluminum, antimony, chromium, cobalt, copper, lead, arsenic, mercury, molybdenum, nickel, silver, thallium, vanadium, and fluorine, although it is understood that other elements and compounds may also be present. Brines can be obtained from natural sources, such as, Chilean brines or Salton Sea brines, geothermal brines, sea water, mineral brines (e.g., lithium chloride or potassium chloride brines), alkali metal salt brines, and industrial brines, for example, industrial brines recovered from ore leaching, mineral dressing, and the like. The method is also equally applicable to artificially prepared brine or salt solutions. Geothermal brines, such as those found in the Salton Sea, can include many dissolved metal salts, including alkaline, alkaline earth, and transition metal salts. In one embodiment, the present invention provides a method for separating manganese, as well as zinc, lead, and silver, from brines, particularly geothermal brines. In certain embodiments, the present invention utilizes the coordination chemistry of the various metals to facilitate separation thereof.

[0042] As shown in Figure 1, process 100 of the present invention first removes silica and iron from the brine solution in an iron/silica removal step 110. In certain embodiments, the iron and silica removal step preferably removes only the iron and silica, while at the same time leaving all other metals and/or ions present in the brine undisturbed. The removal of silica is an important step as the presence of silica can interfere with subsequent processes for the recovery of various other metals. For example, silica frequently clogs pores in filtration media.

[0043] One preferred method for the selective removal of silica and iron includes contacting the solution with iron (III) hydroxide at a pH of between about 4.5 and 6, preferably between about 4.75 and 5.5, more preferably between about 4.9 and 5.3.

[0044] Typically, brine will have an iron (II) salt present naturally. In other embodiments, an iron (II) salt or iron (III) hydroxide can be added to the brine to achieve a certain concentration of iron (II) salt or iron (III) hydroxide relative to the silica or silicon containing compounds present in the brine. In certain embodiments, the molar ratio of the iron (II) salt or iron (III) hydroxide to silica is at least 1:1, preferably at least 4:1, more preferably at least 7:1 and even more preferably at least 10:1.

[0045] When the iron in the brine or silica containing solution is iron (II), for example iron (II) chloride, an oxidant is added to oxidize iron (II) salt to iron (III) hydroxide. The preferred oxidant is air. Thus, in one preferred embodiment, the iron (II) salt present in the brine can be oxidized to iron (III) by sparging the reaction vessel with air. While it is understood that many different oxidants can be used for the oxidation of iron (II) to iron (III), the use of oxygen or air as the oxidant in the pH range of between 4 and 7 is selective for the oxidation of the iron (II) salt to iron (III) hydroxide, and generally does not result in the precipitation or oxidation of other elements or compounds that are present in the brine. Control of the pH of the solution can be achieved with the addition of base (e.g., calcium hydroxide, calcium oxide or the like). As noted previously, it is preferred that the pH is maintained between 4.5 and 6.

[0046] Other exemplary oxidants can include hypohalite compounds, such as hypochlorite, hydrogen peroxide (in the presence of an acid), air, halogens, chlorine dioxide, chlorite, chlorate, perchlorate and other analogous halogen compounds, permanganate salts, chromium compounds, such as chromic and dichromic acids, chromium trioxide, pyridinium chlorochromate (PCC), chromate and dichromate compounds, sulfoxides, persulfuric acid, nitric acid, ozone, and the like. It will be recognized by those skilled in the art that iron (III) hydroxide may also have a significant affinity for arsenic (III) and (V) oxyanions, and these anions, if present in the brine, may be co-deposited with the silica on the iron (III) hydroxide.

[0047] In another embodiment, iron (III) hydroxide can be produced by adding a solution of iron (III) chloride to the brine, which on contact with the more neutral brine solution, will precipitate as iron (III) hydroxide. The brine may require neutralization, such as through the addition of base to initiate precipitation of the iron (III) hydroxide.

[0048] The iron (III) hydroxide contacts the silica present in the brine and forms a precipitate. Without being bound to any specific theory, it is believed that the silica or silicon containing compound attaches to the iron (III) hydroxide. In certain embodiments, the ratio of iron (III) to silica is at least about 1:1, more preferably at least about 4:1. The reaction of the iron (III) hydroxide with silica is capable of removing at least about 80% of the silica present, preferably at least about 90%, and more preferably at least about 95%, and typically depends upon the amount of iron (III) hydroxide present in the solution.

[0049] In certain embodiments, the iron (II) salt containing solution can be sparged with air for a period of at least 15 min., preferably at least 30 min., followed by the addition of a base, such as calcium oxide, calcium hydroxide, sodium hydroxide, or the like, to achieve the desired pH for the solution.

[0050] After silica and iron removal step 110, in precipitation step 120, a base (e.g., calcium oxide, calcium hydroxide or the like) is added to the brine to adjust or maintain a pH of the brine at greater than at least about 6, alternatively between about 6 and 8.5, alternatively between about 6.5 and 8, alternatively between about 6.5 and 7.5. In alternate embodiments, the pH is maintained at about 7. In certain embodiments, the pH is maintained at less than about 9. The base may be in solution or slurry form. Furthermore, the solution is exposed to an oxygen source and manganese and zinc precipitates are formed. In certain embodiments, depending upon the pH of the solution, a lead precipitate may also be formed. To achieve oxidation of the manganese, air is preferably supplied to the solution by sparging or bubbling. Other oxidants suitable for the oxidation of the manganese can include hypohalites, hydrogen peroxide, and ozone. In certain embodiments wherein sodium hydroxide or ammonia is utilized as the base, reduced amounts of calcium and magnesium impurities are present in the precipitates. For example, in certain embodiments, magnesium concentration in the zinc and manganese precipitates is less than about 200 ppm, alternatively less than about 150 ppm, alternatively less than about 100 ppm. Similarly, the concentration of calcium in the zinc and manganese precipitates is less than about 450 ppm, alternatively less than about 400 ppm, alternatively less than about 350 ppm, alternatively less than about 300 ppm, alternatively less than about 250 ppm, alternatively less than about 200 ppm.

[0051] The solids in the brine and base, which solids can include at least manganese and zinc, are separated from the remainder of the mixture, which retains the majority of ions present in the brine. Separation of the solids can be done by conventional filtration means

and can optionally include centrifugation or other known techniques for concentration the solids. In certain embodiments, the remaining brine solution from which the manganese and zinc have been removed can then be reinjected into the geothermal well from which the brine was originally removed.

[0052] In certain embodiments, the preferential precipitation of zinc and manganese from the brine solution results in the precipitation of at least about 80% of the zinc and manganese present in solution, alternatively at least about 85%, alternatively at least about 90%, alternatively at least about 95%. Similarly, the preferential precipitation results in the precipitation of no more than about 10% of other ions present in the brine solution, alternatively less than about 8%, alternatively less than about 5%, alternatively less than about 3%.

[0053] The manganese and zinc solids that are separated from the remaining brine solution can then be dissolved in an acid solution in step 130. Preferred acids include strong mineral acids, such as hydrochloric acid, sulfuric acid, methanesulfonic acid, and the like. The use of hydrochloric acid results in the production of zinc chloride and manganese chloride. Similarly, the use of sulfuric acid results in the production of zinc sulfate and manganese sulfate. In certain embodiments, lead and/or calcium precipitates may be formed during the precipitation of the manganese and zinc. In these embodiments, the selected acid is preferably sulfuric acid, as sulfuric acid is selective for manganese and zinc precipitates, and does not dissolve the lead and/or calcium precipitates that may be present. The acid is preferably added to the solids in greater than approximately a 1:1 molar ratio to the solids. In certain embodiments, it may be beneficial to minimize the amount of excess acid that is utilized for dissolving the manganese and zinc precipitates, for ease of performance of downstream processes, as well as for economic and environmental considerations. In certain embodiments, the solids and acid are mixed to ensure complete dissolution of the solids.

[0054] The acid and dissolved metal solution is then filtered to remove remaining solids, if any, and the solution may then be purified in optional purification step 140 to remove trace metals, which may be present in the acidified solution. It is believed that metals, such as copper, cadmium, nickel, antimony and/or cobalt, as well as other metals or ions, may be present in trace amounts in the acid and dissolved metal solution. These trace metals may interfere with the subsequent separation of manganese and zinc. Purification of the acid and dissolved metal solution can be achieved by known means, such as ion exchange or by

treatment with zinc dust. Zinc dust operates by first displacing other more noble metals from solution and allowing them to precipitate on undissolved zinc dust. For example, copper ions present in the solution will precipitate as copper metal or will deposit on undissolved zinc dust.

[0055] Manganese and zinc can be extracted from the acid and dissolved metal solution using solvent extraction techniques. Suitable solvents for the extraction of manganese and zinc include phosphines, phosphoric acids, and phosphinic acids, such as the following: di(2-ethylhexyl)phosphoric acid (DEHPA) in kerosene or Cyanex® 272 (bis(2,4,4-trimethylpentyl)phosphinic acid); Ionquest 290 (available form Rhodia Inc.) in aliphatic kerosene or the highly branched carboxylic acid extractant (versatic 10)(10–decyl-4-pyridinecarboxylate). In certain embodiments, DEHPA is a suitable extraction solvent, particularly in embodiments where iron has been previously removed. In certain embodiments, manganese and zinc can be extracted with organic amines, such as 1,4-diazabicyclo[2.2.2]octane (DABCO), 2,2'-bipyridyl and piperazine. In certain embodiments, zinc can be preferentially extracted with a functionalized amine, such as polyvinyl pyrrolidone.

[0056] Other exemplary solvents that may be used for the extraction of zinc are discussed in U.S. Pat. No. 5,135,652, the disclosure of which is herein incorporated by reference in its entirety. These exemplary solvents include mono-2-ethylhexylphosphoric acid (M2EHPA), di-2-ethylhexylphosphoric acid (D2EHPA), and mixtures thereof (EHPA). Other exemplary solvents include bis-2,4,4-trimethylpentylmonothiophosphinic acid (Cyanex® 302) and bis-2,4,4-trimethylpentyldithiophosphinic acid (Cyanex® 301). In certain embodiments, the extractant includes both phosphoric acid and phosphinic acid. In certain embodiments, the ratio of phosphoric acid to phosphinic acid is greater than about 1:1, preferably between about 1:1 and 1:6. In certain embodiments, the extractant can be diluted with a hydrocarbon solvent, preferably a dearomatized aliphatic hydrocarbon. Exemplary diluents include Exxsol<sup>TM</sup> D80.

[0057] The pH during the extraction is maintained at less than about 7, alternatively between about 1 and 5, alternatively between about 1 and 3, alternatively in the range of about 1.5 to 3.5, alternatively between about 2 and 4.

[0058] Other solvents suitable for the extraction of zinc from brine solutions are described in "Recovery of Zinc(II) from Acidic Sulfate Solutions. Simulation of Counter-Current

Extraction Stripping Process", Gotfryd, L. and Szymanowski, J.; Physicochemical Problems of Mineral Processing, vol. 38 (2004), pp. 113-120; "New Developments in the Boleo Copper-Cobalt-Zinc-Manganese Project", al.; Dreisinger, available at http://bajamining.com/ resources/Reports/alta paper 2006 boleo final.pdf; "Zinc Solvent Extraction in the Process Industries", Cole, P. and Sole, K.; Mineral Processing and Extractive Metallurgy Review, vol. 24, no. 2 (2003), , pp. 91-137; "Solvent extraction of zinc(II) and manganese(II) with 5,10,15,20-tetraphenyl-21H,23H-porphine(TPP) through the metal exchange reaction of lead(II)-TPP", Kawai, T., Fujiyoshi, R., and Sawamura, S.; Solvent Extr. Res. Dev. Japan, vol. 7 (2000), pp. 36-43, "Solvent Extraction of Zinc from Strong Hydrochloric Acid Solution with Alamine 336", Lee, M. and Nam, S.; Bull. Korean Chem. Soc., vol. 30, no. 7 (2009), pp. 1526-1530, the disclosures of which are incorporated herein by reference.

[0059] Manganese can be isolated by electrolysis or, in step 150, by oxidation to produce manganese dioxide, or by precipitation as a carbonate by reaction with sodium carbonate. In certain preferred embodiments, manganese can be selectively isolated from zinc as manganese dioxide by electrolysis in a sulfate solution, at an anode made of metals, such as titanium or carbon. Alternatively, selective oxidation of manganese to manganese dioxide can be achieved utilizing an oxidant, such as chlorine, hydrogen peroxide, or the like to provide solid manganese dioxide and zinc containing solution. In step 160, precipitated manganese dioxide is separated from the zinc containing solution by known means, such as filtration, centrifugation, or a like process.

