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TRANSMITTAL LETTER T		Attorney Docket No. GBL0007/US
DESIGNATED/ELECTEI CONCERNING A SUBMISS		U.S. Application No. (if known, see 37 CFR 1.5)
International Application No.	International Filing Date	Priority Date Claimed
PCT/GB2016/051688	June 8, 2016	June 9, 2015
Title of Invention COATINGS	***************************************	
First Named Inventor		жи ророн и и од насторина на при пред насторина на пред на пре
Stephen Richard COULSON		
Applicant herewith submits to the United St		
35 U.S.C. 371(f) will not be effective u	ational examination procedures (35 U.S.C. 37 unless the requirements under 35 U.S.C. 371 ion and English translation thereof (If required	1(f)). NOTE: The express request under (c)(1), (2), and (4) for payment of the basic national (), and the oath or declaration of the inventor(s)
	n (35 U.S.C. 371(c)(2)) is attached hereto (not national Bureau or was filed in the United Sta	
 An English language translation of the 	International Application (35 U.S.C. 371(c)(2	2))
a. is attached hereto.		
b. has been previously submitted u		
 An oath or declaration of the inventor(a. \(\bigcap \) is attached. 	(s) (35 U.S.C. 371(c)(4))	
h	ational phase under PCT Rule 4.17(iv).	
Items 5 to 8 below concern amendments ma		
PCT Article 19 and 34 amendments	~~ ~~ ~~ ~~ ~~ ~~ ~~ ~~ ~~ ~~ ~~ ~~ ~~	
	Article 19 are attached (not required if comm	nunicated by the International Bureau) (35 U.S.C.
	19 amendment is attached (35 U.S.C. 371(c)	(3)).
 English translation of annexes (Article attached (35 U.S.C. 371(c)(5)). 	19 and/or 34 amendments only) of the Interr	national Preliminary Examination Report is
Cancellation of amendments made in the intern	national phase	
8a. Do not enter the amendment made in	the international phase under PCT Article 19	
8b. Do not enter the amendment made in	the international phase under PCT Article 34	
NOTE: A proper amendment made in English instruction from applicant not to enter the amer	under Article 19 or 34 will be entered in the Undment(s).	l.S. national phase application absent a clear
The following items 9 to 17 concern a docu	ment(s) or information included.	
9. An Information Disclosure Statement	under 37 CFR 1.97 and 1.98.	
10. A preliminary amendment.		
11. An Application Data Sheet under 37 C	OFR 1.76.	
12. A substitute specification, NOTE: A s	substitute specification cannot include claims.	See 37 CFR 1,125(b).
13. A power of attorney and/or change of	address letter.	
14. A computer-readable form of the sequ	uence listing in accordance with PCT Rule 13	ter.3 and 37 CFR 1.821-1.825.
15. Assignment papers (cover sheet and	document(s)). Name of Assignee:	
16. 37 CFR 3.73(c) Statement (when then	re is an Assignee).	

This collection of information is required by 37 CFR 1.414 and 1.491-1.492. The information is required to obtain or retain a benefit by the public, which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.11 and 1.14. This collection is estimated to take 15 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Mail Stop PCT, Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

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PTC-1390 (06-13)

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U.S. APPLN. No. (if known – see 37 CFR 1.5) INTERNATIONAL APPLICATION No. PCT/GB2016/051688						ATTORNEY GBL0007/US	DOCKET No.
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	t under 37 CFR 1.55 or 1.78 for AIA (First Inv application (1) claims priority to or the benefit				nd (2) als	o contains, or contained at
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Application Da	ta She	et 37 CEF	2 1 76	Attorney Docket Number GE		GBL00	007/US	
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Application la	nform	ation:						
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	Application De	ita Sheet 37 CFR 1.76	Attorney Docket Number	GBL0007/US
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	Title of Invention	COATINGS		

Domestic Benefit/National Stage Information:

This section allows for the applicant to either claim benefit under 35 U.S.C. 119(e), 120, 121, 365(c), or 386(c) or indicate National Stage entry from a PCT application. Providing benefit claim information in the Application Data Sheet constitutes the specific reference required by 35 U.S.C. 119(e) or 120, and 37 CFR 1.78.

When referring to the current application, please leave the "Application Number" field blank.

Prior Application Status	Pending		Remove
Application Number	Continuity Type	Prior Application Number	Filing or 371(c) Date (YYYY-MM-DD)
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Additional Domestic Benefit/National Stage Data may be generated within this form by selecting the Add button.

Foreign Priority Information:

This section allows for the applicant to claim priority to a foreign application. Providing this information in the application data sheet constitutes the claim for priority as required by 35 U.S.C. 119(b) and 37 CFR 1.55. When priority is claimed to a foreign application that is eligible for retrieval under the priority document exchange program (PDX)¹ the information will be used by the Office to automatically attempt retrieval pursuant to 37 CFR 1.55(i)(1) and (2). Under the PDX program, applicant bears the ultimate responsibility for ensuring that a copy of the foreign application is received by the Office from the participating foreign intellectual property office, or a certified copy of the foreign priority application is filed, within the time period specified in 37 CFR 1.55(g)(1).

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Statement under 37 CFR 1.55 or 1.78 for AIA (First Inventor to File) Transition Applications

This application (1) claims priority to or the benefit of an application filed before March 16, 2013 and (2) also
contains, or contained at any time, a claim to a claimed invention that has an effective filing date on or after March
16, 2013 .
NOTE: By providing this statement under 37 CFR 1.55 or 1.78, this application, with a filing date on or after March
16, 2013, will be examined under the first inventor to file provisions of the AIA.

Application Da	ata Sheet 37 CFR 1.76	Attorney Docket Number	GBL0007/US
Application ba	ita Sileet 37 GFK 1.70	Application Number	
Title of Invention	COATINGS		

Authorization or Opt-Out of Authorization to Permit Access:

When this Application Data Sheet is properly signed and filed with the application, applicant has provided written authority to permit a participating foreign intellectual property (IP) office access to the instant application-as-filed (see paragraph A in subsection 1 below) and the European Patent Office (EPO) access to any search results from the instant application (see paragraph B in subsection 1 below).

Should applicant choose not to provide an authorization identified in subsection 1 below, applicant <u>must opt-out</u> of the authorization by checking the corresponding box A or B or both in subsection 2 below.

<u>NOTE</u>: This section of the Application Data Sheet is <u>ONLY</u> reviewed and processed with the <u>INITIAL</u> filing of an application. After the initial filing of an application, an Application Data Sheet cannot be used to provide or rescind authorization for access by a foreign IP office(s). Instead, Form PTO/SB/39 or PTO/SB/69 must be used as appropriate.

- 1. Authorization to Permit Access by a Foreign Intellectual Property Office(s)
- A. <u>Priority Document Exchange (PDX)</u> Unless box A in subsection 2 (opt-out of authorization) is checked, the undersigned hereby <u>grants the USPTO authority</u> to provide the European Patent Office (EPO), the Japan Patent Office (JPO), the Korean Intellectual Property Office (KIPO), the State Intellectual Property Office of the People's Republic of China (SIPO), the World Intellectual Property Organization (WIPO), and any other foreign intellectual property office participating with the USPTO in a bilateral or multilateral priority document exchange agreement in which a foreign application claiming priority to the instant patent application is filed, access to: (1) the instant patent application-as-filed and its related bibliographic data, (2) any foreign or domestic application to which priority or benefit is claimed by the instant application and its related bibliographic data, and (3) the date of filing of this Authorization. See 37 CFR 1.14(h) (1).
- B. <u>Search Results from U.S. Application to EPO</u> Unless box B in subsection 2 (opt-out of authorization) is checked, the undersigned hereby <u>grants the USPTO authority</u> to provide the EPO access to the bibliographic data and search results from the instant patent application when a European patent application claiming priority to the instant patent application is filed. See 37 CFR 1.14(h)(2).

The applicant is reminded that the EPO's Rule 141(1) EPC (European Patent Convention) requires applicants to submit a copy of search results from the instant application without delay in a European patent application that claims priority to the instant application.

- 2. Opt-Out of Authorizations to Permit Access by a Foreign Intellectual Property Office(s)
- A. Applicant <u>DOES NOT</u> authorize the USPTO to permit a participating foreign IP office access to the instant application-as-filed. If this box is checked, the USPTO will not be providing a participating foreign IP office with any documents and information identified in subsection 1A above.
 - B. Applicant <u>DOES NOT</u> authorize the USPTO to transmit to the EPO any search results from the instant patent application. If this box is checked, the USPTO will not be providing the EPO with search results from the instant application.

NOTE: Once the application has published or is otherwise publicly available, the USPTO may provide access to the application in accordance with 37 CFR 1.14.

Application Da	ita Sheet 37 CFR 1.76	Attorney Docket Number	GBL0007/US
Application be	ita oneet 57 OFK 1.70	Application Number	
Title of Invention	COATINGS		

Applicant Information:

Providing assignment information in this section does not substitute for compliance with any requirement of part 3 of Title 37 of CFR to have an assignment recorded by the Office.							
Applicant 1							
1.43; or the name and address of the who otherwise shows sufficient propr applicant under 37 CFR 1.46 (assign	section is the name and address assignee, person to whom the in etary interest in the matter who is se, person to whom the inventor	s of the legal representativ nventor is under an obligat s the applicant under 37 C is obligated to assign, or p	ve who is the applicant under 37 CFR tion to assign the invention, or person				
○ Assignee	C Legal Representative ur	nder 35 U.S.C. 117	Joint Inventor				
Person to whom the inventor is ob	igated to assign.	O Person who show	vs sufficient proprietary interest				
If applicant is the legal representa	tive, indicate the authority to	file the patent application	on, the inventor is:				
Name of the Deceased or Legally	Incapacitated Inventor:						
If the Applicant is an Organization	n check here.						
Organization Name P2i Ltd							
Mailing Address Information F	or Applicant:						
Address 1 9-12	North Central, 127 Olympic Ave	nue					
Address 2 Milto	n Park						
City Abin	gdon	State/Province					
Country GB	Country GB Postal Code OX14 4SA						
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Additional Applicant Data may be	generated within this form by	selecting the Add butto	on.				

Assignee Information including Non-Applicant Assignee Information:

Providing assignment information in this section does not substitute for compliance with any requirement of part 3 of Title 37 of CFR to have an assignment recorded by the Office.

Application Data Sheet 37 CFR 1.76		Attorney Doc	ket Number	et Number GBL0007/US				
		Application N	lumber					
Title of Invention COATINGS								
Assignee 1	Assignee 1							
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Application De	eta Chaot 27 CED 1 76	Attorney Docket Number	GBL0007/US
Application Data Sheet 37 CFR 1.76		Application Number	
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INFORMATION DISCLOSURE STATEMENT BY APPLICANT (Not for submission under 37 CFR 1.99)	Application Number			
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Application Number Filing Date INFORMATION DISCLOSURE First Named Inventor Stephen Richard COULSON STATEMENT BY APPLICANT Art Unit (Not for submission under 37 CFR 1.99) **Examiner Name** Attorney Docket Number GBL0007/US 4 2007/083122 WO A1 2007-07-26 P2I LTD 5 0834352 ΕP Α1 1998-04-08 CIBA-GEIGY AG 6 2014/155099 WO A1 2014-10-02 SEMBLANT LTD 7 2015/128627 WO Α1 2015-09-03 P2I LTD 2016/198855 WO Α1 8 2016-12-15 P2I LTD 9 2016/198856 WO Α1 2016-12-15 P2I LTD If you wish to add additional Foreign Patent Document citation information please click the Add button **NON-PATENT LITERATURE DOCUMENTS** Include name of the author (in CAPITAL LETTERS), title of the article (when appropriate), title of the item Cite Examiner Ţ5 (book, magazine, journal, serial, symposium, catalog, etc), date, pages(s), volume-issue number(s), Initials* No publisher, city and/or country where published. International Search Report and Written Opinion for International Application No. PCT/GB2016/051688, mailed 1 November 8, 2016 (9 pages) If you wish to add additional non-patent literature document citation information please click the Add button **EXAMINER SIGNATURE**

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- (71) Applicant: P2I LTD [GB/GB]; 9-12 North Central, 127 Olympic Avenue, Milton Park, Abingdon Oxfordshire OX14 4SA (GB).
- (72) Inventors: COULSON, Stephen Richard; c/o P2i Ltd, 9-12 North Central, 127 Olympic Avenue, Milton Park, Abingdon Oxfordshire OX14 4SA (GB). EVANS, Delwyn; c/o P2i Ltd, 9-12 North Central, 127 Olympic Avenue, Milton Park, Abingdon Oxfordshire OX14 4SA (GB). SIOKOU, Angeliki; c/o P2i Ltd, 9-12 North Central, 127 Olympic Avenue, Milton Park, Abingdon Oxfordshire OX14 4SA (GB). TELFORD, Clive; c/o P2i Ltd, 9-12 North Central, 127 Olympic Avenue, Milton Park, Abingdon Oxfordshire OX14 4SA (GB).

- (74) Agent: BLAKE, Michael; Greaves Brewster LLP, Copa House, Station Road, Cheddar Somerset BS27 3AH (GB).
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(54) Title: COATINGS

$$R_1$$
 R_2
 R_3

(57) Abstract: The present invention provides an electronic or electrical device or component thereof comprising a cross-linked polymeric coating on a surface of the electronic or electrical device or component thereof; wherein the cross-linked polymeric coating is obtainable by exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow formation of the cross-linked polymeric coating on a surface thereof, wherein the monomer compound has the following formula: where R₁, R₂ and R₄ are each independently selected from hydrogen, optionally substituted branched or straight chain C₁-C₆ alkyl or halo alkyl or aryl optionally substituted by halo, and R₃ is selected from: or where each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C₁-C₆ alkyl, halo alkyl or aryl optionally substituted by halo; and n₁ is an integer from to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moieties and has a boiling point at standard pressure of less than 500°C.

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Coatings

Field of the invention

This invention relates to protective coatings. Specifically, the present invention relates to protective coatings for electronic or electrical devices and components thereof, and methods of forming such coatings. The coatings can protect by being hydrophobic and so resist the ingress of water-based liquid into electronic devices, or they can protect by forming a barrier coating and so provide electrical resistance between the electrical parts of the phone and water based liquid.

Background of the invention

Monounsaturated monomers are used to make barrier coatings using plasma polymerisation processes (see co-pending application).

The perfluoroalkyl chain monomers are also used for generating hydrophobic surfaces from pulsed plasma deposition processes (see WO 9858117 A1).

The power of the plasma initiated polymerisation affects the nature of the polymer produced. The higher average energy inputs of continuous wave plasmas lead to more fragmentation of the monomer, and so the polymer loses structural properties of the monomer. In the case of 1H,1H,2H,2H- perfluorodecyl acrylate (PFAC8), there's less retention of the perfluoroalkyl chain and the contact angle of the surface coating is compromised. Higher plasma energies also lead to more crosslinking. For the lower average energy inputs of pulsed plasmas, there's better retention of monomer structure and less crosslinking. The greater retention of the perfluoro chain under low energy, pulsed plasma conditions leads to best levels of contact angles for the surface coating.

When the perfluoroalkyl chains have eight or more fluorinated carbons (long chain), the polymer made from the monomer has a crystalline structure. When the perfluoroalkyl chains have less than eight fluorinated carbons, the resulting polymer is amorphous and so can be unstable in the presence of water (see "Molecular Aggregation Structure and Surface Properties of Poly(fluoroalkyl acrylate) Thin Films, Marcomolecules, 2005, vol 38, p5699-5705)

When long chain perfluoroalkyl polymers are produced by high average power (continuous wave or CW) or low average power (pulsed wave or PW) plasmas, then because of the crystalline structure of the long chains, the polymers are non-stick to the touch and stable in the presence of water. However, the feel and water stability of shorter chain polymer coatings is affected by the plasma power levels used. For example, when PFAC6 (1H,1H,2H,2H-perfluorooctyl acrylate) is polymerised in low power plasma conditions, the resulting polymer coating can have several disadvantages. For example the coating can cause water drops to spread out a little (slump), be marked by the presence of a water drop on its surface, have a tacky feel, or can be easily smeared (for example on substrates of silicon wafer and ABS plastic).

By increasing the power of the plasma used for polymerisation, the polymer becomes more crosslinked and becomes more resistant to smearing. However, increasing the power has the concomitant effect of decreasing the water contact angle through more monomer fragmentation (as described above). Figure 1 shows the effect of increasing the power to monomer flow ratio for CW plasmas in a 125 litre chamber: at a ratio of 4 W/µl/min, the water contact angle is ~85-95 degrees and the coating is tack-free. However, as the ratio drops, the contact angle increases and the occurrence of tacky/smudgy coatings increases too. Figure 2 shows the same effect for pulsed plasma conditions. These results show that the process window for producing tack and smudge-free coatings have a limited plasma processing range and the final coating has a compromised water contact angle.

Accordingly, it is an aim to solve one or more of the above-mentioned problems with the prior art coatings.

Statement of Invention

An aspect of the present invention provides an electronic or electrical device or component thereof comprising a protective cross-linked polymeric coating on a surface of the electronic or electrical device or component thereof;

wherein the protective cross-linked polymeric coating is obtainable by exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow formation of the protective cross-linked polymeric coating on a surface thereof,

wherein the monomer compound has the following formula:

$$R_1$$
 R_3 R_2 R_4

where R_1 , R_2 and R_4 are each independently selected from hydrogen, optionally substituted branched or straight chain C_1 - C_6 alkyl or halo alkyl or aryl optionally substituted by halo, and R_3 is selected from:

$$\begin{array}{c} O \\ \begin{array}{c} C \\ \end{array} \\ \begin{array}{c} C$$

where each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C_1 - C_6 alkyl, halo alkyl or aryl optionally substituted by halo; and n_1 is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moieties and has a boiling point of less than 500 °C at standard pressure.

High levels of polymer crosslinking (formerly only achievable with high average power continuous wave plasmas) can be achieved by adding a crosslinking molecule to the monomer to produce a cross-linked co-polymer. This has the advantage of increasing the plasma processing range so stable coatings can now be produced in low average energy pulse plasma conditions.

The high degree of retention of the hydrophobic monomer structures from the low energy pulsed plasma gives coatings of the co-polymer a good hydrophobic coating (as evidenced by water contact angles) and a coating that is not sticky or smeary to the touch. Thus, these properties make it suitable for the coating of electronic devices to prevent the ingress of water-based liquids from accidental damage. Such coatings typically have a thickness between 1 and 100 nm, preferably 1 and 50 nm.

The coating may be described as a liquid repellant layer, typically a water repellant layer. The coating may repel water, an aqueous liquid such as rainwater, or oil.

Alternatively, or in addition to providing a liquid repellent layer, the coating may be a physical barrier layer.

The coating may be a physical barrier i.e. it provides a physical barrier to mass and electron transport.

The physical barrier layer restricts diffusion of water, oxygen and ions. When the coating is a physical barrier it typically has a thickness of greater than 50 nm.

The coating is a protective layer i.e. it prevents damage by contact with water or other liquids. The coating may provide is protective functionality by forming a liquid repellent layer and/or a physical barrier layer.

The coating is preferably substantially pin-hole free to enable it to provide a physical barrier. Preferably $\Delta Z/d < 0.15$, where ΔZ is the average height variation on an AFM line scan in nm (as shown in Figure 3) and d is coating thickness in nm.

The value of $\Delta Z/d$ tells us to what extent defects/voids on the surface of the coating extend into the coating, i.e. the percentage value of the depth of defect over total coating thickness. For example, $\Delta Z/d = 0.15$ means that the voids on the surface only extend down to 15% of the coating thickness. A coating with a $\Delta Z/d < 0.15$ is defined herein as being substantially pinhole free. If voids are bigger than this, the desired functionality is unlikely to be achieved.

The coating is preferably conformal, which means that it takes the 3D shape of the electronic or electrical device or component thereof and covers substantially an entire surface of the device. This has the advantage of ensuring that the coating has sufficient thickness to give optimal functionality over an entire surface of the device or component. The meaning of the term "covers substantially an entire surface" will depend to some extent on the type of surface to be covered. For example, for some components, it may be necessary for there to be complete coverage of the surface in order for the component to function after submersion in water. However, for other components or housings, small gaps in coverage may be tolerated.

The coating may have a thickness of 50 to 10,000nm, optionally 50 to 8000nm, 100 to 5000nm, preferably 250nm – 5000nm, most preferably 250nm – 2000nm.

The coating may be electrically insulating and be sufficiently compliant that electrical connectors can be joined to the electronic or electrical device or component thereof and an electrical connection made between the electrical connectors and electronic or electrical device or component thereof without the requirement to first remove the coating. In this case, the force exerted by the electrical connector on the coating is sufficient to alter the structure of or even break through the coating local to the electrical connector, allowing the electrical connection to be made. Electrical connectors can typically be joined to the electronic or electrical device or component in this way for coating thicknesses of under 5000nm, and for high performance coatings below 2000nm.

The electronic or electrical device or component thereof can typically withstand immersion in up to 1m of water for over 30 minutes without failure or corrosion whilst power applied to the electronic or electrical device or component. The effectiveness of the coating can be determined by measuring its electrical resistance at a fixed voltage when submerged in water for a set time period; for example, applying a voltage of 8V to the coating of a device which is submerged in water for 13 minutes. Coatings having a resistance value of 1x10⁷ Ohms or higher in this test are effective barrier coatings and the coated electronic or electrical device or component thereof will pass successfully an IPX7 test. The IPX7 test is the Ingress Protection Marking which classifies and rates the degree of protection provided against water. In the IPX7 test for phones, the device is immersed in water under defined conditions of pressure and time (up to 1 m of submersion) for a duration of 30 minutes. The device has to be powered on during testing and functional after 24 hrs.

In one embodiment, the coating is electrically insulating and has a thickness of less than 1 micron and wherein a force of 5- 20g applied to the coating using a round probe with 1 mm diameter allows an electrical connection to be made with the electronic or electrical device or component thereof in the local area where the force has been applied.

In another embodiment, the coating is electrically insulating and has a thickness of 1-2.5 microns and wherein a force of 20-100g applied to the coating using a round probe with 1 mm diameter allows an electrical connection to be made in the local area of the coating where the force has been applied.

The coating may have a higher density than that of the corresponding monomer from which it is formed. For example, the increase in density may be at least 0.1 g/cm^{3.} The increase in density is explained by the highly crosslinked coating. The high density of the coating improves the barrier properties of the coating.

The coating may form a surface defined by a static water contact angle (WCA) of at least 70°. A coating with a WCA of at least 90° is a liquid repellent, typically a water repellent layer. For fluorinated polymers, the coating may have a static water contact angle of at least 100°. The contact angle of a liquid on a solid substrate gives an indication of the surface energy which in turn illustrates the substrate's liquid repellence. Contact angles were measured on a VCA Optima contact angle analyser, using 3 µl droplets of deionised water at room temperature.

Crosslinker

A basic requirement for the cross-linking reagent is the presence two or more unsaturated bonds, for example, -C=C- or alkyne groups. The unsaturated bonds are attached by one or more linker moieties. The linker moiety is not especially limited provided it joins together the two or more unsaturated bonds. The cross-linking reagent must have a boiling point of less than 500 °C, preferably from -10 to 300°C, optionally 180 to 270°C, most preferably 205 to 260°C at standard pressure. It is a preference, but not a requirement, that the cross-linking reagent is not unduly hazardous for use in plasma processing, that is to say it can be used in a production environment where low levels of vapour do not present any significant health and safety issues (e.g. strongly oxidising, explosive, toxic, or has an offensive odour (e.g. stench reagent)). The crosslinking reagent preferably has one of the following structures:

where Y_1 , Y_2 , Y_3 , Y_4 , Y_5 , Y_6 , Y_7 and Y_8 are each independently selected from hydrogen, optionally substituted cyclic, branched or straight chain C_1 - C_6 alkyl or aryl; and L is a linker moiety.

Most preferably L has the following formula:

where each Y_9 is independently selected from, a bond, -O-, -O-C(O)-, -C(O)-O-, -Y₁₁-O-C(O)-, -C(O)-O-Y₁₁-, and -Y₁₁O-, where Y₁₁ is an optionally substituted cyclic, branched or straight chain C₁-C₈ alkylene; and

 Y_{10} is selected from optionally substituted cyclic, branched or straight chain C_1 - C_8 alkylene and a siloxane containing group.

In one most preferred embodiment Y_{10} has the following formula:

$$\begin{cases} - \begin{bmatrix} Y_{12} \\ C \\ Y_{13} \end{bmatrix}_n \end{cases}$$

wherein each Y_{12} and Y_{13} is independently selected from H, halo, optionally substituted cyclic, branched or straight chain C_1 - C_8 alkyl, or $-OY_{14}$, where Y_{14} is selected from optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl, and n is an integer from 1 to 10.

Optionally, each Y_{12} is H and each Y_{13} is H, i.e. Y_{10} is a linear alkylene chain. For this embodiment Y_9 is preferably a vinyl ester or vinyl ether group.

Optionally, each Y_{12} is fluoro and each Y_{13} is fluoro, i.e. Y_{10} is a linear perfluoroalkylene chain.

Typically, n is from 4 to 6.

In another embodiment Y_{10} has the following formula:

(v)

wherein each Y_{15} is independently selected from optionally substituted branched or straight chain C_1 - C_6 alkyl.

In one embodiment each Y_{15} is methyl, and each Y_9 is a bond.

In a further embodiment Y₁₀ has the following formula:

(vi)

wherein Y_{18} to Y_{19} are each independently selected from H and optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl. A preferred alkenyl group is vinyl. Optionally Y_{18} is H or vinyl, and Y_{16} , Y_{17} and Y_{19} are each H. In one embodiment each of Y_{18} to Y_{19} is H. In another embodiment Y_{18} is vinyl, and Y_{16} , Y_{17} and Y_{19} are each H.

In preferable embodiments for compound (i), L has one of the following structures:

In another embodiment for compound (i) L has one of the following structures:

$$(xi) \qquad V_{10} \qquad (xii) \qquad v_{2} \qquad O \qquad V_{10} \qquad O \qquad v_{2} \qquad (xiii) \qquad v_{3} \qquad O \qquad V_{10} \qquad O \qquad v_{3} \qquad (xiii) \qquad V_{10} \qquad O \qquad v_{3} \qquad V_{10} \qquad V_{1$$

For L according structure (vii), Y_{10} is preferably an alkylene chain or a cycloalkylene, such as those shown in structures iv) and vi) above. The alkylene chain may be a straight chain alkylene chain. When Y_{10} is a cycloalkylene, cyclohexylene is preferred, most preferably 1,4 cyclohexylene.

For L according to structure (viii), Y₁₀ is preferably structure (iv), i.e. an alkylene or fluoroalkylene chain.

For L according to structure (ix), Y_{10} is preferably cycloalkylene, such as the cyclohexylene according to structure (vi).

For L according to structure (x), Y_{10} is preferably structure (iv) wherein each Y_{12} and Y_{13} is F, i.e. a perfluoroalkylene chain.

For L according to structure (xi) or structure (xii) Y_{10} is preferably alkylene or cycloalkylene. Optionally the alkylene or cycloalkylene may be substituted with one or more vinyl groups or alkenyl ether groups, preferably one or more vinyl ether groups.

When each Y_9 is a bond, each Y_{10} may be any of structures (iv), (v) and (vi). Preferably Y_{10} is a straight chain alkylene such that the crosslinking reagent is a diene, such as a heptadiene, octadiene, or nonadiene, most preferably 1,7-octadiene.

When each Y_9 is O, each Y_{10} is preferably a branched or straight chain C_1 - C_6 alkylene, preferably a straight chain alkylene, most preferably a C_4 straight chain alkylene, i.e the crosslinking reagent is 1,4-butanediol divinyl ether

It will be understood that each Y_9 group can be combined with any other Y_9 group and Y_{10} group to form the crosslinking reagent.

The skilled person will be aware of possible substituents for each of the cyclic, branched or straight chain C_1 - C_8 alkylene groups mentioned above. The alkylene groups may be substituted at one or more positions by a suitable chemical group. Halo substituents are preferred, with fluoro substituents most preferred. Each C_1 - C_8 alkylene group may be a C_1 - C_3 , C_2 - C_6 , or C_6 - C_8 alkylene group.

Particularly preferred embodiments of the crosslinker have alkyl chains for Y_{10} and vinyl ester or vinyl ether groups on either side.

A particularly preferred cross-linking reagent is divinyl adipate. (a divinyl ester)

Another preferred cross-linking reagent is 1,4-butanediol divinyl ether (a divinyl ether)

For the most preferred embodiments the crosslinking reagent is selected from Divinyl Adipate (DVA), 1,4 Butanediol divinyl ether (BDVE), 1,4 Cyclohexanedimethanol divinyl ether (CDDE), 1,7 -Octadiene (17OD), 1,2,4 -Trivinylcyclohexane (TVCH), 1,3-Divinyltetramethyldisiloxane (DVTMDS), Diallyl 1,4-cyclohexanedicarboxylate (DCHD), 1,6-Divinylperfluorohexane (DVPFH), 1H,1H,6H,6H-Perfluorohexanediol diacrylate (PFHDA) and glyoxal bis (diallyl acetal) (GBDA).

For the alkyne crosslinking reagents according to compound (ii), L is preferably selected from a branched or straight chain C₁-C₈ alkylene or an ether group. L may be a C₃, C₄, C₅, or C₆ alkylene,

preferably a straight chain alkylene. Chemical structures for the particularly preferred crosslinking reagents are set out below in Table 1:

Divinyl Adipate (DVA)	H_2C O CH_2
1,4 Butanediol divinyl ether (BDVE)	H_2C O CH_2
1,4 Cyclohexanedimethanol	O CH ₂
divinyl ether (CDDE)	H_2C
1,7 -Octadiene (170D)	H_2C CH_2
1,2,4 -Trivinylcyclohexane	CH ₂
(TVCH)	H_2C CH_2
1,3-	ÇH₃ CH₃
Divinyltetramethyldisiloxane	şi—o—şi
(DVTMDS)	H₂C CH₂ CH₃ CH₃
Diallyl 1,4-	O CH ₂
cyclohexanedicarboxylate	G 12
(DCHD)	
	0 CH ₂
1,6-Divinylperfluorohexane	F F F F
(DVPFH)	H_2C F
1H,1H,6H,6H-	0 F F F
Perfluorohexanediol	H_2C O F F O CH_2
diacrylate (PFHDA)	F F O
GBDA	O CH ₂
	H_2C O CH_2
	H_2C O CH_2 H_2C

1,6-heptadiyne	нс=сн
1,7-heptadiyne	НС
1,8-heptadiyne	нс=сн
Propargyl ether	нс=сн

Table 1

Monomer Compound

The monomer compound has the following formula:

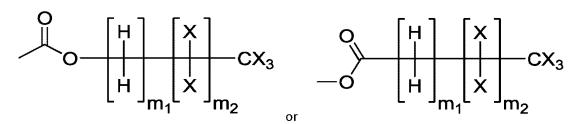
$$R_1$$
 R_3 R_2 R_4

where R_1 , R_2 and R_4 are each independently selected from hydrogen, optionally substituted branched or straight chain C_1 - C_6 alkyl or halo alkyl or aryl optionally substituted by halo, and R_3 is selected from:

$$\begin{array}{c} O \\ \begin{array}{c} X \\ \\ \end{array} \\ \begin{array}{c} C \\ \end{array} \\ \begin{array}{c}$$

where each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C_1 - C_6 alkyl, halo alkyl or aryl optionally substituted by halo; and n_1 is an integer from 1 to 27. Preferably, n_1 is from 1 to 12. Optionally n_1 is from 4 to 12, optionally 6 to 8.

In a preferred embodiment R₃ is selected from:



where m_1 is an integer from 0 to 13 and each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C_1 - C_6 alkyl, halo alkyl or aryl optionally substituted by halo; and m_2 is an integer from 2 to 14;

In a particularly preferred embodiment the monomer is a compound of formula I(a):

I(a)

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wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_6 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1;

; or

the monomer is a compound of formula I(b):

I(b)

$$R_{2} \xrightarrow{R_{1}} O \xrightarrow{R_{5}} R_{7} \begin{bmatrix} R_{9} \\ \dot{C} \end{bmatrix} \begin{bmatrix} X \\ \dot{C} \end{bmatrix} \begin{bmatrix} X \\ \dot{C} \end{bmatrix} C C X_{3}$$

$$R_{4} \xrightarrow{C} C R_{6} R_{8} \begin{bmatrix} R_{10} \\ R_{10} \end{bmatrix}_{a} \begin{bmatrix} X \\ \dot{C} \end{bmatrix}_{b} \begin{bmatrix} X \\ \dot{C} \\ X \end{bmatrix}_{c} C X_{3}$$

wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_6 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1.

The halogen may be chlorine or bromine, but is preferably fluorine for compliance with RoHS regulations (Restriction of Hazardous Substances).

a is from 0 to 10, preferably from 0 to 6, optionally 2 to 4, most preferably 0 or 1. b is from 2 to 14, optionally from 2 to 10, preferably 3 to 7.

Each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or a C_1 - C_6 branched or straight chain alkyl group. The alkyl group may be substituted or unsubstituted, saturated or unsaturated. When the alkyl group is substituted, the location or type of the substituent is not especially limited provided the resultant polymer provides an appropriate liquid repellent and/or barrier layer. The skilled person would be aware of suitable substituents. If the alkyl group is substituted, a preferred substituent is halo, i.e. any of R_1 , R_2 , R_4 , and R_5 to R_{10} may be haloalkyl, preferably fluoro alkyl. Other possible substituents may be hydroxyl or amine groups. If the alkyl group is unsaturated it may comprise one or more alkene or alkyne groups.

Each of R_1 , R_2 , R_4 , and R_5 to R_{10} may be independently selected from hydrogen, methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, tert-butyl, sec-butyl, n-pentyl, neopentyl, n-hexyl, isobexyl, and 3-methylpentyl.

In a preferred embodiment each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or methyl.

In a preferred embodiment, a and c are each independently 0 or 1; and b is from 3 to 7.

In one preferred embodiment each X is H. In an alternative preferred embodiment each X is F.

Optionally R₁ and R₂ are both hydrogen.

Optionally R_4 is hydrogen or methyl. Preferably R_1 and R_2 are both hydrogen and R_4 is hydrogen or methyl.

Optionally R_9 is hydrogen and R_{10} is branched or straight chain C_1 - C_6 alkyl group. In a preferred embodiment R_{10} is methyl.

In one embodiment each of R₅ to R₈ is hydrogen.

In one embodiment each of R₁, R₂, R₄, and R₅ to R₁₀ is hydrogen, each X is H, a=0 and c=0.

In a particularly preferred embodiment the monomer compound has the following formula:

$$H \longrightarrow O (CF_2)_nF$$

where n is from 2 to 10.

In another preferred embodiment the monomer compound has the following formula:

$$H \longrightarrow O (CF_2)_nF$$
 $H \longrightarrow CH_3$

where n is from 2 to 10.

The monomer compound may be selected from 1H,1H,2H,2H-perfluorohexyl acrylate (PFAC4), 1H,1H,2H,2H-perfluorooctyl acrylate (PFAC6), 1H,1H,2H,2H-perfluorodecyl acrylate (PFAC8) and 1H,1H,2H,2H-perfluorododecyl acrylate (PFAC10).

The monomer compound may be selected from 1H,1H,2H,2H-pefluorohexyl methacrylate (PFMAC4), 1H,1H,2H,2H-perfluorooctyl methacrylate (PFMAC6) and 1H,1H,2H,2H-perfluorodecyl methacrylate (PFMAC8).

The monomer compound of formula I(a) may have the following formula:

wherein a and c are each independently 0 or 1, b =3-7 and n is 4 to 10, where n=a+b+c+1.

The monomer ccompound of formula I(a) may have the following formula:

$$H \longrightarrow CH_3$$
 $(CH_2)_nH$

where n is 2 to 12.

The monomer compound may be selected from ethyl hexyl acrylate, hexyl acrylate, decyl acrylate, lauryl dodecyl acrylate and iso decyl acrylate.

The monomer may have the following formula:

where n is from 4 to 14.

The monomer may have the following formula:

$$\begin{array}{c}
H \longrightarrow O \longrightarrow (CH_2)_n \longrightarrow CH_3
\end{array}$$

where n is from 4 to 14.