[0060] In an alternate embodiments, manganese dioxide can be generated at the anode of a divided electrochemical cell by the oxidation of manganese (II) and manganese (III) to generate a manganese dioxide deposited on the surface of the electrode. After the solution is passed through anode compartment, it is then fed to the cathode compartment where zinc metal is electrodeposited. The current density ranges from between about 50 to about 500 A/m². The separator, such as an ion exchange membrane or a porous material that allows the passage of liquids, positioned between the anode and cathode assists in preventing deposition of manganese dioxide on the zinc metal. In certain embodiments, the separator can include a series of baffles. In certain embodiments, it may be advantageous to remove solid manganese dioxide from the electrolytic stream formed in the anode compartment that may be lost from the surface of the anode, such as by filtration, prior to supplying to the cathode compartment. Production of manganese dioxide by electrochemical means and the recovery of zinc metal

by electrowinning preferably includes a conductive solution, such as sulfate, chloride, methanesulfonate, or the like, for improved efficiency. In certain embodiments, it is preferred that the electrochemical cell includes a small amount of free acid in the solution. In alternate embodiments, the electrochemical cell can be operated at a pH ranging from about 0 to 2. Following recovery of the manganese and zinc, the respective solutions can be recycled to the solvent extraction step. Alternatively, the respective solutions can be recycled to the acid solution.

[0061] The zinc containing solution can then be optionally purified in step 170 and then supplied to an electrochemical cell for electrochemical recovery in step 180 by electrowinning (also known as electroextraction). Electrowinning utilizes an electrochemical cell wherein a current is passed from an inert anode, such as lead dioxide, iridium dioxide coated titanium, or other stable substrate, through the zinc containing solution, leading to deposition of the zinc on the cathode. The base cathode can be aluminum, although other metals, such as steel, stainless steel, and titanium, can also be used. The cathode material is selected based upon chemical stability, electrical conductivity, and the ease of removal of zinc from substrate.

[0062] Alternatively, in the process illustrated by Figure 1, the steps for the isolation and recovery of manganese and zinc can be reversed, i.e., the zinc can be separated and isolated from a solution that includes zinc and manganese by electrowinning, followed by the isolation of manganese by either electrowinning or oxidation of the manganese to produce manganese dioxide.

[0063] Optionally, the process may include a step for the recovery of lithium from the geothermal brine. Methods for the recovery are known in the art, such as is described in U.S. Pat. Nos. 4,116,856; 4,116,858; 4,159,311; 4,221,767; 4,291,001; 4,347,327; 4,348,295; 4,348,296; 4,348,297; 4,376,100; 4,430,311; 4,461,714; 4,472,362; 4,540,509; 4,727,167; 5,389,349; 5,599,516; 6,017,500; 6,280,693; and 6,555,078, each of which is incorporated herein by reference in their entirety. Alternatively, methods can be employed utilizing a lithium aluminate intercalate/gibbsite composite material, a resin based lithium aluminate intercalate, and/or a granulated lithium aluminate intercalate. The gibbsite composite is a lithium aluminate intercalate that is grown onto an aluminum trihidrate core. The resin-based lithium aluminate intercalate is formed within the pores of a macroreticular ion exchange resin. The granulated lithium aluminate intercalate consists of fine-grained lithium aluminate

intercalate produced by the incorporation of a small amount of inorganic polymer. The process of contacting the lithium aluminate intercalate material with the geothermal brine is typically carried out in a column that includes the extraction material. The geothermal brine is flowed into the column and lithium ions are captured on the extraction material, while the water and other ions pass through the column. After the column is saturated, the captured lithium is removed by flowing water having a small amount of lithium chloride present through the column. In preferred embodiments, multiple columns are employed for the capture of the lithium.

[0064] In another embodiment of the present invention, in process 200 provided in Figure 2, iron and silica are first removed from the geothermal brine in step 210. Methods for the removal of silica and iron include those methods previously described with respect to Figure 1, and preferably include oxidation of the iron from iron (II) to iron (III), and the control of the pH of the solution with the addition of a base. Preferably, the iron is oxidized with air, and the pH is controlled by the addition of a base, such as calcium oxide or calcium hydroxide, or like compound.

[0065] The brine solution, now having a reduced concentration of silica and iron relative to the initial brine feed, can be supplied to zinc removal process 220 that can include an ion exchange process, for example a basic anionic ion exchange resin like the chloride of a quaternary amine divinylbenzene/stryrene copolymer, or the chloride of trimethylamine functionalized chloromethylated copolymer of styrene and divinylbenzene, such as is described in U.S. Pat. Nos. 6,458,184, which is incorporated herein by reference in its entirety. Zinc separated by ion exchange, existing as zinc chloride or a zinc chloride anions, can then be converted into a saleable zinc product, such as zinc metal, zinc oxide, or zinc sulfate. In certain embodiments, the remaining brine solution from which the manganese and zinc have been removed can then be reinjected into the geothermal well from which the brine was originally removed.

[0066] The remaining solution, which includes manganese, can then optionally be supplied to purification step 230 and purified by ion exchange, solvent extraction, or like process, and the manganese containing phase can be provided to oxidation step 240, such as an electrochemical cell or chemical oxidation process, as described with respect to Figure 1, to facilitate the recovery of manganese dioxide. Purified manganese can be collected in step 250 by filtration. As shown with the dashed line, the liquid phase from step 250 can

optionally be recycled to manganese extraction step 230. As previously discussed, following recovery of the manganese and zinc, the respective solutions can be recycled to the solvent extraction step. Alternatively, the respective solutions can be recycled to the acid solution.

[0067] As noted with respect to Figure 1, in process 200 the lithium can optionally be removed from the brine solution at any point during the process by the means discussed above.

[0068] In yet another embodiment, in process 300 shown in Figure 3, a method for the separation and isolation of manganese and zinc from a brine is provided. As noted with respect to Figures 1 and 2, the first step of the process includes the removal of iron and silica from the brine solution in step 310. Preferably, as discussed above, the iron is oxidized and base is added to the solution to control the pH. Preferably, iron is oxidized with air, and the base is calcium oxide, calcium hydroxide, or a like compound.

[0069] Following removal of a major portion of the silica and iron, the manganese and zinc can be removed by liquid-liquid extraction step 320. Exemplary liquids suitable for the extraction of manganese and zinc are described in U.S. Pat. No. 6,458,184 and U.S. Pub. Pat. App. No. 20030226761, the disclosures of which are incorporated herein by reference in their entirety. The solvents can include, for example, water-immiscible cationic organic solvents, such as di-(2-ethylhexyl) phosphoric acid (D2EHPA), and other similar solvents, as known in the art. In certain embodiments, the remaining brine solution from extraction step 320, from which the manganese and zinc have been removed, can then be reinjected into the geothermal well from which the brine was originally removed.

[0070] Following the liquid-liquid extraction step, the extraction solution that includes the manganese and zinc can be provided to one or more purification steps 330. Purification steps 330 preferably operable to remove calcium and other divalent cations, as well as some metals, such as copper, cadmium, cobalt, molybdenum, and nickel, although the purification steps are not limited to these metals.

[0071] Following purification step 330, the manganese and zinc can be isolated in steps 340 and 350, respectively. Specifically, as previously discussed, manganese dioxide and zinc can each separately be produced from solution by electrowinning. In one embodiment, zinc is recovered first, followed by manganese. In an alternate embodiment, manganese is recovered first, followed by zinc. In certain embodiments, the pH is maintained at less than about 3.5

during the electrowinning process. In alternate embodiments, the temperature is maintained at less than about 60°C during the electrowinning process. In certain embodiments, the pH of the solution supplied to manganese electrochemical recovery step 340 is about 5, and the pH of the solution exiting the electrochemical cell is about 1. The pH of the solution supplied to zinc electrochemical recovery step 350 is about 1.

[0072] In an alternate embodiment, the solution from purification step 330 can be supplied to a single electrochemical recovery step 360 wherein zinc and manganese can be deposited simultaneously as zinc oxide and manganese dioxide.

[0073] As previously discussed, following recovery of the manganese and zinc, the respective solutions can be recycled to either the solvent extraction step or to the acid solution. In certain embodiments, as shown by the dashed line, the solution from zinc electrochemical recovery step 350 can be recycled to purification step 330.

[0074] In another embodiment, as provided in Figure 4, process 400 for the recovery of manganese or zinc from a geothermal brine is provided. As previously discussed, with respect to Figure 1, first step 410 of process 400 includes the removal of iron and silica from the brine solution. In certain embodiments, the iron is oxidized and base is added to control the pH of the solution. In certain embodiments, iron is oxidized with air and the base is calcium oxide or calcium hydroxide, or like compound.

[0075] Following the removal of the iron and silica, in precipitation step 420, additional base, such as lime, slaked lime, limestone, sodium hydroxide, and the like, is added to achieve a pH of between about 6 and 9, preferably up to about 8 when sparged with air, or up to about 9 when it is not sparged with air, to facilitate the precipitation of manganese and zinc. The manganese and zinc precipitates are collected by known means and dissolved in an acid solution in step 430, as previously discussed herein. In certain embodiments, the remaining brine solution from extraction step 420, from which the manganese and zinc have been removed, can then be reinjected into the geothermal well from which the brine was originally removed.

[0076] Optionally, the acid solution, which includes the manganese and zinc, can be purified in step 440, to remove trace metal impurities, such as heavy metals, for example, cobalt, copper, cadmium, nickel, and the like. The acid solution is then extracted in step 450 to recover zinc, as previously provided. Thus, following extraction, a first solution, which

includes zinc and the extraction solvent, is produced and a second solution, which includes manganese, is produced.

[0077] The zinc can then be recovered by electrochemical means in step 460, such as by electrowinning or a like process, as previously discussed. Manganese can be recovered by first oxidizing the manganese in step 470 to produce manganese dioxide, as previously discussed, which can then be recovered electrochemically in step 480 by known means. As previously discussed, as shown by the dashed lines, following recovery of the manganese and zinc, the solutions from steps 460 and 480 can be recycled to solvent extraction step 450 or to the acid solution of dissolution step 430, respectively.

[0078] In another embodiment, as provided in Figure 5, process 500 for the recovery of manganese or zinc from a geothermal brine is provided. As previously discussed, first step 510 of the process includes the removal of iron and silica from the brine solution. In certain embodiments, the iron is oxidized and base is added to control the pH of the solution. In certain embodiments, iron is oxidized with air and the base is calcium oxide or calcium hydroxide.

[0079] Following the removal of the iron and silica, in precipitation step 520, additional base to adjust the pH to at least about 6 is added to facilitate the precipitation of manganese and zinc. The manganese and zinc precipitates are collected by known means, such as by filtration, centrifugation, or a like process, and dissolved in an acid solution in step 530, as previously discussed herein. Optionally, the acid solution, which includes the manganese and zinc, can be purified. In certain embodiments, the remaining brine solution from extraction step 520, from which the manganese and zinc have been removed, can then be reinjected into the geothermal well from which the brine was originally removed.

[0080] The acid solution from step 530 can then be extracted in extraction step 540 to recover manganese and zinc, as previously provided, to provide an extract solution that includes both manganese and zinc. The manganese in the extract solution can be oxidized in step 550 to produce manganese dioxide, which can then be separated by filtration or other known means in step 560. Zinc remaining in the extract solution can then be recovered in step 570 by electrochemical means, such as electrowinning or a like process. In certain embodiments, as shown by the dashed line, the solution from zinc electrochemical recovery step 570 can be recycled to the dissolution step 530.

[0081] In another embodiment, as provided in Figure 6, process 600 for the recovery of manganese or zinc from a geothermal brine is provided. As previously discussed, first step 610 of the process includes the removal of iron and silica from the brine solution. In certain embodiments, the iron is oxidized and base is added to control the pH of the solution, preferably to at least about 5 and up to about 6. In certain embodiments, iron is oxidized with air and the base is calcium oxide or calcium hydroxide.

[0082] Following the removal of the iron and silica, in precipitation step 620, additional base is added to achieve a pH of at least about 6 to cause the precipitation of manganese and zinc. The manganese and zinc precipitates are collected by known means in step 630 and dissolved in an acid solution in step 640, as previously discussed herein. Optionally, the acid solution, which includes the manganese and zinc, can be purified. In certain embodiments, the remaining brine solution from extraction step 630, from which the manganese and zinc have been removed, can then be reinjected into the geothermal well from which the brine was originally removed.

[0083] The acid solution from step 640 is then subjected to a double extraction step 650, wherein the acid solution is contacted with two separate extraction solvents to recover two separate streams, wherein recovery step 670 recovers a first stream that includes manganese and a second stream is recovered includes zinc. Appropriate extraction solvents for the extraction of manganese and zinc have been previously discussed. The manganese in the first stream can be oxidized in oxidation step 680 to produce manganese dioxide, which is then separated by filtration or other known means. The zinc in the second stream can be recovered by electrochemical means, such as electrowinning, in step 660. As previously discussed, as shown by the dashed lines, following recovery of the zinc and manganese in steps 660 and 680, the respective solutions can be recycled to solvent extraction step 650 or to manganese stream recovery step 670, respectively.

[0084] In another embodiment, as provided in Figure 7, process 700 for the recovery of manganese or zinc from a geothermal brine is provided. As previously discussed, a first step of the process includes the removal of iron and silica from the brine solution. In certain embodiments, the iron is oxidized and base is added to control the pH of the solution to about 5 and 6. In certain embodiments, iron is oxidized with air and the base is calcium oxide or calcium hydroxide.

[0085] Following the removal of the iron and silica, in precipitation step 720, additional base is added to achieve a pH of between about 6 and 9, preferably up to about 8 when sparged with air, or up to about 9 when it is not sparged with air, to facilitate the precipitation of manganese and zinc. The manganese and zinc precipitates are separated from a liquid phase in step 730, collected by known means, such as filtration, centrifugation or a like process, and dissolved in an acid solution in step 740, as previously discussed herein. Optionally, the acid solution, which includes the manganese and zinc, can be purified. In certain embodiments, the remaining brine solution from extraction step 730, from which the manganese and zinc have been removed, can then be reinjected into the geothermal well from which the brine was originally removed.

[0086] The acid solution from step 740 is then subjected to double extraction step 750, wherein the acid solution is contacted with two separate extraction solvents to recover two separate streams, wherein the first stream recovered in step 770 includes manganese, and wherein the second stream includes zinc. Appropriate extraction solvents for the extraction of manganese and zinc have been previously discussed. The manganese in the first stream can be electrolytically reduced in step 780, as is known in the art, to produce manganese metal. The zinc in the second stream can be recovered by electrochemical means in step 760, such as by electrowinning or a like process. As previously discussed, as shown by the dashed line, following recovery of the zinc and manganese in steps 760 and 780, the respective solutions can be recycled to the solvent extraction step 750 or to manganese stream recovery step 770, respectively. In certain embodiments, as shown by the dashed line, at least a portion of the non-extraction solvent solution from extraction step 750 can be recycled to dissolution step 740.

[0087] In another embodiment, as provided in Figure 8, process 800 for the recovery of manganese or zinc from a geothermal brine is provided. As previously discussed, a first step of the process includes the removal of iron and silica from the brine solution. In certain embodiments, the iron is oxidized and base is added to control the pH of the solution to between about 4.5 and 6, preferably between about 4.75 and 5.5. In certain embodiments, iron is oxidized with air and the base is calcium oxide or calcium hydroxide.

[0088] Following the removal of the iron and silica, in precipitation step 820, additional base is added to achieve a pH of between about 6 and 9, preferably up to about 8 when sparged with air, or up to about 9 when it is not sparged with air, to facilitate the precipitation of

manganese and zinc. The manganese and zinc precipitates are separated and collected by known means in step 830, such as by filtration, centrifugation or a like process, and dissolved in an acid solution, as previously discussed herein. Optionally, the acid solution that includes the manganese and zinc can be purified. In certain embodiments, the remaining brine solution from extraction step 830, from which the manganese and zinc have been removed, can then be reinjected into the geothermal well from which the brine was originally removed.

[0089] The acid solution is then subjected to a double extraction in step 840, wherein the acid solution is contacted with two separate extraction solvents to recover two separate streams, wherein the first stream includes manganese and the second stream includes zinc. Appropriate extraction solvents for the extraction of manganese and zinc have been previously discussed. The manganese in the first stream can be reacted in step 850 with an acid, such as sulfuric acid, hydrochloric acid, hydrobromic acid, or a like acid to produce a manganese salt, which can then be recovered by precipitation in step 860. The zinc in the second stream can be recovered by electrochemical means, such as electrowinning or like means, or may also be reacted in step 870 with an acid, such as sulfuric acid, hydrochloric acid, hydrobromic acid, or a like acid to produce a salt solution and recovered in step 880 by precipitation, evaporative crystallization, spray drying, or a like process. As previously discussed, as shown by the dashed line, following recovery of the manganese and zinc salts in steps 860 and 880, the respective solutions can be recycled to solvent extraction step 840.

[0090] In another embodiment, as shown in Figure 9, process 900 for the recovery of manganese or zinc from a geothermal brine is provided. As previously discussed, first step 910 of the process includes the removal of iron and silica from the brine solution. In certain embodiments, the iron is oxidized and base is added to control the pH of the solution to between about 4.5 and 6, preferably between about 4.75 and 5.5 In certain embodiments, iron is oxidized with air and the base is calcium oxide or calcium hydroxide.

[0091] Following the removal of the iron and silica, in precipitation step 920 additional base is added to achieve a pH of between about 6 and 9, preferably up to about 8 when sparged with air, or up to about 9 when it is not sparged with air, to facilitate the precipitation of manganese and zinc. The manganese and zinc precipitates are separated in step 930, collected by known means and dissolved in an acid solution in step 940, as previously discussed herein. Optionally, the acid solution, which includes the manganese and zinc, can be purified, as previously discussed. In certain embodiments, the remaining brine solution

from extraction step 930, from which the manganese and zinc have been removed, can then be reinjected into the geothermal well from which the brine was originally removed.

[0092] The acid solution is contacted with an ion exchange resin in step 950, preferably a basic anionic exchange resin, to remove zinc from the solution. In step 960, manganese can be recovered from the solution by electrolytically depositing manganese dioxide from the substantially silica free brine, such as by electrowinning or a like process. In step 970, zinc can then be recovered from the ion exchange resin by known means, and can be converted electrochemically in step 980 to zinc, and the zinc can then be converted to zinc oxide by known means. Optionally, as shown by the dashed line, following removal of the manganese in step 960, the remaining solution can be recycled to dissolution step 940. Similarly, as shown by the dashed line, following zinc recovery step 970, the remaining brine solution can be recycled to ion exchange resin contacting step 950.

[0093] In certain embodiments, an aqueous chloride solution is employed to wash zinc from the ion exchange resin, preferably having a chloride concentration of between about 0.5 and 5%. Optionally, multiple ion exchange resins can be employed. Optionally, at least a portion of a zinc solution produced by washing the ion exchange resin can be recycled to a prior stage of the process. In certain embodiments, the zinc solution produced by washing the ion exchange resin can be extracted with a solvent, wherein the solvent advantageously extracts zinc from the solution. Exemplary extraction solvents have been previously discussed, and can include D2EHPA, polyvinyl pyrrolidone, or the like. Following removal of zinc from the ion exchange resin, a zinc-rich solution is obtained and zinc can then be recovered electrochemically from the zinc-rich solution.

[0094] Referring now to Figure 10, in another aspect, a tenth process for the recovery of manganese and/or zinc from a geothermal brine is provided. First step 1010 of the process includes the removal of iron and silica from the brine solution, as previously described herein. In certain embodiments, the iron is oxidized and base is added to control the pH of the solution to between about 4.5 and 6, preferably between about 4.75 and 5.5. In certain embodiments, iron is oxidized with air and the base is calcium oxide or calcium hydroxide.

[0095] Following the removal of the iron and silica, in precipitation step 1020, additional base is added to cause the precipitation of manganese and zinc. The manganese and zinc precipitates are collected by known means and, in step 1030, dissolved in an acid solution, as previously discussed herein. Optionally, the acid solution that includes the manganese and

zinc can be purified. In certain embodiments, the remaining brine solution from extraction step 1020, from which the manganese and zinc have been removed, can then be reinjected into the geothermal well from which the brine was originally removed.

[0096] The acid solution is filtered in step 1040 to produce a manganese containing solution and zinc precipitates. Zinc precipitates are electrochemically converted to zinc metal in step 1050. The solution is passed to zinc solvent extraction step 1060 to recovery remaining zinc. The manganese containing solution from the filtration step is provided to a reduction step 1070 wherein the manganese containing solution is contacted with a reducing agent, such as SO<sub>2</sub>. In step 1080, the reduced manganese can be recovered from the solution by electrolytically depositing manganese dioxide, such as by electrowinning.

[0097] Optionally, as shown by the dashed line, following recovery of zinc in electrochemical recovery step 1060, a sulfuric acid-rich solution can be recycled to zinc extraction step 1050. Similarly, as shown by the dashed line, following the electrochemical recovery of manganese in step 1080, the remaining brine solution can be recycled to either precipitation step 1020 or dissolution step 1030.

[0098] Referring now to Figure 11, in another aspect, a tenth process for the recovery of manganese and/or zinc from a geothermal brine is provided. First step 1110 of the process includes the removal of iron and silica from the brine solution, as previously described herein. In certain embodiments, the iron is oxidized and base is added to control the pH of the solution to between about 4.5 and 6, preferably between about 4.75 and 5.5. In certain embodiments, iron is oxidized with air and the base is calcium oxide or calcium hydroxide.

[0099] Following the removal of the iron and silica, in precipitation step 1120, additional base is added to cause the precipitation of manganese and zinc. The manganese and zinc precipitates are collected by known means and, in step 1130, dissolved in an ammonium sulfate solution. Optionally, the ammonium sulfate solution that includes the manganese and zinc can be purified. In certain embodiments, the remaining brine solution from extraction step 1020, from which the manganese and zinc have been removed, can then be reinjected into the geothermal well from which the brine was originally removed.

[00100] The ammonium sulfate solution is filtered in step 1040 to produce a manganese containing solution and zinc precipitates. Zinc precipitates are electrochemically converted to zinc metal in step 1050. The solution is passed to zinc solvent extraction step

1060 to recovery remaining zinc. The manganese containing solution from the filtration step is provided to a reduction step 1070 wherein the manganese containing solution is contacted with a reducing agent, such as SO<sub>2</sub>. In step 1080, the reduced manganese can be recovered from the solution by electrolytically depositing manganese dioxide, such as by electrowinning.

[00101] Optionally, as shown by the dashed line, following recovery of zinc in electrochemical recovery step 1060, a sulfuric acid-rich solution can be recycled to zinc extraction step 1050. Similarly, as shown by the dashed line, following the electrochemical recovery of manganese in step 1080, the remaining brine solution can be recycled to either precipitation step 1020 or dissolution step 1030.

[00102] In certain embodiments of the present invention, as described herein, solid zinc oxide produced electrochemically or by ion exchange extraction can be dissolved in various acids for the production of zinc compounds. For example, in one embodiment, zinc oxide can be added to hydrochloric acid to form solid zinc chloride. The solid zinc chloride can then be separated by filtration. In certain embodiments, the zinc chloride can be isolated from solution by removing the liquid by evaporation, spray drying, or other known methods. In an alternate embodiment, zinc oxide can be added to hydrobromic acid to form zinc bromide. Alternatively, zinc oxide can be added to sulfuric acid to form zinc sulfate. Alternatively, zinc oxide can be added to methylsulfonic acid to form zinc methylsulfonate. In certain embodiments, to facilitate precipitation of the various zinc compounds, a portion of the solution can be evaporated, or the zinc compound can be separated by spray drying. In certain embodiments, recovered solid zinc compounds can be washed with minimal water and dried.

## [00103] Use of Amines and Ammonium Salts

[00104] In certain embodiments, the present invention utilizes the coordination chemistry of the various metals to facilitate separation thereof. For example, the binding affinity or binding strength of transition metals with certain amine compounds, including primary, secondary, and tertiary amines, to preferentially form either a solid precipitate or a soluble complex can change, depending upon several experimental factors. Exemplary factors that can affect whether the metal salt will typically form a solid precipitate include basicity of the amine, the hydrophilic/hydrophobic nature of the amine, steric hindrance of the amine, whether the amine coordinates directly with the metal or forms one or more

polymeric coordination complexes with the metal, solution pH, ionic strength of the solution, crystallization kinetics, and solvation properties. Because the formation of metal-amine coordination complexes can be influenced by so many factors, in general, it can be very difficult to customize/optimize an amine to selectively precipitate or dissolve a targeted metal(s) from a geothermal brine or solution that includes a targeted metal merely by identifying the binding characteristics of the metal for a given amine. In this context, ammonia, an inorganic amine, is very unique in that it can act as both base and a ligand simultaneously, depending upon the solution conditions, such as the pH and/or the concentration of metal salts and/or ammonia in the solution.