In an additional aspect, the present invention provides a method for treating an electronic device or component thereof comprising:

exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow a protective cross-linked polymeric coating to form on a surface of the electronic or electrical device or component thereof;

wherein the monomer compound has the following formula:

$$R_1$$
 R_3 R_2 R_4

where R_1 , R_2 and R_4 are each independently selected from hydrogen, optionally substituted branched or straight chain C_1 - C_6 alkyl or halo alkyl or aryl optionally substituted by halo, and R_3 is selected from:

$$\begin{array}{c} \begin{array}{c} O \\ \begin{array}{c} X \\ \end{array} \\ \begin{array}{c} C \\ \end{array} \\ \begin{array}{$$

where each X is independently selected from hydrogen, optionally substituted branched or straight chain C_1 - C_6 alkyl, halo alkyl or aryl optionally substituted by halo; and n_1 is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds and has a boiling point of less than 500 °C at standard pressure.

The monomer compound and the crosslinking reagent utilised in the method are described in further detail above.

Typically the electronic or electrical device or component thereof is placed in a plasma deposition chamber, a glow discharge is ignited within said chamber, and a voltage applied as a pulsed field.

Preferably the applied voltage is at a power of from 30 to 800W. Optionally the voltage is pulsed in a sequence in which the ratio of the time on/time off is in the range of from 0.001 to 1, optionally 0.002 to 0.5. For example, time on may be 10-500 μ s, preferably 35-45 μ s, or 30-40 μ s, such as about 36 μ s and time off may be from 0.1 to 30 ms, preferably 0.1 to 20 ms, optionally from 5 to 15 ms, for example 6ms. Time on may be 35 μ s, 40 μ s, 45 μ s. Time off may be 0.1, 1, 2, 3, 6, 8, 10, 15, 20, 25 or 30 ms.

The term pulsed may mean that the plasma cycles between a state of no (or substantially no) plasma emission (off-state) and a state where a particular amount of plasma is emitted (on-state). Alternatively, pulsed may mean that there is continuous emission of plasma but that the amount of plasma cycles between an upper limit (on-state) and lower limit (off-state).

In a further embodiment, the invention resides in a method of forming a coating on an electronic or electrical device or component thereof as defined above, the method comprising: exposing said substrate in a chamber to a plasma comprising a monomer compound, preferably a continuous plasma, for a sufficient period of time to allow a protective polymeric coating to form on the substrate,

wherein during exposure of the substrate the continuous plasma has a power density of at least 2W/litre, preferably 20W/litre.

Optionally the voltage is applied as a pulsed field for a period of from 30 seconds to 90 minutes. Optionally the voltage is applied as a pulsed field for from 5 to 60 minutes. Optionally in a preliminary step, a continuous power plasma is applied to the electronic or electrical device or component thereof. The preliminary step may be conducted in the presence of an inert gas.

Before the crosslinker and/or the monomer compound enter the deposition chamber each may be in the form of a gas, liquid or a solid (for example a powder) at room temperature. However, it is the preferred that the crosslinking reagent and the monomer compound are both liquid at room temperature, and most preferably that the monomer and crosslinker liquids are miscible.

The monomer compound and/or the crosslinker will suitably be in a gaseous state in the plasma. The plasma may simply comprise a vapour of the monomer compound and the crosslinker. Such a vapour may be formed in-situ, with the compounds being introduced into the chamber in liquid form. The monomer may also be combined with a carrier gas, in particular, an inert gas such as helium or argon.

In preferred embodiments, the monomer and/or the crosslinker may be delivered into the chamber by way of an aerosol device such as a nebuliser or the like, as described for example in WO2003/097245 and WO03/101621, the content of which is incorporated herein by reference. In such an arrangement a carrier gas may not be required, which advantageously assists in achieving high flow rates.

In one embodiment the monomer compound and/or the crosslinking reagent are in gaseous form and are fed into the plasma at a rate of from 10-5000 mg/minute depending on chamber volume, whilst the pulsed voltage is applied.

Optionally the plasma is created with an average power of from 0.001 to 40W/litre.

The crosslinker may be miscible with the monomer and so introduced together or separately into the plasma chamber. Or the crosslinker may be immiscible with the monomer and introduced separately into the plasma chamber. In this context, the term "miscible" means that the crosslinker is soluble in the monomer, and when mixed they form a solution of uniform composition. The term "immiscible" is used to mean that the crosslinker is only partly soluble or insoluble in the monomer, and so either forms an emulsion or separates out into two layers.

The crosslinking reagent is preferably present in an amount of from 10 to 60 (v/v) %, optionally 20 to 40 (v/v) %, optionally 25 to 30 (v/v)%, optionally 30 to 50 (v/v)% of the total volume of monomer compound and crosslinking reagent, depending on the specific crosslinker. The skilled person would understand that the amount will vary to some extent depending on whether the coating is required to

be liquid repellent or provide a barrier to mass and electron transport. The skilled person would understand that the (v/v) percentages are those which give a stable crosslinked polymer coating and the highest water contact angle.

Electronic or electrical device or component thereof

Although the invention is of benefit in the context of a wide variety of substrates, the substrate may, in all aspects of the invention, advantageously be an electronic substrate.

In some embodiments of the invention, the electronic substrate may comprise an electronic or electrical device, i.e. any piece of electrical or electronic equipment. Non-limiting examples of electrical and electronic devices include communications devices such as mobile phones, smartphones and pagers, radios, and sound and audio systems such as loudspeakers, microphones, ringers or buzzers, hearing aids, personal audio equipment such as personal CD, tape cassette or MP3 players, televisions, DVD players including portable DVD players, video recorders, digi and other set-top boxes such as Sky, computers and related components such as laptop, notebook, tablet, phablet or palmtop computers, personal digital assistants (PDAs), keyboards, or instrumentation, games consoles in particular hand-held playstations and the like, data storage devices, outdoor lighting systems or radio antenna and other forms of communication equipment.

In preferred embodiments of the invention, the substrate may comprise or consist of an electronic component, e.g. a printed circuit board (PCB), a printed circuit board array (PCBA), a transistor, resistor, or semi-conductor chip. The electronic component may thus be an internal component of an electronic device, e.g. a mobile phone. The coatings of the invention are particularly valuable in preventing electrochemical migration in such components.

In all aspects of the invention, the precise conditions under which the protective polymeric coating is formed in an effective manner will vary depending upon factors such as, without limitation, the nature of the monomer compound, the crosslinking reagent, the substrate, as well as the desired properties of the coating. These conditions can be determined using routine methods or, preferably, using the techniques and preferred features of the invention described herein, which work in particular synergy with the invention.

The present invention will now be further described with reference to the following non-limiting examples and the accompanying illustrative drawings, of which:

Figure 1 illustrates shows the effect of increasing the power to monomer flow ratio for CW plasmas in a 3 litre chamber in a prior art process.

Figure 2 shows the same effect for pulsed plasma conditions.

Figure 3 shows a tapping mode image over $10x10~\mu m^2$ field of view of a specimen example (thickness d=1230 nm) prepared according to Example 1 (left hand side) and a contour line plot showing the data used for calculation of RMS roughness (right hand side). The ΔZ value indicated on the plot has been taken over an area of the graph that represents the majority of the coating. Peaks that lie above the ΔZ range indicate large particles and troughs that fall below the ΔZ range show voids or pinholes in the coating. The width of the peaks also gives an indication of the particle size.

Figure 4 illustrates the effect of adding the divinyl adipate crosslinker to perfluorooctyl acrylate on water contact angle from a process on a 125 Litre chamber.

Figure 5 shows the effect of adding crosslinkers in accordance with the invention on the electrical resistance per nm of the coatings.

Figure 6 shows a tapping mode image of a 90 nm thick coating, prepared as described in Example 3, over $2x2\mu m^2$ field of view (top left), a representative contour line indicating height variation (z-axis) of the coating (top right) and a phase image indicating full substrate coverage (bottom left) ;RMS roughness of the coating is 1.65 nm and $\Delta z/d = 0.05$

Figure 7 shows a FTIR/ATR spectrum of a coating formed according to Example 5.

Example 1

Example of plasma polymerising PFAC6 with no crosslinker in a 3 litre chamber

PFAC6 was polymerised in a 3 litre glass chamber using a continuous plasma process. In the experimental matrix, the continuous plasma was produced using RF power levels of 50, 100 and 200 Watts. The process pressure was run at 30, 60 and 90 mTorr and the monomer flow rate run at 50, 100 and 200 microlitres of liquid per minute. The duration of the process was either 10 or 40 minutes. The coated substrate was silicon wafer and whether the coating was tacky or not was determined by wiping the substrate with a finger and determining by eye if the coating had smeared. The results in Figure 1 indicate that for lower CW powers, the coating could easily be smeared (i.e. tacky), but by increasing the power to flow ratio, the coating became resistant to smearing (i.e. non-tacky) and this is expected to be due to additional crosslinking caused by more monomer fragmentation. The water contact angle was also seen to decline with increasing power to flow ratio, and this too is indicative of more power increasing the level of monomer fragmentation.

The same 3 litre chamber was used to also look at the effect of power to flow ratios with pulsed plasma polymerisation. In these experiments, the peak power was either 250 or 500 W. The RF power feed was pulsed with an on time of 35 microseconds and off times of 5.5 and 1 millisecond. Process pressure was either 20, 35, 60 or 80 mTorr. The monomer flow was either 50 or 200 microlitres of liquid per minute. Figure 2 shows that low power to flow conditions can generate coatings that can be smeared (tacky) and only by increasing the power could the smear be reduced. This too had a concomitant decline in water contact angle.

The results from Figures 1 and 2 show that plasma processing PFAC6 on its own could only lead to smear –free coatings when the powers were high enough to decrease the other desirable property of the water contact angle. This illustrates the need to add a crosslinker to prevent the coating smearing and maintain the water contact angle.

Example 2

Examples of Perfluoro Alkyl Acrylate co-polymerisation with Crosslinker to show improvements in electrical resistance

Perfluorooctyl acrylate monomer was mixed with a single crosslinker from the following list:

VINYL ESTERS: Divinyl adipate (DVA)

VINYL ETHERS: 1,4-Butanediol divinyl ether (BDVE); 1,4 Cyclohexanedimethanol divinyl ether (CDDE)

DI- or TRI- VINYLS: 1,7-Octadiene (1,7-OD); 1,2,4-Trivinylcyclohexane (TVCH);

DI VINYL with alkyl fluoro group: 1,6-divinylperfluorohexane (DVPFH)

DI VINYL with silicon group: 1,3-Divinyl tetramethyldisiloxane (DVTMDS)

DI VINYL with cyclic ring and carboxylate groups: Diallyl 1,4-cyclohexanedicarboxylate (DCHC)

DI ACRYLATE: 1H1H,6H,6H-perfluorohexanediol diacrylate (PFHDA)

The crosslinker was mixed with the PFAC6 as a percentage by volume for the following percentages: 20, 40, 60, 80, and 100%.

Plasma initiated polymerisation reactions were carried out in a 3 litre glass plasma chamber. The substrates were test circuit boards. The PFAC6/crosslinker mixture was introduced at a rate of 0.04 – 0.06 ul/min and the process pressure was 40 mTorr. The process plasma consisted of a 1 minute continuous plasma (CW) step of 50W followed by 10 - 20 minute pulsed plasma (PW) step of 50W and RF power being delivered by a pulse sequence with a duty cycle of 6.9%. The coated circuit boards were immersed in tap water and an 8V potential was applied for 13 minutes. The final current readings, together with coating thickness measurements, were used to calculate the electrical resistance per nm of coating. Figure 5 below shows the effect of adding a range of crosslinkers on the electrical resistance per nm of the barrier coatings. Figure 5 clearly shows the improvement in resistance that can be gained by adding the crosslinker to the PFAC6 monomer. Furthermore, all of the crosslinkers tested gave non-smear/non-tacky coatings with concentration levels of 20% and above, with the exceptions of DVTMDS, DCHD, PFHDA and DVPFH which were non-smear at levels of 40% and above.

Example 3

Example of crosslinker concentration on water repellent coating.

Water repellent coatings were prepared in a 125 litre volume chamber using PFAC6 with different levels of divinyl adipate (DVA) and using helium as a carrier gas. The deposition process consisted of a 3 minute CW step with 300W power and a pulsed step with 150W power and an RF pulse duty cycle of 0.018%. Silicon wafer was used as the test substrates and the contact angle of the coated wafer was determined by applying a 3ul drop of deionised wafer onto the coated wafer and measuring the contact angle using a VCA Optima (AST products) with image analysis software. The variation of contact angle with (v/v)% DVA crosslinker is shown in Figure 4. It was also noted that for the 10 (v/v) % DVA concentration, the water drop left a mark on the coated wafer where it's circumference had been. This indicated that a higher level of crosslinker was required to give a more stable coating. This observation, in combination with the results in figure 4, suggest an optimum range of 20 – 40 (v/v) % DVA crosslinker, though this is likely to vary, typically within the range of from 10-60 (v/v) % for different monomers and crosslinkers and different chamber sizes.

Pulsed plasma polymerisation of perfluoro alkyl acrylate in the presence of divinyl adipate (DVA) was carried out in a 125 litre chamber. Figure 4 shows the effect of adding different liquid volume percentages of DVA crosslinker to perfluorooctyl acrylate on water contact angle on a silicon wafer substrate.

Example 4

Example of AFM measurement of monomer/crosslinker co-polymer coating

A barrier-style coating was prepared in a 22 litre volume chamber using PFAC6 with 10% DVA. The deposition process consisted of a 1 minute CW step with a CW power to flow ratio of 3.9 (W/microliter/min), and a pulsed step with a PW power/monomer flow ratio of 0.28 (W/microlitres/min). Figure 6 shows the representative topographical and phase contrast images were obtained from all the samples. High spatial resolution images show mainly the structures of areas between the raised features. The RMS roughness of the coating is 1.65 nm and the $\Delta z/d = 0.05$. As the $\Delta z/d$ value is < 0.15, then this indicates that the physical layer is substantially pinhole free.

Example 5

Example of FTIR/ATR measurement of monomer/crosslinker co-polymer coating

Two barrier-style coatings were prepared in a 22 litre volume chamber using PFAC6 only and PFAC6 with 10% DVA as described in example 4. The deposition process consisted of a 1 minute CW step with a CW power to flow ratio of 3.9 (W/microliter/min), and a pulsed step with a PW power/monomer flow ratio of 0.28 (W/microlitres/min). Figure 7 shows the representative FTIR/ATR spectrum obtained from both samples.

The FTIR/ATR intensity ratios of peaks attributed to stretching mode of CF_3 and C=O groups, $CF_3/C=O$, of the coating is indicative of sufficient cross linking in the coating to form a physical barrier. CF_3 refers to the terminal groups in the side chain of PFAC6.

Formation of the barrier coatings is believed to be caused by a mixture of cross linking and controlled fragmentation of the monomers during polymerisation. Cross linking is believed to be predominantly via the CF2-CF3 chain, whilst fragmentation is thought to be predominantly through loss of the C=O group during polymerisation and to a lesser extent shortening of the CF2 chain. Cross linking effects the abundance of –CF3 groups in the coating and controlled fragmentation controls the amount of C=O groups in the coating. The ratio of these two functional groups is an indication that sufficient cross-linking and fragmentation has taken place and can be measured by the ratio of the intensities of the corresponding FTIR/ATR peaks.

It has been shown that coatings with reduced CF₃ / C=O ratios give higher resistance values in the electrical test, as described in Example 2, showing an improved coating performance on increased cross linking (for CF₃) and fragmentation (for C=O). If the FTIR/ATR intensity ratio of the peaks attributed to -CF₃ stretching and C=O stretching, CF₃/C=O, is less than 0.6e^{-0.1n} (where n=6 for PFAC6) the resistance of the coating at the electrical test is expected to be higher than 8MOhms.

The peak intensity ratio CF₃/C=O the coating thickness and the final current readings during the electrical test are shown in table 2:

	Thickness (nm)	Resistance in water at 8V	CF ₃ /C=O ATR peak intensity ratio
PFAC6	1480	2.1x10 ⁷	0.23
PFAC6 and			
(in volume)	1450	9.2x10 ⁹	0.11

Table 2

Claims:

1. An electronic or electrical device or component thereof comprising a protective cross-linked polymeric coating on a surface of the electronic or electrical device or component thereof; wherein the protective cross-linked polymeric coating is obtainable by exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow formation of the protective cross-linked polymeric coating on a surface thereof,

wherein the monomer compound has the following formula:

$$R_1$$
 R_2
 R_4

where R₁, R₂ and R₄ are each independently selected from hydrogen, optionally substituted branched or straight chain C₁-C₀ alkyl or halo alkyl or aryl optionally substituted by halo, and R₃ is selected from:

where each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C₁-C₆ alkyl, halo alkyl or aryl optionally substituted by halo; and n₁ is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moieties and has a boiling point of less than 500 °C at standard pressure.

- 2. An electronic or electrical device or component thereof according to claim 1, wherein the protective cross-linked polymeric coating is a physical barrier to mass and electron transport.
- 3. An electronic or electrical device or component thereof according to claim 1 or claim 2, wherein the protective cross-linked polymeric coating forms a liquid repellent surface defined by a static water contact angle (WCA) of at least 90°.
- An electronic or electrical device or component thereof according to any preceding claim, wherein the crosslinking reagent has one of the following structures:

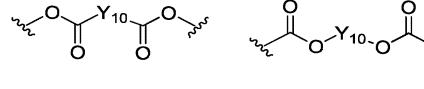
where Y_1 , Y_2 , Y_3 , Y_4 , Y_5 , Y_6 , Y_7 and Y_8 are each independently selected from hydrogen, optionally substituted cyclic, branched or straight chain C_1 - C_6 alkyl or aryl; and L is a linker moiety.

5. An electronic or electrical device or component thereof according to claim 4, wherein for compound (i) L has the following formula:

each Y_9 is independently selected from, a bond, -O-, -O-C(O)-, -C(O)-O-, -Y₁₁-O-C(O)-, -C(O)-O-Y₁₁-, -OY₁₁-, and -Y₁₁O-, where Y₁₁ is an optionally substituted cyclic, branched or straight chain C₁-C₈ alkylene; and

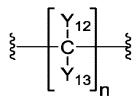
 Y_{10} is selected from optionally substituted cyclic, branched or straight chain C_1 - C_8 alkylene and a siloxane group.

6. An electronic or electrical device or component thereof according to claim 5, wherein for compound (i) L has one of the following structures:



7. An electronic or electrical device or component thereof according to claim 5, wherein for compound (i) L has one of the following structures:

8. An electronic or electrical device or component thereof according to any of claims 5 to 7, wherein Y_{10} has the following formula:



wherein each Y_{12} and Y_{13} is independently selected from H, halo, optionally substituted cyclic, branched or straight chain alkyl, or $-OY_{14}$, where Y_{14} is selected from optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl, and n is an integer from 1 to 10.

- 9. An electronic or electrical device or component thereof according to claim 8, wherein each Y_{12} is H and each Y_{13} is H.
- 10. An electronic or electrical device or component thereof according to claim 8, wherein each Y_{12} is fluoro and each Y_{13} is fluoro.
- 11. An electronic or electrical device or component thereof according to any of claims 8 to 10, wherein n is from 4 to 6.
- 12. An electronic or electrical device or component thereof according to any of claims 5 to 7, wherein Y_{10} has the following formula:

Wherein each Y_{15} is independently selected from optionally substituted branched or straight chain C_1 - C_5 alkyl.

- 13. An electronic or electrical device or component thereof according to claim 12, wherein each Y_{15} is methyl, and each Y_9 is a bond.
- 14. An electronic or electrical device or component thereof according to any of claims 5 to 7, wherein Y_{10} has the following formula:

wherein Y_{16} to Y_{19} are each independently selected from H and optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl.

- 15. An electronic or electrical device or component thereof according to claim 14, wherein Y_{18} is H or vinylene, and Y_{16} , Y_{17} and Y_{19} are each H.
- 16. An electronic or electrical device or component thereof according to any preceding claim wherein the crosslinking reagent is selected from Divinyl Adipate (DVA), 1,4 Butanediol divinyl ether (BDVE), 1,4 Cyclohexanedimethanol divinyl ether (CDDE), 1,7 -Octadiene (17OD), 1,2,4 Trivinylcyclohexane (TVCH), 1,3-Divinyltetramethyldisiloxane (DVTMDS), Diallyl 1,4-cyclohexanedicarboxylate (DCHD), 1,6-Divinylperfluorohexane (DVPFH), 1H,1H,6H,6H-Perfluorohexanediol diacrylate (PFHDA) and glyoxal bis (diallyl acetal) (GBDA).
- 17. An electronic or electrical device or component thereof according to claim 4, wherein for compound (ii), L is selected from a branched or straight chain C₁-C₈ alkylene or an ether group.
- 18. An electronic or electrical device or component thereof according to any preceding claim wherein the monomer is a compound of formula I(a):

I(a)

wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_6 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1;

; or

wherein the monomer is a compound of formula I(b):

I(b)

$$R_{2} \xrightarrow[R_{4}]{R_{1}} \xrightarrow[O]{R_{5}} \xrightarrow[R_{6}]{R_{7}} \left[\begin{array}{c} R_{9} \\ \dot{C} \\ R_{10} \end{array} \right]_{a} \left[\begin{array}{c} X \\ \dot{C} \\ X \end{array} \right]_{b} \left[\begin{array}{c} X \\ \dot{C} \\ \dot{C} \\ X_{3} \end{array} \right]_{c} CX_{3}$$

wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_6 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1.

- 19. An electronic or electrical device or component thereof according to claim 18, wherein the halogen is fluorine.
- 20. An electronic or electrical device or component thereof according to claim 18 or claim 19, wherein each of R₁, R₂, R₄, and R₅ to R₁₀ is independently selected from hydrogen, methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, tert-butyl, sec-butyl, n-pentyl, neopentyl, n-hexyl, isohexyl, and 3-methylpentyl.
- 21. An electronic or electrical device or component thereof according to claim 20, wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or methyl.
- 22. An electronic or electrical device or component thereof according to any of claims 18 to 20, wherein a and c are each independently 0 or 1; and b is from 3 to 7.
- 23. An electronic or electrical device or component thereof according to any preceding claim wherein each X is H.
- 24. An electronic or electrical device or component thereof according to any of claims 1 to 22, wherein each X is F.
- 25. An electronic or electrical device or component thereof according to any of claims 18 to 24, wherein R_1 and R_2 are both hydrogen.
- 26. An electronic or electrical device or component thereof according to any of claims 18 to 25, wherein R₄ is hydrogen or methyl.
- 27. An electronic or electrical device or component thereof according to any of claims 18 to 26, wherein R_9 is hydrogen and R_{10} is branched or straight chain C_1 - C_6 alkyl group.
- 28. An electronic or electrical device or component thereof according to claim 27, wherein R₁₀ is methyl.
- 29. An electronic or electrical device or component thereof according to any of claims 18 to 28, wherein each of R_5 to R_8 is hydrogen.
- 30. An electronic or electrical device or component thereof according to any of claims 18 to 29 wherein each of each of R₁, R₂, R₄, and R₅ to R₁₀ is hydrogen, each X is H, a=0 and c=0.
- 31. An electronic or electrical device or component thereof according to any of claims 24 to 30, wherein the compound of formula I(a) has the following formula:

$$H \longrightarrow O$$
 $(CF_2)_nF$

where n is from 2 to 10.

32. An electronic or electrical device or component thereof according to any of claims 24 to 30, wherein the compound of formula I(a) has the following formula:

$$H \longrightarrow O (CF_2)_n F$$
 CH_3

where n is from 2 to 10.

- 33. An electronic or electrical device or component thereof according to claim 31, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-perfluorohexyl acrylate (PFAC4), 1H,1H,2H,2H-perfluorooctyl acrylate (PFAC6), 1H,1H,2H,2H-perfluorodecyl acrylate (PFAC8) and 1H,1H,2H,2H-perfluorododecyl acrylate (PFAC10).
- 34. An electronic or electrical device or component thereof according to claim 32, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-pefluorohexyl methacrylate (PFMAC4), 1H,1H,2H,2H-perfluorooctyl methacrylate (PFMAC6) and 1H,1H,2H,2H-perfluorodecyl methacrylate (PFMAC8).
- 35. An electronic or electrical device or component thereof according to any of claims 18 to 23, wherein the compound of formula I(a) has the following formula:

wherein a and c are each independently 0 or 1, b =3-7 and n is 4 to 10, where n=a+b+c+1.

36. An electronic or electrical device or component thereof according to any of claims 18 to 23, wherein the compound of formula I(a) has the following formula:

$$H$$
 O
 $CH_2)_nH$
 CH_3

where n is 2 to 12.

- 37. An electronic or electrical device or component thereof according to claim 35 or claim 36, wherein the compound of formula I(a) is selected from ethyl hexyl acrylate, hexyl acrylate, decyl acrylate, lauryl dodecyl acrylate and iso decyl acrylate.
- 38. An electronic or electrical device or component thereof according to any of claims 18 to 23, wherein the monomer is a compound of formula I(b) and has the following formula:

$$\begin{array}{c}
R_1 & O \\
R_2 & R_3
\end{array} (CH_2)_n - CH_3$$

where n is from 4 to 14, optionally where R_1 , R_2 and R_3 are each H.

- 39. An electronic or electrical device or component thereof according to any of the preceding claims, wherein the electronic or electrical device or component thereof is selected from mobile phones, smartphones, pagers, radios, sound and audio systems such as loudspeakers, microphones, ringers and/or buzzers, hearing aids, personal audio equipment such as personal CD, tape cassette or MP3 players, televisions, DVD players including portable DVD players, video recorders, digi and other set-top boxes, computers and related components such as laptop, notebook, tablet, phablet or palmtop computers, personal digital assistants (PDAs), keyboards, or instrumentation, games consoles, data storage devices, outdoor lighting systems, radio antennae and other forms of communication equipment, and printed circuit boards.
- 40. A method for treating an electronic or electrical device or component thereof comprising:

exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow a protective cross-linked polymeric coating to form on a surface of the electronic or electrical device or component thereof;

wherein the monomer compound has the following formula:

$$R_1$$
 R_3 R_2 R_4

where R_1 , R_2 and R_4 are each independently selected from hydrogen, optionally substituted branched or straight chain C_1 - C_6 alkyl or halo alkyl or aryl optionally substituted by halo, and R_3 is selected from:

$$CX_3 = CX_3$$

$$CX_3 = CX_3$$

$$CX_3 = CX_3$$

$$CX_3 = CX_3$$

where each X is independently selected from hydrogen, optionally substituted branched or straight chain C₁-C₆ alkyl, halo alkyl or aryl optionally substituted by halo; and n₁ is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moieties and has a boiling point at standard pressure of less than 500°C.

41. A method according to claim 40, wherein the crosslinking reagent has one of the following structures:

$$Y_2$$
 Y_1 Y_4 Y_5 Y_5 Y_6 Y_7 Y_7 Y_7 Y_8

where Y_1 , Y_2 , Y_3 , Y_4 , Y_5 , Y_6 , Y_7 and Y_8 are each independently selected from hydrogen, optionally substituted cyclic, branched or straight chain C_1 - C_6 alkyl or aryl; and L is a linker moiety.

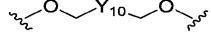
42. A method according to claim 41, wherein for compound (i) L has the following formula:

each Y_9 is independently selected from, a bond, -O-, -O-C(O)-, -C(O)-O-, -Y₁₁-O-C(O)-, -C(O)-O-Y₁₁-, -OY₁₁-, and -Y₁₁O-, where Y₁₁ is an optionally substituted cyclic, branched or straight chain C₁-C₈ alkylene; and

 Y_{10} is selected from optionally substituted cyclic, branched or straight chain C_1 - C_8 alkylene and a siloxane group.

43. A method according to claim 42, wherein for compound (i) L has one of the following structures:

44. A method according to claim 42, wherein for compound (i) L has one of the following structures:



45. An method according to any of claims 42 to 44, wherein Y₁₀ has the following formula:

$$\begin{cases} -\begin{bmatrix} Y_{12} \\ C \end{bmatrix} \\ Y_{13} \end{bmatrix}_{n} \end{cases}$$

wherein each Y_{12} and Y_{13} is independently selected from H, halo, optionally substituted cyclic, branched or straight chain alkyl, or -OY₁₄, where Y₁₄ is selected from optionally substituted branched or straight chain C₁-C₈ alkyl or alkenyl, and n is an integer from 1 to 10.

- A method according to claim 45, wherein each Y₁₂ is H and each Y₁₃ is H. 46.
- 47. A method according to claim 45, wherein each Y_{12} is fluoro and each Y_{13} is fluoro.
- 48. A method according to any of claims 45 to 47, wherein n is from 4 to 6.
- A method according to any of claims 42 to 44, wherein Y₁₀ has the following formula: 49.

$$\begin{cases}
- \begin{cases}
Y_{15} \\
S_{1} \\
Y_{15}
\end{cases}
- \begin{cases}
Y_{15} \\
Y_{15}
\end{cases}$$

Wherein each Y₁₅ is independently selected from optionally substituted branched or straight chain C₁-C₈ alkyl.

- 50. A method according to claim 49, wherein each Y_{15} is methyl, and each Y_{9} is a bond.
- 51. A method according to any of claims 42 to 44, wherein Y_{10} has the following formula:

wherein Y_{16} to Y_{19} are each independently selected from H and optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl.

- 52. An method according to claim 51, wherein Y_{18} is vinylene or H, and Y_{16} , Y_{17} and Y_{19} are each H.
- 53. A method according to any of claims 40 to 52, wherein the crosslinking reagent is selected from Divinyl Adipate (DVA), 1,4 Butanediol divinyl ether (BDVE), 1,4 Cyclohexanedimethanol divinyl ether (CDDE), 1,7 -Octadiene (17OD), 1,2,4 -Trivinylcyclohexane (TVCH), 1,3-Divinyltetramethyldisiloxane (DVTMDS), Diallyl 1,4-cyclohexanedicarboxylate (DCHD), 1,6-Divinylperfluorohexane (DVPFH), 1H,1H,6H,6H-Perfluorohexanediol diacrylate (PFHDA) and glyoxal bis (diallyl acetal) (GBDA).
- 54. A method according to claim 41, wherein for compound (ii), L is selected from a branched or straight chain C₁-C₈ alkylene or an ether group.
- 55. A method according to any of claims 40 to 54 wherein the monomer is a compound of formula l(a):

I(a)

wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_6 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1;

; or

wherein the monomer is a compound of formula I(b):

I(b)

$$R_{2} \xrightarrow[R_{4}]{R_{1}} \xrightarrow[O]{R_{5}} \xrightarrow[R_{6}]{R_{7}} \left[\begin{array}{c} R_{9} \\ \dot{C} \\ \dot{R}_{10} \end{array} \right]_{a} \left[\begin{array}{c} X \\ \dot{C} \\ \dot{X} \end{array} \right]_{b} \left[\begin{array}{c} X \\ \dot{C} \\ \dot{C} \\ \dot{X} \end{array} \right]_{c} CX_{3}$$

wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_6 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1.

- 56. A method according to claim 55, wherein the halogen is fluorine.
- 57. A method according to claim 55 or claim 56, wherein each of R₁, R₂, R₄, and R₅ to R₁₀ is independently selected from hydrogen, methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, tert-butyl, sec-butyl, n-pentyl, neopentyl, n-hexyl, isohexyl, and 3-methylpentyl.
- 58. A method according to claim 57, wherein each of R₁, R₂, R₄, and R₅ to R₁₀ is independently selected from hydrogen or methyl.
- 59. A method according to any of claims 55 to 58, wherein a and c are each independently 0 or 1; and b is from 3 to 7.
- 60. A method according to any preceding claim wherein each X is H.
- 61. A method according to any of claims 41 to 59, wherein each X is F.
- 62. A method according to any of claims 41 to 61, wherein R_1 and R_2 are both hydrogen.
- 63. A method according to any of claims 41 to 62, wherein R₄ is hydrogen or methyl.
- 64. A method according to any of claims 41 to 63, wherein R_9 is hydrogen and R_{10} is branched or straight chain C_1 - C_8 alkyl group.
- 65. A method according to claim 64, wherein R₁₀ is methyl.
- 66. A method to any of claims 41 to 65, wherein each of R_5 to R_8 is hydrogen.
- 67. A method according to any of claims 41 to 66 wherein each of each of R_1 , R_2 , R_4 , and R_5 to R_{10} is hydrogen, each X is H, a=0 and c=0.
- 68. A method according to any of claims 61 to 67, wherein the compound of formula I(a) has the following formula:

where n is from 2 to 10.

69. A method according to any of claims 61 to 67, wherein the compound of formula I(a) has the following formula:

$$H \longrightarrow O$$
 CH_3
 $(CF_2)_nF$

where n is from 2 to 10.

- 70. A method according to claim 68, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-perfluorohexyl acrylate (PFAC4), 1H,1H,2H,2H-perfluorooctyl acrylate (PFAC6), 1H,1H,2H,2H-perfluorodecyl acrylate (PFAC8) and 1H,1H,2H,2H-perfluorododecyl acrylate (PFAC10).
- 71. A method according to claim 69, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-perfluorohexyl methacrylate (PFMAC4), 1H,1H,2H,2H-perfluorooctyl methacrylate (PFMAC6) and 1H,1H,2H,2H-perfluorodecyl methacrylate (PFMAC8).
- 72. A method according to any of claims 41 to 61, wherein the compound of formula I(a) has the following formula:

wherein a and c are each independently 0 or 1, b =3-7 and n is 4 to 10, where n=a+b+c+1.

73. A method according to any of claims 41 to 61, wherein the compound of formula I(a) has the following formula:

$$\begin{array}{c} H \\ \\ \\ \\ H \end{array} \begin{array}{c} O \\ \\ CH_3 \end{array} (CH_2)_n H \\ \end{array}$$

where n is 2 to 12.

- 74. A method according to claim 72 or claim 73, wherein the compound of formula I(a) is selected from ethyl hexyl acrylate, hexyl acrylate, decyl acrylate, lauryl dodecyl acrylate and iso decyl acrylate.
- 75. A method according to any of claims 41 to 60, wherein the monomer is a compound of formula I(b) and has the following formula:

where n is from 4 to 14.

76. A method according to claim 75 wherein the monomer is a compound of formula I(b) and has the following formula:

$$\begin{array}{c}
H \longrightarrow O \longrightarrow (CH_2)_n \longrightarrow CH_3
\end{array}$$

where n is from 4 to 14.

- 77. A method according to any of claims 40 to 76 wherein the electronic or electrical device or component thereof is placed in a plasma deposition chamber, a glow discharge is ignited within said chamber, and a voltage applied as a pulsed field.
- 78. A method according to claim 77 wherein applied voltage is at a power of from 40 to 500W.
- 79. A method according to claim 77 or claim 78 wherein the voltage is pulsed in a sequence in which the ratio of the time on : time off is in the range of from 1:500 to 1:1500.
- 80. A method according to any of claims 77 to 79 wherein the voltage is pulsed in a sequence where power is on for 20-50μs, and off for from 1000μs to 30000μs.
- 81. A method according to any of claims 77 to 80 wherein the voltage is applied as a pulsed field for a period of from 30 seconds to 90 minutes.
- 82. A method according to claim 81 wherein the voltage is applied as a pulsed field for from 5 to 60 minutes.
- 83. A method according to any one of claims 77 to 82, wherein in a preliminary step, a continuous power plasma is applied to the electronic or electrical device or component thereof.

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Tack FreeTacky

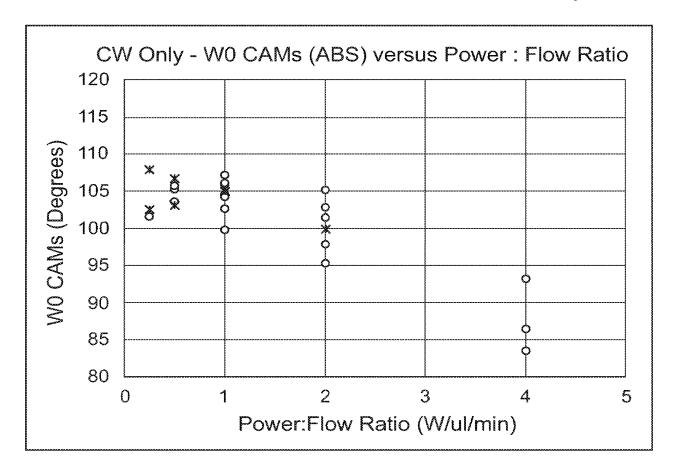


Figure 1

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Tack Free* Tacky

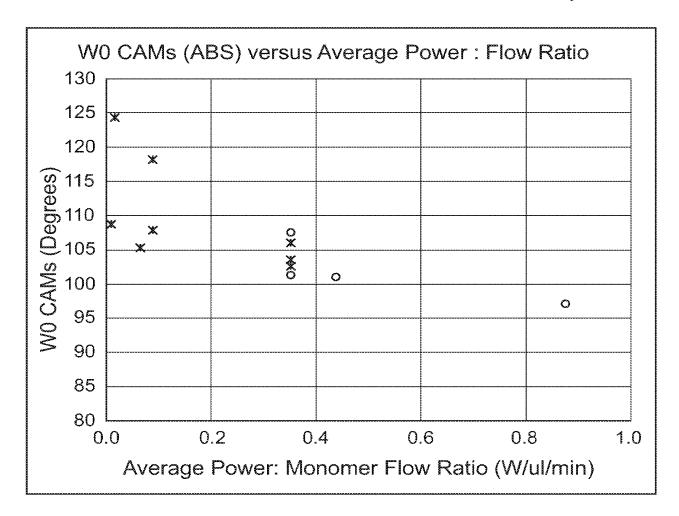
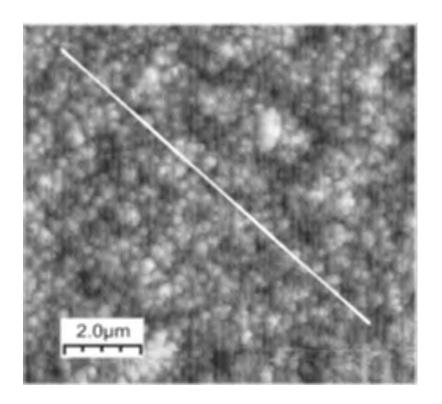


Figure 2

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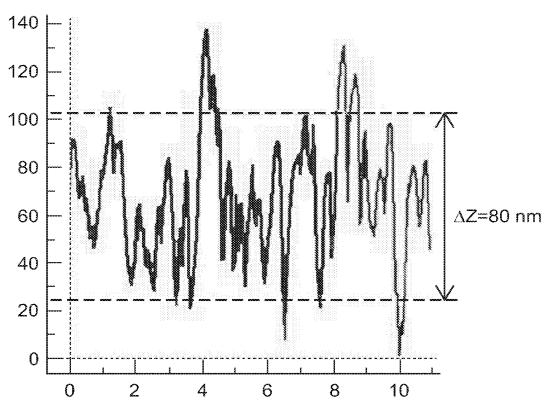


Figure 3

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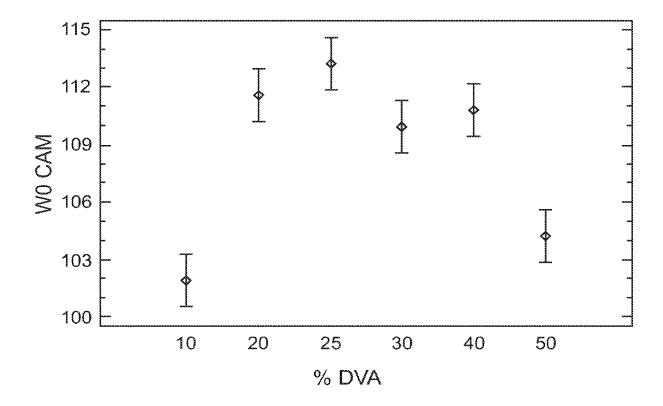
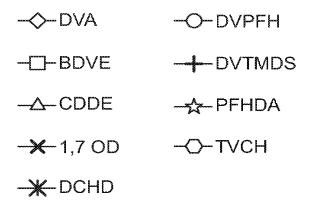


Figure 4

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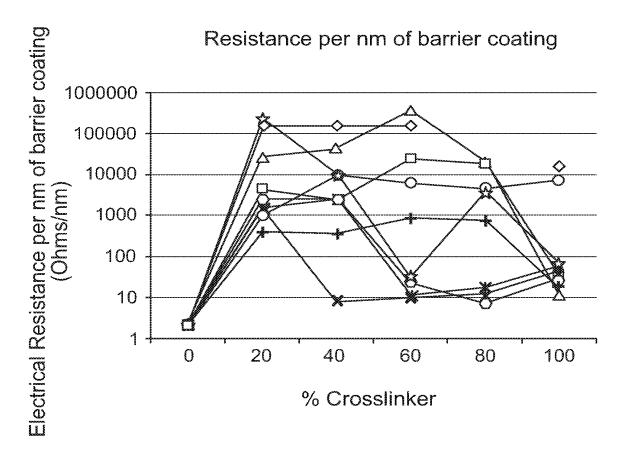
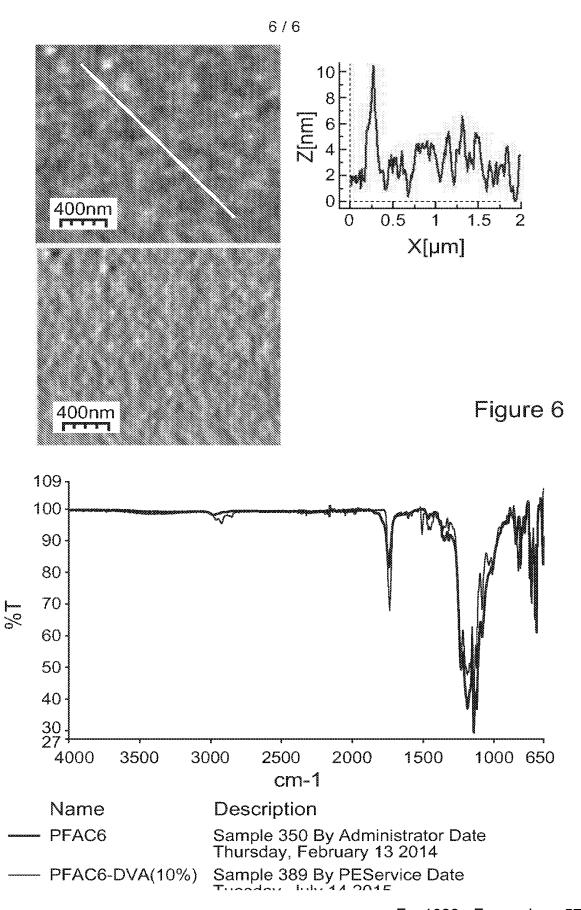


Figure 5



INTERNATIONAL SEARCH REPORT

International application No

PCT/GB2016/051688 a. classification of subject matter INV. C09D5/00 ADD. According to International Patent Classification (IPC) or to both national classification and IPC B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) C09D Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) EPO-Internal, WPI Data C. DOCUMENTS CONSIDERED TO BE RELEVANT Relevant to claim No. Category* Citation of document, with indication, where appropriate, of the relevant passages WO 2007/083122 A1 (P2I LTD [GB]; COULSON STEPHEN [GB]) 26 July 2007 (2007-07-26) 1-91 Υ claims 10, 11, 27; example 1 Υ EP 0 834 352 A1 (CIBA GEIGY AG [CH]) 1-91 8 April 1998 (1998-04-08) claim 8; page 5 line 59 - page 6 lines 1-2; example B-14 WO 2014/155099 A1 (SEMBLANT LTD [GB]) 1-91 Α 2 October 2014 (2014-10-02) claims 1, 5 together with examples 1, 2

X

See patent family annex.

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Date of the actual completion of the international search Date of mailing of the international search report 28 October 2016 08/11/2016 Name and mailing address of the ISA/ Authorized officer

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Kositza, Matthias

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INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No
PCT/GB2016/051688

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V		rson to Whom the Inventor is Und				applicant is a iuris	tic entity)
Ō	Person Who Ot	herwise Shows Sufficient Propriet concurrently being filed with this	ary Interest (e.g.,	a petition und	er 37 CFR 1.46(b)	(2) was granted in	
			URE of Applica	444			
The t	undersigned (who	se title is supplied below) is authoriz	zed to act on beha	If of the applica	nt (e.g., where the	applicant is a juristi	c entity).
Signa	ature	K.C.		Date (Optional) 26 Octob	er 2016	
Nam		Dr. Kenneth Armour					
Title		In-House Counsel			, P2l l	.TD	
NOT and c	E: Signature - The certifications. If mo	is form must be signed by the applic ore than one applicant, use multiple	cant in accordance forms.	with 37 CFR 1	.33. See 37 CFR 1	.4 for signature req	uirements
Total	of 1	forms are submitted					*, *

This collection of information is required by 37 CFR 1.131, 1.32, and 1.33. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.11 and 1.14. This collection is estimated to take 3 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

If you need assistance in completing the form, call 1-800-PTO-9199 and select option 2.

PATENT COOPERATION TREATY

PCT

INTERNATIONAL SEARCH REPORT

(PCT Article 18 and Rules 43 and 44)

Applicant's or agent's file reference	FOR FURTHER	see Form PCT/ISA/220
P2494PC00	7.01.01.	ell as, where applicable, item 5 below.
International application No.	International filing date (day/month/year)	(Earliest) Priority Date (day/month/year)
PCT/GB2016/051688	8 June 2016 (08-06-2016)	9 June 2015 (09-06-2015)
Applicant	-	-
P2I LTD		
This international search report has been according to Article 18. A copy is being tra	prepared by this International Searching Aut Insmitted to the International Bureau.	hority and is transmitted to the applicant
This international search report consists o	f a total ofsheets.	
X It is also accompanied by	a copy of each prior art document cited in th	is report.
Basis of the report		
	nternational search was carried out on the b	
	upplication in the language in which it was file international application into	eq , which is the language
	rnished for the purposes of international sea	
	report has been established taking into acco o this Authority under Rule 91 (Rule 43.6 <i>bis</i> i	unt the rectification of an obvious mistake a)).
c. With regard to any nucle c	otide and/or amino acid sequence disclose	ed in the international application, see Box No. I.
2. Certain claims were fou	nd unsearchable (See Box No. II)	
3. Unity of invention is lack	king (see Box No III)	
4. With regard to the title ,		
X the text is approved as su	bmitted by the applicant	
the text has been establis	hed by this Authority to read as follows:	
5. With regard to the abstract,		
X the text is approved as su	• • • • • • • • • • • • • • • • • • • •	
	hed, according to Rule 38.2, by this Authorit in the date of mailing of this international se	y as it appears in Box No. IV. The applicant arch report, submit comments to this Authority
6. With regard to the drawings ,		
3	ublished with the abstract is Figure No	
as suggested by t	he applicant	
as selected by this	s Authority, because the applicant failed to s	uggest a figure
as selected by this	s Authority, because this figure better charac	sterizes the invention
b. X none of the figures is to be	e published with the abstract	

INTERNATIONAL SEARCH REPORT

International application No

PCT/GB2016/051688 A. CLASSIFICATION OF SUBJECT MATTER INV. C09D5/00 ADD. According to International Patent Classification (IPC) or to both national classification and IPC **B. FIELDS SEARCHED** Minimum documentation searched (classification system followed by classification symbols) C09D Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) EPO-Internal, WPI Data C. DOCUMENTS CONSIDERED TO BE RELEVANT Category* Citation of document, with indication, where appropriate, of the relevant passages Relevant to claim No. WO 2007/083122 A1 (P2I LTD [GB]; COULSON STEPHEN [GB]) 26 July 2007 (2007-07-26) Υ 1-91 claims 10, 11, 27; example 1 EP 0 834 352 A1 (CIBA GEIGY AG [CH]) Υ 1-91 8 April 1998 (1998-04-08) claim 8; page 5 line 59 - page 6 lines 1-2; example B-14 WO 2014/155099 A1 (SEMBLANT LTD [GB]) 1-91 Α 2 October 2014 (2014-10-02) claims 1, 5 together with examples 1, 2 and 3 IX I Further documents are listed in the continuation of Box C. See patent family annex. Special categories of cited documents:

- "A" document defining the general state of the art which is not considered to be of particular relevance
- "E" earlier application or patent but published on or after the international filing date
- "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- "O" document referring to an oral disclosure, use, exhibition or other
- document published prior to the international filing date but later than the priority date claimed
- "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
- "&" document member of the same patent family

Kositza, Matthias

Date of the actual completion of the international search Date of mailing of the international search report 28 October 2016 08/11/2016 Name and mailing address of the ISA/ Authorized officer European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Fax: (+31-70) 340-3016

3

INTERNATIONAL SEARCH REPORT

Information on patent family members

International application No
PCT/GB2016/051688

		· · · · · · · · · · · · · · · · · · ·	
Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO 2007083122 A	26-07-2007	AU 2007206779 A1 CA 2637625 A1 CN 101370877 A CN 103396688 A EP 1976942 A1 GB 2434369 A IL 192905 A JP 2009523914 A KR 20080097201 A KR 20140063887 A NZ 570462 A TW I400014 B US 2010189914 A1 WO 2007083122 A1 ZA 200806846 B	26-07-2007 26-07-2007 18-02-2009 20-11-2013 08-10-2008 25-07-2007 27-06-2013 25-06-2009 04-11-2008 27-05-2014 26-11-2010 21-06-2013 29-07-2010 26-07-2007 25-11-2009
EP 0834352 A	08-04-1998	AR 004877 A1 EP 0834352 A1 US 6169127 B1	10-03-1999 08-04-1998 02-01-2001
WO 2014155099 A	02-10-2014	CN 105075408 A EP 2979524 A1 JP 2016521296 A KR 20150135420 A TW 201446890 A WO 2014155099 A1	18-11-2015 03-02-2016 21-07-2016 02-12-2015 16-12-2014 02-10-2014

Information on Search Strategy - Pilot phase (see OJ 2015, A86) The type of information contained in this sheet may change during the pilot for improving the usefulness of this new service.

Application Number

PCT/GB2016/051688

TITLE: COATINGS

APPLICANT: P2I LTD

IPC CLASSIFICATION: CO9D5/00

EXAMINER: Kositza, Matthias

CONSULTED DATABASES: EPODOC, WPI, NPL

CLASSIFICATION SYMBOLS DEFINING EXTENT OF THE SEARCH:

IPC:

CPC: C09D5/++

FI/F-TERMS:

KEYWORDS OR OTHER ELEMENTS FEATURING THE INVENTION:

PO FORM P04A42

PATENT COOPERATION TREATY

From the INTERNATIONAL SEARCHING AUTHORITY

То:		PCT				
see form PCT/ISA/220		WRITTEN OPINION OF THE INTERNATIONAL SEARCHING AUTHORITY (PCT Rule 43 <i>bis</i> .1)				
			Date of mailing (day/month/yea	•	sheet)	
Applicant's or agent's file resee form PCT/ISA/220			FOR FURT	HER ACTION n 2 below		
International application No PCT/GB2016/051688).	International filing date (day/month/year)	Priority date (day/month/ye 09.06.2015	ear)	
International Patent Classif INV. C09D5/00	ication (IPC) or l	both national classification	and IPC	1		
Applicant P2I LTD						
Box No. I BOX No. II BOX No. II BOX No. III BOX No. IV BOX No. VI BOX No. VI BOX No. VII BOX No. VIII BOX NO.	Basis of the operiority Non-establishmatic ack of unity of the assumed state applicability; circertain docum. Certain defects are actional prelimed in the actional prelimed and actional actio	ment of opinion with regif invention ement under Rule 43bistations and explanations ents cited is in the international approactions on the internation all Preliminary Examination is all Preliminary Examininity other than this one to 66.1 bis(b) that written cover, considered to be a y together, where approach PCT/ISA/220 or before	ard to novelty, is.1(a)(i) with regs supporting sublication and application application be the IPEA application opinions of this written opinion priate, with am	nventive step and industrial apports of the IPEA, the applicant is invited from the priority date of 22 months from the priority date of 22 months from the priority date of the	be a oply where the vertee to of 3 months	
Name and mailing address	of the ISA:	Date of c	ompletion of	Authorized Officer	weines Petenton,	
European Pa D-80298 Mul Tel. +49 89 2 Fax: +49 89	nich 2399 - 0	see form PCT/ISA/		Kositza, Matthias Telephone No. +49 89 2399-0	Total	

	Box	No. I	Basis of the opinion
1.	With	n regai	rd to the language, this opinion has been established on the basis of:
	\boxtimes	the in	ternational application in the language in which it was filed.
			slation of the international application into , which is the language of a translation furnished for the ses of international search (Rules 12.3(a) and 23.1 (b)).
2.			pinion has been established taking into account the rectification of an obvious mistake authorized notified to this Authority under Rule 91 (Rule 43 <i>bis</i> .1(a))
3.			regard to any nucleotide and/or amino acid sequence disclosed in the international application, this on has been established on the basis of a sequence listing:
		a. 🗆	forming part of the international application as filed:
			☐ in the form of an Annex C/ST.25 text file.
			□ on paper or in the form of an image file.
		b. □	furnished together with the international application under PCT Rule 13 <i>ter</i> .1(a) for the purposes of international search only in the form of an Annex C/ST.25 text file.
		c. 🗆	furnished subsequent to the international filing date for the purposes of international search only:
			☐ in the form of an Annex C/ST.25 text file (Rule 13ter.1(a)).
			□ on paper or in the form of an image file (Rule 13 <i>ter</i> .1(b) and Administrative Instructions, Section 713).
4.		the re	dition, in the case that more than one version or copy of a sequence listing has been filed or furnished, quired statements that the information in the subsequent or additional copies is identical to that application as filed or does not go beyond the application as filed, as appropriate, were hed.
5.	Add	litional	comments:

Box No. V Reasoned statement under Rule 43*bis*.1(a)(i) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

1. Statement

Novelty (N) Yes: Claims <u>1-91</u>

No: Claims

Inventive step (IS) Yes: Claims

No: Claims <u>1-91</u>

Industrial applicability (IA) Yes: Claims <u>1-91</u>

No: Claims

2. Citations and explanations

see separate sheet

item V

1. Documents

D1	WO.	2007	70831	22	Δ1
D I	VV	2007	/ 000 1	~~	\neg

D2 EP 0 834 352 A1

D3 WO 2014/155099 A1

2. Independent claims

Claim 1 relates to an electronic or electrical device or component thereof having a protective cross-linked polymeric coating made by plasma polymerization of a combination of components (1) + (2):

- (1) unsaturated monomer having the formula detailed in the claim
- (2) (poly)unsaturated cross-linking agent.

Claim 40 relates to a method for providing an electronic or electrical device or component thereof having a protective cross-linked polymeric coating made by plasma polymerization of a combination of components (1) + (2) according to claim 1.

3. Novelty

Claims 1 and 40, and the claims dependent thereon, are considered novel.

4. Inventive step

Claims 1 and 40 do not involve an inventive step. Starting from document D1 (see claims 10, 11, 27; example 1) describing the plasma polymerization of component (1) for making protective coatings for electronic / electrical components the skilled person would easily recognize by combining document D2 (see claim 8; p. 5 l. 59 - p. 6, l. 1-2, example B-14) that adding an additional cross-linking component (2) allows to vary / control the degree of cross-linking of the coating. See the application page 2, lines 1-10. At present it cannot be seen that the addition of the cross-linking component (2) is connected with any surprising beneficial technical effect which could substantiate the presence of an inventive step. The dependent claims represent trivial modifications of claims 1 and 40 which are obvious to the skilled person familiar with the present technical field.

5. Industrial applicability

The claimed subject-matter is industrially applicable.

Electronic Patent A	۱pp	lication Fee	Transmit	ttal	
Application Number:					
Filing Date:					
Title of Invention:	co	ATINGS			
First Named Inventor/Applicant Name:	Ste	phen Richard COUI	LSON		
Filer:	Pa	ul John Parins/Sawy	er Imrie		
Attorney Docket Number:	GB	L0007/US			
Filed as Large Entity					
Filing Fees for U.S. National Stage under 35 USC 371					
Description		Fee Code	Quantity	Amount	Sub-Total in USD(\$)
Basic Filing:					
NATIONAL STAGE FEE		1631	1	280	280
NATL STAGE SEARCH FEE - REPORT PROVIDED		1642	1	480	480
NATIONAL STAGE EXAM - ALL OTHER CASES		1633	1	720	720
Pages:					
Claims:					
Miscellaneous-Filing:					
OATH/DECL > 30 MOS FROM 371 COMMENCEMENT		1617	1	140	140
Petition:			F _Y 1	028 - Favore	ed - p. 71
					

Description	Fee Code	Quantity	Amount	Sub-Total in USD(\$)
Patent-Appeals-and-Interference:				
Post-Allowance-and-Post-Issuance:				
Extension-of-Time:				
Miscellaneous:				
	Tot	al in USD	(\$)	1620

Electronic Acknowledgement Receipt		
EFS ID:	31178417	
Application Number:	15735130	
International Application Number:	PCT/GB2016/051688	
Confirmation Number:	7101	
Title of Invention:	COATINGS	
First Named Inventor/Applicant Name:	Stephen Richard COULSON	
Customer Number:	33072	
Filer:	Paul John Parins/Sawyer Imrie	
Filer Authorized By:	Paul John Parins	
Attorney Docket Number:	GBL0007/US	
Receipt Date:	08-DEC-2017	
Filing Date:		
Time Stamp:	19:01:32	
Application Type:	U.S. National Stage under 35 USC 371	

Payment information:

Submitted with Payment	yes
Payment Type	CARD
Payment was successfully received in RAM	\$1620
RAM confirmation Number	121117INTEFSW19015300
Deposit Account	
Authorized User	

The Director of the USPTO is hereby authorized to charge indicated fees and credit any overpayment as follows:

File Listin	g:				
Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl
			2757957		
1	Transmittal of New Application	GBL0007US_371_Transmittal. pdf	7997e6c01eb01f494443d52e5af6ff14aea46 cc4	no	3
Warnings:	-			·	
Information:					
			4711371		
2	Application Data Sheet	GBL0007US_ADS.pdf	a71b5dee05f78c5930ebe218717a81f9aab 1238c	no	8
Warnings:	•				
Information:					
This is not an U	SPTO supplied ADS fillable form				
	Information Disclosure Statement (IDS)		1212053		
3	Form (SB08)	GBL0007US_IDS.pdf	425baf35e333440977faf12283507d2f1bee 0d65	no	4
Warnings:					
Information:					
This is not an U	SPTO supplied IDS fillable form				
			1871945		
4	Documents submitted with 371 Applications	GBL0007US_PCT_Publication. pdf	161e1fe6ab54042e3e24a799d4de1d0a60a f17f9	no	44
Warnings:	-				
Information:					
			1103599		
5	Power of Attorney	GBL0007US_POA.pdf	cad992be926ca3b3f2722ee99b5b6cb003a 6a943	no	2
Warnings:	<u> </u>		<u> </u>		
Information:					
			1526775		
6	Foreign Reference	EP0834352A1.pdf	b97ca728d51a3a6c53022fda81cfdbf67da6 c6d9	no	22
			Ex. 1028 - Fav	ored - p. 	74

Warnings:					
Information:					
7	Foreign Reference	GBL0002US_PCT_Publication. pdf	1278890 22a5c3738381a2ef0e835e442b170c7ad5b afa53	no	33
Warnings:					
Information:					
			3484345		
8	Foreign Reference	GBL0005US_PCT_Publication. pdf	69eba55ebd18d46a8fba59b48e329a6a039 698ea	no	92
Warnings:					
Information:					
			1826295		
9	Foreign Reference	GBL0006US_PCT_Publication. pdf	d9ba60957218b228167b15711b4d878414 ea725b	no	35
Warnings:		•			
Information:					
			249320	no	9
10	Non Patent Literature	NPL_ISRWO.pdf	900d3143e1f29562af8f384181cce8ff7047a fbe		
Warnings:		•			
Information:					
11	Foreign Reference	WO1998058117.pdf	984480 528f1689796e2cb1923f459e39436fdb5775	no	33
			cd98		
Warnings:					
Information:					Г
			1001835		
12	Foreign Reference	WO2003097245.pdf	fef5e661f02a1201f650b30901128ee137f30 d96	no	29
Warnings:		-			
Information:					
13	Foreign Reference	WO2003101621.pdf	808052 2e5a81f330b397c713ca98edae718fa31307 5222	no	19
Warnings:		+		<u> </u>	
Information:			Ex. 1028 - Fav	ored - p.	75

			892913		
14	Foreign Reference	WO2007083122.pdf		no	24
			4f215476b9dd0e30101f2024b1aeae47e28 95573		
Warnings:					<u> </u>
Information					
			1122008		
15	Foreign Reference	WO2014155099.pdf		no	28
			60ce32c3ec3352fa4aea72aabaf866551eea 9fd0		
Warnings:					
Information	1				
			37014		
16	Fee Worksheet (SB06)	fee-info.pdf		no	2
			7df29816925b054d8b5559a10f8c728ecce5 bf78		
Warnings:					l
Information	:				
		Total Files Size (in bytes)	248	368852	

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New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

0	For receiving Office use only		
0-1	International Application No.	PCT/GB2016/051688	
0-2	International Filing Date	08 June 2016 (08.06.2016)	
0-3	Name of receiving Office and "PCT International Application"	RO/GB	
0-4	Form PCT/RO/101 PCT Request		
0-4-1	Prepared Using	PCT Online Filing Version 3.5.000.244e MT/FOP 20141031/0.20.5.20	
0-5	Petition The undersigned requests that the present international application be processed according to the Patent Cooperation Treaty		
0-6	Receiving Office (specified by the applicant)	UK Intellectual Property Office (RO/GB)	
0-7	Applicant's or agent's file reference	P2494PC00	
i	Title of Invention	COATINGS	
П	Applicant		
II-1	This person is	Applicant only	
II-2	Applicant for	All designated States	
II -4	Name	P2i Ltd	
II-5	Address	9-12 North Central 127 Olympic Avenue Milton Park Abingdon Oxfordshire OX14 4SA United Kingdom	
II-6	State of nationality	GB	
II-7	State of residence	GB	
III-1	Applicant and/or inventor		
III-1-1	This person is	Inventor only	
III-1-3	Inventor for	All designated States	
III-1- 4	Name (LAST, First)	COULSON, Stephen Richard	
III-1-5	Address	c/o P2i Ltd 9-12 North Central, 127 Olympic Avenue, Milton Park Abingdon Oxfordshire OX14 4SA United Kingdom	

III-2	Applicant and/or inventor	
III-2-1	This person is	Inventor only
III-2-3	Inventor for	All designated States
III-2- 4	Name (LAST, First)	EVANS, Delwyn
III-2-5	Address	c/o P2i Ltd 9-12 North Central, 127 Olympic Avenue, Milton Park Abingdon Oxfordshire OX14 4SA United Kingdom
111-3	Applicant and/or inventor	
III-3-1	This person is	Inventor only
111-3-3	Inventor for	All designated States
111-3-4	Name (LAST, First)	SIOKOU, Angeliki
III-3-5	Address	c/o P2i Ltd 9-12 North Central, 127 Olympic Avenue, Milton Park Abingdon Oxfordshire OX14 4SA United Kingdom
III-4	Applicant and/or inventor	
III -4 -1	This person is	Inventor only
111-4-3	Inventor for	All designated States
-4-4	Name (LAST, First)	TELFORD, Clive
III- 4 -5	Address	c/o P2i Ltd 9-12 North Central, 127 Olympic Avenue, Milton Park Abingdon Oxfordshire OX14 4SA United Kingdom

IV-1	Agent or common representative; or address for correspondence	
	The person identified below is hereby/ has been appointed to act on behalf of the applicant(s) before the competent International Authorities as:	Agent
IV-1-1	Name (LAST, First)	BLAKE, Michael
IV-1-2	Address	Greaves Brewster LLP Copa House Station Road Cheddar Somerset BS27 3AH United Kingdom
IV-1-3	Telephone No.	+ 44 (0)1934 745880
IV-1- 4	Facsimile No.	+ 44 (0)1934 745881
v	DESIGNATIONS	
V-1	The filing of this request constitutes under Rule 4.9(a), the designation of all Contracting States bound by the PCT on the international filing date, for the grant of every kind of protection available and, where applicable, for the grant of both regional and national patents.	
VI-1	Priority claim of earlier regional application	
VI-1-1	Filing date	09 June 2015 (09.06.2015)
VI-1-2	Number	15386017.6
VI-1-3	Regional Office	EP
VI-2	Incorporation by reference :	
	where an element of the international application referred to in Article 11(1)(iii)(d) or (e) or a part of the description, claims or drawings referred to in Rule 20.5(a) is not otherwise contained in this international application but is completely contained in an earlier application whose priority is claimed on the date on which one or more elements referred to in Article 11(1)(iii) were first received by the receiving Office, that element or part is, subject to confirmation under Rule 20.6, incorporated by reference in this international application for the purposes of Rule 20.6.	
VII-1	International Searching Authority Chosen	European Patent Office (EPO) (ISA/EP)

VIII	Declarations	Number of declarations		
VIII-1	Declaration as to the identity of the inventor	_		
VIII-2	Declaration as to the applicant's entitlement, as at the international filing date, to apply for and be granted a patent	_		
VIII-3	Declaration as to the applicant's entitlement, as at the international filing date, to claim the priority of the earlier application	_		
VIII-4	Declaration of inventorship (only for the purposes of the designation of the United States of America)	_		
VIII-5	Declaration as to non-prejudicial disclosures or exceptions to lack of novelty	-		
IX	Check list	Number of sheets	Electronic file(s) attached	
IX-1	Request (including declaration sheets)	5	✓	
IX-2	Description	21	√	
IX-3	Claims	14	√	
IX-4	Abstract	1	√	
IX-5	Drawings	6		
IX-7	TOTAL	47		
	Accompanying Items	Paper document(s) attached	Electronic file(s) attached	
IX-8	Fee calculation sheet	-	✓	
IX-19	Other	Pre-conversion archive	✓	
IX-19	Other	List of additional agents		
IX-20	Figure of the drawings which should accompany the abstract			
IX-21	application	English		
X-1	Signature of applicant, agent or common representative	(PKCS7 Digital Signa	ture)	
X-1-1	Name (LAST, First)	Blake, Michael		
X-1-3	Capacity (if such capacity is not obvious from reading the request)	(Representative)		

Print Out (Original in Electronic Form)

FOR RECEIVING OFFICE USE ONLY

10-1	Date of actual receipt of the purported international application	08	June	2016	(08.06.2016)
10-2	Drawings:				
10-2-1	Received				
10-2-2	Not received				
10-3	Corrected date of actual receipt due to later but timely received papers or drawings completing the purported international application				
10-4	Date of timely receipt of the required corrections under PCT Article 11(2)				
10-5	International Searching Authority	ISA	A/EP		
10-6	Transmittal of search copy delayed until search fee is paid				

FOR INTERNATIONAL BUREAU USE ONLY

Continuation of Box IV-1 Additional Agents

Our reference: P2494PC00-QR8AQ

TURNER, Rhiannon Rosalind

WALLIS, Naomi Rachel

BUMKE, Jakob Wenzel

LELAND, Emma Clare

GLANSDORP, Freija Gwendolyn

BLAKE, Michael Edward

SPINNER, David Richard

DUTTON, Erica

of Greaves Brewster LLP

Copa House

Station Road

Cheddar

Somerset

BS27 3AH

United Kingdom.

International Bureau



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- (71) Applicant: P2I LTD [GB/GB]; 9-12 North Central, 127 Olympic Avenue, Milton Park, Abingdon Oxfordshire OX14 4SA (GB).
- (72) Inventors: COULSON, Stephen Richard; c/o P2i Ltd, 9-12 North Central, 127 Olympic Avenue, Milton Park, Abingdon Oxfordshire OX14 4SA (GB). EVANS, Delwyn; c/o P2i Ltd, 9-12 North Central, 127 Olympic Avenue, Milton Park, Abingdon Oxfordshire OX14 4SA (GB). SIOKOU, Angeliki; c/o P2i Ltd, 9-12 North Central, 127 Olympic Avenue, Milton Park, Abingdon Oxfordshire OX14 4SA (GB). TELFORD, Clive; c/o P2i Ltd, 9-12 North Central, 127 Olympic Avenue, Milton Park, Abingdon Oxfordshire OX14 4SA (GB).

- (74) Agent: BLAKE, Michael; Greaves Brewster LLP, Copa House, Station Road, Cheddar Somerset BS27 3AH (GB).
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(54) Title: COATINGS

$$R_1$$
 R_2
 R_4

(57) Abstract: The present invention provides an electronic or electrical device or component thereof comprising a cross-linked polymeric coating on a surface of the electronic or electrical device or component thereof; wherein the cross-linked polymeric coating is obtainable by exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow formation of the cross-linked polymeric coating on a surface thereof, wherein the monomer compound has the following formula: where R₁, R₂ and R₄ are each independently selected from hydrogen, optionally substituted branched or straight chain C₁-C₆ alkyl or halo alkyl or aryl optionally substituted by halo, and R₃ is selected from: or where each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C₁-C₆ alkyl, halo alkyl or aryl optionally substituted by halo; and n₁ is an integer from to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moieties and has a boiling point at standard pressure of less than 500°C.

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Coatings

Field of the invention

This invention relates to protective coatings. Specifically, the present invention relates to protective coatings for electronic or electrical devices and components thereof, and methods of forming such coatings. The coatings can protect by being hydrophobic and so resist the ingress of water-based liquid into electronic devices, or they can protect by forming a barrier coating and so provide electrical resistance between the electrical parts of the phone and water based liquid.

Background of the invention

Monounsaturated monomers are used to make barrier coatings using plasma polymerisation processes (see co-pending application).

The perfluoroalkyl chain monomers are also used for generating hydrophobic surfaces from pulsed plasma deposition processes (see WO 9858117 A1).

The power of the plasma initiated polymerisation affects the nature of the polymer produced. The higher average energy inputs of continuous wave plasmas lead to more fragmentation of the monomer, and so the polymer loses structural properties of the monomer. In the case of 1H,1H,2H,2H- perfluorodecyl acrylate (PFAC8), there's less retention of the perfluoroalkyl chain and the contact angle of the surface coating is compromised. Higher plasma energies also lead to more crosslinking. For the lower average energy inputs of pulsed plasmas, there's better retention of monomer structure and less crosslinking. The greater retention of the perfluoro chain under low energy, pulsed plasma conditions leads to best levels of contact angles for the surface coating.

When the perfluoroalkyl chains have eight or more fluorinated carbons (long chain), the polymer made from the monomer has a crystalline structure. When the perfluoroalkyl chains have less than eight fluorinated carbons, the resulting polymer is amorphous and so can be unstable in the presence of water (see "Molecular Aggregation Structure and Surface Properties of Poly(fluoroalkyl acrylate) Thin Films, Marcomolecules, 2005, vol 38, p5699-5705)

When long chain perfluoroalkyl polymers are produced by high average power (continuous wave or CW) or low average power (pulsed wave or PW) plasmas, then because of the crystalline structure of the long chains, the polymers are non-stick to the touch and stable in the presence of water. However, the feel and water stability of shorter chain polymer coatings is affected by the plasma power levels used. For example, when PFAC6 (1H,1H,2H,2H-perfluorooctyl acrylate) is polymerised in low power plasma conditions, the resulting polymer coating can have several disadvantages. For example the coating can cause water drops to spread out a little (slump), be marked by the presence of a water drop on its surface, have a tacky feel, or can be easily smeared (for example on substrates of silicon wafer and ABS plastic).

By increasing the power of the plasma used for polymerisation, the polymer becomes more crosslinked and becomes more resistant to smearing. However, increasing the power has the concomitant effect of decreasing the water contact angle through more monomer fragmentation (as described above). Figure 1 shows the effect of increasing the power to monomer flow ratio for CW plasmas in a 125 litre chamber: at a ratio of 4 W/µl/min, the water contact angle is ~85-95 degrees and the coating is tack-free. However, as the ratio drops, the contact angle increases and the occurrence of tacky/smudgy coatings increases too. Figure 2 shows the same effect for pulsed plasma conditions. These results show that the process window for producing tack and smudge-free coatings have a limited plasma processing range and the final coating has a compromised water contact angle.