[00105] For example, in certain embodiments, ammonia reacts with certain hexaaqua metal ions in solution to form metal hydroxide (see, eq. 1 and 2) precipitates or soluble metal ammonium coordination complexes (see, eq. 3), depending upon ammonia concentration. In equations 1 and 2, ammonia acts as a base to form the metal hydroxide precipitates. In equation 3, ammonia acts as a ligand, resulting in a clear solution having the metal complex dissolved therein.

eq. 1. 
$$[M(H_2O)_6]^{2+} + NH_3 \Rightarrow [M(H_2O)_5(OH)]^+ + NH_4^+$$

eq. 2. 
$$[M(H_2O)_6]^{2+} + 2NH_3 \rightleftharpoons [M(H_2O)_4(OH)_2] + 2NH_4^+$$

eq. 3. 
$$[M(H_2O)_6]^{2+} + 6NH_3 = [M(NH_3)_6]^{2+}$$

[00106] Furthermore, it certain embodiments, the metal ion and ammonia can form one of several possible intermediate complex species that may be isolated, wherein the metal ion coordination sphere can include ammonia, water and hydroxyl groups, depending upon the composition of the salt solution, temperature, pH, and ammonia concentration. The chemical equilibrium involving the precipitation and dissolution of metals salts can thus be advantageously used to selectively isolate certain transition metals from brines and metal containing solutions.

[00107] Referring now to Figure 12, one embodiment of the present invention is provided. Process 100 for the selective removal of manganese from a manganese containing solution, such as a geothermal brine, is provided. Brine 1202 is provided via line 1204 to manganese reaction tank 1206. Ammonia is supplied via line 1208 to manganese reaction tank 1206, where it contacts the manganese containing solution to selectively precipitate

manganese having a purity of greater than 95%, alternatively greater than about 97%, alternatively greater than about 99%. The manganese reaction tank 1206 is maintained at a pH of at least about 6.8, alternatively at least about 8.2, alternatively at least about 8.4, to limit co-precipitation of other metal ions. A manganese-oxide/hydroxide precipitate can be collected from reaction tank 1206 via line 1212. In certain embodiments, it is believed that the manganese-oxide/hydroxide may include a high percentage of Mn<sub>3</sub>O<sub>4</sub>. The brine, having a reduced concentration of manganese, also referred to as a manganese depleted brine solution, can optionally be supplied via line 1210 to a holding tank 1214. In certain embodiments, the brine can be supplied via line 1210 into an injection well (not shown). Air is supplied to produce a reduced pH brine solution having a pH of less than about 7, alternatively less than about 6, alternatively between about 5 and 6. The reduced pH solution can be supplied from holding tank 1214 via line 1217 to zinc reaction tank 1218, which can also be supplied with lime supplied via line 1220, to increase the pH to greater than about 7, alternatively be a pH of between about 7.2 and 7.7, alternatively about 7.5, thereby causing the zinc to precipitate. The zinc precipitate can be collected via line 1222, and the remaining brine solution, having a reduced concentration of both manganese and zinc, also referred to as a manganese and zinc depleted brine solution, can be removed via line 1220. Brine removed via line 1224 can be supplied to an alternate process for the recovery of additional metal ions, or alternatively can be supplied to an injection well (not shown).

[00108] Referring now to Figure 13, another embodiment of the present invention is provided. Process 1300 for the selective removal of manganese and zinc from a manganese and zinc containing solution, such as a geothermal brine, is provided. Brine that includes manganese and zinc is provided from tank 1302 via line 1304 to precipitation tank 1306, where the brine is combined with lime supplied via line 1307 to provide a pH of between about 7.5 and 8, thereby precipitating zinc and manganese. The remaining brine solution, have a decreased concentration of manganese and zinc, can be supplied via line 1308 to injection well 1310, or alternatively supplied to an alternate process for the removal of additional metal ions (not shown). The solid manganese and zinc can be supplied from tank 1306 via line 1312 to a manganese separation process 1314, where the solids are contacted with an ammonium salt that is supplied via line 1315, until a pH of at least about 8.5, alternatively about 9.0 is achieved, to dissolve zinc precipitates, while the manganese remains as a solid. The solid manganese is collected via line 1318, and the zinc containing solution 1316 is supplied to a zinc precipitation process 1320, it is contacted with air supplied via line

1321, preferably supplied via a bubbler, and lime supplied via line 1323, to produce a pH of less than 8, preferably between about 7.2 and 7.7, more preferably about 7.5. Lowering the pH is effective to produce a zinc precipitate, which can be collected via line 1322. Waste solution can be removed via line 1324.

[00109] It is understood that various means can be employed for isolating precipitated solids, including filters, settling tanks, centrifuges, and the like. It is also understood that purification of collected solids can include means for washing solids with water.

[00110]Referring now to Figure 14, one embodiment of the present invention is provided. Brine is supplied from a holding tank or directly from the source, such as a geothermal well, via line 1402 to brine tank 1404. To brine tank 1404, ammonia can be supplied via line 1408 from ammonia tank 1406. Ammonia is supplied in known amounts to selectively precipitate manganese present in the brine, which other ions remain in solution. The solid manganese precipitate can be collected from brine tank 1404 via line 1410. The solution in brine tank 1404, which includes brine (having a lower manganese concentration than originally supplied) and ammonia are supplied via line 1412 to tank 1414, which can include stirring means, such as a mechanical stirrer, and can be supplied with air via line 1416. Air can be added via line 1416 to reduce the pH of the solution selectively to less than 7, preferably between 5 and 6. Ammonia can be removed as a gas from tank 1414 via line 1420 and ammonia-free brine can be removed from the tank via line 1418. Ammonia removed via line 1420 can be supplied to a separation tank wherein air is separated via line 1424 and ammonia is separated via line 1426, and can be recycled back to ammonia tank 1406. In certain embodiments, ammonia tank 1406 can be supplied with fresh make up ammonia, as needed or desired.

### [00111] Extraction of Lead

[00112] In certain aspects of the invention, the process for selectively removing manganese and zinc from brine solutions can also include steps for the removal and recovery of lead from brine solutions. In addition to providing a source for the production of lead, the process also results in higher purity manganese, as a portion of the lead can be present in the recovered manganese.

[00113] As shown in Figure 16, a process for the extraction of various metals from brine, according to one aspect of the present invention, is provided. Brine is supplied via line

1602 and is combined with air supplied via line 1604, lime (20% solution by volume) supplied via line 1606, and flocculant (0.025% by volume in water) supplied via line 1608 in silica removal reactor 1610. Exemplary flow rates are as follows: brine (6 gal/min), flocculant (0.01 gal/min), air (100 cfm), and lime (0.5 lb/min). In silica removal reactor, a solid silica precipitate is formed, which is then removed via line 1612. Air and water vapor can be removed from silica removal reactor 1610 vai line 1614. Under the exemplary flow conditions, for a brine solution having a silica concentration of about 10 ppm, production of the wet silica cake is about 15 lb/hr. A silica lean brine solution can then be supplied via line 1616 to a lithium extraction reactor 1618. Lithium extraction reactor 1618 can include a lithium aluminum intercalate or other extraction medium that has been prepared for the purposes of extracting lithium. Lithium extraction reactor 1618 can include a water inlet 1620, a lithium salt extraction line 1622 for removal of lithium, typically as a chloride salt, after the salt has extracted from the silica lean brine. Water vapor can be removed from reactor 1618 via line 1624.

[00114] A brine solution that is lean in both silica and lithium can be supplied via line 1626 to zinc extraction process 1628, which in certain embodiments can be an ion exchange resin that is designed to extract zinc ions, while allowing other ions to pass through the membrane. Zinc chloride can be collected from the extraction process via line 1630. The remaining brine solution, having had silica, lithium and zinc extracted therefrom, can be supplied via line 1632 to lead extraction reactor 1634. In certain embodiments, the brine solution supplied via line 1632 can have a pH of between about 5 and 6, and a temperature of between about 75°C and 105°C In lead extraction reactor 1634, the brine is contacted with a sulfide compound, such as hydrogen sulfide, sodium sulfide, sodium hydrogen sulfide, calcium sulfide, and the like, which can be supplied to the reactor via line 1636, to form lead sulfide. The lead sulfide precipitate can optionally be filtered or centrifuged, and then can be removed from reactor 1634 via line 1638.

[00115] Following the lead extraction, the remaining solution is supplied to manganese extraction reactor 1642, which can include any of the several different examples of manganese extraction that have been described herein. A remaining brine solution, having reduced concentration of silica, lithium, zinc, lead and manganese, can be collected via line 1644 and either injected into a geothermal well, or supplied to further extraction or other processes. A solid manganese oxide and/or manganese hydroxide precipitate (which can

include MnO<sub>4</sub>, MnO<sub>2</sub>, and/or Mn(OH)<sub>2</sub>) can be collected via line 1646. Preferably, air is excluded during the manganese precipitation.

[00116]Referring now to Figure 17, a process for the extraction of various metals from brine, according to one aspect of the present invention, is provided. Brine is supplied via line 1702 and is combined with air supplied via line 1704, lime (20% solution by volume) supplied via line 1706, and flocculant (0.025% by volume in water) supplied via line 1708 in silica removal reactor 1710. Exemplary flow rates are as follows: brine (6 gal/min), flocculant (0.01 gal/min), air (100 cfm), and lime (0.5 lb/min). In silica removal reactor, a solid silica precipitate is formed, which is then removed via line 1712. Air and water vapor can be removed from silica removal reactor 1710 via line 1714. Under the exemplary flow conditions, for a brine solution having a silica concentration of about 10 ppm, production of the wet silica cake is about 15 lb/hr. A silica lean brine solution can then be supplied via line 1716 zinc extraction reactor 1718, which in certain embodiments can be an ion exchange resin that is designed to extract zinc ions, while allowing other ions to pass through the membrane. Zinc chloride can be collected from the extraction process via line 1720. The remaining brine solution, having had silica and zinc extracted therefrom, can be supplied via line 1722 to lead extraction reactor 1724. In certain embodiments, the brine solution supplied via line 1722 can have a pH of between about 5 and 6, and a temperature of between about 75°C and 105°C In lead extraction reactor 1724, the brine is contacted with a sulfide compound, such as sodium sulfide, hydrogen sulfide, sodium hydrogen sulfide, calcium sulfide, and the like, which can be supplied to the reactor via line 1726, to form a lead sulfide precipitate. The lead sulfide precipitate can be removed from reactor 1724 via line 1728.

[00117] Following the lead extraction, the remaining solution is supplied via line 1730 to manganese extraction reactor 1732, which can include any of the several different examples of manganese extraction that have been described herein. A remaining brine solution, having reduced concentration of silica, zinc, lead and manganese, can be collected via line 1734 and either injected into a geothermal well, or supplied to further extraction or other processes. A solid manganese oxide and/or manganese hydroxide precipitate can be collected via line 1736.

[00118] In certain embodiments, the silica removal process can also include the addition of one or more NORM inhibitors, such as Nalco 9355 and Nalco 1387, which is supplied to the silica removal reactor, along with the lime, air, brine and flocculent.

[00119] In certain embodiments, the lithium extraction process can be an ion exchange process. Additionally, in certain embodiments, the extraction of lithium may result in the co-extraction of trace amounts of other salts present in the brine solution, such as sodium, potassium, calcium, manganese, and zinc.

[00120] Following the manganese extraction, in the embodiments of the present invention exemplified by Figures 16 and 17, the remaining brine solution can have a pH of between about 4.9 and about 5.5, at a temperature of between about 90°C and 100°C. Generally, in processes that include a lithium extraction step, the lithium concentration will be less than about 250 ppm, preferably less than about 100 ppm. Similarly, the concentrations of zinc, silica, lead and manganese, will all be decreased relative to the feed solution.

## [00121] Examples

[00122] For testing purposes, a synthetic brine was employed for examples 1-3 having metal concentrations of approximately the following: 1600 mg/L Fe; 96 mg/L Si; 2500 mg/L Mn; 790 mg/L Zn; 290 mg/L Li; 41,000 mg/L Ca; 27,000 mg/L K; 85,500 mg/L Na; and 185 mg/L Sr.

Example 1. Approximately 1.22 L of the synthetic brine was placed in a 2L [00123]reactor and maintained at a temperature of between about 90-95°C and sparged with air at a rate of about 2.25 L/minute. The initial pH of the brine was about 4.89. To the reaction approximately 14 g of a 20% slurry of calcium hydroxide added. After addition of the slurry, a pH of about 2.85 was achieved, which gradually increased to approximately 3.56 after about 10 minutes. After 40 minutes, at which time the pH was about 2.9, approximately 5.33 g of a 20% slurry of calcium hydroxide was added, which raised the pH to about 4.07. The brine and the calcium hydroxide slurry were mixed for approximately 30 min, during which time the pH decreased to approximately 4.0, at which time approximately 21.22 g of the 20% slurry of calcium hydroxide was added. The addition of the calcium hydroxide slurry increased the pH to approximately 4.5. The mixture was stirred for about another 20 minutes, after which approximately 28.54 g of the calcium hydroxide slurry was again added, and the pH increased to approximately 5.18. The reaction was allowed to stir for an about additional 30 minutes, and the solid was collected and weighed. The solid includes approximately 99.6% of the iron present in the brine and approximately 99.9% of the silica. Additionally, approximately 49.2% of the manganese present in the brine was removed.

[00124] Example 2. Approximately 1.32 L of the synthetic brine was placed in a 2L reactor and maintained at a temperature of between about 90-95°C and sparged with air at a rate of about 2.25 L/minute. The reaction was stirred for approximately 60 minutes and the pH of the solution was monitored. After about 60 minutes, a pH of about 2.05 was achieved. To the brine solution was added approximately 9.73 g of a 20% slurry of calcium hydroxide, which raised the pH to about 5.4. The brine and the calcium hydroxide slurry were mixed for approximately 30 min, during which time the pH decreased to approximately 3.4, at which time approximately 2.56 g of the 20% slurry of calcium hydroxide was added. The addition of the slurry increased the pH to approximately 4.9. The mixture was stirred for about another 20 minutes, after which approximately 1.21 g of the calcium hydroxide slurry was again added, and the pH increased to approximately 5.3. The reaction was allowed to stir for about an additional 70 minutes, and the solid was collected and weighed. The solid includes approximately 98% of the iron present in the brine and approximately 99% of the silica. Additionally, approximately 2% of the manganese present in the brine was removed.

[00125] Example 3. Approximately 1.32 L of the synthetic brine was placed in a 2L reactor and maintained at a temperature of between about 90-95°C and sparged with air at a rate of about 2.25 L/minute. The reaction was stirred for approximately 60 minutes and the pH of the solution was monitored. After about 22 minutes, a pH of about 2.52 was achieved. To the brine solution was added approximately 9.7 g of a 20% slurry of calcium hydroxide, which raised the pH to about 5.56. The brine and the calcium hydroxide slurry were mixed for approximately 13 min, during which time the pH decreased to approximately 4.27, at which time approximately 1.9 g of the 20% slurry of calcium hydroxide was added. The addition of the calcium hydroxide slurry increased the pH to approximately 5.2. The mixture was stirred for about another 5 minutes, during which time the pH decreased to approximately 4.49. Approximately 2.25 g of the calcium hydroxide slurry was again added, and the pH increased to approximately 5.17. The reaction was allowed to stir for about an additional 110 minutes, during which time the pH was maintained at between about 5.13 and 5.17, and the solid was collected and weighed. The solid includes approximately 95.6% of the iron present in the brine and approximately 88.5% of the silica. Additionally, approximately 2% of the manganese present in the brine was removed.

[00126] Example 4. A synthetic brine having a composition that includes about 330 mg/L Li; 2400 mg/L Mn; 740 mg/L Zn; 40,000 mg/L Ca; 26,000 mg/L K; 91,000 mg/L Na; 180 mg/L Sr and 0.8 mg/L Fe was placed in a 2L reactor and maintained at a temperature of

between about 90-95°C and sparged with air at a rate of about 2.25 L/minute. The initial pH was approximately 5.5. After sparging the reactor with air, a calcium hydroxide slurry was added sufficient to bring the pH to approximately 6.6. Additional calcium hydroxide slurry was added over about the next 180 minutes at various intervals. During the addition of the calcium hydroxide slurry, the pH increased from an initial value of about 6.6 to 8. A precipitate was collected which included zinc and manganese. The process recovered about 95.2% of the manganese present in the brine, about 94.6% of the zinc present in the brine, about 0.8% of the calcium present in the brine, and about 75% of the iron present in the brine. Due to the high recovery of iron by this process, the need for removal is confirmed.

[00127] Example 5. A synthetic brine having a composition that includes about 326 mg/L Li; 2640 mg/L Mn; 886 mg/L Zn; 41,000 mg/L Ca; 28,000 mg/L K; 84,000 mg/L Na; 180 mg/L Sr and 0.3 mg/L Fe was placed in a 2L reactor and maintained at a temperature of between about 90-95°C and sparged with air at a rate of about 2.25 L/minute. After sparging the reactor with air, a calcium hydroxide slurry was added in a single dosage sufficient that the pH of the brine solution was measured immediately after addition of the calcium hydroxide slurry and was about 7.6. During the stirring and sparging of the reaction, the pH increased from an initial value of about 7.6 to 7.9 after approximately 15 minutes, and then decreased gradually to about 7.5. A precipitate was collected which included zinc and manganese. The process recovered about 100% of the manganese present in the brine, about 99.9% of the zinc present in the brine, and about 8% of the lithium present in the brine.

## [**00128**] Example 5

Approximately 10 g of a synthetic geothermal brine having an approximate pH of 5.2 and a composition that mimics the composition of Salton Sea (generally, the simulated brine has a composition of about 260 ppm lithium, 63,000 ppm sodium, 20,100 ppm potassium, 33,000 ppm calcium, 130 ppm strontium, 700 ppm zinc, 1700 ppm iron, 450 ppm boron, 54 ppm sulfate, 3 ppm fluoride, 450 ppm ammonium ion, 180 ppm barium, 160 ppm silicon dioxide, and 181,000 ppm chloride) was titrated with a solution that contains about 28-30% by volume ammonia to a maximum pH of about 8.5. The solids began precipitating when the pH of solution was about 6.5. A portion of the brine was decanted and analyzed at various pH levels to identify and estimate the precipitated metal salts (see, Table 1 and Figure 4). Table 1 shows that in the presence of ammonia, manganese precipitated from the brine solution with highest selectivity, and the amount of zinc that is co-precipitated

with the manganese varied from between about 0 to 10%, depending upon the pH of the solution. Furthermore, the solids that precipitated at a pH of about 8 were washed, dried (at 100°C) and digested to analyze the components of the precipitate and purity of the manganese solids. The digested sample revealed the presence of only two metal elements were present, specifically Mn (366.4 mg/g) and Zn (8.06 mg/g). The remainder of metal elements, if present, were below detection limits.

[00130] The results of the analysis at various pH values is provided in both Figure 15 and Table 1, which shows the composition of the synthetic brine before contacting with ammonia, and the composition of the decanted brine that has been separated from the precipitated solids at various different pH levels. As shown in the table, at a pH of about 6.8, approximately 67% of manganese and 13% of zinc that was initially present in the brine solution precipitated around pH 6.8, however, as the pH is increased to about 7.8, the percentage of manganese that was precipitated increased up to a maximum of almost 99%, while the amount of zinc that is precipitated decreases to about 2%.

Table 1. Brine composition after precipitation using 28-30% ammonia solution

	Ba,	Ca,	K,	Li,	Mg,	Mn,	Na,	Sr,	Zn,	В,
	mg/L									
Control	194.4	41120	23060	283.7	11.96	2311	73650	418.1	777.3	511.1
pH 6.8	234.2	48610	27420	341.8	20.51	746.9	86860	497.4	676.5	566.5
pH 7.5	212.4	44050	24540	305.1	18.39	108.1	75880	432.9	734.4	537.8
pH 7.8	194.7	40790	22670	279.9	15.88	28.56	71040	403.7	762.1	509.4
pH 8	192.1	40280	22370	275.8	13.81	32.94	69490	395.3	734.1	498.4
pH 8.4	217.2	43560	25130	317	5.551	84.03	78470	446.5	797.3	539.5

[00131] As shown in Figure 15, these results indicate that at higher pH values, i.e., at a pH of about 8.5, zinc forms a soluble coordination complex, with no measurable precipitate formed, while manganese forms a metal hydroxide/oxide precipitate. In certain embodiments, it is believed that the precipitated solids may be MnO<sub>2</sub> or Mn<sub>3</sub>O<sub>4</sub> and ZnO. The manganese oxides purity from the digestion studies indicated the purity was about 98%. Further optimization of pH and experimental conditions could increase the manganese oxide purity to significantly higher levels.

[00132] Example 7

[00133] To show improvement in the purity of subsequently precipitated manganese when lead is removed by precipitation with sodium sulfide, the following manganese precipitation experiments were done using varying amounts of Ca(OH)<sub>2</sub>. Actual brines were used in the Example 7.

Table 2. Lead Concentration in Manganese Oxides Precipitates.

	Process	Ca(OH) <sub>2</sub>	28% NH <sub>3</sub>	Air	%B	%Ca	%Mg	%Mn	%Pb	%Zn
		g/L	mL/L							
1	Mn ppt	3	1	No	0.07	0.3	0	71	0.2	0.03
1	$(NH_3 + Ca(OH)_2)$								:	
2	Mn ppt	4	1	No	0.08	0.3	0.7	69	0.6	0.1
4	$(NH_3 + Ca(OH)_2)$									
3	Mn ppt	6	1	No	0.06	4.1	3	59	0.07	0.07
]	$(NH_3 + Ca(OH)_2)$									
4	Mn ppt (NH <sub>3</sub> +	~ 6	0	Yes	0.06	2.1	2.5	58	0.02	0
~	Ca(OH) <sub>2</sub> ) after Pb									
	sep. with Na <sub>2</sub> S									
5	Mn ppt (Ca(OH) <sub>2</sub> )	~ 6	0	Yes	1	1.6	0.1	65	0.3	0.31

[00134] In the first trial, for the manganese precipitation, approximately 3 g of Ca(OH)<sub>2</sub> is added per liter of brine, along with approximately 1 mL of NH<sub>3</sub> per liter of brine. Trials 2 and 3 subsequently add additional Ca(OH)<sub>2</sub>. Trial 4 utilizes sodium sulfide for the removal of lead, prior to the manganese precipitation and utilizes Ca(OH)<sub>2</sub> (but does not utilize ammonia) for the manganese precipitation. Trial 5 utilizes only Ca(OH)<sub>2</sub> for the precipitation of manganese, and does not use ammonia.

[00135] As shown in Table 2, trial 4, wherein the lead is removed by precipitation with sodium sulfide prior to the manganese precipitation, results in a manganese oxide product having a significantly reduced lead concentration. Lead concentration can be reduced in the manganese precipitate to less than 200 ppm, as compared with upwards of about 30,000 ppm when lead is not removed prior to precipitation.

[00136] As is understood in the art, not all equipment or apparatuses are shown in the figures. For example, one of skill in the art would recognize that various holding tanks and/or pumps may be employed in the present method.

[00137] The singular forms "a", "an" and "the" include plural referents, unless the context clearly dictates otherwise.

[00138] Optional or optionally means that the subsequently described event or circumstances may or may not occur. The description includes instances where the event or circumstance occurs and instances where it does not occur.

[00139] Ranges may be expressed herein as from about one particular value, and/or to about another particular value. When such a range is expressed, it is to be understood that another embodiment is from the one particular value and/or to the other particular value, along with all combinations within said range.

[00140] Throughout this application, where patents or publications are referenced, the disclosures of these references in their entireties are intended to be incorporated by reference into this application, in order to more fully describe the state of the art to which the invention pertains, except when these reference contradict the statements made herein.

[00141] As used herein, recitation of the term about and approximately with respect to a range of values should be interpreted to include both the upper and lower end of the recited range.

[00142] Although the present invention has been described in detail, it should be understood that various changes, substitutions, and alterations can be made hereupon without departing from the principle and scope of the invention. Accordingly, the scope of the present invention should be determined by the following claims and their appropriate legal equivalents.