Accordingly, it is an aim to solve one or more of the above-mentioned problems with the prior art coatings.

Statement of Invention

An aspect of the present invention provides an electronic or electrical device or component thereof comprising a protective cross-linked polymeric coating on a surface of the electronic or electrical device or component thereof;

wherein the protective cross-linked polymeric coating is obtainable by exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow formation of the protective cross-linked polymeric coating on a surface thereof,

wherein the monomer compound has the following formula:

$$R_1$$
 R_3 R_2 R_4

where R_1 , R_2 and R_4 are each independently selected from hydrogen, optionally substituted branched or straight chain C_1 - C_6 alkyl or halo alkyl or aryl optionally substituted by halo, and R_3 is selected from:

where each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C_1 - C_6 alkyl, halo alkyl or aryl optionally substituted by halo; and n_1 is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moieties and has a boiling point of less than 500 °C at standard pressure.

High levels of polymer crosslinking (formerly only achievable with high average power continuous wave plasmas) can be achieved by adding a crosslinking molecule to the monomer to produce a cross-linked co-polymer. This has the advantage of increasing the plasma processing range so stable coatings can now be produced in low average energy pulse plasma conditions.

The high degree of retention of the hydrophobic monomer structures from the low energy pulsed plasma gives coatings of the co-polymer a good hydrophobic coating (as evidenced by water contact angles) and a coating that is not sticky or smeary to the touch. Thus, these properties make it suitable for the coating of electronic devices to prevent the ingress of water-based liquids from accidental damage. Such coatings typically have a thickness between 1 and 100 nm, preferably 1 and 50 nm.

The coating may be described as a liquid repellant layer, typically a water repellant layer. The coating may repel water, an aqueous liquid such as rainwater, or oil.

Alternatively, or in addition to providing a liquid repellent layer, the coating may be a physical barrier layer.

The coating may be a physical barrier i.e. it provides a physical barrier to mass and electron transport.

The physical barrier layer restricts diffusion of water, oxygen and ions. When the coating is a physical barrier it typically has a thickness of greater than 50 nm.

The coating is a protective layer i.e. it prevents damage by contact with water or other liquids. The coating may provide is protective functionality by forming a liquid repellent layer and/or a physical barrier layer.

The coating is preferably substantially pin-hole free to enable it to provide a physical barrier. Preferably $\Delta Z/d < 0.15$, where ΔZ is the average height variation on an AFM line scan in nm (as shown in Figure 3) and d is coating thickness in nm.

The value of $\Delta Z/d$ tells us to what extent defects/voids on the surface of the coating extend into the coating, i.e. the percentage value of the depth of defect over total coating thickness. For example, $\Delta Z/d = 0.15$ means that the voids on the surface only extend down to 15% of the coating thickness. A coating with a $\Delta Z/d < 0.15$ is defined herein as being substantially pinhole free. If voids are bigger than this, the desired functionality is unlikely to be achieved.

The coating is preferably conformal, which means that it takes the 3D shape of the electronic or electrical device or component thereof and covers substantially an entire surface of the device. This has the advantage of ensuring that the coating has sufficient thickness to give optimal functionality over an entire surface of the device or component. The meaning of the term "covers substantially an entire surface" will depend to some extent on the type of surface to be covered. For example, for some components, it may be necessary for there to be complete coverage of the surface in order for the component to function after submersion in water. However, for other components or housings, small gaps in coverage may be tolerated.

The coating may have a thickness of 50 to 10,000nm, optionally 50 to 8000nm, 100 to 5000nm, preferably 250nm – 5000nm, most preferably 250nm – 2000nm.

The coating may be electrically insulating and be sufficiently compliant that electrical connectors can be joined to the electronic or electrical device or component thereof and an electrical connection made between the electrical connectors and electronic or electrical device or component thereof without the requirement to first remove the coating. In this case, the force exerted by the electrical connector on the coating is sufficient to alter the structure of or even break through the coating local to the electrical connector, allowing the electrical connection to be made. Electrical connectors can typically be joined to the electronic or electrical device or component in this way for coating thicknesses of under 5000nm, and for high performance coatings below 2000nm.

The electronic or electrical device or component thereof can typically withstand immersion in up to 1m of water for over 30 minutes without failure or corrosion whilst power applied to the electronic or electrical device or component. The effectiveness of the coating can be determined by measuring its electrical resistance at a fixed voltage when submerged in water for a set time period; for example, applying a voltage of 8V to the coating of a device which is submerged in water for 13 minutes. Coatings having a resistance value of 1x10⁷ Ohms or higher in this test are effective barrier coatings and the coated electronic or electrical device or component thereof will pass successfully an IPX7 test. The IPX7 test is the Ingress Protection Marking which classifies and rates the degree of protection provided against water. In the IPX7 test for phones, the device is immersed in water under defined conditions of pressure and time (up to 1 m of submersion) for a duration of 30 minutes. The device has to be powered on during testing and functional after 24 hrs.

In one embodiment, the coating is electrically insulating and has a thickness of less than 1 micron and wherein a force of 5- 20g applied to the coating using a round probe with 1 mm diameter allows an electrical connection to be made with the electronic or electrical device or component thereof in the local area where the force has been applied.

In another embodiment, the coating is electrically insulating and has a thickness of 1-2.5 microns and wherein a force of 20-100g applied to the coating using a round probe with 1 mm diameter allows an electrical connection to be made in the local area of the coating where the force has been applied.

The coating may have a higher density than that of the corresponding monomer from which it is formed. For example, the increase in density may be at least 0.1 g/cm^{3.} The increase in density is explained by the highly crosslinked coating. The high density of the coating improves the barrier properties of the coating.

The coating may form a surface defined by a static water contact angle (WCA) of at least 70°. A coating with a WCA of at least 90° is a liquid repellent, typically a water repellent layer. For fluorinated polymers, the coating may have a static water contact angle of at least 100°. The contact angle of a liquid on a solid substrate gives an indication of the surface energy which in turn illustrates the substrate's liquid repellence. Contact angles were measured on a VCA Optima contact angle analyser, using 3 µl droplets of deionised water at room temperature.

Crosslinker

A basic requirement for the cross-linking reagent is the presence two or more unsaturated bonds, for example, -C=C- or alkyne groups. The unsaturated bonds are attached by one or more linker moieties. The linker moiety is not especially limited provided it joins together the two or more unsaturated bonds. The cross-linking reagent must have a boiling point of less than 500 °C, preferably from -10 to 300°C, optionally 180 to 270°C, most preferably 205 to 260°C at standard pressure. It is a preference, but not a requirement, that the cross-linking reagent is not unduly hazardous for use in plasma processing, that is to say it can be used in a production environment where low levels of vapour do not present any significant health and safety issues (e.g. strongly oxidising, explosive, toxic, or has an offensive odour (e.g. stench reagent)). The crosslinking reagent preferably has one of the following structures:

where Y_1 , Y_2 , Y_3 , Y_4 , Y_5 , Y_6 , Y_7 and Y_8 are each independently selected from hydrogen, optionally substituted cyclic, branched or straight chain C_1 - C_6 alkyl or aryl; and L is a linker moiety.

Most preferably L has the following formula:

where each Y_9 is independently selected from, a bond, -O-, -O-C(O)-, -C(O)-O-, -Y₁₁-O-C(O)-, -C(O)-O-Y₁₁-, -OY₁₁-, and -Y₁₁O-, where Y₁₁ is an optionally substituted cyclic, branched or straight chain C₁-C₈ alkylene; and

 Y_{10} is selected from optionally substituted cyclic, branched or straight chain C_1 - C_8 alkylene and a siloxane containing group.

In one most preferred embodiment Y_{10} has the following formula:

$$\begin{cases}
-\begin{cases}
Y_{12} \\
C \\
Y_{13}
\end{cases}_{n}
\end{cases}$$

wherein each Y_{12} and Y_{13} is independently selected from H, halo, optionally substituted cyclic, branched or straight chain C_1 - C_8 alkyl, or $-OY_{14}$, where Y_{14} is selected from optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl, and n is an integer from 1 to 10.

Optionally, each Y_{12} is H and each Y_{13} is H, i.e. Y_{10} is a linear alkylene chain. For this embodiment Y_9 is preferably a vinyl ester or vinyl ether group.

Optionally, each Y_{12} is fluoro and each Y_{13} is fluoro, i.e. Y_{10} is a linear perfluoroalkylene chain.

Typically, n is from 4 to 6.

In another embodiment Y₁₀ has the following formula:

(v)

wherein each Y_{15} is independently selected from optionally substituted branched or straight chain C_1 - C_6 alkyl.

In one embodiment each Y_{15} is methyl, and each Y_{9} is a bond.

In a further embodiment Y₁₀ has the following formula:

Υ₁₆ Υ₁₇ ξ

(vi)

wherein Y_{16} to Y_{19} are each independently selected from H and optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl. A preferred alkenyl group is vinyl. Optionally Y_{18} is H or vinyl, and Y_{18} , Y_{17} and Y_{19} are each H. In one embodiment each of Y_{18} to Y_{19} is H. In another embodiment Y_{18} is vinyl, and Y_{16} , Y_{17} and Y_{19} are each H.

In preferable embodiments for compound (i), L has one of the following structures:

In another embodiment for compound (i) L has one of the following structures:

For L according structure (vii), Y_{10} is preferably an alkylene chain or a cycloalkylene, such as those shown in structures iv) and vi) above. The alkylene chain may be a straight chain alkylene chain. When Y_{10} is a cycloalkylene, cyclohexylene is preferred, most preferably 1,4 cyclohexylene.

For L according to structure (viii), Y_{10} is preferably structure (iv), i.e. an alkylene or fluoroalkylene chain.

For L according to structure (ix), Y_{10} is preferably cycloalkylene, such as the cyclohexylene according to structure (vi).

For L according to structure (x), Y_{10} is preferably structure (iv) wherein each Y_{12} and Y_{13} is F, i.e. a perfluoroalkylene chain.

For L according to structure (xi) or structure (xii) Y_{10} is preferably alkylene or cycloalkylene. Optionally the alkylene or cycloalkylene may be substituted with one or more vinyl groups or alkenyl ether groups, preferably one or more vinyl ether groups.

When each Y_9 is a bond, each Y_{10} may be any of structures (iv), (v) and (vi). Preferably Y_{10} is a straight chain alkylene such that the crosslinking reagent is a diene, such as a heptadiene, octadiene, or nonadiene, most preferably 1,7-octadiene.

When each Y_9 is O, each Y_{10} is preferably a branched or straight chain C_1 - C_6 alkylene, preferably a straight chain alkylene, i.e the crosslinking reagent is 1,4-butanediol divinyl ether

It will be understood that each Y_9 group can be combined with any other Y_9 group and Y_{10} group to form the crosslinking reagent.

The skilled person will be aware of possible substituents for each of the cyclic, branched or straight chain C_1 - C_8 alkylene groups mentioned above. The alkylene groups may be substituted at one or more positions by a suitable chemical group. Halo substituents are preferred, with fluoro substituents most preferred. Each C_1 - C_8 alkylene group may be a C_1 - C_8 , or C_8 - C_8 alkylene group.

Particularly preferred embodiments of the crosslinker have alkyl chains for Y₁₀ and vinyl ester or vinyl ether groups on either side.

A particularly preferred cross-linking reagent is divinyl adipate. (a divinyl ester)

Another preferred cross-linking reagent is 1,4-butanediol divinyl ether (a divinyl ether)

For the most preferred embodiments the crosslinking reagent is selected from Divinyl Adipate (DVA), 1,4 Butanediol divinyl ether (BDVE), 1,4 Cyclohexanedimethanol divinyl ether (CDDE), 1,7 -Octadiene (17OD), 1,2,4 -Trivinylcyclohexane (TVCH), 1,3-Divinyltetramethyldisiloxane (DVTMDS), Diallyl 1,4-cyclohexanedicarboxylate (DCHD), 1,6-Divinylperfluorohexane (DVPFH), 1H,1H,6H,6H-Perfluorohexanediol diacrylate (PFHDA) and glyoxal bis (diallyl acetal) (GBDA).

For the alkyne crosslinking reagents according to compound (ii), L is preferably selected from a branched or straight chain C₁-C₈ alkylene or an ether group. L may be a C₃, C₄, C₅, or C₆ alkylene,

preferably a straight chain alkylene. Chemical structures for the particularly preferred crosslinking reagents are set out below in Table 1:

Divinul Adinata (DVA)	0
Divinyl Adipate (DVA)	H ₂ C O CH ₂
	II O
1,4 Butanediol divinyl ether	0 CH ₂
(BDVE)	H ₂ C O
1,4 Cyclohexanedimethanol	O CH ₂
divinyl ether (CDDE)	H ₂ C O
1,7 -Octadiene (170D)	H_2C CH_2
1,2,4 -Trivinylcyclohexane	CH ₂
(TVCH)	
	H_2C CH_2
1,3-	CH₃ CH₃
Divinyltetramethyldisiloxane	Si—O—Si————————————————————————————————
(DVTMDS)	H_2C' CH_2 CH_3 CH_3
Diallyl 1,4-	O CH ₂
cyclohexanedicarboxylate	G112
(DCHD)	
	CH₂
	0 0 0 0
1,6-Divinylperfluorohexane	F F F
(DVPFH)	H ₂ C F F F CH ₂
1H,1H,6H,6H-	0 F F
Perfluorohexanediol	H ₂ C
diacrylate (PFHDA)	O F F F CH ₂
	·
GBDA	O CH ₂
	H_2C O O CH_2
	H₂C Ó

1,6-heptadiyne	нс=
1,7-heptadiyne	HC===CH
1,8-heptadiyne	нс=сн
	^
Propargyl ether	нс=Сн

Table 1

Monomer Compound

The monomer compound has the following formula:

$$R_1 R_3$$
 $R_2 R_4$

where R_1 , R_2 and R_4 are each independently selected from hydrogen, optionally substituted branched or straight chain C_1 - C_6 alkyl or halo alkyl or aryl optionally substituted by halo, and R_3 is selected from:

$$\begin{array}{c|c}
O & \begin{bmatrix} X \\ C \\ X \end{bmatrix}_{n_1} CX_3 & O & \begin{bmatrix} X \\ C \\ X \end{bmatrix}_{n_1} CX_3
\end{array}$$

where each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C_1 - C_6 alkyl, halo alkyl or aryl optionally substituted by halo; and n_1 is an integer from 1 to 27. Preferably, n_1 is from 1 to 12. Optionally n_1 is from 4 to 12, optionally 6 to 8.

In a preferred embodiment R₃ is selected from:

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where m_1 is an integer from 0 to 13 and each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C_1 - C_6 alkyl, halo alkyl or aryl optionally substituted by halo; and m_2 is an integer from 2 to 14;

In a particularly preferred embodiment the monomer is a compound of formula I(a):

I(a)

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wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_6 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1;

; or

the monomer is a compound of formula I(b):

I(b)

wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_6 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1.

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The halogen may be chlorine or bromine, but is preferably fluorine for compliance with RoHS regulations (Restriction of Hazardous Substances).

a is from 0 to 10, preferably from 0 to 6, optionally 2 to 4, most preferably 0 or 1. b is from 2 to 14, optionally from 2 to 10, preferably 3 to 7.

Each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or a C_1 - C_6 branched or straight chain alkyl group. The alkyl group may be substituted or unsubstituted, saturated or unsaturated. When the alkyl group is substituted, the location or type of the substituent is not especially limited provided the resultant polymer provides an appropriate liquid repellent and/or barrier layer. The skilled person would be aware of suitable substituents. If the alkyl group is substituted, a preferred substituent is halo, i.e. any of R_1 , R_2 , R_4 , and R_5 to R_{10} may be haloalkyl, preferably fluoro alkyl. Other possible substituents may be hydroxyl or amine groups. If the alkyl group is unsaturated it may comprise one or more alkene or alkyne groups.

Each of R₁, R₂, R₄, and R₅ to R₁₀ may be independently selected from hydrogen, methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, tert-butyl, sec-butyl, n-pentyl, neopentyl, n-hexyl, isohexyl, and 3-methylpentyl.

In a preferred embodiment each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or methyl.

In a preferred embodiment, a and c are each independently 0 or 1; and b is from 3 to 7.

In one preferred embodiment each X is H. In an alternative preferred embodiment each X is F.

Optionally R₁ and R₂ are both hydrogen.

Optionally R_4 is hydrogen or methyl. Preferably R_1 and R_2 are both hydrogen and R_4 is hydrogen or methyl.

Optionally R_9 is hydrogen and R_{10} is branched or straight chain C_1 - C_6 alkyl group. In a preferred embodiment R_{10} is methyl.

In one embodiment each of R₅ to R₈ is hydrogen.

In one embodiment each of R₁, R₂, R₄, and R₅ to R₁₀ is hydrogen, each X is H, a=0 and c=0.

In a particularly preferred embodiment the monomer compound has the following formula:

$$H \longrightarrow O$$
 $(CF_2)_nF$

where n is from 2 to 10.

In another preferred embodiment the monomer compound has the following formula:

$$H \longrightarrow O (CF_2)_nF$$
 $H \longrightarrow CH_3$

where n is from 2 to 10.

The monomer compound may be selected from 1H,1H,2H,2H-perfluorohexyl acrylate (PFAC4), 1H,1H,2H,2H-perfluorooctyl acrylate (PFAC6), 1H,1H,2H,2H-perfluorodecyl acrylate (PFAC8) and 1H,1H,2H,2H-perfluorododecyl acrylate (PFAC10).

The monomer compound may be selected from 1H,1H,2H,2H-pefluorohexyl methacrylate (PFMAC4), 1H,1H,2H,2H-perfluorooctyl methacrylate (PFMAC6) and 1H,1H,2H,2H-perfluorodecyl methacrylate (PFMAC8).

The monomer compound of formula I(a) may have the following formula:

wherein a and c are each independently 0 or 1, b =3-7 and n is 4 to 10, where n=a+b+c+1.

The monomer ccompound of formula I(a) may have the following formula:

$$H \longrightarrow CH_3$$
 $(CH_2)_nH$

where n is 2 to 12.

The monomer compound may be selected from ethyl hexyl acrylate, hexyl acrylate, decyl acrylate, lauryl dodecyl acrylate and iso decyl acrylate.

The monomer may have the following formula:

$$\begin{array}{c}
R_1 & O \\
R_2 & R_3
\end{array}$$

$$\begin{array}{c}
(CH_2)_n - CH_3$$

where n is from 4 to 14.

The monomer may have the following formula:

$$H O \longrightarrow (CH_2)_n \longrightarrow CH_3$$

where n is from 4 to 14.

In an additional aspect, the present invention provides a method for treating an electronic device or component thereof comprising:

exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow a protective cross-linked polymeric coating to form on a surface of the electronic or electrical device or component thereof;

wherein the monomer compound has the following formula:

$$R_1$$
 R_2
 R_2

where R_1 , R_2 and R_4 are each independently selected from hydrogen, optionally substituted branched or straight chain C_1 - C_6 alkyl or halo alkyl or aryl optionally substituted by halo, and R_3 is selected from:

where each X is independently selected from hydrogen, optionally substituted branched or straight chain C_1 - C_8 alkyl, halo alkyl or aryl optionally substituted by halo; and n_1 is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds and has a boiling point of less than 500 °C at standard pressure.

The monomer compound and the crosslinking reagent utilised in the method are described in further detail above.

Typically the electronic or electrical device or component thereof is placed in a plasma deposition chamber, a glow discharge is ignited within said chamber, and a voltage applied as a pulsed field.

Preferably the applied voltage is at a power of from 30 to 800W. Optionally the voltage is pulsed in a sequence in which the ratio of the time on/time off is in the range of from 0.001 to 1, optionally 0.002 to 0.5. For example, time on may be 10-500 μ s, preferably 35-45 μ s, or 30-40 μ s, such as about 36 μ s and time off may be from 0.1 to 30 ms, preferably 0.1 to 20 ms, optionally from 5 to 15 ms, for example 6ms. Time on may be 35 μ s, 40 μ s, 45 μ s. Time off may be 0.1, 1, 2, 3, 6, 8, 10, 15, 20, 25 or 30 ms.

The term pulsed may mean that the plasma cycles between a state of no (or substantially no) plasma emission (off-state) and a state where a particular amount of plasma is emitted (on-state). Alternatively, pulsed may mean that there is continuous emission of plasma but that the amount of plasma cycles between an upper limit (on-state) and lower limit (off-state).

In a further embodiment, the invention resides in a method of forming a coating on an electronic or electrical device or component thereof as defined above, the method comprising: exposing said substrate in a chamber to a plasma comprising a monomer compound, preferably a continuous plasma, for a sufficient period of time to allow a protective polymeric coating to form on the substrate,

wherein during exposure of the substrate the continuous plasma has a power density of at least 2W/litre, preferably 20W/litre.

Optionally the voltage is applied as a pulsed field for a period of from 30 seconds to 90 minutes. Optionally the voltage is applied as a pulsed field for from 5 to 60 minutes. Optionally in a preliminary step, a continuous power plasma is applied to the electronic or electrical device or component thereof. The preliminary step may be conducted in the presence of an inert gas.

Before the crosslinker and/or the monomer compound enter the deposition chamber each may be in the form of a gas, liquid or a solid (for example a powder) at room temperature. However, it is the preferred that the crosslinking reagent and the monomer compound are both liquid at room temperature, and most preferably that the monomer and crosslinker liquids are miscible.

The monomer compound and/or the crosslinker will suitably be in a gaseous state in the plasma. The plasma may simply comprise a vapour of the monomer compound and the crosslinker. Such a vapour may be formed in-situ, with the compounds being introduced into the chamber in liquid form. The monomer may also be combined with a carrier gas, in particular, an inert gas such as helium or argon.

In preferred embodiments, the monomer and/or the crosslinker may be delivered into the chamber by way of an aerosol device such as a nebuliser or the like, as described for example in WO2003/097245 and WO03/101621, the content of which is incorporated herein by reference. In such an arrangement a carrier gas may not be required, which advantageously assists in achieving high flow rates.

In one embodiment the monomer compound and/or the crosslinking reagent are in gaseous form and are fed into the plasma at a rate of from 10-5000 mg/minute depending on chamber volume, whilst the pulsed voltage is applied.

Optionally the plasma is created with an average power of from 0.001 to 40W/litre.

The crosslinker may be miscible with the monomer and so introduced together or separately into the plasma chamber. Or the crosslinker may be immiscible with the monomer and introduced separately into the plasma chamber. In this context, the term "miscible" means that the crosslinker is soluble in the monomer, and when mixed they form a solution of uniform composition. The term "immiscible" is used to mean that the crosslinker is only partly soluble or insoluble in the monomer, and so either forms an emulsion or separates out into two layers.

The crosslinking reagent is preferably present in an amount of from 10 to 60 (v/v) %, optionally 20 to 40 (v/v) %, optionally 25 to 30 (v/v)%, optionally 30 to 50 (v/v)% of the total volume of monomer compound and crosslinking reagent, depending on the specific crosslinker. The skilled person would understand that the amount will vary to some extent depending on whether the coating is required to

be liquid repellent or provide a barrier to mass and electron transport. The skilled person would understand that the (v/v) percentages are those which give a stable crosslinked polymer coating and the highest water contact angle.

Electronic or electrical device or component thereof

Although the invention is of benefit in the context of a wide variety of substrates, the substrate may, in all aspects of the invention, advantageously be an electronic substrate.

In some embodiments of the invention, the electronic substrate may comprise an electronic or electrical device, i.e. any piece of electrical or electronic equipment. Non-limiting examples of electrical and electronic devices include communications devices such as mobile phones, smartphones and pagers, radios, and sound and audio systems such as loudspeakers, microphones, ringers or buzzers, hearing aids, personal audio equipment such as personal CD, tape cassette or MP3 players, televisions, DVD players including portable DVD players, video recorders, digi and other set-top boxes such as Sky, computers and related components such as laptop, notebook, tablet, phablet or palmtop computers, personal digital assistants (PDAs), keyboards, or instrumentation, games consoles in particular hand-held playstations and the like, data storage devices, outdoor lighting systems or radio antenna and other forms of communication equipment.

In preferred embodiments of the invention, the substrate may comprise or consist of an electronic component, e.g. a printed circuit board (PCB), a printed circuit board array (PCBA), a transistor, resistor, or semi-conductor chip. The electronic component may thus be an internal component of an electronic device, e.g. a mobile phone. The coatings of the invention are particularly valuable in preventing electrochemical migration in such components.

In all aspects of the invention, the precise conditions under which the protective polymeric coating is formed in an effective manner will vary depending upon factors such as, without limitation, the nature of the monomer compound, the crosslinking reagent, the substrate, as well as the desired properties of the coating. These conditions can be determined using routine methods or, preferably, using the techniques and preferred features of the invention described herein, which work in particular synergy with the invention.

The present invention will now be further described with reference to the following non-limiting examples and the accompanying illustrative drawings, of which:

Figure 1 illustrates shows the effect of increasing the power to monomer flow ratio for CW plasmas in a 3 litre chamber in a prior art process.

Figure 2 shows the same effect for pulsed plasma conditions.

Figure 3 shows a tapping mode image over $10x10~\mu m^2$ field of view of a specimen example (thickness d=1230 nm) prepared according to Example 1 (left hand side) and a contour line plot showing the data used for calculation of RMS roughness (right hand side). The ΔZ value indicated on the plot has been taken over an area of the graph that represents the majority of the coating. Peaks that lie above the ΔZ range indicate large particles and troughs that fall below the ΔZ range show voids or pinholes in the coating. The width of the peaks also gives an indication of the particle size.

Figure 4 illustrates the effect of adding the divinyl adipate crosslinker to perfluorooctyl acrylate on water contact angle from a process on a 125 Litre chamber.

Figure 5 shows the effect of adding crosslinkers in accordance with the invention on the electrical resistance per nm of the coatings.

Figure 6 shows a tapping mode image of a 90 nm thick coating, prepared as described in Example 3, over $2x2\mu m^2$ field of view (top left), a representative contour line indicating height variation (z-axis) of the coating (top right) and a phase image indicating full substrate coverage (bottom left) ;RMS roughness of the coating is 1.65 nm and $\Delta z/d = 0.05$

Figure 7 shows a FTIR/ATR spectrum of a coating formed according to Example 5.

Example 1

Example of plasma polymerising PFAC6 with no crosslinker in a 3 litre chamber

PFAC6 was polymerised in a 3 litre glass chamber using a continuous plasma process. In the experimental matrix, the continuous plasma was produced using RF power levels of 50, 100 and 200 Watts. The process pressure was run at 30, 60 and 90 mTorr and the monomer flow rate run at 50, 100 and 200 microlitres of liquid per minute. The duration of the process was either 10 or 40 minutes. The coated substrate was silicon wafer and whether the coating was tacky or not was determined by wiping the substrate with a finger and determining by eye if the coating had smeared. The results in Figure 1 indicate that for lower CW powers, the coating could easily be smeared (i.e. tacky), but by increasing the power to flow ratio, the coating became resistant to smearing (i.e. non-tacky) and this is expected to be due to additional crosslinking caused by more monomer fragmentation. The water contact angle was also seen to decline with increasing power to flow ratio, and this too is indicative of more power increasing the level of monomer fragmentation.

The same 3 litre chamber was used to also look at the effect of power to flow ratios with pulsed plasma polymerisation. In these experiments, the peak power was either 250 or 500 W. The RF power feed was pulsed with an on time of 35 microseconds and off times of 5.5 and 1 millisecond. Process pressure was either 20, 35, 60 or 80 mTorr. The monomer flow was either 50 or 200 microlitres of liquid per minute. Figure 2 shows that low power to flow conditions can generate coatings that can be smeared (tacky) and only by increasing the power could the smear be reduced. This too had a concomitant decline in water contact angle.

The results from Figures 1 and 2 show that plasma processing PFAC6 on its own could only lead to smear –free coatings when the powers were high enough to decrease the other desirable property of the water contact angle. This illustrates the need to add a crosslinker to prevent the coating smearing and maintain the water contact angle.

Example 2

Examples of Perfluoro Alkyl Acrylate co-polymerisation with Crosslinker to show improvements in electrical resistance

Perfluorooctyl acrylate monomer was mixed with a single crosslinker from the following list:

VINYL ESTERS: Divinyl adipate (DVA)

VINYL ETHERS: 1,4-Butanediol divinyl ether (BDVE); 1,4 Cyclohexanedimethanol divinyl ether (CDDE)

DI- or TRI- VINYLS: 1,7-Octadiene (1,7-OD); 1,2,4-Trivinylcyclohexane (TVCH);

DI VINYL with alkyl fluoro group: 1,6-divinylperfluorohexane (DVPFH)

DI VINYL with silicon group: 1,3-Divinyl tetramethyldisiloxane (DVTMDS)

DI VINYL with cyclic ring and carboxylate groups: Diallyl 1,4-cyclohexanedicarboxylate (DCHC)

DI ACRYLATE: 1H1H,6H,6H-perfluorohexanediol diacrylate (PFHDA)

The crosslinker was mixed with the PFAC6 as a percentage by volume for the following percentages: 20, 40, 60, 80, and 100%.

Plasma initiated polymerisation reactions were carried out in a 3 litre glass plasma chamber. The substrates were test circuit boards. The PFAC6/crosslinker mixture was introduced at a rate of 0.04 – 0.06 ul/min and the process pressure was 40 mTorr. The process plasma consisted of a 1 minute continuous plasma (CW) step of 50W followed by 10 - 20 minute pulsed plasma (PW) step of 50W and RF power being delivered by a pulse sequence with a duty cycle of 6.9%. The coated circuit boards were immersed in tap water and an 8V potential was applied for 13 minutes. The final current readings, together with coating thickness measurements, were used to calculate the electrical resistance per nm of coating. Figure 5 below shows the effect of adding a range of crosslinkers on the electrical resistance per nm of the barrier coatings. Figure 5 clearly shows the improvement in resistance that can be gained by adding the crosslinker to the PFAC6 monomer. Furthermore, all of the crosslinkers tested gave non-smear/non-tacky coatings with concentration levels of 20% and above, with the exceptions of DVTMDS, DCHD, PFHDA and DVPFH which were non-smear at levels of 40% and above.

Example 3

Example of crosslinker concentration on water repellent coating.

Water repellent coatings were prepared in a 125 litre volume chamber using PFAC6 with different levels of divinyl adipate (DVA) and using helium as a carrier gas. The deposition process consisted of a 3 minute CW step with 300W power and a pulsed step with 150W power and an RF pulse duty cycle of 0.018%. Silicon wafer was used as the test substrates and the contact angle of the coated wafer was determined by applying a 3ul drop of deionised wafer onto the coated wafer and measuring the contact angle using a VCA Optima (AST products) with image analysis software. The variation of contact angle with (v/v)% DVA crosslinker is shown in Figure 4. It was also noted that for the 10 (v/v) % DVA concentration, the water drop left a mark on the coated wafer where it's circumference had been. This indicated that a higher level of crosslinker was required to give a more stable coating. This observation, in combination with the results in figure 4, suggest an optimum range of 20 – 40 (v/v) % DVA crosslinker, though this is likely to vary, typically within the range of from 10-60 (v/v) % for different monomers and crosslinkers and different chamber sizes.

Pulsed plasma polymerisation of perfluoro alkyl acrylate in the presence of divinyl adipate (DVA) was carried out in a 125 litre chamber. Figure 4 shows the effect of adding different liquid volume percentages of DVA crosslinker to perfluorooctyl acrylate on water contact angle on a silicon wafer substrate.

Example 4

Example of AFM measurement of monomer/crosslinker co-polymer coating

A barrier-style coating was prepared in a 22 litre volume chamber using PFAC6 with 10% DVA. The deposition process consisted of a 1 minute CW step with a CW power to flow ratio of 3.9 (W/microliter/min), and a pulsed step with a PW power/monomer flow ratio of 0.28 (W/microlitres/min). Figure 6 shows the representative topographical and phase contrast images were obtained from all the samples. High spatial resolution images show mainly the structures of areas between the raised features. The RMS roughness of the coating is 1.65 nm and the $\Delta z/d = 0.05$. As the $\Delta z/d$ value is < 0.15, then this indicates that the physical layer is substantially pinhole free.

Example 5

Example of FTIR/ATR measurement of monomer/crosslinker co-polymer coating

Two barrier-style coatings were prepared in a 22 litre volume chamber using PFAC6 only and PFAC6 with 10% DVA as described in example 4. The deposition process consisted of a 1 minute CW step with a CW power to flow ratio of 3.9 (W/microliter/min), and a pulsed step with a PW power/monomer flow ratio of 0.28 (W/microlitres/min). Figure 7 shows the representative FTIR/ATR spectrum obtained from both samples.

The FTIR/ATR intensity ratios of peaks attributed to stretching mode of CF_3 and C=O groups, $CF_3/C=O$, of the coating is indicative of sufficient cross linking in the coating to form a physical barrier. CF_3 refers to the terminal groups in the side chain of PFAC6.

Formation of the barrier coatings is believed to be caused by a mixture of cross linking and controlled fragmentation of the monomers during polymerisation. Cross linking is believed to be predominantly via the CF2-CF3 chain, whilst fragmentation is thought to be predominantly through loss of the C=O group during polymerisation and to a lesser extent shortening of the CF2 chain. Cross linking effects the abundance of –CF3 groups in the coating and controlled fragmentation controls the amount of C=O groups in the coating. The ratio of these two functional groups is an indication that sufficient cross-linking and fragmentation has taken place and can be measured by the ratio of the intensities of the corresponding FTIR/ATR peaks.

It has been shown that coatings with reduced CF₃ / C=O ratios give higher resistance values in the electrical test, as described in Example 2, showing an improved coating performance on increased cross linking (for CF₃) and fragmentation (for C=O). If the FTIR/ATR intensity ratio of the peaks attributed to -CF₃ stretching and C=O stretching, CF₃/C=O, is less than 0.6e^{-0.1n} (where n=6 for PFAC6) the resistance of the coating at the electrical test is expected to be higher than 8MOhms.

The peak intensity ratio CF₃/C=O the coating thickness and the final current readings during the electrical test are shown in table 2:

	Thickness (nm)	Resistance in water at 8V (Ohms)	CF ₃ /C=O ATR peak intensity ratio
PFAC6	1480	2.1×10 ⁷	0.23
PFAC6 and 10%DVA			
(in volume)	1450	9.2x10 ⁹	0.11

Table 2

Claims:

1. An electronic or electrical device or component thereof comprising a protective cross-linked polymeric coating on a surface of the electronic or electrical device or component thereof; wherein the protective cross-linked polymeric coating is obtainable by exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow formation of the protective cross-linked polymeric coating on a surface thereof,

wherein the monomer compound has the following formula:

$$\stackrel{R_1}{\searrow} = \stackrel{R_3}{\swarrow}$$
 $\stackrel{R_2}{\searrow} = \stackrel{R_4}{\searrow}$

where R_1 , R_2 and R_4 are each independently selected from hydrogen, optionally substituted branched or straight chain C_1 - C_6 alkyl or halo alkyl or aryl optionally substituted by halo, and R_3 is selected from:

where each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C_1 - C_6 alkyl, halo alkyl or aryl optionally substituted by halo; and n_1 is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moieties and has a boiling point of less than 500 °C at standard pressure.

- 2. An electronic or electrical device or component thereof according to claim 1, wherein the protective cross-linked polymeric coating is a physical barrier to mass and electron transport.
- 3. An electronic or electrical device or component thereof according to claim 1 or claim 2, wherein the protective cross-linked polymeric coating forms a liquid repellent surface defined by a static water contact angle (WCA) of at least 90°.
- 4. An electronic or electrical device or component thereof according to any preceding claim, wherein the crosslinking reagent has one of the following structures:

$$Y_2 \xrightarrow{Y_1} Y_4 Y_5$$
 $Y_3 Y_6 \qquad Y_7 = -L - Y_8$

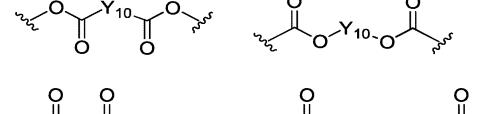
where Y_1 , Y_2 , Y_3 , Y_4 , Y_5 , Y_6 , Y_7 and Y_8 are each independently selected from hydrogen, optionally substituted cyclic, branched or straight chain C_1 - C_6 alkyl or aryl; and L is a linker moiety.