### We claim:

1. A method for recovering zinc and manganese from a geothermal brine, the method comprising the steps of:

providing a geothermal brine, said geothermal brine comprising manganese and zinc; selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine;

adjusting the pH of the substantially silica free brine to a pH suitable to form precipitates of zinc and manganese, such that precipitates of zinc and manganese are selectively formed and other metal precipitates are not formed; and separating the zinc and manganese precipitates from the brine.

- 2. The method of claim 1, further comprising the steps of:
  - dissolving the precipitates of zinc and manganese to produce a zinc and manganese containing solution;
  - extracting the zinc and manganese containing solution with an extraction solvent to produce a first liquid phase that includes the extraction solvent and zinc and a second liquid phase that includes manganese.
- 3. The method of claim 2 further comprising the steps of separating the first and second liquid phases and electrochemically recovering zinc from the first liquid phase.
- 4. The method of claim 2 further comprising the steps of separating the first and second liquid phases:

reducing the second liquid phase to form  $\mbox{Mn}^{2+}$ ; and

- supplying the Mn<sup>2+</sup> to an electrochemical cell and recovering the manganese by electrochemical means.
- 5. The method of claim 2 wherein the precipitates of zinc and manganese are dissolved in an acid.

- 6. The method of claim 2 wherein the precipitates of zinc and manganese are dissolved in ammonium sulfate.
- 7. The method of claim 1 wherein the step of selectively removing silica and iron from the geothermal brine comprises providing iron (III) at a pH of between about 4.5 and 6 and precipitating the silica and iron from the brine.
- 8. The method of claim 1 wherein the step of precipitating the zinc and manganese comprises adding sufficient base to adjust the pH to between 6 and 8 and providing an air oxidant to the substantially silica free brine.
- 9. The method of claim 3 further comprising contacting the zinc with hydrochloric acid to produce zinc chloride.
- 10. The method of claim 3, further comprising contacting the zinc with sulfuric acid to produce zinc sulfate.
- 11. A method for recovering zinc and manganese from a geothermal brine, the method comprising the steps of:

- selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine;
- adjusting the pH of the substantially silica free brine to a pH suitable to form precipitates of zinc and manganese, such that precipitates of zinc and manganese are selectively formed and other metal precipitates are not formed; and

separating the zinc and manganese precipitates from the brine.

- 12. The method of claim 11 further comprising the steps of:
  - dissolving the precipitates of zinc and manganese to produce a zinc manganese solution and oxidizing the manganese to form a manganese precipitate and a zinc solution;

and separating the manganese precipitate from the zinc solution.

- 13. The method of claim 12 recovering the zinc by electrochemical means.
- 14. The method of claim 11 wherein the step of selectively removing silica and iron from the geothermal brine comprises providing iron (III) at a pH of between about 4.5 and 6 and precipitating the silica and iron from the brine.
- 15. The method of claim 11 wherein the step of precipitating the zinc and manganese comprises adding sufficient base to adjust the pH to between 6 and 8 and providing an air oxidant to the substantially silica free brine.
- 16. The method of claim 12 wherein the step of dissolving the zinc and manganese comprises providing a mineral acid sufficient to dissolve the zinc and manganese precipitate.
- 17. The method of claim 13 wherein the step of recovering the zinc by electrochemical means comprises plating an electrode with zinc metal from the zinc solution.
- 18. The method of claim 12 further comprising contacting the zinc with hydrochloric acid to produce zinc chloride.
- 19. method of claim 12 further comprising contacting the zinc with sulfuric acid to produce zinc sulfate.
- 20. A method for recovering zinc and manganese from a geothermal brine, the method comprising the steps of:

selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine that includes manganese and zinc;

removing the zinc from the substantially silica free brine;

extracting manganese from the substantially silica free brine;

oxidizing the manganese to produce a manganese dioxide precipitate; and

recovering the magnesium dioxide precipitate.

- 21. The method of claim 20 wherein the step of selectively removing silica and iron from the geothermal brine comprises providing iron (III) at a pH of between about 4.5 and 6 and precipitating the silica.
- 22. The method of claim 20 wherein the zinc is removed from the substantially silica free brine by ion exchange.
- 23. The method of claim 20 wherein the step of oxidizing the manganese to produce manganese dioxide comprises electrolytic deposition.
- 24. A method for recovering zinc and manganese from a geothermal brine, the method comprising the steps of:

- selectively removing silica and iron from the geothermal brine to produce a substantially silica free brine that comprises manganese and zinc;
- extracting manganese and zinc from the substantially silica free brine to produce a manganese zinc solution;
- electrochemically removing manganese from the manganese zinc solution to produce a residual solution that includes zinc; and

electrochemically removing zinc from the residual solution.

- 25. The method of claim 24 wherein the step of selectively removing silica and iron from the geothermal brine comprises providing iron (III) at a pH of between about 4.5 and 6 and precipitating the silica.
- 26. The method of claim 24 wherein the step extracting manganese and zinc from the substantially silica free brine comprises extracting the manganese and zinc with a solvent selected from the group consisting of phosphines, phosphoric acids, and phosphinic acids.
- 27. The method of claim 24 wherein the step of recovering the manganese by electrochemical means comprises plating an electrode with manganese metal from the manganese zinc solution.

- 28. The method of claim 24 wherein the step of electrochemically removing the zinc means comprises plating an electrode with zinc metal from the zinc solution.
- 29. A method for recovering zinc and manganese from a geothermal brine, the method comprising the steps of:

selectively removing silica and iron from the geothermal brine to produce a substantially silica and iron free brine that comprises manganese and zinc;

adding a basic solution containing an amine or ammonia up to about pH 8 to 9 to selectively precipitate manganese from the brine;

producing a manganese precipitate with less than 0.15% Zn and 4% Ca and 4% Mg; and

recovering the precipitate.

- 30. The method of claim 29, further comprising the step of dissolving said precipitate in an acidic solution to recover the manganese salt.
- 31. The method of claim 30, further comprising the step of adding a reducing agent to ensure dissolution of at least about 95% of the manganese.

SELECTIVE RECOVERY OF MANGANESE AND ZINC FROM GEOTHERMAL BRINES

## **ABSTRACT**

This invention relates to a method for the selective recovery of manganese and zinc from geothermal brines that includes the steps of removing silica and iron from the brine, oxidizing the manganese and zinc to form precipitates thereof, recovering the manganese and zinc precipitates, solubilizing the manganese and zinc precipitates, purifying the manganese and zinc, and forming a manganese precipitate, and recovering the zinc by electrochemical means.

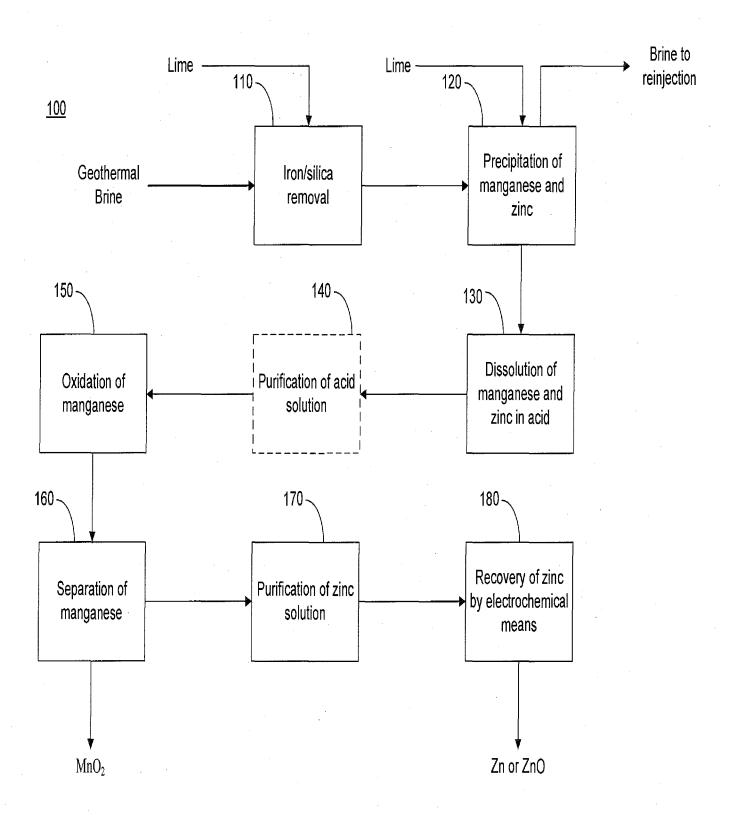


Figure 1

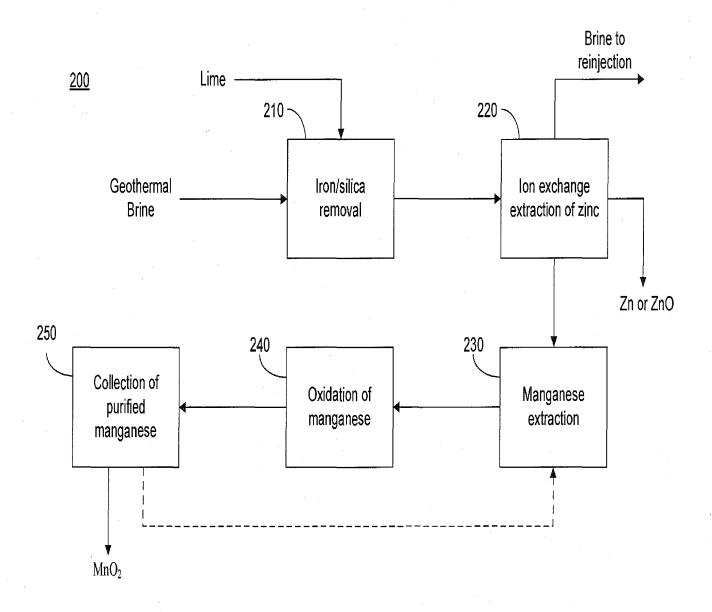


Figure 2

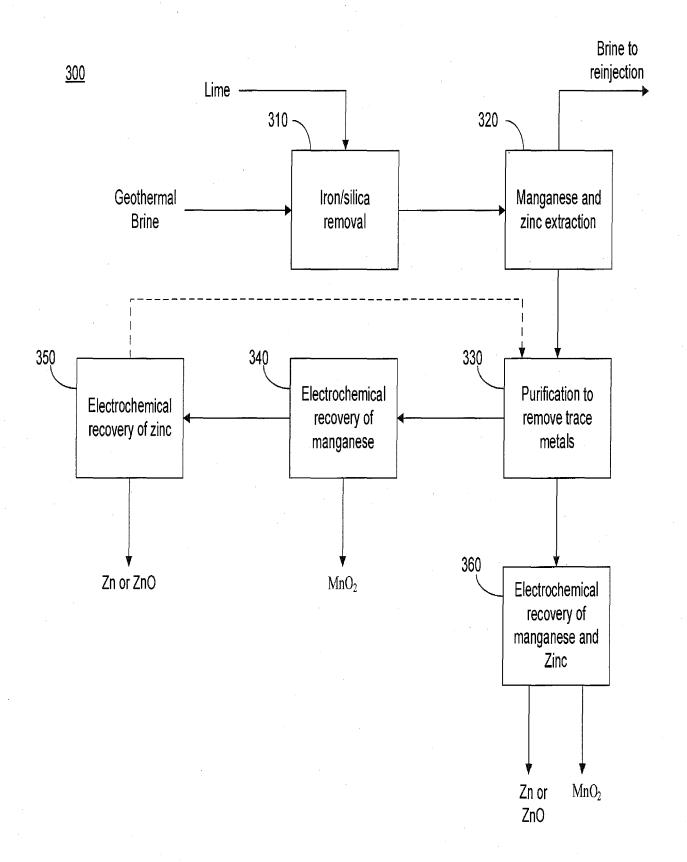


Figure 3

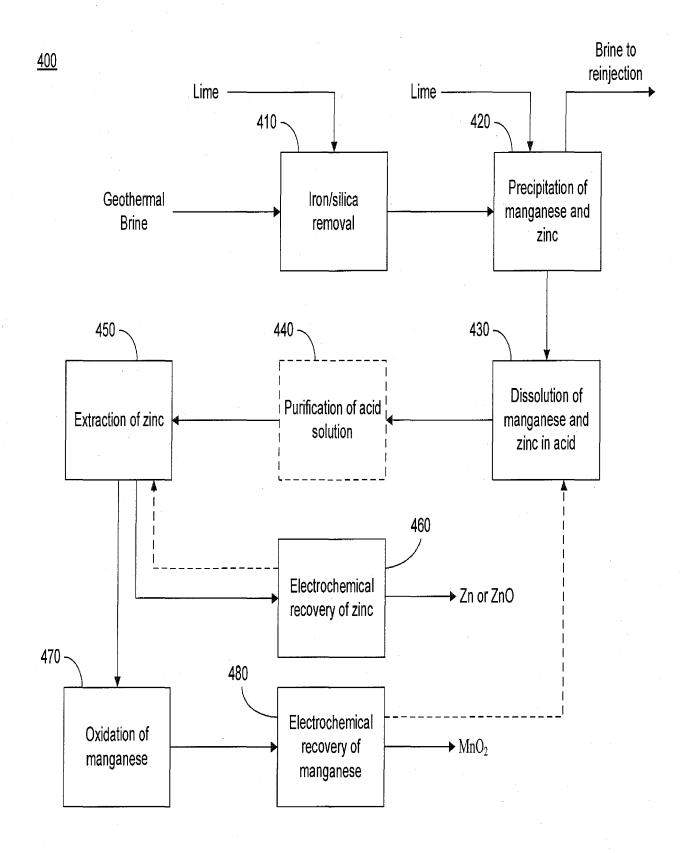


Figure 4

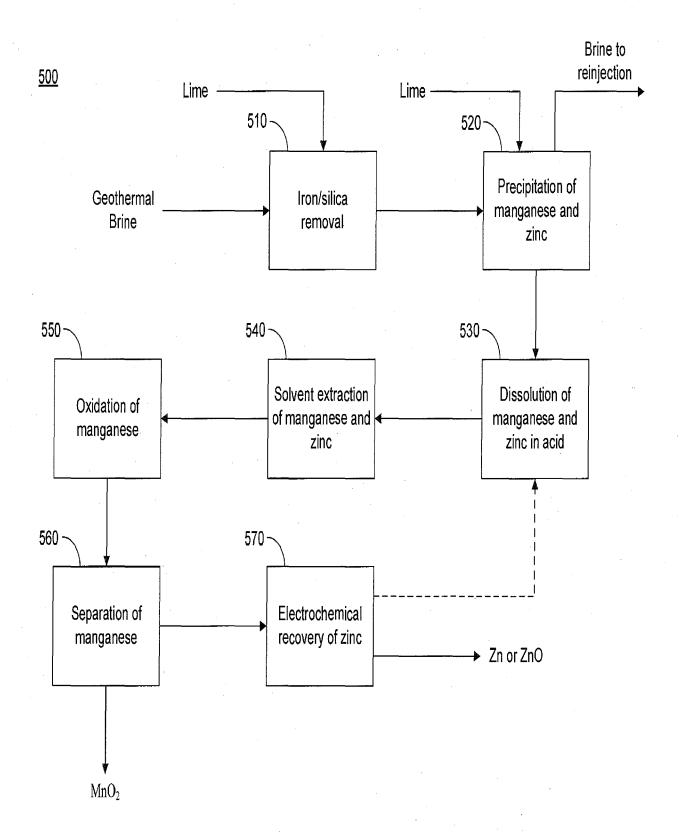


Figure 5

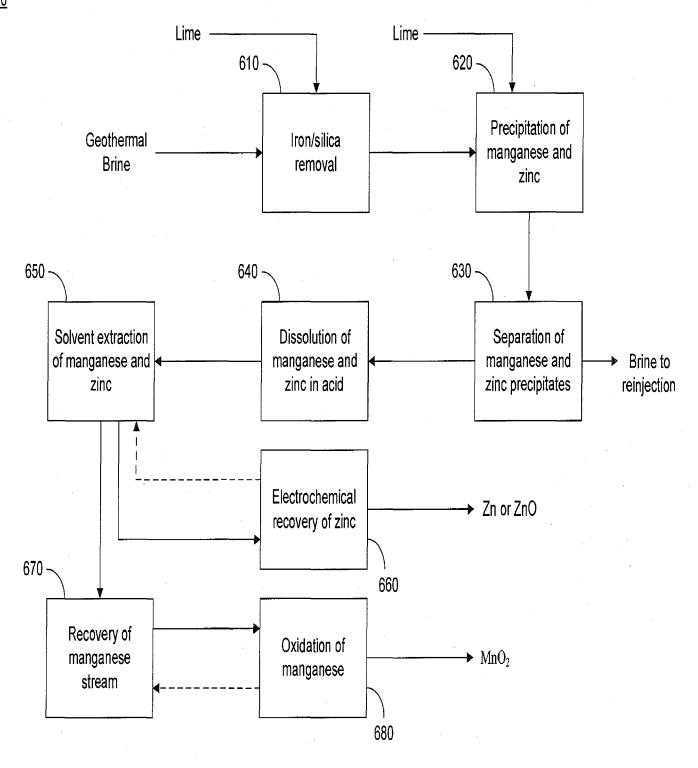


Figure 6

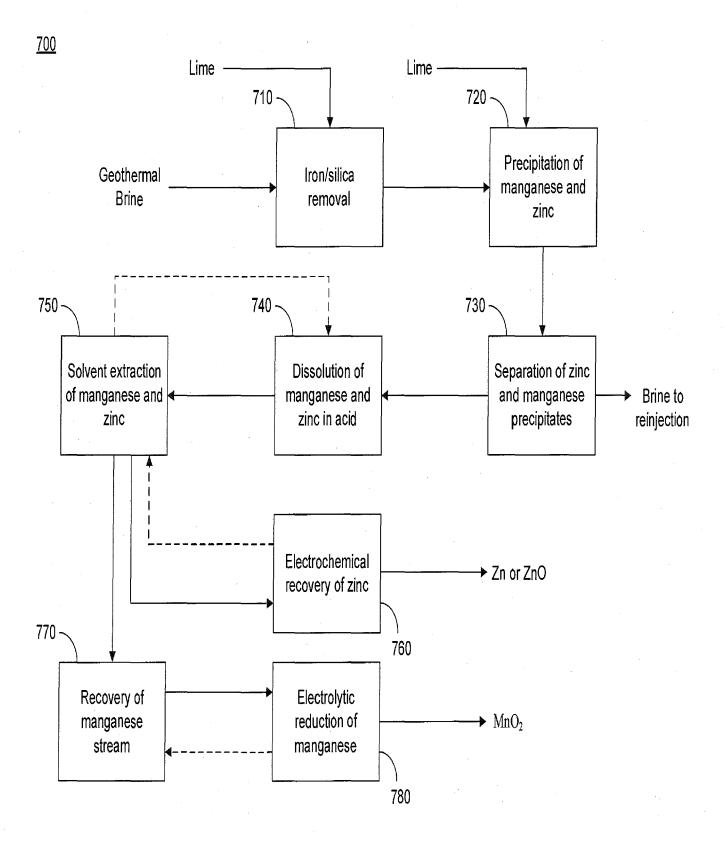


Figure 7

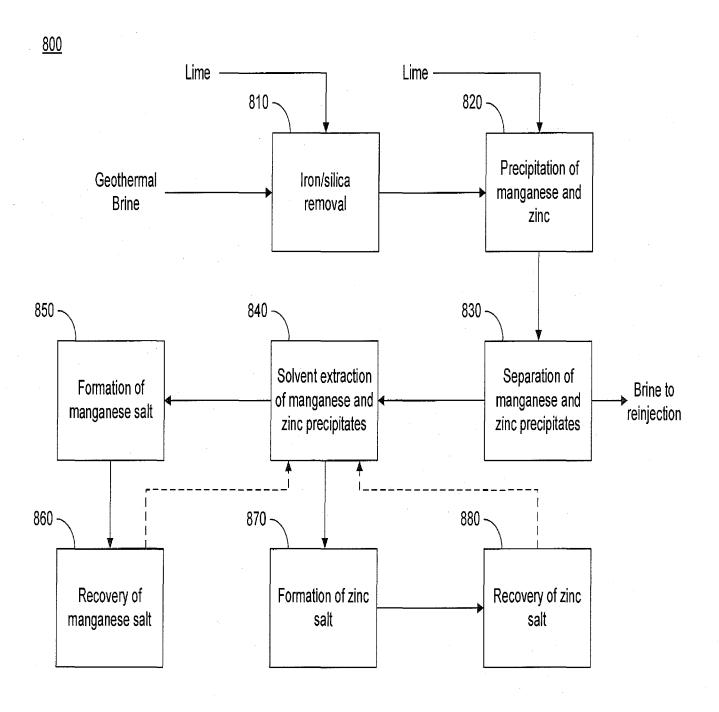


Figure 8

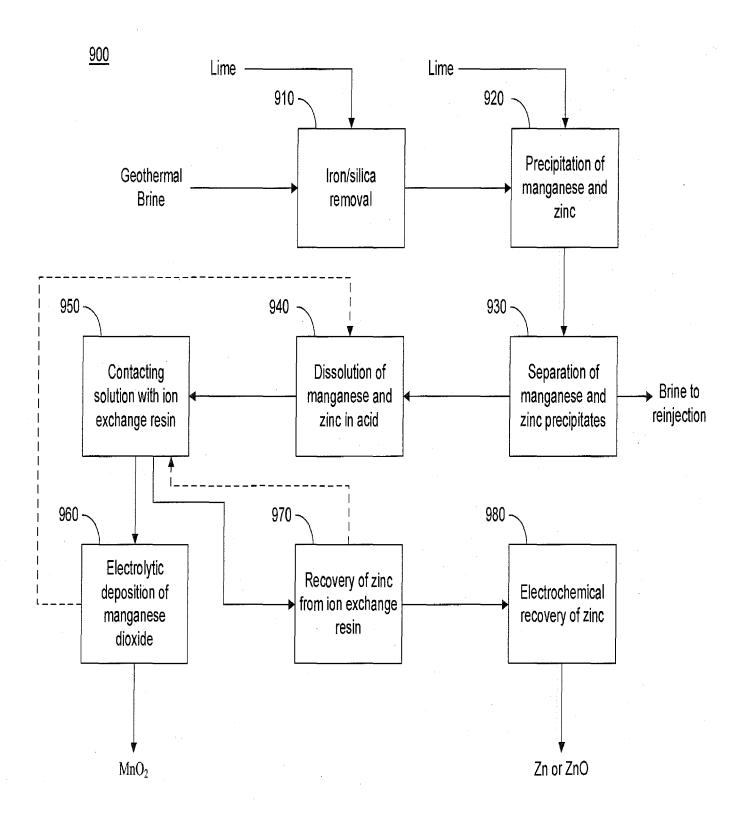


Figure 9

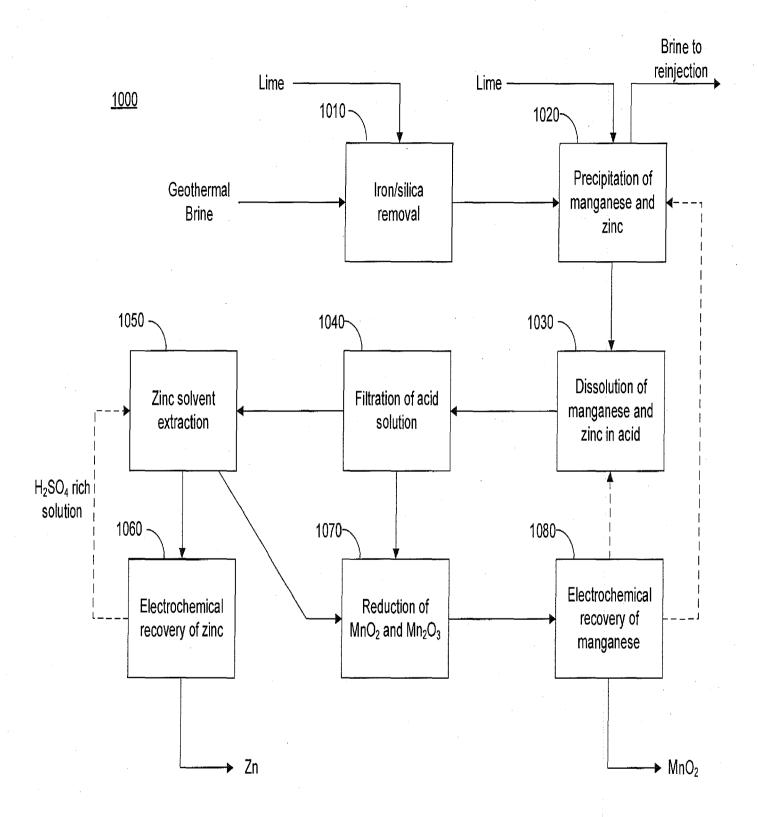


Figure 10

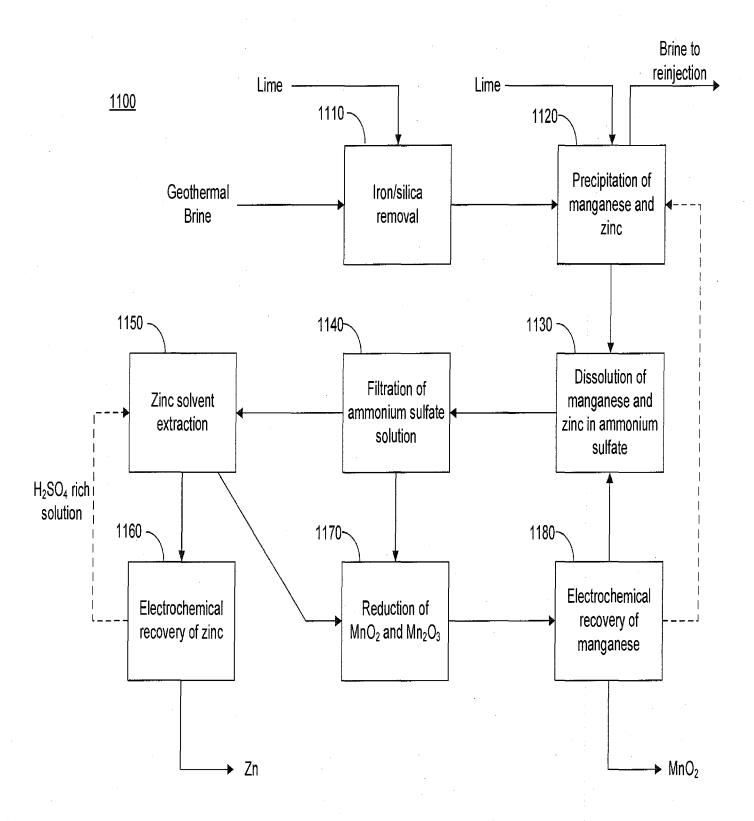


Figure 11

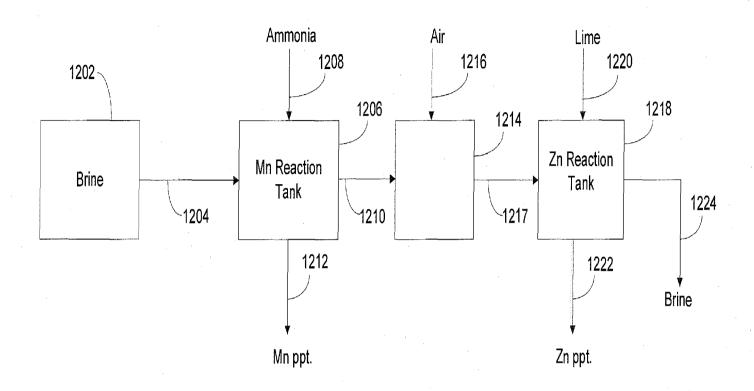


Figure 12

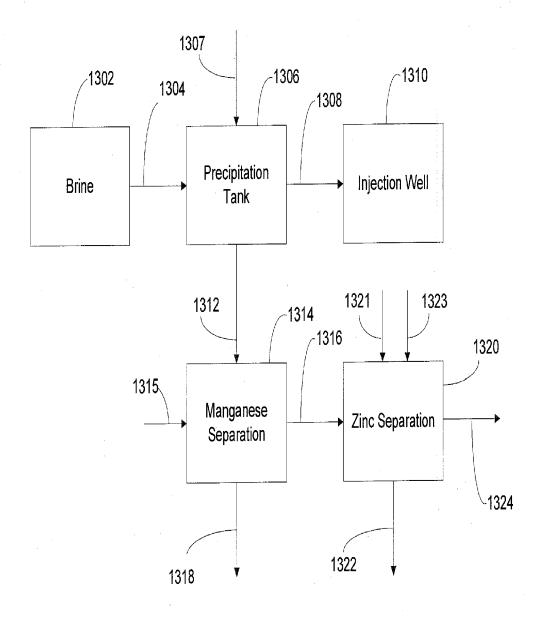


Figure 13

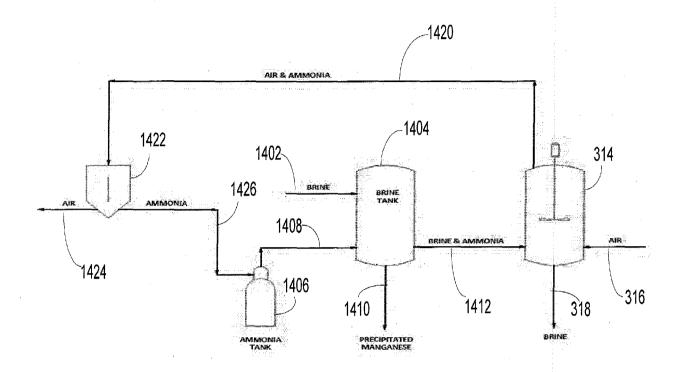


Figure 14

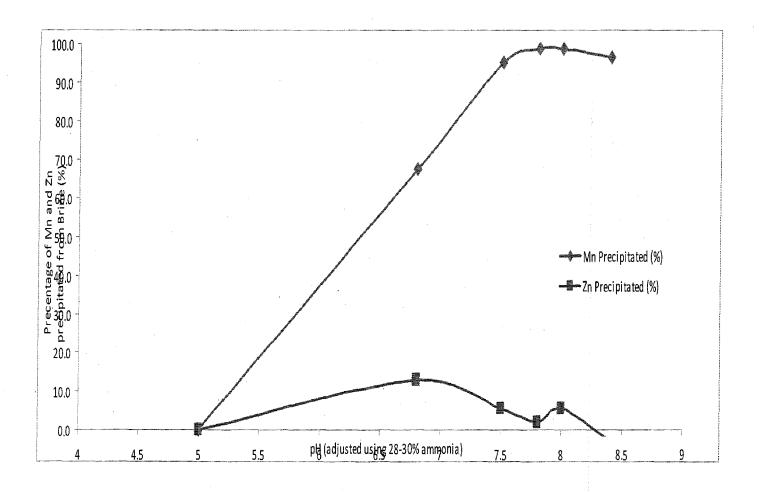


Figure 15

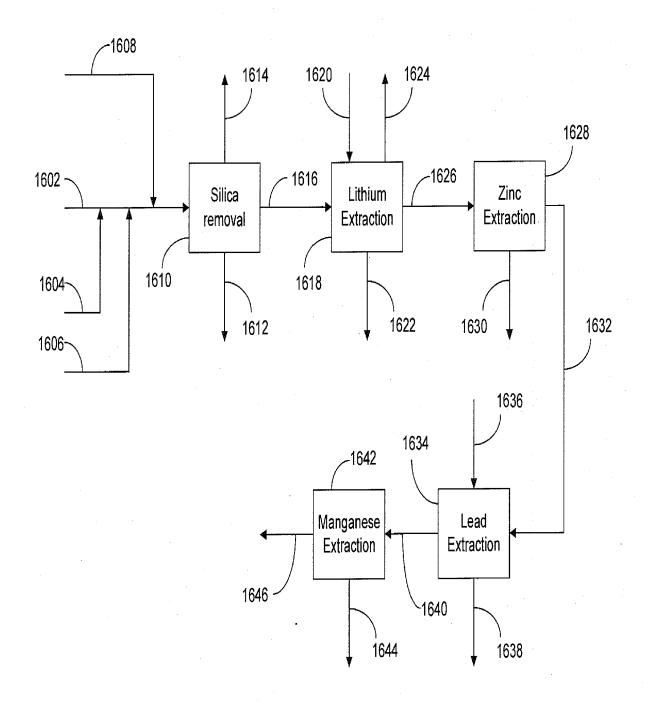


Figure 16

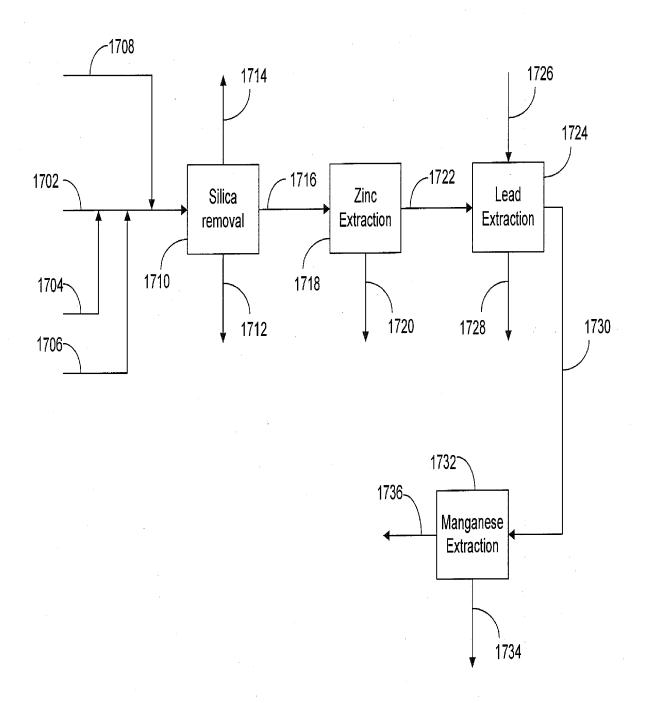


Figure 17

Electronic Patent A	<b>App</b>	lication Fee	Transmit	tal	
Application Number:					
Filing Date:					
Title of Invention:	Sel	ective Recovery of I	Manganese, Lea	d and Zinc	
First Named Inventor/Applicant Name:	Ste	phen Harrison			
Filer:	Kar	thika Perumal/Deb	ra Odum		
Attorney Docket Number:					
Filed as Small Entity					
Utility under 35 USC 111(a) Filing Fees					
Description		Fee Code	Quantity	Amount	Sub-Total in USD(\$)
Basic Filing:					
Utility filing Fee (Electronic filing)		4011	1	70	70
Utility Search Fee		2111	1	300	300
Utility Examination Fee		2311	1	360	360
Pages:					
Claims:					
Claims in excess of 20		2202	11	40	440
Independent Claims in Excess of 3		2201	2	210	420
Miscellaneous-Filing:					

Description	Fee Code	Quantity	Amount	Sub-Total in USD(\$)				
Late Filing Fee for Oath or Declaration	2051	1	70	70				
Petition:								