5. An electronic or electrical device or component thereof according to claim 4, wherein for compound (i) L has the following formula:

each Y_9 is independently selected from, a bond, -O-, -O-C(O)-, -C(O)-O-, -Y₁₁-O-C(O)-, -C(O)-O-Y₁₁-, -OY₁₁-, and -Y₁₁O-, where Y₁₁ is an optionally substituted cyclic, branched or straight chain C₁-C₈ alkylene; and

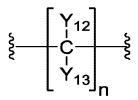
 Y_{10} is selected from optionally substituted cyclic, branched or straight chain C_1 - C_8 alkylene and a siloxane group.

6. An electronic or electrical device or component thereof according to claim 5, wherein for compound (i) L has one of the following structures:



7. An electronic or electrical device or component thereof according to claim 5, wherein for compound (i) L has one of the following structures:

8. An electronic or electrical device or component thereof according to any of claims 5 to 7, wherein Y_{10} has the following formula:



wherein each Y_{12} and Y_{13} is independently selected from H, halo, optionally substituted cyclic, branched or straight chain alkyl, or $-OY_{14}$, where Y_{14} is selected from optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl, and n is an integer from 1 to 10.

- 9. An electronic or electrical device or component thereof according to claim 8, wherein each Y_{12} is H and each Y_{13} is H.
- 10. An electronic or electrical device or component thereof according to claim 8, wherein each Y_{12} is fluoro and each Y_{13} is fluoro.
- 11. An electronic or electrical device or component thereof according to any of claims 8 to 10, wherein n is from 4 to 6.
- 12. An electronic or electrical device or component thereof according to any of claims 5 to 7, wherein Y_{10} has the following formula:

Wherein each Y_{15} is independently selected from optionally substituted branched or straight chain C_1 - C_8 alkyl.

- 13. An electronic or electrical device or component thereof according to claim 12, wherein each Y_{15} is methyl, and each Y_9 is a bond.
- 14. An electronic or electrical device or component thereof according to any of claims 5 to 7, wherein Y_{10} has the following formula:

wherein Y_{16} to Y_{19} are each independently selected from H and optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl.

- 15. An electronic or electrical device or component thereof according to claim 14, wherein Y_{18} is H or vinylene, and Y_{18} , Y_{17} and Y_{19} are each H.
- 16. An electronic or electrical device or component thereof according to any preceding claim wherein the crosslinking reagent is selected from Divinyl Adipate (DVA), 1,4 Butanediol divinyl ether (BDVE), 1,4 Cyclohexanedimethanol divinyl ether (CDDE), 1,7 -Octadiene (17OD), 1,2,4 Trivinylcyclohexane (TVCH), 1,3-Divinyltetramethyldisiloxane (DVTMDS), Diallyl 1,4-cyclohexanedicarboxylate (DCHD), 1,6-Divinylperfluorohexane (DVPFH), 1H,1H,6H,6H-Perfluorohexanediol diacrylate (PFHDA) and glyoxal bis (diallyl acetal) (GBDA).
- 17. An electronic or electrical device or component thereof according to claim 4, wherein for compound (ii), L is selected from a branched or straight chain C₁-C₈ alkylene or an ether group.
- 18. An electronic or electrical device or component thereof according to any preceding claim wherein the monomer is a compound of formula I(a):

I(a)

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wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_6 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1;

; or

wherein the monomer is a compound of formula I(b):

l(b)

$$R_{2} \xrightarrow{R_{1}} O \xrightarrow{R_{5}} R_{7} \begin{bmatrix} R_{9} \\ \dot{C} \end{bmatrix} \begin{bmatrix} X \\ \dot{C} \end{bmatrix} \begin{bmatrix} X \\ \dot{C} \end{bmatrix} C C X_{3}$$

$$R_{4} \xrightarrow{C} R_{6} R_{8} \begin{bmatrix} R_{9} \\ R_{10} \end{bmatrix}_{a} \begin{bmatrix} X \\ \dot{C} \end{bmatrix}_{b} \begin{bmatrix} X \\ \dot{C} \\ X \end{bmatrix}_{c} C X_{3}$$

wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_6 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1.

- 19. An electronic or electrical device or component thereof according to claim 18, wherein the halogen is fluorine.
- 20. An electronic or electrical device or component thereof according to claim 18 or claim 19, wherein each of R₁, R₂, R₄, and R₅ to R₁₀ is independently selected from hydrogen, methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, tert-butyl, sec-butyl, n-pentyl, neopentyl, n-hexyl, isohexyl, and 3-methylpentyl.
- 21. An electronic or electrical device or component thereof according to claim 20, wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or methyl.
- 22. An electronic or electrical device or component thereof according to any of claims 18 to 20, wherein a and c are each independently 0 or 1; and b is from 3 to 7.
- 23. An electronic or electrical device or component thereof according to any preceding claim wherein each X is H.
- 24. An electronic or electrical device or component thereof according to any of claims 1 to 22, wherein each X is F.
- 25. An electronic or electrical device or component thereof according to any of claims 18 to 24, wherein R_1 and R_2 are both hydrogen.
- 26. An electronic or electrical device or component thereof according to any of claims 18 to 25, wherein R_4 is hydrogen or methyl.
- 27. An electronic or electrical device or component thereof according to any of claims 18 to 26, wherein R_9 is hydrogen and R_{10} is branched or straight chain C_1 - C_6 alkyl group.
- 28. An electronic or electrical device or component thereof according to claim 27, wherein R₁₀ is methyl.
- 29. An electronic or electrical device or component thereof according to any of claims 18 to 28, wherein each of R_5 to R_8 is hydrogen.
- 30. An electronic or electrical device or component thereof according to any of claims 18 to 29 wherein each of each of R₁, R₂, R₄, and R₅ to R₁₀ is hydrogen, each X is H, a=0 and c=0.
- 31. An electronic or electrical device or component thereof according to any of claims 24 to 30, wherein the compound of formula I(a) has the following formula:

$$H \longrightarrow O$$
 $(CF_2)_nF$

where n is from 2 to 10.

32. An electronic or electrical device or component thereof according to any of claims 24 to 30, wherein the compound of formula I(a) has the following formula:

$$H \longrightarrow O (CF_2)_nF$$
 $H \longrightarrow CH_3$

where n is from 2 to 10.

- 33. An electronic or electrical device or component thereof according to claim 31, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-perfluorohexyl acrylate (PFAC4), 1H,1H,2H,2H-perfluorooctyl acrylate (PFAC6), 1H,1H,2H,2H-perfluorodecyl acrylate (PFAC8) and 1H,1H,2H,2H-perfluorododecyl acrylate (PFAC10).
- 34. An electronic or electrical device or component thereof according to claim 32, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-pefluorohexyl methacrylate (PFMAC4), 1H,1H,2H,2H-perfluorooctyl methacrylate (PFMAC6) and 1H,1H,2H,2H-perfluorodecyl methacrylate (PFMAC8).
- 35. An electronic or electrical device or component thereof according to any of claims 18 to 23, wherein the compound of formula I(a) has the following formula:

wherein a and c are each independently 0 or 1, b =3-7 and n is 4 to 10, where n=a+b+c+1.

36. An electronic or electrical device or component thereof according to any of claims 18 to 23, wherein the compound of formula I(a) has the following formula:

$$H \longrightarrow O$$
 $CH_2)_nH$
 CH_3

where n is 2 to 12.

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- 37. An electronic or electrical device or component thereof according to claim 35 or claim 36, wherein the compound of formula I(a) is selected from ethyl hexyl acrylate, hexyl acrylate, decyl acrylate, lauryl dodecyl acrylate and iso decyl acrylate.
- 38. An electronic or electrical device or component thereof according to any of claims 18 to 23, wherein the monomer is a compound of formula I(b) and has the following formula:

$$\begin{array}{c}
R_1 & O \\
R_2 & R_3
\end{array}$$

$$\begin{array}{c}
(CH_2)_n - CH_3$$

where n is from 4 to 14, optionally where R_1 , R_2 and R_3 are each H.

- 39. An electronic or electrical device or component thereof according to any of the preceding claims, wherein the electronic or electrical device or component thereof is selected from mobile phones, smartphones, pagers, radios, sound and audio systems such as loudspeakers, microphones, ringers and/or buzzers, hearing aids, personal audio equipment such as personal CD, tape cassette or MP3 players, televisions, DVD players including portable DVD players, video recorders, digi and other set-top boxes, computers and related components such as laptop, notebook, tablet, phablet or palmtop computers, personal digital assistants (PDAs), keyboards, or instrumentation, games consoles, data storage devices, outdoor lighting systems, radio antennae and other forms of communication equipment, and printed circuit boards.
- 40. A method for treating an electronic or electrical device or component thereof comprising:

exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow a protective cross-linked polymeric coating to form on a surface of the electronic or electrical device or component thereof;

wherein the monomer compound has the following formula:

$$R_1$$
 R_2
 R_4

where R_1 , R_2 and R_4 are each independently selected from hydrogen, optionally substituted branched or straight chain C_1 - C_6 alkyl or halo alkyl or aryl optionally substituted by halo, and R_3 is selected from:

$$\begin{array}{c} O \\ \begin{array}{c} X \\ \\ \end{array} \\ \begin{array}{c} O \\ \end{array} \\ \begin{array}{c} X \\ \\ \end{array} \\ \begin{array}{c} O \\ \end{array} \\ \begin{array}{c} X \\ \\ \end{array} \\ \begin{array}{c} O \\ \end{array} \\ \begin{array}{c} X \\ \\ \end{array} \\ \begin{array}{c} C \\ \end{array} \\ \\ \begin{array}{c} C \\ \end{array} \\ \begin{array}{c} C \\ \end{array}$$

where each X is independently selected from hydrogen, optionally substituted branched or straight chain C_1 - C_6 alkyl, halo alkyl or aryl optionally substituted by halo; and n_1 is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moieties and has a boiling point at standard pressure of less than 500°C.

41. A method according to claim 40, wherein the crosslinking reagent has one of the following structures:

$$Y_2 \xrightarrow{Y_1} Y_4 Y_5 Y_6 Y_7 = L = Y_8$$

where Y_1 , Y_2 , Y_3 , Y_4 , Y_5 , Y_6 , Y_7 and Y_8 are each independently selected from hydrogen, optionally substituted cyclic, branched or straight chain C_1 - C_8 alkyl or aryl; and L is a linker moiety.

42. A method according to claim 41, wherein for compound (i) L has the following formula:

each Y_9 is independently selected from, a bond, -O-, -O-C(O)-, -C(O)-O-, -Y₁₁-O-C(O)-, -C(O)-O-Y₁₁-, and -Y₁₁O-, where Y₁₁ is an optionally substituted cyclic, branched or straight chain C₁-C₈ alkylene; and

 Y_{10} is selected from optionally substituted cyclic, branched or straight chain C_1 - C_8 alkylene and a siloxane group.

43. A method according to claim 42, wherein for compound (i) L has one of the following structures:

44. A method according to claim 42, wherein for compound (i) L has one of the following structures:

45. An method according to any of claims 42 to 44, wherein Y₁₀ has the following formula:

$$\xi = \begin{bmatrix} Y_{12} \\ C \\ Y_{13} \end{bmatrix}_n$$

wherein each Y_{12} and Y_{13} is independently selected from H, halo, optionally substituted cyclic, branched or straight chain alkyl, or $-OY_{14}$, where Y_{14} is selected from optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl, and n is an integer from 1 to 10.

- 46. A method according to claim 45, wherein each Y₁₂ is H and each Y₁₃ is H.
- 47. A method according to claim 45, wherein each Y₁₂ is fluoro and each Y₁₃ is fluoro.
- 48. A method according to any of claims 45 to 47, wherein n is from 4 to 6.
- 49. A method according to any of claims 42 to 44, wherein Y₁₀ has the following formula:

Wherein each Y_{15} is independently selected from optionally substituted branched or straight chain C_1 - C_8 alkyl.

- 50. A method according to claim 49, wherein each Y_{15} is methyl, and each Y_{9} is a bond.
- 51. A method according to any of claims 42 to 44, wherein Y_{10} has the following formula:

wherein Y_{16} to Y_{19} are each independently selected from H and optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl.

- 52. An method according to claim 51, wherein Y_{18} is vinylene or H, and Y_{16} , Y_{17} and Y_{19} are each H.
- 53. A method according to any of claims 40 to 52, wherein the crosslinking reagent is selected from Divinyl Adipate (DVA), 1,4 Butanediol divinyl ether (BDVE), 1,4 Cyclohexanedimethanol divinyl ether (CDDE), 1,7 -Octadiene (17OD), 1,2,4 -Trivinylcyclohexane (TVCH), 1,3-Divinyltetramethyldisiloxane (DVTMDS), Diallyl 1,4-cyclohexanedicarboxylate (DCHD), 1,6-Divinylperfluorohexane (DVPFH), 1H,1H,6H,6H-Perfluorohexanediol diacrylate (PFHDA) and glyoxal bis (diallyl acetal) (GBDA).
- 54. A method according to claim 41, wherein for compound (ii), L is selected from a branched or straight chain C₁-C₈ alkylene or an ether group.
- 55. A method according to any of claims 40 to 54 wherein the monomer is a compound of formula l(a):

I(a)

wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_6 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1;

; or

wherein the monomer is a compound of formula I(b):

I(b)

$$R_{2} \xrightarrow[R_{4}]{R_{1}} \xrightarrow[O]{R_{5}} \xrightarrow[R_{6}]{R_{7}} \left[\begin{array}{c} R_{9} \\ C \\ R_{10} \end{array} \right]_{a} \left[\begin{array}{c} X \\ C \\ X \end{array} \right]_{b} \left[\begin{array}{c} X \\ C \\ C \\ X \end{array} \right]_{c} CX_{3}$$

wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_6 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1.

- 56. A method according to claim 55, wherein the halogen is fluorine.
- 57. A method according to claim 55 or claim 56, wherein each of R₁, R₂, R₄, and R₅ to R₁₀ is independently selected from hydrogen, methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, tert-butyl, sec-butyl, n-pentyl, neopentyl, n-hexyl, isohexyl, and 3-methylpentyl.
- 58. A method according to claim 57, wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or methyl.
- 59. A method according to any of claims 55 to 58, wherein a and c are each independently 0 or 1; and b is from 3 to 7.
- 60. A method according to any preceding claim wherein each X is H.
- 61. A method according to any of claims 41 to 59, wherein each X is F.
- 62. A method according to any of claims 41 to 61, wherein R_1 and R_2 are both hydrogen.
- 63. A method according to any of claims 41 to 62, wherein R₄ is hydrogen or methyl.
- 64. A method according to any of claims 41 to 63, wherein R_9 is hydrogen and R_{10} is branched or straight chain C_1 - C_6 alkyl group.
- 65. A method according to claim 64, wherein R_{10} is methyl.
- 66. A method to any of claims 41 to 65, wherein each of R_5 to R_8 is hydrogen.
- 67. A method according to any of claims 41 to 66 wherein each of each of R_1 , R_2 , R_4 , and R_5 to R_{10} is hydrogen, each X is H, a=0 and c=0.
- 68. A method according to any of claims 61 to 67, wherein the compound of formula I(a) has the following formula:

$$H \longrightarrow O (CF_2)_nF$$

where n is from 2 to 10.

69. A method according to any of claims 61 to 67, wherein the compound of formula I(a) has the following formula:

$$H$$
 O
 CH_3
 $(CF_2)_nF$

where n is from 2 to 10.

- 70. A method according to claim 68, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-perfluorohexyl acrylate (PFAC4), 1H,1H,2H,2H-perfluorooctyl acrylate (PFAC6), 1H,1H,2H,2H-perfluorodecyl acrylate (PFAC8) and 1H,1H,2H,2H-perfluorododecyl acrylate (PFAC10).
- 71. A method according to claim 69, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-perfluorohexyl methacrylate (PFMAC4), 1H,1H,2H,2H-perfluorooctyl methacrylate (PFMAC6) and 1H,1H,2H,2H-perfluorodecyl methacrylate (PFMAC8).
- 72. A method according to any of claims 41 to 61, wherein the compound of formula I(a) has the following formula:

wherein a and c are each independently 0 or 1, b =3-7 and n is 4 to 10, where n=a+b+c+1.

73. A method according to any of claims 41 to 61, wherein the compound of formula I(a) has the following formula:

where n is 2 to 12.

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- 74. A method according to claim 72 or claim 73, wherein the compound of formula I(a) is selected from ethyl hexyl acrylate, hexyl acrylate, decyl acrylate, lauryl dodecyl acrylate and iso decyl acrylate.
- 75. A method according to any of claims 41 to 60, wherein the monomer is a compound of formula I(b) and has the following formula:

where n is from 4 to 14.

76. A method according to claim 75 wherein the monomer is a compound of formula I(b) and has the following formula:

where n is from 4 to 14.

- 77. A method according to any of claims 40 to 76 wherein the electronic or electrical device or component thereof is placed in a plasma deposition chamber, a glow discharge is ignited within said chamber, and a voltage applied as a pulsed field.
- 78. A method according to claim 77 wherein applied voltage is at a power of from 40 to 500W.
- 79. A method according to claim 77 or claim 78 wherein the voltage is pulsed in a sequence in which the ratio of the time on : time off is in the range of from 1:500 to 1:1500.
- 80. A method according to any of claims 77 to 79 wherein the voltage is pulsed in a sequence where power is on for 20-50μs, and off for from 1000μs to 30000μs.
- 81. A method according to any of claims 77 to 80 wherein the voltage is applied as a pulsed field for a period of from 30 seconds to 90 minutes.
- 82. A method according to claim 81 wherein the voltage is applied as a pulsed field for from 5 to 60 minutes.
- 83. A method according to any one of claims 77 to 82, wherein in a preliminary step, a continuous power plasma is applied to the electronic or electrical device or component thereof.

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Tack Free* Tacky

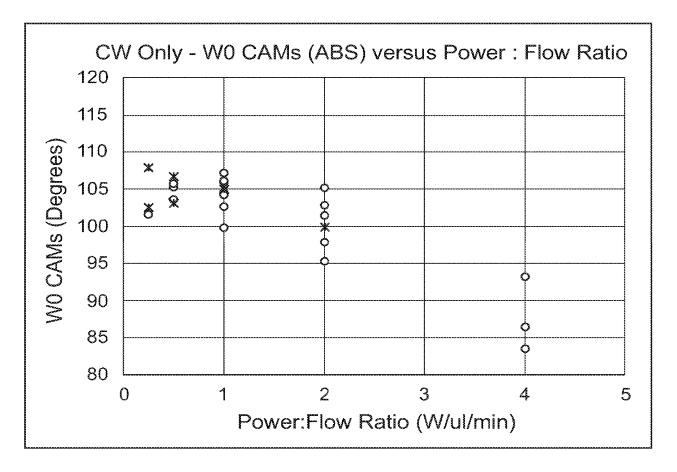


Figure 1

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Tack Free* Tacky

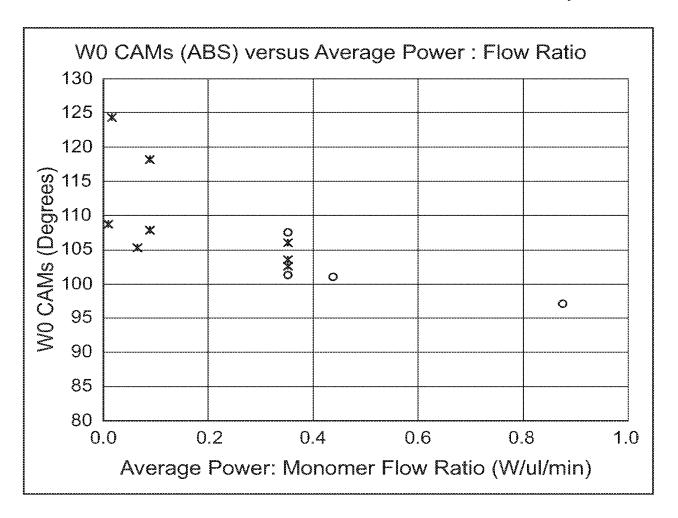
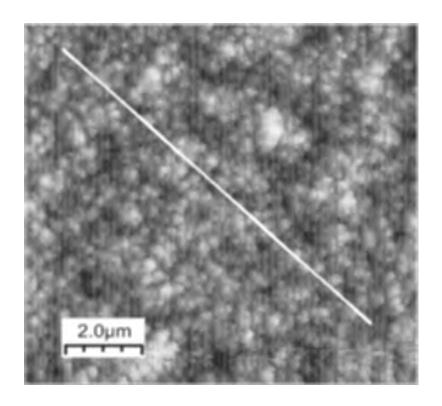


Figure 2

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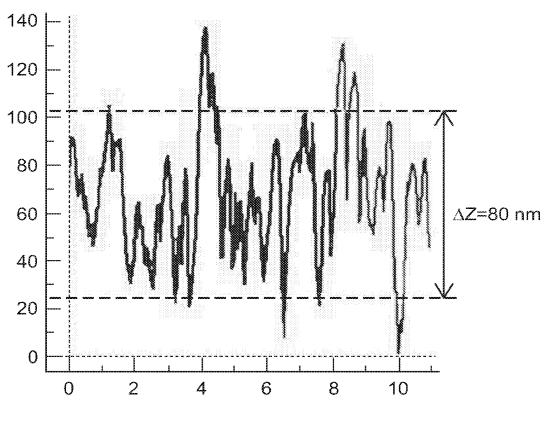


Figure 3

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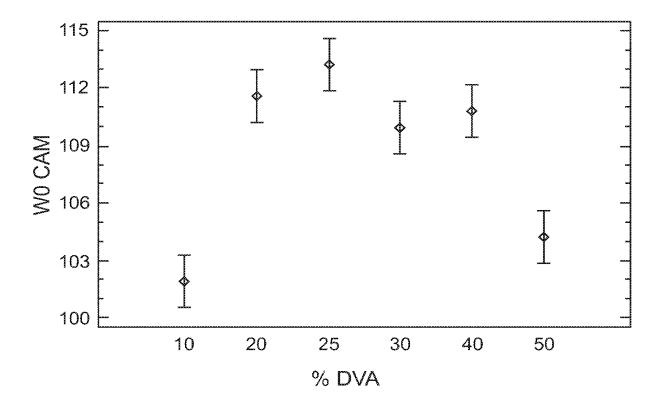
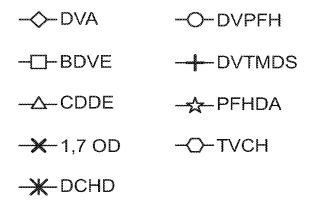


Figure 4

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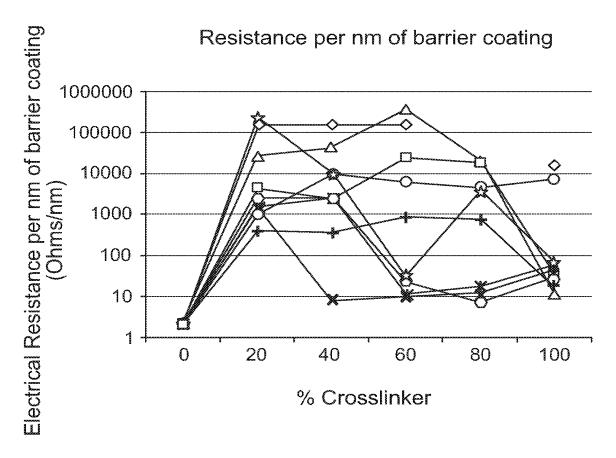
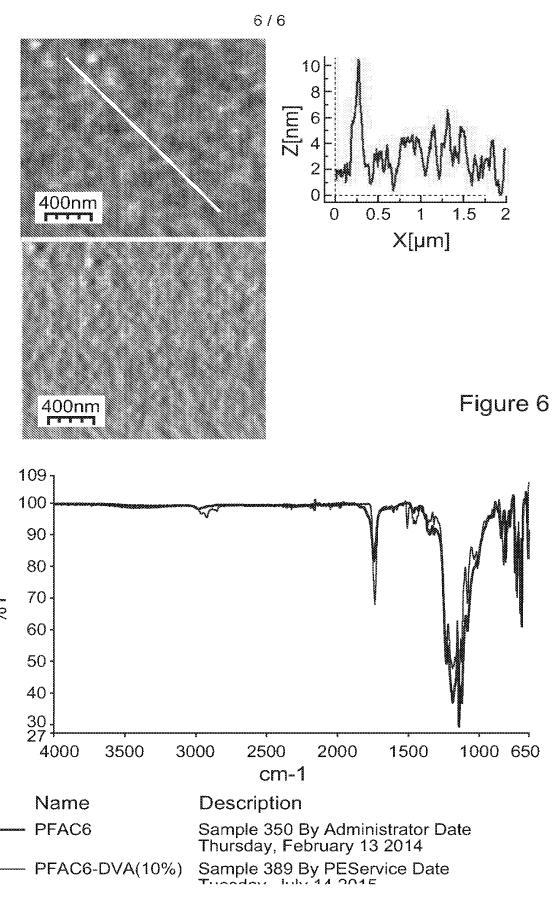


Figure 5



Ex. 1028 - Favored - p. 124 SUBSTITUTE SHEET (RULE 26)

PCT/GB2016/051688

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IPC CLASSIFICATION: CO9D5/00

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Bescheinigung

Certificate

Attestation

Die angehefteten Unterlagen stimmen mit der als ursprünglich eingereicht geltenden Fassung der auf dem nächsten Blatt bezeichneten europäischen Patentanmeldung überein. The attached documents are exact copies of the text in which the European patent application described on the following page is deemed to have been filed.

Les documents joints à la présente attestation sont conformes au texte, considéré comme initialement déposé, de la demande de brevet européen qui est spécifiée à la page suivante.

Patentanmeldung Nr.

Patent application No.

Demande de brevet nº

15386017.6 / EP15386017

The organization code and number of your priority application, to be used for filing abroad under the Paris Convention, is EP15386017.

Der Präsident des Europäischen Patentamts; Im Auftrag

For the President of the European Patent Office Le President de l'Office européen des brevets p.o.

V. Joseph

Anmeldung Nr: Application no.:

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Demande no :

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p2i Limited Global Headquarters 9-12 North Central 127 Olympic Avenue Milton Park Abingdon Oxfordshire OX14 4SA/GB

Bezeichnung der Erfindung / Title of the invention / Titre de l'invention: (Falls die Bezeichnung der Erfindung nicht angegeben ist, oder falls die Anmeldung in einer Nicht-Amtssprache des EPA eingereicht wurde, siehe Beschreibung bezüglich ursprünglicher Bezeichnung. If no title is shown, or if the application has been filed in a non-EPO language, please refer to the description for the original title. Si aucun titre n'est indiqué, ou si la demande a été déposée dans une langue autre qu'une langue officielle de l'OEB, se référer à la description pour le titre original.)

COATINGS

In Anspruch genommene Prioritat(en) / Priority(Priorities) claimed / Priorité(s) revendiquée(s) Staat/Tag/Aktenzeichen / State/Date/File no. / Pays/Date/Numéro de dépôt:

Am Anmeldetag benannte Vertragstaaten / Contracting States designated at date of filing / Etats contractants désignées lors du dépôt:

AL AT BE BG CH CY CZ DE DK EE ES FI FR GB GR HR HU IE IS IT LI LT LU LV MC MK MT NL NO PL PT RO RS SE SI SK SM TR

Coatings

Field of the invention

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This invention relates to protective coatings. Specifically, the present invention relates to protective coatings for electronic or electrical devices and components thereof, and methods of forming such coatings. The coatings can protect by being hydrophobic and so resist the ingress of water-based liquid into electronic devices, or they can protect by forming a barrier coating and so provide electrical resistance between the electrical parts of the phone and water based liquid.

Background of the invention

Monounsaturated monomers are used to make barrier coatings using plasma polymerisation processes (see co-pending application).

The perfluoroalkyl chain monomers are also used for generating hydrophobic surfaces from pulsed plasma deposition processes (see WO 9858117 A1).

The power of the plasma initiated polymerisation affects the nature of the polymer produced. The higher average energy inputs of continuous wave plasmas lead to more fragmentation of the monomer, and so the polymer loses structural properties of the monomer. In the case of 1H,1H,2H,2H- perfluorodecyl acrylate (PFAC8), there's less retention of the perfluoroalkyl chain and the contact angle of the surface coating is compromised. Higher plasma energies also lead to more crosslinking. For the lower average energy inputs of pulsed plasmas, there's better retention of monomer structure and less crosslinking. The greater retention of the perfluoro chain under low energy, pulsed plasma conditions leads to best levels of contact angles for the surface coating.

When the perfluoroalkyl chains have eight or more fluorinated carbons (long chain), the polymer made from the monomer has a crystalline structure. When the perfluoroalkyl chains have less than eight fluorinated carbons, the resulting polymer is amorphous and so can be unstable in the presence of water (see "Molecular Aggregation Structure and Surface Properties of Poly(fluoroalkyl acrylate) Thin Films, Marcomolecules, 2005, vol 38, p5699-5705)

When long chain perfluoroalkyl polymers are produced by high average power (continuous wave or CW) or low average power (pulsed wave or PW) plasmas, then because of the crystalline structure of the long chains, the polymers are non-stick to the touch and stable in the presence of water. However, the feel and water stability of shorter chain polymer coatings is affected by the plasma power levels used. For example, when PFAC6 (1H,1H,2H,2H-perfluorooctyl acrylate) is polymerised in low power plasma conditions, the resulting polymer coating can have several disadvantages. For example the coating can cause water drops to spread out a little (slump), be marked by the presence of a water drop on its surface, have a tacky feel, or can be easily smeared (for example on substrates of silicon wafer and ABS plastic).

By increasing the power of the plasma used for polymerisation, the polymer becomes more crosslinked and becomes more resistant to smearing. However, increasing the power has the concomitant effect of decreasing the water contact angle through more monomer fragmentation (as described above). Figure 1 shows the effect of increasing the power to monomer flow ratio for CW plasmas in a 125 litre chamber: at a ratio of 4 W/µl/min, the water contact angle is ~85-95 degrees and the coating is tack-free. However, as the ratio drops, the contact angle increases and the occurrence of tacky/smudgy coatings increases too. Figure 2 shows the same effect for pulsed plasma conditions. These results show that the process window for producing tack and smudge-free coatings have a limited plasma processing range and the final coating has a compromised water contact angle.

Accordingly, it is an aim to solve one or more of the above-mentioned problems with the prior art coatings.

Statement of Invention

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An aspect of the present invention provides an electronic or electrical device or component thereof comprising a protective cross-linked polymeric coating on a surface of the electronic or electrical device or component thereof;

wherein the protective cross-linked polymeric coating is obtainable by exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow formation of the protective cross-linked polymeric coating on a surface thereof,

wherein the monomer compound has the following formula:

where R_1 , R_2 and R_4 are each independently selected from hydrogen, optionally substituted branched or straight chain C_1 – C_6 alkyl or halo alkyl or aryl optionally substituted by halo, and R_3 is selected from:

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where each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C₁-C₈ alkyl, halo alkyl or anyl optionally substituted by halo; and n₁ is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moleties and has a boiling point of less than 500 °C at standard pressure.

High levels of polymer crosslinking (formerly only achievable with high average power continuous wave plasmas) can be achieved by adding a crosslinking molecule to the monomer to produce a cross-linked co-polymer. This has the advantage of increasing the plasma processing range so stable coatings can now be produced in low average energy pulse plasma conditions.

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- The high degree of retention of the hydrophobic monomer structures from the low energy pulsed plasma gives coatings of the co-polymer a good hydrophobic coating (as evidenced by water contact angles) and a coating that is not sticky or smeary to the touch. Thus, these properties make it suitable for the coating of electronic devices to prevent the ingress of water-based liquids from accidental damage. Such coatings typically have a thickness between 1 and 100 nm, preferably 1 and 50 nm.
- The coating may be described as a liquid repellant layer, typically a water repellant layer. The coating may repel water, an aqueous liquid such as rainwater, or oil.

Alternatively, or in addition to providing a liquid repellent layer, the coating may be a physical barrier layer.

The coating may be a physical barrier i.e. it provides a physical barrier to mass and electron transport.

The physical barrier layer restricts diffusion of water, oxygen and ions. When the coating is a physical barrier it typically has a thickness of greater than 50 nm.

The coating is a protective layer i.e. it prevents damage by contact with water or other liquids. The coating may provide is protective functionality by forming a liquid repellent layer and/or a physical barrier layer.

The coating is preferably substantially pin-hole free to enable it to provide a physical barrier.

Preferably ΔZ/d <0.15, where ΔZ is the average height variation on an AFM line scan in nm (as shown in Figure 3) and d is coating thickness in nm.

The value of $\Delta Z/d$ tells us to what extent defects/voids on the surface of the coating extend into the coating, i.e. the percentage value of the depth of defect over total coating thickness. For example, $\Delta Z/d = 0.15$ means that the voids on the surface only extend down to 15% of the coating thickness. A coating with a $\Delta Z/d < 0.15$ is defined herein as being substantially pinhole free. If voids are bigger than this, the desired functionality is unlikely to be achieved.

The coating is preferably conformal, which means that it takes the 3D shape of the electronic or electrical device or component thereof and covers substantially an entire surface of the device. This has the advantage of ensuring that the coating has sufficient thickness to give optimal functionality over an entire surface of the device or component. The meaning of the term "covers substantially an entire surface" will depend to some extent on the type of surface to be covered. For example, for some components, it may be necessary for there to be complete coverage of the surface in order for the component to function after submersion in water. However, for other components or housings, small gaps in coverage may be tolerated.

The coating may have a thickness of 50 to 10,000nm, optionally 50 to 8000nm, 100 to 5000nm, preferably 250nm – 5000nm, most preferably 250nm – 2000nm.

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The coating may be electrically insulating and be sufficiently compliant that electrical connectors can be joined to the electronic or electrical device or component thereof and an electrical connection made between the electrical connectors and electronic or electrical device or component thereof without the requirement to first remove the coating. In this case, the force exerted by the electrical connector on the coating is sufficient to alter the structure of or even break through the coating local to the electrical connector, allowing the electrical connection to be made. Electrical connectors can typically be joined to the electronic or electrical device or component in this way for coating thicknesses of under 5000nm, and for high performance coatings below 2000nm.

The electronic or electrical device or component thereof can typically withstand immersion in up to 1m of water for over 30 minutes without failure or corrosion whilst power applied to the electronic or electrical device or component. The effectiveness of the coating can be determined by measuring its electrical resistance at a fixed voltage when submerged in water for a set time period; for example, applying a voltage of 8V to the coating of a device which is submerged in water for 13 minutes. Coatings having a resistance value of 1x10⁷ Ohms or higher in this test are effective barrier coatings and the coated electronic or electrical device or component thereof will pass successfully an IPX7 test. The IPX7 test is the Ingress Protection Marking which classifies and rates the degree of protection provided against water. In the IPX7 test for phones, the device is immersed in water under defined conditions of pressure and time (up to 1 m of submersion) for a duration of 30 minutes. The device has to be powered on during testing and functional after 24 hrs.

In one embodiment, the coating is electrically insulating and has a thickness of less than 1 micron and wherein a force of 5- 20g applied to the coating using a round probe with 1 mm diameter allows an electrical connection to be made with the electronic or electrical device or component thereof in the local area where the force has been applied.

In another embodiment, the coating is electrically insulating and has a thickness of 1-2.5 microns and wherein a force of 20-100g applied to the coating using a round probe with 1 mm diameter allows an electrical connection to be made in the local area of the coating where the force has been applied.

The coating may have a higher density than that of the corresponding monomer from which it is formed. For example, the increase in density may be at least 0.1 g/cm³. The increase in density is explained by the highly crosslinked coating. The high density of the coating improves the barrier properties of the coating.

The coating may form a surface defined by a static water contact angle (WCA) of at least 70°. A coating with a WCA of at least 90° is a liquid repellent, typically a water repellent layer. For fluorinated polymers, the coating may have a static water contact angle of at least 100°. The contact angle of a liquid on a solid substrate gives an indication of the surface energy which in turn illustrates the substrate's liquid repellence. Contact angles were measured on a VCA Optima contact angle analyser, using 3 µl droplets of deionised water at room temperature.