Patent-Appeals-and-Interference:								
Post-Allowance-and-Post-Issuance:								
Extension-of-Time:	Extension-of-Time:							
Miscellaneous:								
	Tot	al in USD	(\$)	1660				

Electronic Acknowledgement Receipt					
EFS ID:	16332453				
Application Number:	13943395				
International Application Number:					
Confirmation Number:	2954				
Title of Invention:	Selective Recovery of Manganese, Lead and Zinc				
First Named Inventor/Applicant Name:	Stephen Harrison				
Customer Number:	35979				
Filer:	Karthika Perumal/Debra Odum				
Filer Authorized By:	Karthika Perumal				
Attorney Docket Number:					
Receipt Date:	16-JUL-2013				
Filing Date:					
Time Stamp:	16:46:29				
Application Type:	Utility under 35 USC 111(a)				

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## File Listing:

Document Number	Document Description File Name		File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)
1	Application Data Sheet	20130716ApplicationDataShee	102826	no	4
	, pp. leation batta sincet	t Continuation.pdf	19a46832ea05de48f4d15fae17fc89085ec2 aa12	0	
Warnings:					
Information:					
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2		20130716 Non provisional Conti	2654134	yes	59
-		nuationPatentApplication.pdf	af5fe983887cecdec5abe1fab6f2801dfd6aa 73b	,	
	Multip	part Description/PDF files in .	zip description		
	Document De	Start	End		
	Specificat	1		36	
	Claims	37	2	41	
	Abstrac	42	4	42	
	Drawings-only black and v	43	43 59		
Warnings:					
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3	Fee Worksheet (SB06)	fee-info.pdf	39358	no	2
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		Total Files Size (in bytes)	27	96318	

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#### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.