Crosslinker

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A basic requirement for the cross-linking reagent is the presence two or more unsaturated bonds, for example, -C=C- or alkyne groups. The unsaturated bonds are attached by one or more linker moieties. The linker moiety is not especially limited provided it joins together the two or more unsaturated bonds. The cross-linking reagent must have a boiling point of less than 500 °C, preferably from -10 to 300°C, optionally 180 to 270°C, most preferably 205 to 260°C at standard pressure. It is a preference, but not a requirement, that the cross-linking reagent is not unduly hazardous for use in plasma processing, that is to say it can be used in a production environment where low levels of vapour do not present any significant health and safety issues (e.g. strongly oxidising, explosive, toxic, or has an offensive odour (e.g. stench reagent)). The crosslinking reagent preferably has one of the following structures:

where Y_1 , Y_2 , Y_3 , Y_4 , Y_5 , Y_8 , Y_7 and Y_8 are each independently selected from hydrogen, optionally substituted cyclic, branched or straight chain C_1 - C_8 alkyl or aryl; and L is a linker molety.

Most preferably L has the following formula:

where each Y_8 is independently selected from, a bond, -O-, -O-C(O)-, -C(O)-O-, -Y₁₁-O-C(O)-, -C(O)-O-Y₁₁-, -OY₁₁-, and -Y₁₁O-, where Y₁₁ is an optionally substituted cyclic, branched or straight chain C₁-C₈ alkylene; and

 Y_{10} is selected from optionally substituted cyclic, branched or straight chain C_1 - C_8 alkylene and a siloxane containing group.

In one most preferred embodiment Y₁₀ has the following formula:

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wherein each Y_{12} and Y_{13} is independently selected from H, halo, optionally substituted cyclic, branched or straight chain C_1 - C_8 alkyl, or - OY_{14} , where Y_{14} is selected from optionally substituted branched or straight chain C_1 - C_8 alkyl or alkeryl, and n is an integer from 1 to 10.

Optionally, each Y_{12} is H and each Y_{13} is H, i.e. Y_{10} is a linear alkylene chain. For this embodiment Y_9 is preferably a vinyl ester or vinyl ether group.

Optionally, each Y₁₂ is fluoro and each Y₁₃ is fluoro, i.e. Y₁₀ is a linear perfluoroalkylene chain.

Typically, n is from 4 to 6.

In another embodiment Y₁₀ has the following formula:

15 (v)

wherein each Y_{16} is independently selected from optionally substituted branched or straight chain C_t - C_6 alkyl.

In one embodiment each Y_{15} is methyl, and each Y_{9} is a bond.

In a further embodiment Y₁₀ has the following formula:

(vi)

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wherein Y_{18} to Y_{19} are each Independently selected from H and optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl. A preferred alkenyl group is vinyl. Optionally Y_{18} is H or vinyl, and Y_{18} , Y_{17} and Y_{19} are each H. In one embodiment each of Y_{18} to Y_{19} is H. In another embodiment Y_{18} is vinyl, and Y_{19} , Y_{17} and Y_{19} are each H.

In preferable embodiments for compound (I), L has one of the following structures:

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In another embodiment for compound (i) L has one of the following structures:

For L according structure (vii), Y₁₀ is preferably an alkylene chain or a cycloalkylene, such as those shown in structures Iv) and vi) above. The alkylene chain may be a straight chain alkylene chain.

When Y₁₀ is a cycloalkylene, cyclohexylene is preferred, most preferably 1,4 cyclohexylene.

For L according to structure (viii), Y₁₀ is preferably structure (iv), i.e. an alkylene or fluoroalkylene chain.

For L according to structure (ix), Y₁₀ is preferably cycloalkylene, such as the cyclohexylene according to structure (vI).

For L according to structure (x), Y_{10} is preferably structure (iv) wherein each Y_{12} and Y_{13} is F, i.e. a perfluoroalkylene chain.

For L according to structure (xi) or structure (xii) Y_{10} is preferably alkylene or cycloalkylene. Optionally the alkylene or cycloalkylene may be substituted with one or more vinyl groups or alkenyl ether groups, preferably one or more vinyl ether groups.

When each Y₀ is a bond, each Y₁₀ may be any of structures (iv), (v) and (vi). Preferably Y₁₀ is a straight chain alkylene such that the crosslinking reagent is a diene, such as a heptadiene, octadiene, or nonadiene, most preferably 1,7-octadiene.

When each Y_9 is O, each Y_{10} is preferably a branched or straight chain C_1 - C_6 alkylene, preferably a straight chain alkylene, most preferably a C_4 straight chain alkylene, i.e the crosslinking reagent is 1,4-butanediol divinyl ether

It will be understood that each Y_{θ} group can be combined with any other Y_{θ} group and Y_{10} group to form the crosslinking reagent.

The skilled person will be aware of possible substituents for each of the cyclic, branched or straight chain C₁-C₈ alkylene groups mentioned above. The alkylene groups may be substituted at one or more positions by a suitable chemical group. Halo substituents are preferred, with fluoro substituents most preferred. Each C₁-C₈ alkylene group may be a C₁-C₃, C₂-C₆, or C₆-C₈ alkylene group.

Particularly preferred embodiments of the crosslinker have alkyl chains for Y₁₀ and vinyl ester or vinyl ether groups on either side.

A particularly preferred cross-linking reagent is divinyl adipate. (a divinyl ester)

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Another preferred cross-linking reagent is 1,4-butanediol divinyl ether (a divinyl ether)

For the most preferred embodiments the crosslinking reagent is selected from Divinyl Adipate (DVA), 1,4 Butanediol divinyl ether (BDVE), 1,4 Cyclohexanedimethanol divinyl ether (CDDE), 1,7 -Octadlene (17OD), 1,2,4 -Trivinylcyclohexane (TVCH), 1,3-Divinyltetramethyldisiloxane (DVTMDS), Dlallyl 1,4-cyclohexanedicarboxylate (DCHD), 1,6-Divinylperfluorohexane (DVPFH), 1H,1H,6H,6H-Perfluorohexanediol diacrylate (PFHDA) and glyoxal bls (diallyl acetal) (GBDA).

For the alkyne crosslinking reagents according to compound (ii), L is preferably selected from a branched or straight chain C_1 - C_8 alkylene or an ether group. L may be a C_3 , C_4 , C_5 , or C_8 alkylene,

preferably a straight chain alkylene. Chemical structures for the particularly preferred crosslinking reagents are set out below in Table 1:

Divinyl Adipate (DVA)	H ₂ C, O, O
	O CHF
1,4 Butanediol divinyl ether (BDVE)	H ₂ C O O O O O O O O O O O O O O O O O O O
1,4 Cyclohexanedimethanol divinyl ether (CDDE)	H ₂ C_O CH ₂
1,7 -Octadiene (17OD)	H ₂ C CH ₂
1,2,4 -Trivinylcyclohexane (TVCH)	H ₂ C CH ₂
1,3- Divinyltetramethyldisiloxane (DVTMDS)	CH ₃ CH ₃ Si O Si CH ₂ CH ₃ CH ₂
Diallyl 1,4- cyclohexanedicarboxylate (DCHD)	CH ₂
1,6-Divinylperfluorohexane (DVPFH)	H ₂ C F F F CH ₂
1H,1H,6H,6H- Perfluorohexanediol diacrylate (PFHDA)	H ₂ C CH ₂
GBDA	H ₂ C CH ₂

1,6-heptadiyne	нс
1,7-heptadiyne	нс
1,8-heptadiyne	нс
Propargyl ether	нс

Table 1

Monomer Compound

The monomer compound has the following formula:

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where R_1 , R_2 and R_4 are each independently selected from hydrogen, optionally substituted branched or straight chain C_1 – C_6 alkyl or halo alkyl or aryl optionally substituted by halo, and R_3 is selected from:

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where each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C_1 - C_6 alkyl, halo alkyl or anyl optionally substituted by halo; and n_1 is an integer from 1 to 27. Preferably, n_1 is from 1 to 12. Optionally n_1 is from 4 to 12, optionally 6 to 8.

In a preferred embodiment R₃ is selected from:

where m_1 is an integer from 0 to 13 and each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C_1 - C_6 alkyl, halo alkyl or anyl optionally substituted by halo; and m_2 is an integer from 2 to 14;

In a particularly preferred embodiment the monomer is a compound of formula I(a):

Ka)

$$\begin{array}{c|c} R_{1} & O & R_{5} & R_{7} & \begin{bmatrix} R_{0} & X \\ X & X \end{bmatrix} & \begin{bmatrix} X & X \\ X & X \end{bmatrix} & CX_{0} & CX_$$

wherein each of R₁, R₂, R₄, and R₅ to R₁₀ is independently selected from hydrogen or an optionally substituted C₁-C₅ branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1;

Or

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the monomer is a compound of formula I(b):

I(b)

$$R_{2} \longrightarrow \begin{bmatrix} R_{1} & C & R_{5} & R_{7} & R_{5} & R_{7} & R_{5} & R_{5} & R_{7} & R_{5} & R_{5}$$

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wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_6 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1.

The halogen may be chlorine or bromine, but is preferably fluorine for compliance with RoHS regulations (Restriction of Hazardous Substances).

a is from 0 to 10, preferably from 0 to 6, optionally 2 to 4, most preferably 0 or 1. b is from 2 to 14, optionally from 2 to 10, preferably 3 to 7.

Each of R₁, R₂, R₄, and R₅ to R₁₀ is Independently selected from hydrogen or a C₁-C₈ branched or straight chain alkyl group. The alkyl group may be substituted or unsubstituted, saturated or unsaturated. When the alkyl group is substituted, the location or type of the substituent is not especially limited provided the resultant polymer provides an appropriate liquid repellent and/or barrier layer. The skilled person would be aware of suitable substituents. If the alkyl group is substituted, a preferred substituent is halo, i.e. any of R₁, R₂, R₄, and R₅ to R₁₀ may be haloalkyl, preferably fluoro alkyl. Other possible substituents may be hydroxyl or amine groups. If the alkyl group is unsaturated it may comprise one or more alkene or alkyne groups.

Each of R₁, R₂, R₄, and R₅ to R₁₀ may be independently selected from hydrogen, methyl, ethyl, npropyl, isopropyl, n-butyl, isobutyl, tert-butyl, sec-butyl, n-pentyl, neopentyl, n-hexyl, isohexyl, and 3methylpentyl.

In a preferred embodiment each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or methyl.

20 In a preferred embodiment, a and c are each independently 0 or 1; and b is from 3 to 7.

In one preferred embodiment each X is H. In an alternative preferred embodiment each X is F.

Optionally R₁ and R₂ are both hydrogen.

Optionally R_4 is hydrogen or methyl. Preferably R_1 and R_2 are both hydrogen and R_4 is hydrogen or methyl.

Optionally R₉ is hydrogen and R₁₀ is branched or straight chain C₁-C₆ alkyl group. In a preferred embodiment R₁₀ is methyl.

In one embodiment each of R₅ to R₈ is hydrogen.

30 In one embodiment each of R₁, R₂, R₄, and R₅ to R₁₀ is hydrogen, each X is H, a=0 and c=0.

In a particularly preferred embodiment the monomer compound has the following formula:

where n is from 2 to 10.

in another preferred embodiment the monomer compound has the following formula:

$$H = C_{CH_3}^{O} (CF_2)_n F$$

5 where n is from 2 to 10.

The monomer compound may be selected from 1H,1H,2H,2H-perfluorohexyl acrylate (PFAC4), 1H,1H,2H,2H-perfluorooctyl acrylate (PFAC6), 1H,1H,2H,2H-perfluorodoecyl acrylate (PFAC8) and 1H,1H,2H,2H-perfluorodoecyl acrylate (PFAC10).

The monomer compound may be selected from 1H,1H,2H,2H-perfluorohexyl methacrylate (PFMAC4),

1H,1H,2H,2H-perfluorooctyl methacrylate (PFMAC8) and 1H,1H,2H,2H-perfluorodecyl methacrylate (PFMAC8).

The monomer compound of formula I(a) may have the following formula:

wherein a and c are each independently 0 or 1, b =3-7 and n is 4 to 10, where n=a+b+c+1.

15 The monomer compound of formula I(a) may have the following formula:

where n is 2 to 12.

The monomer compound may be selected from ethyl hexyl acrylate, hexyl acrylate, decyl acrylate, lauryl dodecyl acrylate and iso decyl acrylate.

5 The monomer may have the following formula:

where n is from 4 to 14.

The monomer may have the following formula:

where n is from 4 to 14.

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In an additional aspect, the present invention provides a method for treating an electronic device or component thereof comprising:

exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow a protective cross-linked polymeric coating to form on a surface of the electronic or electrical device or component thereof;

wherein the monomer compound has the following formula:

where R_1 , R_2 and R_4 are each independently selected from hydrogen, optionally substituted branched or straight chain C_1 – C_0 alkyl or halo alkyl or aryl optionally substituted by halo, and R_3 is selected from:

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where each X is independently selected from hydrogen, optionally substituted branched or straight chain C₁-C₆ alkyl, halo alkyl or aryl optionally substituted by halo; and n₁ is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds and has a boiling point of less than 500 °C at standard pressure.

The monomer compound and the crosslinking reagent utilised in the method are described in further detail above.

Typically the electronic or electrical device or component thereof is placed in a plasma deposition chamber, a glow discharge is ignited within said chamber, and a voltage applied as a pulsed field.

Preferably the applied voltage is at a power of from 30 to 800W. Optionally the voltage is pulsed in a sequence in which the ratio of the time on/time off is in the range of from 0.001 to 1, optionally 0.002 to 0.5. For example, time on may be 10-500 μ s, preferably 35-45 μ s, or 30-40 μ s, such as about 36 μ s and time off may be from 0.1 to 30 ms, preferably 0.1 to 20 ms, optionally from 5 to 15 ms, for example 6ms. Time on may be 35 μ s, 40 μ s, 45 μ s. Time off may be 0.1, 1, 2, 3, 6, 8, 10, 15, 20, 25 or 30 ms.

The term pulsed may mean that the plasma cycles between a state of no (or substantially no) plasma emission (off-state) and a state where a particular amount of plasma is emitted (on-state).

Alternatively, pulsed may mean that there is continuous emission of plasma but that the amount of plasma cycles between an upper limit (on-state) and lower limit (off-state).

In a further embodiment, the invention resides in a method of forming a coating on an electronic or electrical device or component thereof as defined above, the method comprising: exposing said substrate in a chamber to a plasma comprising a monomer compound, preferably a continuous plasma, for a sufficient period of time to allow a protective polymeric coating to form on the substrate,

wherein during exposure of the substrate the continuous plasma has a power density of at least 2W/litre, preferably 20W/litre.

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Optionally the voltage is applied as a pulsed field for a period of from 30 seconds to 90 minutes. Optionally the voltage is applied as a pulsed field for from 5 to 60 minutes. Optionally in a preliminary step, a continuous power plasma is applied to the electronic or electrical device or component thereof. The preliminary step may be conducted in the presence of an inert gas.

Before the crosslinker and/or the monomer compound enter the deposition chamber each may be in the form of a gas, liquid or a solid (for example a powder) at room temperature. However, it is the preferred that the crosslinking reagent and the monomer compound are both liquid at room temperature, and most preferably that the monomer and crosslinker liquids are miscible.

The monomer compound and/or the crosslinker will suitably be in a gaseous state in the plasma. The plasma may simply comprise a vapour of the monomer compound and the crosslinker. Such a vapour may be formed in-situ, with the compounds being introduced into the chamber in liquid form. The monomer may also be combined with a carrier gas, in particular, an inert gas such as helium or argon.

In preferred embodiments, the monomer and/or the crosslinker may be delivered into the chamber by way of an aerosol device such as a nebuliser or the like, as described for example in WO2003/097245 and WO03/101621, the content of which is incorporated herein by reference. In such an arrangement a carrier gas may not be required, which advantageously assists in achieving high flow rates.

In one embodiment the monomer compound and/or the crosslinking reagent are in gaseous form and are fed into the plasma at a rate of from 10-5000 mg/minute depending on chamber volume, whilst the pulsed voltage is applied.

25 Optionally the plasma is created with an average power of from 0.001 to 40W/litre.

The crosslinker may be miscible with the monomer and so introduced together or separately into the plasma chamber. Or the crosslinker may be immiscible with the monomer and introduced separately into the plasma chamber. In this context, the term "miscible" means that the crosslinker is soluble in the monomer, and when mixed they form a solution of uniform composition. The term "immiscible" is used to mean that the crosslinker is only partly soluble or insoluble in the monomer, and so either forms an emulsion or separates out into two layers.

The crosslinking reagent is preferably present in an amount of from 10 to 60 (v/v) %, optionally 20 to 40 (v/v) %, optionally 25 to 30 (v/v)%, optionally 30 to 50 (v/v)% of the total volume of monomer compound and crosslinking reagent, depending on the specific crosslinker. The skilled person would understand that the amount will vary to some extent depending on whether the coating is required to

be liquid repellent or provide a barrier to mass and electron transport. The skilled person would understand that the (v/v) percentages are those which give a stable crosslinked polymer coating and the highest water contact angle.

Electronic or electrical device or component thereof

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Although the invention is of benefit in the context of a wide variety of substrates, the substrate may, in all aspects of the invention, advantageously be an electronic substrate.

In some embodiments of the invention, the electronic substrate may comprise an electronic or electrical device, i.e. any piece of electrical or electronic equipment. Non-limiting examples of electrical and electronic devices include communications devices such as mobile phones, smartphones and pagers, radios, and sound and audio systems such as loudspeakers, microphones, ringers or buzzers, hearing aids, personal audio equipment such as personal CD, tape cassette or MP3 players, televisions, DVD players including portable DVD players, video recorders, digi and other set-top boxes such as Sky, computers and related components such as laptop, notebook, tablet, phablet or paimtop computers, personal digital assistants (PDAs), keyboards, or instrumentation, games consoles in particular hand-held playstations and the like, data storage devices, outdoor lighting systems or radio antenna and other forms of communication equipment.

In preferred embodiments of the invention, the substrate may comprise or consist of an electronic component, e.g. a printed circuit board (PCB), a printed circuit board array (PCBA), a transistor, resistor, or semi-conductor chip. The electronic component may thus be an internal component of an electronic device, e.g. a mobile phone. The coatings of the invention are particularly valuable in preventing electrochemical migration in such components.

In all aspects of the invention, the precise conditions under which the protective polymenic coating is formed in an effective manner will vary depending upon factors such as, without limitation, the nature of the monomer compound, the crosslinking reagent, the substrate, as well as the desired properties of the coating. These conditions can be determined using routine methods or, preferably, using the techniques and preferred features of the invention described herein, which work in particular synergy with the invention.

The present invention will now be further described with reference to the following non-limiting examples and the accompanying illustrative drawings, of which:

Figure 1 illustrates shows the effect of increasing the power to monomer flow ratio for CW plasmas in a 3 litre chamber in a prior art process.

Figure 2 shows the same effect for pulsed plasma conditions.

Figure 3 shows a tapping mode image over $10x10~\mu\text{m}^2$ field of view of a specimen example (thickness d=1230 nm) prepared according to Example 1 (left hand side) and a contour line plot showing the data used for calculation of RMS roughness (right hand side). The ΔZ value indicated on the plot has been taken over an area of the graph that represents the majority of the coating. Peaks that lie above the ΔZ range indicate large particles and troughs that fall below the ΔZ range show voids or pinholes in the coating. The width of the peaks also gives an indication of the particle size.

Figure 4 illustrates the effect of adding the divinyl adipate crosslinker to perfluorooctyl acrylate on water contact angle from a process on a 125 Litre chamber.

Figure 5 shows the effect of adding crosslinkers in accordance with the invention on the electrical resistance per nm of the coatings.

Figure 6 shows a tapping mode image of a 90 nm thick coating, prepared as described in Example 3, over $2x2\mu m^2$ field of view (top left), a representative contour line indicating height variation (z-axis) of the coating (top right) and a phase image indicating full substrate coverage (bottom left) ;RMS roughness of the coating is 1.65 nm and $\Delta z/d = 0.05$

15 Example 1

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Example of plasma polymerising PFAC6 with no crosslinker in a 3 litre chamber

PFAC6 was polymerised in a 3 litre glass chamber using a continuous plasma process. In the experimental matrix, the continuous plasma was produced using RF power levels of 50, 100 and 200 Watts. The process pressure was run at 30, 60 and 90 mTorr and the monomer flow rate run at 50, 100 and 200 microlitres of liquid per minute. The duration of the process was either 10 or 40 minutes. The coated substrate was silicon wafer and whether the coating was tacky or not was determined by wiping the substrate with a finger and determining by eye if the coating had smeared. The results in Figure 1 indicate that for lower CW powers, the coating could easily be smeared (i.e. tacky), but by increasing the power to flow ratio, the coating became resistant to smearing (i.e. non-tacky) and this is expected to be due to additional crosslinking caused by more monomer fragmentation. The water contact angle was also seen to decline with increasing power to flow ratio, and this too is indicative of more power increasing the level of monomer fragmentation.

The same 3 litre chamber was used to also look at the effect of power to flow ratios with pulsed plasma polymerisation. In these experiments, the peak power was either 250 or 500 W. The RF power feed was pulsed with an on time of 35 microseconds and off times of 5.5 and 1 millisecond. Process pressure was either 20, 35, 60 or 80 mTorr. The monomer flow was either 50 or 200 microlitres of liquid per minute. Figure 2 shows that low power to flow conditions can generate coatings that can be smeared (tacky) and only by increasing the power could the smear be reduced. This too had a concomitant decilne in water contact angle.

The results from Figures 1 and 2 show that plasma processing PFAC6 on its own could only lead to smear –free coatings when the powers were high enough to decrease the other desirable property of the water contact angle. This illustrates the need to add a crosslinker to prevent the coating smearing and maintain the water contact angle.

5 Example 2

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Examples of Perfluoro Alkyl Acrylate co-polymerisation with Crosslinker to show improvements in electrical resistance

Perfluorocctyl acrylate monomer was mixed with a single crosslinker from the following list:

VINYL ESTERS: Divinyl adipate (DVA)

10 VINYL ETHERS: 1,4-Butanediol divinyl ether (BDVE); 1,4 Cyclohexanedimethanol divinyl ether (CDDE)

DI- or TRI- VINYLS: 1,7-Octadiene (1,7-OD); 1,2,4-Trivinylcyclohexane (TVCH);

DI VINYL with alkyl fluoro group: 1,6-divinylperfluorohexane (DVPFH)

DI VINYL with silicon group: 1,3-Divinyl tetramethyldisiloxane (DVTMDS)

15 DI VINYL with cyclic ring and carboxylate groups: Diallyl 1,4-cyclohexanedicarboxylate (DCHC)

DI ACRYLATE: 1H1H,6H,6H-perfluorohexanediol diacrylate (PFHDA)

The crosslinker was mixed with the PFAC6 as a percentage by volume for the following percentages: 20, 40, 60, 80, and 100%.

Plasma initiated polymerisation reactions were carried out in a 3 litre glass plasma chamber. The substrates were test circuit boards. The PFAC6/crosslinker mixture was introduced at a rate of 0.04 — 0.06 ul/min and the process pressure was 40 mTorr. The process plasma consisted of a 1 minute continuous plasma (CW) step of 50W followed by 10 - 20 minute pulsed plasma (PW) step of 50W and RF power being delivered by a pulse sequence with a duty cycle of 6.9%. The coated circuit boards were immersed in tap water and an 8V potential was applied for 13 minutes. The final current readings, together with coating thickness measurements, were used to calculate the electrical resistance per nm of coating. Figure 5 below shows the effect of adding a range of crosslinkers on the electrical resistance per nm of the barrier coatings. Figure 5 clearly shows the improvement in resistance that can be gained by adding the crosslinker to the PFAC6 monomer. Furthermore, all of the crosslinkers tested gave non-smear/non-tacky coatings with concentration levels of 20% and above, with the exceptions of DVTMDS, DCHD, PFHDA and DVPFH which were non-smear at levels of 40% and above.

Example 3

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Example of crosslinker concentration on water repellent coating.

Water repellent coatings were prepared in a 125 litre volume chamber using PFAC6 with different levels of divinyl adipate (DVA) and using helium as a carrier gas. The deposition process consisted of a 3 minute CW step with 300W power and a pulsed step with 150W power and an RF pulse duty cycle of 0.018%. Silicon wafer was used as the test substrates and the contact angle of the coated wafer was determined by applying a 3ul drop of deionised wafer onto the coated wafer and measuring the contact angle using a VCA Optima (AST products) with image analysis software. The variation of contact angle with (v/v)% DVA crosslinker is shown in Figure 4. It was also noted that for the 10 (v/v)% DVA concentration, the water drop left a mark on the coated wafer where it's circumference had been. This indicated that a higher level of crosslinker was required to give a more stable coating. This observation, in combination with the results in figure 4, suggest an optimum range of 20 – 40 (v/v)% DVA crosslinker, though this is likely to vary, typically within the range of from 10-60 (v/v)% for different monomers and crosslinkers and different chamber sizes.

Pulsed plasma polymerisation of perfluoro alkyl acrylate in the presence of divinyl adipate (DVA) was carried out in a 125 litre chamber. Figure 4 shows the effect of adding different liquid volume percentages of DVA crosslinker to perfluorooctyl acrylate on water contact angle on a silicon wafer substrate.

Example 4

20 Example of AFM measurement of monomer/crosslinker co-polymer coating

A barrier-style coating was prepared in a 22 litre volume chamber using PFAC6 with 10% DVA. The deposition process consisted of a 1 minute CW step with a CW power to flow ratio of 3.9 (W/microliter/min), and a pulsed step with a PW power/monomer flow ratio of 0.28 (W/microlitres/min). Figure 6 shows the representative topographical and phase contrast images were obtained from all the samples. High spatial resolution images show mainly the structures of areas between the raised features. The RMS roughness of the coating is 1.65 nm and the $\Delta z/d = 0.05$. As the $\Delta z/d$ value is < 0.15, then this indicates that the physical layer is substantially pinhole free.

Claims:

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1. An electronic or electrical device or component thereof comprising a protective cross-linked polymeric coating on a surface of the electronic or electrical device or component thereof; wherein the protective cross-linked polymeric coating is obtainable by exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow formation of the protective cross-linked polymeric coating on a surface thereof,

wherein the monomer compound has the following formula:

where R₁, R₂ and R₄ are each independently selected from hydrogen, optionally substituted branched or straight chain C₁-C₈ alkyl or halo alkyl or aryl optionally substituted by halo, and R₃ is selected from:

where each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C₁-C₈ alkyl, halo alkyl or aryl optionally substituted by halo; and n₁ is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moleties and has a boiling point of less than 500 °C at standard pressure.

- An electronic or electrical device or component thereof according to claim 1, wherein the
 protective cross-linked polymeric coating is a physical barrier to mass and electron transport.
- 20 3. An electronic or electrical device or component thereof according to claim 1 or claim 2, wherein the protective cross-linked polymeric coating forms a liquid repellent surface defined by a static water contact angle (WCA) of at least 90°.
 - 4. An electronic or electrical device or component thereof according to any preceding claim, wherein the crosslinking reagent has one of the following structures:

where Y_1 , Y_2 , Y_3 , Y_4 , Y_5 , Y_6 , Y_7 and Y_8 are each independently selected from hydrogen, optionally substituted cyclic, branched or straight chain C_1 - C_6 alkyl or anyl; and L is a linker moiety.

5 5. An electronic or electrical device or component thereof according to claim 4, wherein for compound (i) L has the following formula:

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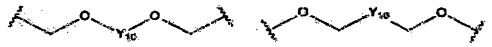
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each Y_9 is independently selected from, a bond, -O-, -O-C(O)-, -C(O)-O-, -Y₁₁-O-C(O)-, -C(O)-O-Y₁₁-, -OY₁₁-, and -Y₁₁O-, where Y₁₁ is an optionally substituted cyclic, branched or straight chain C₁-C₈ alkylene; and

 Y_{10} is selected from optionally substituted cyclic, branched or straight chain C_1 - C_8 alkylene and a slloxane group.

6. An electronic or electrical device or component thereof according to claim 5, wherein for compound (i) L has one of the following structures:

7. An electronic or electrical device or component thereof according to claim 5, wherein for compound (i) L has one of the following structures:

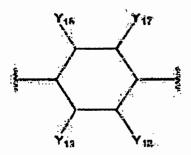


8. An electronic or electrical device or component thereof according to any of claims 5 to 7, wherein Y_{10} has the following formula:

wherein each Y_{12} and Y_{13} is independently selected from H, halo, optionally substituted cyclic, branched or straight chain alkyl, or $-OY_{14}$, where Y_{14} is selected from optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl, and n is an integer from 1 to 10.

- An electronic or electrical device or component thereof according to claim 8, wherein each Y₁₂ is H and each Y₁₃ is H.
- 10. An electronic or electrical device or component thereof according to claim 8, wherein each Y_{12} is fluoro and each Y_{13} is fluoro.
- 10 11. An electronic or electrical device or component thereof according to any of claims 8 to 10, wherein n is from 4 to 6.
 - 12. An electronic or electrical device or component thereof according to any of claims 5 to 7, wherein Y₁₀ has the following formula:

- Wherein each Y₁₅ is independently selected from optionally substituted branched or straight chain C₁-C₆ alkyl.
 - An electronic or electrical device or component thereof according to claim 12, wherein each Y₁₅ is methyl, and each Y₉ is a bond.
 - 14. An electronic or electrical device or component thereof according to any of claims 5 to 7, wherein Y₁₀ has the following formula:



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wherein Y_{18} to Y_{19} are each independently selected from H and optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl.

- 15. An electronic or electrical device or component thereof according to claim 14, wherein Y₁₈ is H or vinylene, and Y₁₈, Y₁₇ and Y₁₉ are each H.
- 16. An electronic or electrical device or component thereof according to any preceding claim wherein the crosslinking reagent is selected from Divinyl Adipate (DVA), 1,4 Butanediol divinyl ether (BDVE), 1,4 Cyclohexanedimethanol divinyl ether (CDDE), 1,7 -Octadiene (17OD), 1,2,4 Trivinylcyclohexane (TVCH), 1,3-Divinyltetramethyldisiloxane (DVTMDS), Diallyl 1,4-cyclohexanedicarboxylate (DCHD), 1,6-Divinylperfluorohexane (DVPFH), 1H,1H,6H,6H-Perfluorohexanediol diacrylate (PFHDA) and glyoxal bis (diallyl acetal) (GBDA).
- 10 17. An electronic or electrical device or component thereof according to claim 4, wherein for compound (ii), L is selected from a branched or straight chain C₁-C₈ alkylene or an ether group.
 - 18. An electronic or electrical device or component thereof according to any preceding claim wherein the monomer is a compound of formula I(a):

i(a)

$$\begin{array}{c|c} R_1 & C & R_2 & R_3 & R_4 & R_{10} & R_{20} & R_$$

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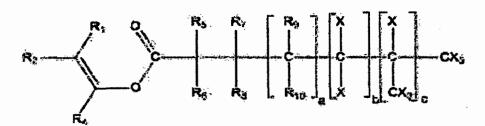
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wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_6 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1;

; or

20 wherein the monomer is a compound of formula I(b):

I(b)



wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_8 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1.

- An electronic or electrical device or component thereof according to claim 18, wherein the halogen is fluorine.
- 20. An electronic or electrical device or component thereof according to claim 18 or claim 19, wherein each of R₁, R₂, R₄, and R₅ to R₁₀ is independently selected from hydrogen, methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, tert-butyl, sec-butyl, n-pentyl, neopentyl, n-hexyl, isobexyl, and 3-methylpentyl.
- 21. An electronic or electrical device or component thereof according to claim 20, wherein each of R₁, R₂, R₄, and R₅ to R₁₀ is independently selected from hydrogen or methyl.
 - 22. An electronic or electrical device or component thereof according to any of claims 18 to 20, wherein a and c are each independently 0 or 1; and b is from 3 to 7.
 - 23. An electronic or electrical device or component thereof according to any preceding claim wherein each X is H.
 - 24. An electronic or electrical device or component thereof according to any of claims 1 to 22, wherein each X is F.
 - 25. An electronic or electrical device or component thereof according to any of claims 18 to 24, wherein R₁ and R₂ are both hydrogen.
- 20 26. An electronic or electrical device or component thereof according to any of claims 18 to 25, wherein R₄ is hydrogen or methyl.
 - 27. An electronic or electrical device or component thereof according to any of claims 18 to 26, wherein R₆ is hydrogen and R₁₀ is branched or straight chain C₁-C₅ alkyl group.
 - 28. An electronic or electrical device or component thereof according to claim 27, wherein R₁₀ is methyl.
 - 29. An electronic or electrical device or component thereof according to any of claims 18 to 28, wherein each of R₅ to R₅ is hydrogen.
 - 30. An electronic or electrical device or component thereof according to any of claims 18 to 29 wherein each of each of R₁, R₂, R₄, and R₅ to R₁₀ is hydrogen, each X is H, a=0 and c=0.
- 30 31. An electronic or electrical device or component thereof according to any of claims 24 to 30, wherein the compound of formula I(a) has the following formula:

where n is from 2 to 10.

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32. An electronic or electrical device or component thereof according to any of claims 24 to 30, wherein the compound of formula I(a) has the following formula:

where n is from 2 to 10.

- 5 33. An electronic or electrical device or component thereof according to claim 31, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-perfluorohexyl acrylate (PFAC4), 1H,1H,2H,2H-perfluorooctyl acrylate (PFAC6), 1H,1H,2H,2H-perfluorodecyl acrylate (PFAC8) and 1H,1H,2H,2H-perfluorododecyl acrylate (PFAC10).
- 34. An electronic or electrical device or component thereof according to claim 32, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-perfluorohexyl methacrylate (PFMAC4), 1H,1H,2H,2H-perfluorooctyl methacrylate (PFMAC6) and 1H,1H,2H,2H-perfluorodecyl methacrylate (PFMAC8).
- 15 35. An electronic or electrical device or component thereof according to any of claims 18 to 23, wherein the compound of formula I(a) has the following formula:

wherein a and c are each independently 0 or 1, b =3-7 and n is 4 to 10, where n=a+b+c+1.

36. An electronic or electrical device or component thereof according to any of claims 18 to 23, wherein the compound of formula I(a) has the following formula:

$$H = CH_3$$
 $(CH_2)_nH$

where n is 2 to 12.

- 37. An electronic or electrical device or component thereof according to claim 35 or claim 36, wherein the compound of formula I(a) is selected from ethyl hexyl acrylate, hexyl acrylate, decyl acrylate, lauryl dodecyl acrylate and iso decyl acrylate.
- 5 38. An electronic or electrical device or component thereof according to any of claims 18 to 23, wherein the monomer is a compound of formula I(b) and has the following formula:

where n is from 4 to 14, optionally where R₁, R₂ and R₃ are each H.

- 39. An electronic or electrical device or component thereof according to any of the preceding claims, wherein the electronic or electrical device or component thereof is selected from mobile phones, smartphones, pagers, radios, sound and audio systems such as loudspeakers, microphones, ringers and/or buzzers, hearing aids, personal audio equipment such as personal CD, tape cassette or MP3 players, televisions, DVD players including portable DVD players, video recorders, digi and other set-top boxes, computers and related components such as laptop, notebook, tablet, phablet or palmtop computers, personal digital assistants (PDAs), keyboards, or instrumentation, games consoles, data storage devices, outdoor lighting systems, radio antennae and other forms of communication equipment, and printed circuit boards.
 - 40. A method for treating an electronic or electrical device or component thereof comprising:
- 20 exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow a protective cross-linked polymeric coating to form on a surface of the electronic or electrical device or component thereof;

wherein the monomer compound has the following formula:

where R_1 , R_2 and R_4 are each independently selected from hydrogen, optionally substituted branched or straight chain C_1 - C_6 alkyl or halo alkyl or aryl optionally substituted by halo, and R_3 is selected from:

where each X is independently selected from hydrogen, optionally substituted branched or straight chain C₁-C₆ alkyl, halo alkyl or aryl optionally substituted by halo; and n₁ is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moieties and has a boiling point at standard pressure of less than 500°C.

41. A method according to claim 40, wherein the crosslinking reagent has one of the following structures:

where Y_1 , Y_2 , Y_3 , Y_4 , Y_5 , Y_6 , Y_7 and Y_8 are each independently selected from hydrogen, optionally substituted cyclic, branched or straight chain C_1 - C_8 alkyl or aryl; and L is a linker moiety.

42. A method according to claim 41, wherein for compound (i) L has the following formula:

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each Y_9 is independently selected from, a bond, -O-, -O-C(O)-, --C(O)-O-, -Y₁₁-O-C(O)-, -C(O)-O-Y₁₁-, and -Y₁₁O-, where Y₁₁ is an optionally substituted cyclic, branched or straight chain C₁-C₈ alkylene; and

Y₁₀ is selected from optionally substituted cyclic, branched or straight chain C₁-C₈ alkylene and a 20 siloxane group.

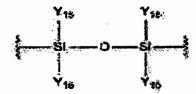
43. A method according to claim 42, wherein for compound (i) has one of the following structures:

44. A method according to claim 42, wherein for compound (i) L has one of the following structures:

5 45. An method according to any of claims 42 to 44, wherein Y₁₀ has the following formula:

wherein each Y₁₂ and Y₁₃ is independently selected from H, halo, optionally substituted cyclic, branched or straight chain alkyl, or –OY₁₄, where Y₁₄ is selected from optionally substituted branched or straight chain C₁-C₈ alkyl or alkenyl, and n is an integer from 1 to 10.

- 46. A method according to claim 45, wherein each Y12 is H and each Y13 is H.
- 47. A method according to claim 45, wherein each Y12 is fluoro and each Y13 is fluoro.
- 48. A method according to any of claims 45 to 47, wherein n is from 4 to 6.
- 15 49. A method according to any of claims 42 to 44, wherein Y₁₀ has the following formula:



Wherein each Y_{15} is independently selected from optionally substituted branched or straight chain C_1 - C_6 alkyl.

- A method according to claim 49, wherein each Y₁₅ is methyl, and each Y₉ is a bond.
- 20 51. A method according to any of claims 42 to 44, wherein Y₁₀ has the following formula:

wherein Y_{18} to Y_{19} are each independently selected from H and optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl.

52. An method according to claim 51, wherein Y_{18} is vinylene or H, and Y_{18} , Y_{17} and Y_{19} are each H.

53. A method according to any of claims 40 to 52, wherein the crosslinking reagent is selected from Divinyl Adipate (DVA), 1,4 Butanediol divinyl ether (BDVE), 1,4 Cyclohexanedimethanol divinyl ether (CDDE), 1,7 -Octadiene (17OD), 1,2,4 -Trivinylcyclohexane (TVCH), 1,3-Divinyltetramethyldisiloxane (DVTMDS), Diallyl 1,4-cyclohexanedicarboxylate (DCHD), 1,6-Divinylperfluorohexane (DVPFH), 1H,1H,6H,6H-Perfluorohexanediol diacrylate (PFHDA) and glyoxal bis (diallyl acetal) (GBDA).

54. A method according to claim 41, wherein for compound (ii), it is selected from a branched or straight chain C₁-C₈ alkylene or an ether group.

55. A method according to any of claims 40 to 54 wherein the monomer is a compound of formula I(a):

i(a)

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$$\begin{array}{c|c}
R_1 & O & R_5 & R_7 & R_9 \\
\hline
O & R_5 & R_9 & R_{50} & R_7 \\
\hline
\end{array}$$

wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_6 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1;

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wherein the monomer is a compound of formula I(b):

l(b)

$$R_{2} = \begin{bmatrix} R_{1} & O & R_{3} & R_{7} & R_{4} & X \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ &$$

wherein each of R_1 , R_2 , R_4 , and R_6 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_0 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1.

- 5 56. A method according to claim 55, wherein the halogen is fluorine.
 - 57. A method according to claim 55 or claim 56, wherein each of R₁, R₂, R₄, and R₅ to R₁₀ is independently selected from hydrogen, methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, tert-butyl, sec-butyl, n-pentyl, neopentyl, n-hexyl, isobexyl, and 3-methylpentyl.
 - 58. A method according to claim 57, wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or methyl.
 - 59. A method according to any of claims 55 to 58, wherein a and c are each independently 0 or 1; and b is from 3 to 7.
 - 60. A method according to any preceding claim wherein each X is H.
 - 61. A method according to any of claims 41 to 59, wherein each X is F.
- 15 62. A method according to any of claims 41 to 61, wherein R₁ and R₂ are both hydrogen.
 - A method according to any of claims 41 to 62, wherein R₄ is hydrogen or methyl.
 - 64. A method according to any of claims 41 to 63, wherein R₉ is hydrogen and R₁₀ is branched or straight chain C₁-C₆ alkyl group.
 - 65. A method according to claim 64, wherein R₁₀ is methyl.
- 20 66. A method to any of claims 41 to 65, wherein each of R₅ to R₅ is hydrogen.
 - 67. A method according to any of claims 41 to 66 wherein each of each of R₁, R₂, R₄, and R₅ to R₁₀ is hydrogen, each X is H, a=0 and c=0.
 - 68. A method according to any of claims 61 to 67, wherein the compound of formula I(a) has the following formula:

where n is from 2 to 10.

69. A method according to any of claims 61 to 67, wherein the compound of formula !(a) has the following formula:

$$H \longrightarrow O (CF_2)_nF$$

$$CH_3$$

where n is from 2 to 10.

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- 70. A method according to claim 68, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-perfluorohexyl acrylate (PFAC4), 1H,1H,2H,2H-perfluorodecyl acrylate (PFAC8) and 1H,1H,2H,2H-perfluorodecyl acrylate (PFAC10).
- 71. A method according to claim 69, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-perfluorohexyl methacrylate (PFMAC4), 1H,1H,2H,2H-perfluorooctyl methacrylate (PFMAC6) and 1H,1H,2H,2H-perfluorodecyl methacrylate (PFMAC8).
 - 72. A method according to any of claims 41 to 61, wherein the compound of formula I(a) has the following formula:

- 15 wherein a and c are each independently 0 or 1, b =3-7 and n is 4 to 10, where n=a+b+c+1.
 - 73. A method according to any of claims 41 to 61, wherein the compound of formula i(a) has the following formula:

where n is 2 to 12.

- 74. A method according to claim 72 or claim 73, wherein the compound of formula I(a) is selected from ethyl hexyl acrylate, hexyl acrylate, decyl acrylate, lauryl dodecyl acrylate and iso decyl acrylate.
- 75. A method according to any of claims 41 to 60, wherein the monomer is a compound of formula I(b) and has the following formula:

where n is from 4 to 14.

76. A method according to claim 75 wherein the monomer is a compound of formula I(b) and has the following formula:

where n is from 4 to 14.

- **77**. A method according to any of claims 40 to 76 wherein the electronic or electrical device or component thereof is placed in a plasma deposition chamber, a glow discharge is ignited within said chamber, and a voltage applied as a pulsed field.
- A method according to claim 77 wherein applied voltage is at a power of from 40 to 500W. 15 78.
 - A method according to claim 77 or claim 78 wherein the voltage is pulsed in a sequence in 79. which the ratio of the time on: time off is in the range of from 1:500 to 1:1500.
 - 80. A method according to any of claims 77 to 79 wherein the voltage is pulsed in a sequence where power is on for 20-50µs, and off for from 1000µs to 30000µs.
- A method according to any of claims 77 to 80 wherein the voltage is applied as a pulsed field 20 81. for a period of from 30 seconds to 90 minutes.
 - A method according to claim 81 wherein the voltage is applied as a pulsed field for from 5 to 60 82. minutes.
- 83. A method according to any one of claims 77 to 82, wherein in a preliminary step, a continuous power plasma is applied to the electronic or electrical device or component thereof.

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- 84. A method according to claim 83, wherein the preliminary step is conducted in the presence of an inert gas.
- 85. A method according to any one of claims 77 to 84, wherein the monomer compound and/or the crosslinking reagent in gaseous form is fed into the plasma at a rate of from 80-300 mg/minute, whilst the pulsed voltage is applied.
- 86. A method according to any one of claims 77 to 85, wherein the plasma is created with a voltage at an average power of from 0.001 to 500w/m³.
- 87. A method according to any of claims 40 to 86, wherein the monomer compound is miscible with the crosslinking reagent.
- 10 88. A method according to any of claims 40 to 86, wherein the monomer compound is immiscible with the crosslinking reagent.
 - 89. A method according to claim 87, wherein the step of exposing the electronic or electrical device or component thereof to a plasma occurs in a plasma chamber; and the monomer compound and crosslinking reagent are mixed prior to entering the chamber.
- 15 90. A method according to claim 87 or claim 88 wherein the step of exposing the electronic or electrical device or component thereof to a plasma occurs in a plasma chamber; and the monomer compound and crosslinking reagent are introduced separately into the chamber.
 - 91. A method according to any of claims 40 to 90, wherein the crosslinking reagent is present in an amount of from 10 to 60 (v/v) % of the total volume of monomer compound and crosslinking reagent.

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Coatings

Abstract

The present invention provides an electronic or electrical device or component thereof comprising a cross-linked polymeric coating on a surface of the electronic or electrical device or component thereof; wherein the cross-linked polymeric coating is obtainable by exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a cross-linking reagent for a period of time sufficient to allow formation of the cross-linked polymeric coating on a surface thereof,

wherein the monomer compound has the following formula:

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where R₁, R₂ and R₄ are each independently selected from hydrogen, optionally substituted branched or straight chain C₁-C₅ alkyl or halo alkyl or aryl optionally substituted by halo, and R₃ is selected from:

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where each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C₁-C₆ alkyl, halo alkyl or aryl optionally substituted by halo; and n₁ is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moieties and has a boiling point at standard pressure of less than 500°C.

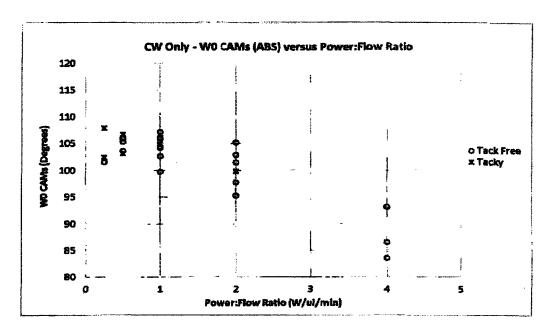


Figure 1

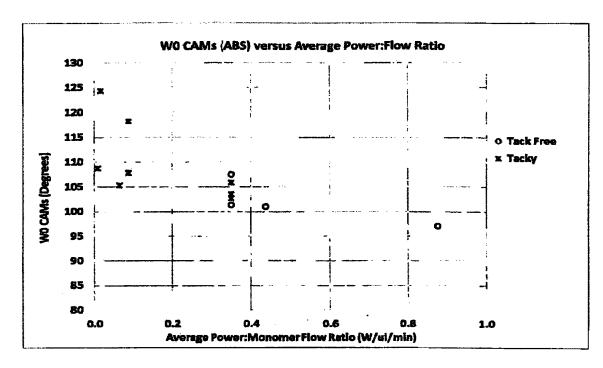


Figure 2

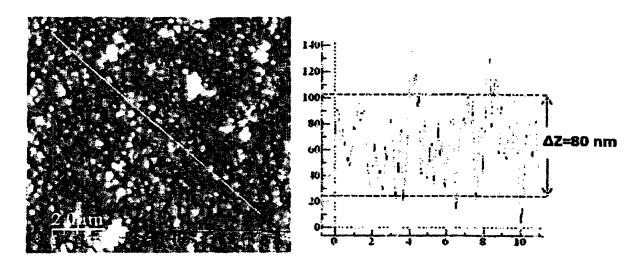


Figure 3

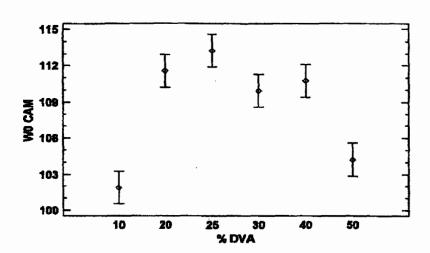


Figure 4

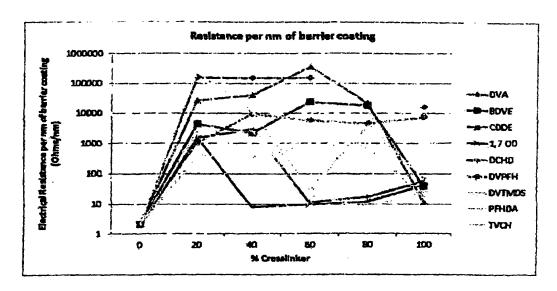
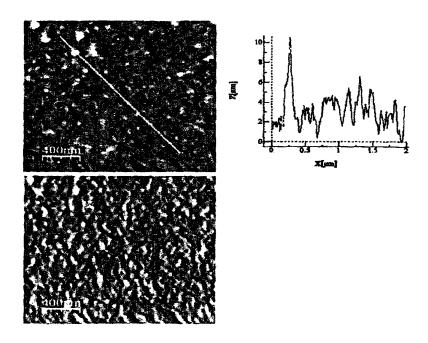


Figure 5

Figure 6





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CONFIRMATION NO. 7101
FILING RECEIPT

33072 KAGAN BINDER, PLLC SUITE 200, MAPLE ISLAND BUILDING 221 MAIN STREET NORTH STILLWATER. MN 55082

Date Mailed: 03/16/2018

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Inventor(s)

Stephen Richard COULSON, Abingdon, UNITED KINGDOM;

Delwyn EVANS, Abingdon, UNITED KINGDOM; Angeliki SIOKOU, Abingdon, UNITED KINGDOM; Clive TELFORD, Abingdon, UNITED KINGDOM;

Applicant(s)

P2i Ltd, Abingdon, UNITED KINGDOM

Power of Attorney: The patent practitioners associated with Customer Number 33072

Domestic Priority data as claimed by applicant

This application is a 371 of PCT/GB2016/051688 06/08/2016

Foreign Applications (You may be eligible to benefit from the **Patent Prosecution Highway** program at the USPTO. Please see http://www.uspto.gov for more information.) EUROPEAN PATENT OFFICE (EPO) 15386017.6 06/09/2015

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page 1 of 3

The country code and number of your priority application, to be used for filing abroad under the Paris Convention, is **US 15/735,130**

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Title

COATINGS

Preliminary Class

Statement under 37 CFR 1.55 or 1.78 for AIA (First Inventor to File) Transition Applications: No

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PATENT APPLICATION FEE DETERMINATION RECORD Substitute for Form PTO-875									Application or Docket Number 15/735,130		
APPLICATION AS FILED - PART I (Column 1) (Column 2) SMALL ENTITY									OTHER THAN OR SMALL ENTITY		
	FOR	NUMBE	NUMBER FILED		NUMBER EXTRA		FEE(\$)	1	RATE(\$)	FEE(\$)	
BASIC FEE (37 CFR 1.16(a), (b), or (c))		N	N/A		N/A			1	N/A	300	
SEARCH FEE (37 CFR 1.16(k), (i), or (m))		N	N/A		N/A			1	N/A	520	
EXAMINATION FEE (37 CFR 1.16(o), (p), or (q))		N	N/A		J/A	N/A		1	N/A	760	
TOTAL CLAIMS (37 CFR 1.16(i))		90	90 minus 20=		* 70			OR	x 100 =	7000	
INDEPENDENT CLAIMS (37 CFR 1.16(h))		^{/S} 2	2 minus 3 =					1	x 460 =	0.00	
FEE	PLICATION SIZE E CFR 1.16(s))	\$310 (\$15) 50 sheets	If the specification and drawings exceed 100 sheets of paper, the application size fee due is \$310 (\$155 for small entity) for each additional 50 sheets or fraction thereof. See 35 U.S.C. 41(a)(1)(G) and 37 CFR 1.16(s).							0.00	
MUL	TIPLE DEPE N DE	NT CLAIM PRE	SENT (37	7 CFR 1.16(j))						820	
* If th	ne difference in co	lumn 1 is less th	an zero,	enter "0" in colur	nn 2.	TOTAL		1	TOTAL	9400	
NTA		(Column 1) CLAIMS REMAINING AFTER AMENDMENT		(Column 2) HIGHEST NUMBER PREVIOUSLY PAID FOR	(Column 3) PRESENT EXTRA	SMALL RATE(\$)	ADDITIONAL FEE(\$)	OR	SMALL RATE(\$)	ADDITIONAL FEE(\$)	
	Total (37 CFR 1.16(i))	*	Minus	**	=	X =		OR	X =		
AMENDMENT	Independent (37 CFR 1.16(h))	*	Minus	***	=	x =		OR	x =		
₽	Application Size Fee (37 CFR 1.16(s))]			
	FIRST PRESENTATION OF MULTIPLE DEPENDENT CLAIM (37 CFR 1.16(j))							OR			
TOTAL ADD'L FEE								OR	TOTAL ADD'L FEE		
	Т	(Column 1)	<u> </u>	(Column 2) HIGHEST	(Column 3)		1	7			
NT B		REMAINING AFTER AMENDMENT		NUMBER PREVIOUSLY PAID FOR	PRESENT EXTRA	RATE(\$)	ADDITIONAL FEE(\$)		RATE(\$)	ADDITIONAL FEE(\$)	
AMENDMENT	Total (37 CFR 1.16(i))	*	Minus	**	=	X =		OR	x =		
	Independent (37 CFR 1.16(h))	*	Minus	***	=	x =		OR	x =		
¥	Application Size Fee (37 CFR 1.16(s))]			
	FIRST PRESENTATION OF MULTIPLE DEPENDENT CLAIM (37 CFR 1.16(j))							OR			
						TOTAL ADD'L FEE		OR	TOTAL ADD'L FEE		
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KAGAN BINDER, PLLC SUITE 200, MAPLE ISLAND BUILDING 221 MAIN STREET NORTH

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STILLWATER, MN 55082

Notification of Insufficiency Under 37 CFR 1.492 and/or 1.495 (DO/EO/US)

The following items have been submitted by the applicant or the International Bureau to the United States Patent and Trademark Office as a Designated / Elected Office (37 CFR 1.495).

- Priority Document
- Copy of the International Application filed on 12/08/2017
- Copy of the International Search Report filed on 12/08/2017
- Information Disclosure Statements filed on 12/08/2017
- Request for Immediate Examination filed on 12/08/2017
- U.S. Basic National Fees filed on 12/08/2017
- Authorize Access to Search Results filed on 12/08/2017
- Priority Documents filed on 12/08/2017
- Power of Attorney filed on 12/08/2017
- Authorization to Permit Access filed on 12/08/2017
- Application Data Sheet (37 CFR 1.76) filed on 12/08/2017

The following items **MUST** be furnished within the period set forth below:

 Additional claim fees of \$7820 as a undiscounted entity, including any required multiple dependent claim fee, are required. Applicant must submit the additional claim fees or cancel the additional claims for which fees are due.

SUMMARY OF FEES DUE:

Total additional fees required for this application is \$7820 for an Undiscounted Entity:

Applicant is notified that the above-identified application contains the deficiencies noted below. No period for reply is set forth in this notice for correction of these deficiencies. However, if a deficiency relates to the inventor's oath or declaration, the applicant must file an oath or declaration in compliance with 37 CFR 1.63, or a substitute statement in compliance with 37 CFR 1.64, executed by or with respect to each actual inventor no later than the expiration of the time period set in the "Notice of Allowability" to avoid abandonment. See 37 CFR 1.495(c).

 Properly executed inventor's oath or declaration for the following inventor(s) has not been submitted: Stephen Richard COULSON, Delwyn EVANS, Angeliki SIOKOU, and Clive TELFORD

Total additional claim fee(s) for this application is \$7820

- \$7000 for 70 total claims over 20.
- \$820 for multiple dependent claim surcharge.

page 1 of 2

FORM PCT/DO/EO/923 (371 Formalities Notice)

ALL OF THE ITEMS SET FORTH ABOVE MUST BE SUBMITTED WITHIN TWO (2) MONTHS FROM THE DATE OF THIS NOTICE. FAILURE TO PROPERLY RESPOND WILL RESULT IN ABANDONMENT.

The time period set above may be extended by filing a petition and fee for extension of time under the provisions of 37 CFR 1.136(a).

Applicant is reminded that any communications to the United States Patent and Trademark Office must be mailed to the address given in the heading and include the U.S. application no. shown above (37 CFR 1.5)

Registered users of EFS-Web may alternatively submit their reply to this notice via EFS-Web. https://sportal.uspto.gov/authenticate/AuthenticateUserLocalEPF.html

For more information about EFS-Web please call the USPTO Electronic Business Center at **1-866-217-9197** or visit our website at http://www.uspto.gov/ebc.

If you are not using EFS-Web to submit your reply, you must include a copy of this notice.

MAMYE I WAGSTAFF	
Telephone: (571) 272-1057	

MULTIPLE DEPENDENT CLAIM Application Number Filing Date FEE CALCULATION SHEET 15735130 Substitute for Form PTO-1360 (For use with Form PTO/SB/06) Applicant(s) Stephen COULSON * May be used for additional claims or amendments AFTER FIRST AMENDMENT AFTER SECOND AMENDMENT CLAIMS AS FILED Indep Depend Indep Depend Indep Depend Indep Depend Indep Depend Indep Depend 51 (1)2 52 (1) 3 53 (1) 54 4 (1) (1) 55 5 (1) (1)56 57 (1)(1) (1) 58 (1) 8 59 9 (1) (1) 10 (1) 60 (1) (1)61 (1) 12 (1) 62 (1) 13 (1) 63 (1) 14 64 15 (1) 65 (1) 16 (1) 66 (1) 17 (1) 67 (1) 18 (1) 68 (1) 19 (1) 69 (1) 20 70 (1) (1) 71 21 (1) (1) 22 72 (1) (1) 24 74 (1) (1)25 (1) 75 (1) 26 (1) 76 (1) (1) (1) 78 28 (1) (1) 29 79 (1) (1) 80 81 (1) (1) 82 32 (1) (1) 33 (1) 83 (1) 34 (1) 84 35 (1) 85 86 36 (1) 37 87 (1) 38 88 (1)(1) 89 40 90 41 91 1 92 42 1 43 93 44 94 45 95 3 96 46 3 48 98 (1)49 (1) 99 100 50 (1) Total Inde 0 0

0

Total Depend Total Claims

In The United States Patent And Trademark Office

Inventor: Stephen Richard COULSON Examiner: To Be Assigned

Serial No. 15/735,130 Group Art Unit: 1762

International Filing Date: June 8, 2016 Confirmation No. 7101

International App. No. PCT/GB2016/051688 Docket No. GBL0007/US

371(c): December 8, 2017

For: COATINGS

Commissioner for Patents P.O. Box 1450 Alexandria, Virginia 22313-1450 I CERTIFY THAT ON <u>06/15/2018</u>, THIS CORRESPONDENCE IS BEING ELECTRONICALLY TRANSMITTED TO THE UNITED STATES PATENT AND TRADEMARK OFFICE VIA THE OFFICE'S EFS-WEB

/Sonja A. Talley/ Sonja A. Talley

Preliminary Amendment

Dear Sir or Madam:

This is a Preliminary Amendment in the above-captioned application. Kindly enter this Amendment prior to calculating the filing fee for this application.

No fee is believed to be required in connection with the filing of this Preliminary Amendment.

Claim Amendments

Please amend claims 1-3, 5-16, and 31-34; and cancel claims 4, 17-30, and 35-91 as follows:

1. (Currently Amended) An electronic or electrical device or <u>electronic or electrical</u> component thereof comprising a protective cross-linked polymeric coating on a surface of the electronic or <u>electrical said</u> device or component thereof;

wherein the protective cross-linked polymeric coating is obtainable by exposing the electronic or electrical <u>said</u> device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow formation of the protective cross-linked polymeric coating on a surface thereof,

wherein the monomer compound is a compound of formula I(a):

$$\begin{array}{c|c} I(a) \\ \hline \\ R_1 \\ \hline \\ R_2 \\ \hline \\ R_4 \\ \end{array} \begin{array}{c|c} R_5 \\ \hline \\ R_6 \\ \hline \\ R_8 \\ \hline \\ R_{10} \\ \hline \\ \end{array} \begin{array}{c|c} X \\ \hline \\ \hline \\ C \\ \hline \\ C \\ \end{array} \begin{array}{c} C \\ \hline \\ C \\ C \\ \end{array} \begin{array}{c} C \\ \hline \\ C \\ C \\ \end{array}$$

wherein each of R_1 , R_2 , and R_5 to R_8 is hydrogen; each of R_4 , R_9 and R_{10} is independently hydrogen or methyl; each X is independently hydrogen or halogen; a is from 0-10; b is from 3 to 7; and c is 0 or 1-has the following formula:

where R₁, R₂ and R₄ are each independently selected from hydrogen, optionally substituted branched or straight chain C₁-C₆ alkyl or halo alkyl or aryl optionally substituted by halo, and R₃ is selected from:

$$\begin{array}{c|c} & & & & \\ & & & \\ \hline & & & \\ \hline & & \\ \hline & & \\ \hline & & \\ \end{array}$$

where each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C₁-C₆ alkyl, halo alkyl or aryl optionally substituted by halo; and n₁ is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moieties and has a boiling point of less than 500 °C at standard pressure, the crosslinking reagent having one of the following structures:

where Y₁, Y₂, Y₃, Y₄, Y₅, Y₆, Y₇ and Y₈ are each hydrogen; and L is a linker moiety.

- 2.(Currently Amended) An electronic or electrical The device or component thereof according to claim 1, wherein the protective cross-linked polymeric coating is a physical barrier to mass and electron transport.
- 3.(Currently Amended) An electronic or electrical <u>The</u> device or component thereof according to claim 1 or elaim 2, wherein the protective cross-linked polymeric coating forms a liquid repellent surface defined by a static water contact angle (WCA) of at least 90°.

4. (Cancelled)

5. (Currently Amended) An electronic or electrical <u>The</u> device or component thereof according to elaim 4 claim 1, wherein for compound (i) L has the following formula:

each Y_9 is independently selected from, a bond, -O-, -O-C(O)-, -C(O)-O-, -Y₁₁-O-C(O)-, -C(O)-O-Y₁₁-, -OY₁₁-, and -Y₁₁O-, where Y₁₁ is an optionally substituted cyclic, branched or straight chain C₁-C₈ alkylene; and

Y₁₀ is selected from optionally substituted cyclic, branched or straight chain C₁-C₈ alkylene and a siloxane group.

6. (Currently Amended) An electronic or electrical <u>The</u> device or component thereof according to claim 5, wherein for compound (i) L has one of the following structures:

7. (Currently Amended) An electronic or electrical The device or component thereof according to claim 5, wherein for compound (i) L has one of the following structures:

8. (Currently Amended) An electronic or electrical The device or component thereof according to any of claims 5 to 7 claim 5, wherein Y_{10} has the following formula:

wherein each Y_{12} and Y_{13} is independently selected from H, halo, optionally substituted cyclic, branched or straight chain alkyl, or $-OY_{14}$, where Y_{14} is selected from optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl, and n is an integer from 1 to 10.

9. (Currently Amended) An electronic or electrical The device or component thereof according to claim 8, wherein each Y_{12} is H and each Y_{13} is H.

10. (Currently Amended) An electronic or electrical The device or component thereof according to claim 8, wherein each Y_{12} is fluoro and each Y_{13} is fluoro.

11. (Currently Amended) An electronic or electrical <u>The</u> device or component thereof according to any of claims 8 to 10 claim 8, wherein n is from 4 to 6.

12. (Currently Amended) An electronic or electrical <u>The</u> device or component thereof according to any of claims 5 to 7 claim 5, wherein Y_{10} has the following formula:

Wherein each Y_{15} is independently selected from optionally substituted branched or straight chain C_1 - C_6 alkyl.

- 13. (Currently Amended) An electronic or electrical The device or component thereof according to claim 12, wherein each Y_{15} is methyl, and each Y_9 is a bond.
- 14. (Currently Amended) An electronic or electrical <u>The</u> device or component thereof according to any of claims 5 to 7 claim 5, wherein Y_{10} has the following formula:

wherein Y_{16} to Y_{19} are each independently selected from H and optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl.

- 15. (Currently Amended) An electronic or electrical The device or component thereof according to claim 14, wherein Y₁₈ is H or vinylene, and Y₁₆, Y₁₇ and Y₁₉ are each H.
- 16. (Currently Amended) An electronic or electrical The device or component thereof according to any preceding claim claim 1, wherein the crosslinking reagent is selected from Divinyl Adipate (DVA), 1,4 Butanediol divinyl ether (BDVE), 1,4 Cyclohexanedimethanol divinyl ether (CDDE), 1,7 -Octadiene (170D), 1,2,4 -Trivinylcyclohexane (TVCH), 1,3-Divinyltetramethyldisiloxane (DVTMDS), Diallyl 1,4-cyclohexanedicarboxylate (DCHD), 1,6-Divinylperfluorohexane (DVPFH), 1H,1H,6H,6H-Perfluorohexanediol diacrylate (PFHDA) and glyoxal bis (diallyl acetal) (GBDA).

17-30. (Cancelled)

31. (Currently Amended) An electronic or electrical The device or component thereof according to any of claims 24 to 30 claim 1, wherein the compound of formula I(a) has the following formula:

where n is from 2 to 10.

32. (Currently Amended) An electronic or electrical <u>The</u> device or component thereof according to any of claims 24 to 30 claim 1, wherein the compound of formula I(a) has the following formula:

$$\begin{array}{c}
H \\
CH_3
\end{array}$$
(CF₂)_nF

where n is from 2 to 10.

- 33. (Currently Amended) An electronic or electrical The device or component thereof according to claim 31, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-perfluorohexyl acrylate (PFAC4), 1H,1H,2H,2H-perfluoroctyl acrylate (PFAC6), 1H,1H,2H,2H-perfluorodecyl acrylate (PFAC8) and 1H,1H,2H,2H-perfluorododecyl acrylate (PFAC10).
- 34. (Currently Amended) An electronic or electrical <u>The</u> device or component thereof according to claim 32, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-pefluorohexyl

methacrylate (PFMAC4), 1H,1H,2H,2H-perfluorooctyl methacrylate (PFMAC6) and 1H,1H,2H,2H-perfluorodecyl methacrylate (PFMAC8).

35-91. (Cancelled)

Remarks

Applicant respectfully requests that the preliminary amendment be entered prior to examination of the above-identified patent application.

Claims 1-91 are pending in the above-identified patent application. With this Preliminary Amendment, claims 1-3, 5-16, and 31-34 are amended, and claims 4, 17-30, and 35-91 are canceled. Upon entry of the current amendment, claims 1-3, 5-16, and 31-34 are pending, which include 19 total claims and 1 independent claim.

Claims Pending Upon National Stage Entry

Applicant notes that the PCT Publication of PCT application PCT/GB2016/051688 published with 34 pages of text, which included 83 claims. The "Claims Worksheet (PTO-2022)" on PAIR identifies 83 claims as well. But, Applicant points out that the PCT Publication of PCT application PCT/GB2016/051688 erroneously left out page 35, which was originally filed with PCT application PCT/GB2016/051688 and included claims 84-91. A copy of PCT application PCT/GB2016/051688 as filed was downloaded from http://www.wipo.int/patentscope/en/ and is submitted herewith as an attachment, which shows page 35 and claims 84-91. Applicant's undersigned representative, Mr. Paul J. Parins, contacted the IB on June 15, 2018, and the IB concurred with respect to page 35 as explained above. The IB indicated that PCT application PCT/GB2016/051688 would be re-published in July, 2018

Accordingly, the present Preliminary Amendment references claims 84-91 as canceled.

Support for Claim Amendments

Applicant submits that the claim amendments are fully supported by the application as originally filed and do not present new matter.

with page 35 and provided to the USPTO as a Designated Office.

Support for the amendment to claim 1 can be found at, e. g., page 5, lines 23-27 and page 11, lines 6-11. Claim 1 is also amended for clarification purposes.

Claims 2-3, 5-16, and 31-34 are amended to conform to the amendment of claim 1, from which they depend.

Claims 3, 8, 11, 12, 14, 16, 31, and 32 are amended to remove multiple dependencies.

Docket No. GBL0007/US Page 10

Approval of the application and allowance of the claims are earnestly solicited. In the event that a phone conference between the Examiner and the Applicant's undersigned representative would help resolve any issues in the application, the Examiner is invited to contact said attorney at (651) 275-9831.

Respectfully Submitted,

Date: June 15, 2018 By: /Paul J. Parins/

Paul J. Parins, Reg. No. 54,358 Customer Number 33072

Phone: 651-275-9831 Fax: 651-351-2954

#119892

Coatings

Field of the invention

This invention relates to protective coatings. Specifically, the present invention relates to protective coatings for electronic or electrical devices and components thereof, and methods of forming such coatings. The coatings can protect by being hydrophobic and so resist the ingress of water-based liquid into electronic devices, or they can protect by forming a barrier coating and so provide electrical resistance between the electrical parts of the phone and water based liquid.

Background of the invention

Monounsaturated monomers are used to make barrier coatings using plasma polymerisation processes (see co-pending application).

The perfluoroalkyl chain monomers are also used for generating hydrophobic surfaces from pulsed plasma deposition processes (see WO 9858117 A1).

The power of the plasma initiated polymerisation affects the nature of the polymer produced. The higher average energy inputs of continuous wave plasmas lead to more fragmentation of the monomer, and so the polymer loses structural properties of the monomer. In the case of 1H,1H,2H,2H- perfluorodecyl acrylate (PFAC8), there's less retention of the perfluoroalkyl chain and the contact angle of the surface coating is compromised. Higher plasma energies also lead to more crosslinking. For the lower average energy inputs of pulsed plasmas, there's better retention of monomer structure and less crosslinking. The greater retention of the perfluoro chain under low energy, pulsed plasma conditions leads to best levels of contact angles for the surface coating.

When the perfluoroalkyl chains have eight or more fluorinated carbons (long chain), the polymer made from the monomer has a crystalline structure. When the perfluoroalkyl chains have less than eight fluorinated carbons, the resulting polymer is amorphous and so can be unstable in the presence of water (see "Molecular Aggregation Structure and Surface Properties of Poly(fluoroalkyl acrylate) Thin Films, Marcomolecules, 2005, vol 38, p5699-5705)

When long chain perfluoroalkyl polymers are produced by high average power (continuous wave or CW) or low average power (pulsed wave or PW) plasmas, then because of the crystalline structure of the long chains, the polymers are non-stick to the touch and stable in the presence of water. However, the feel and water stability of shorter chain polymer coatings is affected by the plasma power levels used. For example, when PFAC6 (1H,1H,2H,2H-perfluorooctyl acrylate) is polymerised in low power plasma conditions, the resulting polymer coating can have several disadvantages. For example the coating can cause water drops to spread out a little (slump), be marked by the presence of a water drop on its surface, have a tacky feel, or can be easily smeared (for example on substrates of silicon wafer and ABS plastic).

By increasing the power of the plasma used for polymerisation, the polymer becomes more crosslinked and becomes more resistant to smearing. However, increasing the power has the concomitant effect of decreasing the water contact angle through more monomer fragmentation (as described above). Figure 1 shows the effect of increasing the power to monomer flow ratio for CW plasmas in a 125 litre chamber: at a ratio of 4 W/µl/min, the water contact angle is ~85-95 degrees and the coating is tack-free. However, as the ratio drops, the contact angle increases and the occurrence of tacky/smudgy coatings increases too. Figure 2 shows the same effect for pulsed plasma conditions. These results show that the process window for producing tack and smudge-free coatings have a limited plasma processing range and the final coating has a compromised water contact angle.

Accordingly, it is an aim to solve one or more of the above-mentioned problems with the prior art coatings.

Statement of Invention

An aspect of the present invention provides an electronic or electrical device or component thereof comprising a protective cross-linked polymeric coating on a surface of the electronic or electrical device or component thereof;

wherein the protective cross-linked polymeric coating is obtainable by exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow formation of the protective cross-linked polymeric coating on a surface thereof,

wherein the monomer compound has the following formula:

where R_1 , R_2 and R_4 are each independently selected from hydrogen, optionally substituted branched or straight chain C_1 - C_6 alkyl or halo alkyl or aryl optionally substituted by halo, and R_3 is selected from:

where each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C_1 - C_6 alkyl, halo alkyl or aryl optionally substituted by halo; and n_1 is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moieties and has a boiling point of less than 500 °C at standard pressure.

High levels of polymer crosslinking (formerly only achievable with high average power continuous wave plasmas) can be achieved by adding a crosslinking molecule to the monomer to produce a cross-linked co-polymer. This has the advantage of increasing the plasma processing range so stable coatings can now be produced in low average energy pulse plasma conditions.

The high degree of retention of the hydrophobic monomer structures from the low energy pulsed plasma gives coatings of the co-polymer a good hydrophobic coating (as evidenced by water contact angles) and a coating that is not sticky or smeary to the touch. Thus, these properties make it suitable for the coating of electronic devices to prevent the ingress of water-based liquids from accidental damage. Such coatings typically have a thickness between 1 and 100 nm, preferably 1 and 50 nm.

The coating may be described as a liquid repellant layer, typically a water repellant layer. The coating may repel water, an aqueous liquid such as rainwater, or oil.

Alternatively, or in addition to providing a liquid repellent layer, the coating may be a physical barrier layer.

The coating may be a physical barrier i.e. it provides a physical barrier to mass and electron transport.

The physical barrier layer restricts diffusion of water, oxygen and ions. When the coating is a physical barrier it typically has a thickness of greater than 50 nm.

The coating is a protective layer i.e. it prevents damage by contact with water or other liquids. The coating may provide is protective functionality by forming a liquid repellent layer and/or a physical barrier layer.

The coating is preferably substantially pin-hole free to enable it to provide a physical barrier. Preferably $\Delta Z/d < 0.15$, where ΔZ is the average height variation on an AFM line scan in nm (as shown in Figure 3) and d is coating thickness in nm.

The value of $\Delta Z/d$ tells us to what extent defects/voids on the surface of the coating extend into the coating, i.e. the percentage value of the depth of defect over total coating thickness. For example, $\Delta Z/d = 0.15$ means that the voids on the surface only extend down to 15% of the coating thickness. A coating with a $\Delta Z/d < 0.15$ is defined herein as being substantially pinhole free. If voids are bigger than this, the desired functionality is unlikely to be achieved.

The coating is preferably conformal, which means that it takes the 3D shape of the electronic or electrical device or component thereof and covers substantially an entire surface of the device. This has the advantage of ensuring that the coating has sufficient thickness to give optimal functionality over an entire surface of the device or component. The meaning of the term "covers substantially an entire surface" will depend to some extent on the type of surface to be covered. For example, for some components, it may be necessary for there to be complete coverage of the surface in order for the component to function after submersion in water. However, for other components or housings, small gaps in coverage may be tolerated.

The coating may have a thickness of 50 to 10,000nm, optionally 50 to 8000nm, 100 to 5000nm, preferably 250nm – 5000nm, most preferably 250nm – 2000nm.

The coating may be electrically insulating and be sufficiently compliant that electrical connectors can be joined to the electronic or electrical device or component thereof and an electrical connection made between the electrical connectors and electronic or electrical device or component thereof without the requirement to first remove the coating. In this case, the force exerted by the electrical connector on the coating is sufficient to alter the structure of or even break through the coating local to the electrical connector, allowing the electrical connection to be made. Electrical connectors can typically be joined to the electronic or electrical device or component in this way for coating thicknesses of under 5000nm, and for high performance coatings below 2000nm.

The electronic or electrical device or component thereof can typically withstand immersion in up to 1m of water for over 30 minutes without failure or corrosion whilst power applied to the electronic or electrical device or component. The effectiveness of the coating can be determined by measuring its electrical resistance at a fixed voltage when submerged in water for a set time period; for example, applying a voltage of 8V to the coating of a device which is submerged in water for 13 minutes. Coatings having a resistance value of 1x10⁷ Ohms or higher in this test are effective barrier coatings and the coated electronic or electrical device or component thereof will pass successfully an IPX7 test. The IPX7 test is the Ingress Protection Marking which classifies and rates the degree of protection provided against water. In the IPX7 test for phones, the device is immersed in water under defined conditions of pressure and time (up to 1 m of submersion) for a duration of 30 minutes. The device has to be powered on during testing and functional after 24 hrs.

In one embodiment, the coating is electrically insulating and has a thickness of less than 1 micron and wherein a force of 5- 20g applied to the coating using a round probe with 1 mm diameter allows an electrical connection to be made with the electronic or electrical device or component thereof in the local area where the force has been applied.

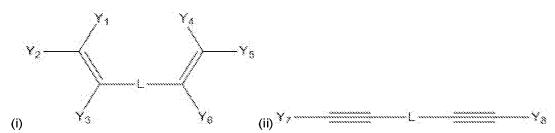
In another embodiment, the coating is electrically insulating and has a thickness of 1-2.5 microns and wherein a force of 20-100g applied to the coating using a round probe with 1 mm diameter allows an electrical connection to be made in the local area of the coating where the force has been applied.

The coating may have a higher density than that of the corresponding monomer from which it is formed. For example, the increase in density may be at least 0.1 g/cm^{3.} The increase in density is explained by the highly crosslinked coating. The high density of the coating improves the barrier properties of the coating.

The coating may form a surface defined by a static water contact angle (WCA) of at least 70°. A coating with a WCA of at least 90° is a liquid repellent, typically a water repellent layer. For fluorinated polymers, the coating may have a static water contact angle of at least 100°. The contact angle of a liquid on a solid substrate gives an indication of the surface energy which in turn illustrates the substrate's liquid repellence. Contact angles were measured on a VCA Optima contact angle analyser, using 3 µl droplets of deionised water at room temperature.

Crosslinker

A basic requirement for the cross-linking reagent is the presence two or more unsaturated bonds, for example, -C=C- or alkyne groups. The unsaturated bonds are attached by one or more linker moieties. The linker moiety is not especially limited provided it joins together the two or more unsaturated bonds. The cross-linking reagent must have a boiling point of less than 500 °C, preferably from -10 to 300°C, optionally 180 to 270°C, most preferably 205 to 260°C at standard pressure. It is a preference, but not a requirement, that the cross-linking reagent is not unduly hazardous for use in plasma processing, that is to say it can be used in a production environment where low levels of vapour do not present any significant health and safety issues (e.g. strongly oxidising, explosive, toxic, or has an offensive odour (e.g. stench reagent)). The crosslinking reagent preferably has one of the following structures:



where Y_1 , Y_2 , Y_3 , Y_4 , Y_5 , Y_6 , Y_7 and Y_8 are each independently selected from hydrogen, optionally substituted cyclic, branched or straight chain C_1 - C_6 alkyl or aryl; and L is a linker moiety.

Most preferably L has the following formula:

where each Y_9 is independently selected from, a bond, -O-, -O-C(O)-, -C(O)-O-, -Y₁₁-O-C(O)-, -C(O)-O-Y₁₁-, -OY₁₁-, and -Y₁₁O-, where Y₁₁ is an optionally substituted cyclic, branched or straight chain C₁-C₈ alkylene; and

Y₁₀ is selected from optionally substituted cyclic, branched or straight chain C₁-C₈ alkylene and a siloxane containing group.

In one most preferred embodiment Y₁₀ has the following formula:

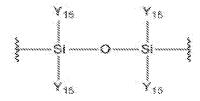
wherein each Y_{12} and Y_{13} is independently selected from H, halo, optionally substituted cyclic, branched or straight chain C_1 - C_8 alkyl, or $-OY_{14}$, where Y_{14} is selected from optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl, and n is an integer from 1 to 10.

Optionally, each Y_{12} is H and each Y_{13} is H, i.e. Y_{10} , is a linear alkylene chain. For this embodiment Y_9 is preferably a vinyl ester or vinyl ether group.

Optionally, each Y₁₂ is fluoro and each Y₁₃ is fluoro, i.e. Y₁₀ is a linear perfluoroalkylene chain.

Typically, n is from 4 to 6.

In another embodiment Y₁₀ has the following formula:



(v)

wherein each Y₁₅ is independently selected from optionally substituted branched or straight chain C₁-C₆ alkyl.

In one embodiment each Y_{15} is methyl, and each Y_{9} is a bond.

In a further embodiment Y_{10} has the following formula:

(vi)

wherein Y_{16} to Y_{19} are each independently selected from H and optionally substituted branched or straight chain C_{1} - C_{8} alkyl or alkenyl. A preferred alkenyl group is vinyl. Optionally Y_{18} is H or vinyl, and Y_{16} , Y_{17} and Y_{19} are each H. In one embodiment each of Y_{16} to Y_{19} is H. In another embodiment Y_{18} is vinyl, and Y_{16} , Y_{17} and Y_{19} are each H.

In preferable embodiments for compound (i), L has one of the following structures:

In another embodiment for compound (i) L has one of the following structures:

For L according structure (vii), Y_{10} is preferably an alkylene chain or a cycloalkylene, such as those shown in structures iv) and vi) above. The alkylene chain may be a straight chain alkylene chain. When Y_{10} is a cycloalkylene, cyclohexylene is preferred, most preferably 1,4 cyclohexylene.

For L according to structure (viii), Y_{10} is preferably structure (iv), i.e. an alkylene or fluoroalkylene chain.

For L according to structure (ix), Y₁₀ is preferably cycloalkylene, such as the cyclohexylene according to structure (vi).

For L according to structure (x), Y_{10} is preferably structure (iv) wherein each Y_{12} and Y_{13} is F, i.e. a perfluoroalkylene chain.

For L according to structure (xi) or structure (xii) Y_{10} is preferably alkylene or cycloalkylene. Optionally the alkylene or cycloalkylene may be substituted with one or more vinyl groups or alkenyl ether groups, preferably one or more vinyl ether groups.

When each Y_9 is a bond, each Y_{10} may be any of structures (iv), (v) and (vi). Preferably Y_{10} is a straight chain alkylene such that the crosslinking reagent is a diene, such as a heptadiene, octadiene, or nonadiene, most preferably 1,7-octadiene.

When each Y_9 is O, each Y_{10} is preferably a branched or straight chain C_{1-} C_6 alkylene, preferably a straight chain alkylene, most preferably a C_4 straight chain alkylene, i.e the crosslinking reagent is 1,4-butanediol divinyl ether

It will be understood that each Y_9 group can be combined with any other Y_9 group and Y_{10} group to form the crosslinking reagent.

The skilled person will be aware of possible substituents for each of the cyclic, branched or straight chain C_1 - C_8 alkylene groups mentioned above. The alkylene groups may be substituted at one or more positions by a suitable chemical group. Halo substituents are preferred, with fluoro substituents most preferred. Each C_1 - C_8 alkylene group may be a C_1 - C_8 , or C_6 - C_8 alkylene group.

Particularly preferred embodiments of the crosslinker have alkyl chains for Y_{10} and vinyl ester or vinyl ether groups on either side.

A particularly preferred cross-linking reagent is divinyl adipate. (a divinyl ester)

Another preferred cross-linking reagent is 1,4-butanediol divinyl ether (a divinyl ether)

For the most preferred embodiments the crosslinking reagent is selected from Divinyl Adipate (DVA), 1,4 Butanediol divinyl ether (BDVE), 1,4 Cyclohexanedimethanol divinyl ether (CDDE), 1,7 -Octadiene (17OD), 1,2,4 -Trivinylcyclohexane (TVCH), 1,3-Divinyltetramethyldisiloxane (DVTMDS), Diallyl 1,4-cyclohexanedicarboxylate (DCHD), 1,6-Divinylperfluorohexane (DVPFH), 1H,1H,6H,6H-Perfluorohexanediol diacrylate (PFHDA) and glyoxal bis (diallyl acetal) (GBDA).

For the alkyne crosslinking reagents according to compound (ii), L is preferably selected from a branched or straight chain C₁-C₈ alkylene or an ether group. L may be a C₃, C₄, C₅, or C₆ alkylene,

preferably a straight chain alkylene. Chemical structures for the particularly preferred crosslinking reagents are set out below in Table 1:

Divinyl Adipate (DVA)	H ₂ C, O,
	CH ₂
1,4 Butanediol divinyl ether (BDVE)	H_2C O CH_2
1,4 Cyclohexanedimethanol	O CH ₂
divinyl ether (CDDE)	H ₂ C O
1,7 -Octadiene (170D)	H_2C CH_2
1,2,4 -Trivinylcyclohexane	CH ₂
(TVCH)	
	H_2C CH_2
1,3-	CH ₃ CH ₃
Divinyltetramethyldisiloxane	Si—O—Si————————————————————————————————
(DVTMDS)	H_2C' CH_2 CH_3 CH_3
Diallyl 1,4-	OOCH ₂
cyclohexanedicarboxylate	G12
(DCHD)	
	0 CH ₂
1,6-Divinylperfluorohexane	F F F
(DVPFH)	H_2C F
1H,1H,6H,6H-	0 F F 4 C . F F 0 ~~
Perfluorohexanediol	H_2C O F F O CH_2
diacrylate (PFHDA)	F F O
GBDA	QCH ₂
	H_2C O O CH_2
	H ₂ C ∕Ó

1,6-heptadiyne	НС=СН
1,7-heptadiyne	НС
1,8-heptadiyne	нс=сн
Propargyl ether	нс=сн

Table 1

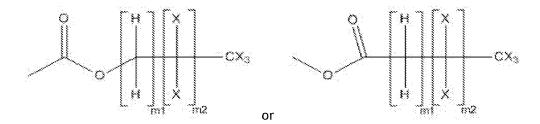
Monomer Compound

The monomer compound has the following formula:

where R_1 , R_2 and R_4 are each independently selected from hydrogen, optionally substituted branched or straight chain C_1 - C_6 alkyl or halo alkyl or aryl optionally substituted by halo, and R_3 is selected from:

where each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C_1 - C_6 alkyl, halo alkyl or aryl optionally substituted by halo; and n_1 is an integer from 1 to 27. Preferably, n_1 is from 1 to 12. Optionally n_1 is from 4 to 12, optionally 6 to 8.

In a preferred embodiment R₃ is selected from:



where m_1 is an integer from 0 to 13 and each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C_1 - C_6 alkyl, halo alkyl or aryl optionally substituted by halo; and m_2 is an integer from 2 to 14;

In a particularly preferred embodiment the monomer is a compound of formula I(a):

l(a)

wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_6 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1;

; or

the monomer is a compound of formula I(b):

I(b)

$$R_{2} \longrightarrow \left\{ \begin{array}{c} R_{1} \\ \\ \\ \\ \\ \\ \\ \\ \\ \end{array} \right\} \left\{ \begin{array}{c} R_{2} \\ \\ \\ \\ \\ \\ \end{array} \right\} \left\{ \begin{array}{c} R_{2} \\ \\ \\ \\ \\ \end{array} \right\} \left\{ \begin{array}{c} X \\ \\ \\ \\ \\ \\ \end{array} \right\} \left\{ \begin{array}{c} X \\ \\ \\ \\ \\ \end{array} \right\} \left\{ \begin{array}{c} X \\ \\ \\ \\ \\ \end{array} \right\} \left\{ \begin{array}{c} X \\ \\ \\ \\ \\ \end{array} \right\} \left\{ \begin{array}{c} X \\ \\ \\ \\ \\ \end{array} \right\} \left\{ \begin{array}{c} X \\ \\ \\ \\ \\ \end{array} \right\} \left\{ \begin{array}{c} X \\ \\ \\ \\ \\ \end{array} \right\} \left\{ \begin{array}{c} X \\ \\ \\ \end{array} \right\} \left\{ \begin{array}{c} X \\ \\ \\ \end{array} \right\} \left\{ \begin{array}{c} X \\ \\ \\ \\$$

wherein each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or an optionally substituted C_1 - C_6 branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1.

The halogen may be chlorine or bromine, but is preferably fluorine for compliance with RoHS regulations (Restriction of Hazardous Substances).

a is from 0 to 10, preferably from 0 to 6, optionally 2 to 4, most preferably 0 or 1. b is from 2 to 14, optionally from 2 to 10, preferably 3 to 7.

Each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or a C_1 - C_6 branched or straight chain alkyl group. The alkyl group may be substituted or unsubstituted, saturated or unsaturated. When the alkyl group is substituted, the location or type of the substituent is not especially limited provided the resultant polymer provides an appropriate liquid repellent and/or barrier layer. The skilled person would be aware of suitable substituents. If the alkyl group is substituted, a preferred substituent is halo, i.e. any of R_1 , R_2 , R_4 , and R_5 to R_{10} may be haloalkyl, preferably fluoro alkyl. Other possible substituents may be hydroxyl or amine groups. If the alkyl group is unsaturated it may comprise one or more alkene or alkyne groups.

Each of R₁, R₂, R₄, and R₅ to R₁₀ may be independently selected from hydrogen, methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, tert-butyl, sec-butyl, n-pentyl, neopentyl, n-hexyl, isohexyl, and 3-methylpentyl.

In a preferred embodiment each of R_1 , R_2 , R_4 , and R_5 to R_{10} is independently selected from hydrogen or methyl.

In a preferred embodiment, a and c are each independently 0 or 1; and b is from 3 to 7.

In one preferred embodiment each X is H. In an alternative preferred embodiment each X is F.

Optionally R₁ and R₂ are both hydrogen.

Optionally R_4 is hydrogen or methyl. Preferably R_1 and R_2 are both hydrogen and R_4 is hydrogen or methyl.

Optionally R_9 is hydrogen and R_{10} is branched or straight chain C_1 - C_6 alkyl group. In a preferred embodiment R_{10} is methyl.

In one embodiment each of R₅ to R₈ is hydrogen.

In one embodiment each of R₁, R₂, R₄, and R₅ to R₁₀ is hydrogen, each X is H, a=0 and c=0.

In a particularly preferred embodiment the monomer compound has the following formula:

where n is from 2 to 10.

In another preferred embodiment the monomer compound has the following formula:

$$\begin{array}{c}
H \\
CH_{3}
\end{array}$$
(CF₂)_nF

where n is from 2 to 10.

The monomer compound may be selected from 1H,1H,2H,2H-perfluorohexyl acrylate (PFAC4), 1H,1H,2H,2H-perfluorooctyl acrylate (PFAC6), 1H,1H,2H,2H-perfluorodecyl acrylate (PFAC8) and 1H,1H,2H,2H-perfluorododecyl acrylate (PFAC10).

The monomer compound may be selected from 1H,1H,2H,2H-perfluorohexyl methacrylate (PFMAC4), 1H,1H,2H,2H-perfluorooctyl methacrylate (PFMAC6) and 1H,1H,2H,2H-perfluorodecyl methacrylate (PFMAC8).

The monomer compound of formula I(a) may have the following formula:

wherein a and c are each independently 0 or 1, b =3-7 and n is 4 to 10, where n=a+b+c+1.

The monomer ccompound of formula I(a) may have the following formula:

$$H \longrightarrow CH_3$$
 (CH₂)_nF

where n is 2 to 12.

The monomer compound may be selected from ethyl hexyl acrylate, hexyl acrylate, decyl acrylate, lauryl dodecyl acrylate and iso decyl acrylate.

The monomer may have the following formula:

where n is from 4 to 14.

The monomer may have the following formula:

where n is from 4 to 14.

In an additional aspect, the present invention provides a method for treating an electronic device or component thereof comprising:

exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow a protective cross-linked polymeric coating to form on a surface of the electronic or electrical device or component thereof;

wherein the monomer compound has the following formula:

where R₁, R₂ and R₄ are each independently selected from hydrogen, optionally substituted branched or straight chain C₁-C₆ alkyl or halo alkyl or aryl optionally substituted by halo, and R₃ is selected from:

$$\begin{array}{c|c} & & & \\ & & &$$

where each X is independently selected from hydrogen, optionally substituted branched or straight chain C_1 - C_6 alkyl, halo alkyl or aryl optionally substituted by halo; and n_1 is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds and has a boiling point of less than 500 °C at standard pressure.

The monomer compound and the crosslinking reagent utilised in the method are described in further detail above.

Typically the electronic or electrical device or component thereof is placed in a plasma deposition chamber, a glow discharge is ignited within said chamber, and a voltage applied as a pulsed field.

Preferably the applied voltage is at a power of from 30 to 800W. Optionally the voltage is pulsed in a sequence in which the ratio of the time on/time off is in the range of from 0.001 to 1, optionally 0.002 to 0.5. For example, time on may be 10-500 μ s, preferably 35-45 μ s, or 30-40 μ s, such as about 36 μ s and time off may be from 0.1 to 30 ms, preferably 0.1 to 20 ms, optionally from 5 to 15 ms, for example 6ms. Time on may be 35 μ s, 40 μ s, 45 μ s. Time off may be 0.1, 1, 2, 3, 6, 8, 10, 15, 20, 25 or 30 ms.

The term pulsed may mean that the plasma cycles between a state of no (or substantially no) plasma emission (off-state) and a state where a particular amount of plasma is emitted (on-state). Alternatively, pulsed may mean that there is continuous emission of plasma but that the amount of plasma cycles between an upper limit (on-state) and lower limit (off-state).

In a further embodiment, the invention resides in a method of forming a coating on an electronic or electrical device or component thereof as defined above, the method comprising: exposing said substrate in a chamber to a plasma comprising a monomer compound, preferably a continuous plasma, for a sufficient period of time to allow a protective polymeric coating to form on the substrate,

wherein during exposure of the substrate the continuous plasma has a power density of at least 2W/litre, preferably 20W/litre.

Optionally the voltage is applied as a pulsed field for a period of from 30 seconds to 90 minutes. Optionally the voltage is applied as a pulsed field for from 5 to 60 minutes. Optionally in a preliminary step, a continuous power plasma is applied to the electronic or electrical device or component thereof. The preliminary step may be conducted in the presence of an inert gas.

Before the crosslinker and/or the monomer compound enter the deposition chamber each may be in the form of a gas, liquid or a solid (for example a powder) at room temperature. However, it is the preferred that the crosslinking reagent and the monomer compound are both liquid at room temperature, and most preferably that the monomer and crosslinker liquids are miscible.

The monomer compound and/or the crosslinker will suitably be in a gaseous state in the plasma. The plasma may simply comprise a vapour of the monomer compound and the crosslinker. Such a vapour may be formed in-situ, with the compounds being introduced into the chamber in liquid form. The monomer may also be combined with a carrier gas, in particular, an inert gas such as helium or argon.

In preferred embodiments, the monomer and/or the crosslinker may be delivered into the chamber by way of an aerosol device such as a nebuliser or the like, as described for example in WO2003/097245 and WO03/101621, the content of which is incorporated herein by reference. In such an arrangement a carrier gas may not be required, which advantageously assists in achieving high flow rates.

In one embodiment the monomer compound and/or the crosslinking reagent are in gaseous form and are fed into the plasma at a rate of from 10-5000 mg/minute depending on chamber volume, whilst the pulsed voltage is applied.

Optionally the plasma is created with an average power of from 0.001 to 40W/litre.

The crosslinker may be miscible with the monomer and so introduced together or separately into the plasma chamber. Or the crosslinker may be immiscible with the monomer and introduced separately into the plasma chamber. In this context, the term "miscible" means that the crosslinker is soluble in the monomer, and when mixed they form a solution of uniform composition. The term "immiscible" is used to mean that the crosslinker is only partly soluble or insoluble in the monomer, and so either forms an emulsion or separates out into two layers.

The crosslinking reagent is preferably present in an amount of from 10 to 60 (v/v) %, optionally 20 to 40 (v/v) %, optionally 25 to 30 (v/v)%, optionally 30 to 50 (v/v)% of the total volume of monomer compound and crosslinking reagent, depending on the specific crosslinker. The skilled person would understand that the amount will vary to some extent depending on whether the coating is required to

be liquid repellent or provide a barrier to mass and electron transport. The skilled person would understand that the (v/v) percentages are those which give a stable crosslinked polymer coating and the highest water contact angle.

Electronic or electrical device or component thereof

Although the invention is of benefit in the context of a wide variety of substrates, the substrate may, in all aspects of the invention, advantageously be an electronic substrate.

In some embodiments of the invention, the electronic substrate may comprise an electronic or electrical device, i.e. any piece of electrical or electronic equipment. Non-limiting examples of electrical and electronic devices include communications devices such as mobile phones, smartphones and pagers, radios, and sound and audio systems such as loudspeakers, microphones, ringers or buzzers, hearing aids, personal audio equipment such as personal CD, tape cassette or MP3 players, televisions, DVD players including portable DVD players, video recorders, digi and other set-top boxes such as Sky, computers and related components such as laptop, notebook, tablet, phablet or palmtop computers, personal digital assistants (PDAs), keyboards, or instrumentation, games consoles in particular hand-held playstations and the like, data storage devices, outdoor lighting systems or radio antenna and other forms of communication equipment.

In preferred embodiments of the invention, the substrate may comprise or consist of an electronic component, e.g. a printed circuit board (PCB), a printed circuit board array (PCBA), a transistor, resistor, or semi-conductor chip. The electronic component may thus be an internal component of an electronic device, e.g. a mobile phone. The coatings of the invention are particularly valuable in preventing electrochemical migration in such components.

In all aspects of the invention, the precise conditions under which the protective polymeric coating is formed in an effective manner will vary depending upon factors such as, without limitation, the nature of the monomer compound, the crosslinking reagent, the substrate, as well as the desired properties of the coating. These conditions can be determined using routine methods or, preferably, using the techniques and preferred features of the invention described herein, which work in particular synergy with the invention.

The present invention will now be further described with reference to the following non-limiting examples and the accompanying illustrative drawings, of which:

Figure 1 illustrates shows the effect of increasing the power to monomer flow ratio for CW plasmas in a 3 litre chamber in a prior art process.

Figure 2 shows the same effect for pulsed plasma conditions.

Figure 3 shows a tapping mode image over $10x10~\mu m^2$ field of view of a specimen example (thickness d=1230 nm) prepared according to Example 1 (left hand side) and a contour line plot showing the data used for calculation of RMS roughness (right hand side). The ΔZ value indicated on the plot has been taken over an area of the graph that represents the majority of the coating. Peaks that lie above the ΔZ range indicate large particles and troughs that fall below the ΔZ range show voids or pinholes in the coating. The width of the peaks also gives an indication of the particle size.

Figure 4 illustrates the effect of adding the divinyl adipate crosslinker to perfluorooctyl acrylate on water contact angle from a process on a 125 Litre chamber.

Figure 5 shows the effect of adding crosslinkers in accordance with the invention on the electrical resistance per nm of the coatings.

Figure 6 shows a tapping mode image of a 90 nm thick coating, prepared as described in Example 3, over $2x2\mu m^2$ field of view (top left), a representative contour line indicating height variation (z-axis) of the coating (top right) and a phase image indicating full substrate coverage (bottom left) ;RMS roughness of the coating is 1.65 nm and $\Delta z/d = 0.05$

Figure 7 shows a FTIR/ATR spectrum of a coating formed according to Example 5.

Example 1

Example of plasma polymerising PFAC6 with no crosslinker in a 3 litre chamber

PFAC6 was polymerised in a 3 litre glass chamber using a continuous plasma process. In the experimental matrix, the continuous plasma was produced using RF power levels of 50, 100 and 200 Watts. The process pressure was run at 30, 60 and 90 mTorr and the monomer flow rate run at 50, 100 and 200 microlitres of liquid per minute. The duration of the process was either 10 or 40 minutes. The coated substrate was silicon wafer and whether the coating was tacky or not was determined by wiping the substrate with a finger and determining by eye if the coating had smeared. The results in Figure 1 indicate that for lower CW powers, the coating could easily be smeared (i.e. tacky), but by increasing the power to flow ratio, the coating became resistant to smearing (i.e. non-tacky) and this is expected to be due to additional crosslinking caused by more monomer fragmentation. The water contact angle was also seen to decline with increasing power to flow ratio, and this too is indicative of more power increasing the level of monomer fragmentation.

The same 3 litre chamber was used to also look at the effect of power to flow ratios with pulsed plasma polymerisation. In these experiments, the peak power was either 250 or 500 W. The RF power feed was pulsed with an on time of 35 microseconds and off times of 5.5 and 1 millisecond. Process pressure was either 20, 35, 60 or 80 mTorr. The monomer flow was either 50 or 200 microlitres of liquid per minute. Figure 2 shows that low power to flow conditions can generate coatings that can be smeared (tacky) and only by increasing the power could the smear be reduced. This too had a concomitant decline in water contact angle.

The results from Figures 1 and 2 show that plasma processing PFAC6 on its own could only lead to smear –free coatings when the powers were high enough to decrease the other desirable property of the water contact angle. This illustrates the need to add a crosslinker to prevent the coating smearing and maintain the water contact angle.

Example 2

Examples of Perfluoro Alkyl Acrylate co-polymerisation with Crosslinker to show improvements in electrical resistance

Perfluorooctyl acrylate monomer was mixed with a single crosslinker from the following list:

VINYL ESTERS: Divinyl adipate (DVA)

VINYL ETHERS: 1,4-Butanediol divinyl ether (BDVE); 1,4 Cyclohexanedimethanol divinyl ether (CDDE)

DI- or TRI- VINYLS: 1,7-Octadiene (1,7-OD); 1,2,4-Trivinylcyclohexane (TVCH);

DI VINYL with alkyl fluoro group: 1,6-divinylperfluorohexane (DVPFH)

DI VINYL with silicon group: 1,3-Divinyl tetramethyldisiloxane (DVTMDS)

DI VINYL with cyclic ring and carboxylate groups: Diallyl 1,4-cyclohexanedicarboxylate (DCHC)

DI ACRYLATE: 1H1H,6H,6H-perfluorohexanediol diacrylate (PFHDA)

The crosslinker was mixed with the PFAC6 as a percentage by volume for the following percentages: 20, 40, 60, 80, and 100%.

Plasma initiated polymerisation reactions were carried out in a 3 litre glass plasma chamber. The substrates were test circuit boards. The PFAC6/crosslinker mixture was introduced at a rate of 0.04 – 0.06 ul/min and the process pressure was 40 mTorr. The process plasma consisted of a 1 minute continuous plasma (CW) step of 50W followed by 10 - 20 minute pulsed plasma (PW) step of 50W and RF power being delivered by a pulse sequence with a duty cycle of 6.9%. The coated circuit boards were immersed in tap water and an 8V potential was applied for 13 minutes. The final current readings, together with coating thickness measurements, were used to calculate the electrical resistance per nm of coating. Figure 5 below shows the effect of adding a range of crosslinkers on the electrical resistance per nm of the barrier coatings. Figure 5 clearly shows the improvement in resistance that can be gained by adding the crosslinker to the PFAC6 monomer. Furthermore, all of the crosslinkers tested gave non-smear/non-tacky coatings with concentration levels of 20% and above, with the exceptions of DVTMDS, DCHD, PFHDA and DVPFH which were non-smear at levels of 40% and above.

Example 3

Example of crosslinker concentration on water repellent coating.

Water repellent coatings were prepared in a 125 litre volume chamber using PFAC6 with different levels of divinyl adipate (DVA) and using helium as a carrier gas. The deposition process consisted of a 3 minute CW step with 300W power and a pulsed step with 150W power and an RF pulse duty cycle of 0.018%. Silicon wafer was used as the test substrates and the contact angle of the coated wafer was determined by applying a 3ul drop of deionised wafer onto the coated wafer and measuring the contact angle using a VCA Optima (AST products) with image analysis software. The variation of contact angle with (v/v)% DVA crosslinker is shown in Figure 4. It was also noted that for the 10 (v/v) % DVA concentration, the water drop left a mark on the coated wafer where it's circumference had been. This indicated that a higher level of crosslinker was required to give a more stable coating. This observation, in combination with the results in figure 4, suggest an optimum range of 20 – 40 (v/v) % DVA crosslinker, though this is likely to vary, typically within the range of from 10-60 (v/v) % for different monomers and crosslinkers and different chamber sizes.

Pulsed plasma polymerisation of perfluoro alkyl acrylate in the presence of divinyl adipate (DVA) was carried out in a 125 litre chamber. Figure 4 shows the effect of adding different liquid volume percentages of DVA crosslinker to perfluorooctyl acrylate on water contact angle on a silicon wafer substrate.

Example 4

Example of AFM measurement of monomer/crosslinker co-polymer coating

A barrier-style coating was prepared in a 22 litre volume chamber using PFAC6 with 10% DVA. The deposition process consisted of a 1 minute CW step with a CW power to flow ratio of 3.9 (W/microliter/min), and a pulsed step with a PW power/monomer flow ratio of 0.28 (W/microlitres/min). Figure 6 shows the representative topographical and phase contrast images were obtained from all the samples. High spatial resolution images show mainly the structures of areas between the raised features. The RMS roughness of the coating is 1.65 nm and the $\Delta z/d = 0.05$. As the $\Delta z/d$ value is < 0.15, then this indicates that the physical layer is substantially pinhole free.

Example 5

Example of FTIR/ATR measurement of monomer/crosslinker co-polymer coating

Two barrier-style coatings were prepared in a 22 litre volume chamber using PFAC6 only and PFAC6 with 10% DVA as described in example 4. The deposition process consisted of a 1 minute CW step with a CW power to flow ratio of 3.9 (W/microliter/min), and a pulsed step with a PW power/monomer flow ratio of 0.28 (W/microlitres/min). Figure 7 shows the representative FTIR/ATR spectrum obtained from both samples.

The FTIR/ATR intensity ratios of peaks attributed to stretching mode of CF₃ and C=O groups, CF₃/C=O, of the coating is indicative of sufficient cross linking in the coating to form a physical barrier. CF₃ refers to the terminal groups in the side chain of PFAC6.

Formation of the barrier coatings is believed to be caused by a mixture of cross linking and controlled fragmentation of the monomers during polymerisation. Cross linking is believed to be predominantly via the CF2-CF3 chain, whilst fragmentation is thought to be predominantly through loss of the C=O group during polymerisation and to a lesser extent shortening of the CF2 chain. Cross linking effects the abundance of –CF3 groups in the coating and controlled fragmentation controls the amount of C=O groups in the coating. The ratio of these two functional groups is an indication that sufficient cross-linking and fragmentation has taken place and can be measured by the ratio of the intensities of the corresponding FTIR/ATR peaks.

It has been shown that coatings with reduced CF₃ / C=O ratios give higher resistance values in the electrical test, as described in Example 2, showing an improved coating performance on increased cross linking (for CF₃) and fragmentation (for C=O). If the FTIR/ATR intensity ratio of the peaks attributed to -CF₃ stretching and C=O stretching, CF₃/C=O, is less than 0.6e^{-0.1n} (where n=6 for PFAC6) the resistance of the coating at the electrical test is expected to be higher than 8MOhms.

The peak intensity ratio CF₃/C=O the coating thickness and the final current readings during the electrical test are shown in table 2:

	Thickness (nm)	Resistance in water at 8V	CF ₃ /C=O ATR peak intensity ratio
PFAC6	1480	2.1x10 ⁷	0.23
10%DVA	nd		
(in volume)	1450	9.2x10 ⁹	0.11

Table 2

Claims:

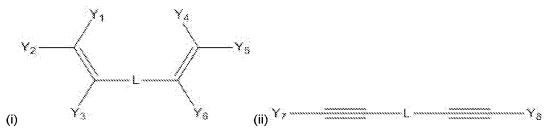
1. An electronic or electrical device or component thereof comprising a protective cross-linked polymeric coating on a surface of the electronic or electrical device or component thereof; wherein the protective cross-linked polymeric coating is obtainable by exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow formation of the protective cross-linked polymeric coating on a surface thereof,

wherein the monomer compound has the following formula:

where R₁, R₂ and R₄ are each independently selected from hydrogen, optionally substituted branched or straight chain C₁-C₅ alkyl or halo alkyl or aryl optionally substituted by halo, and R₃ is selected from:

where each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C_1 - C_6 alkyl, halo alkyl or aryl optionally substituted by halo; and n_1 is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moieties and has a boiling point of less than 500 °C at standard pressure.

- 2. An electronic or electrical device or component thereof according to claim 1, wherein the protective cross-linked polymeric coating is a physical barrier to mass and electron transport.
- 3. An electronic or electrical device or component thereof according to claim 1 or claim 2, wherein the protective cross-linked polymeric coating forms a liquid repellent surface defined by a static water contact angle (WCA) of at least 90°.
- 4. An electronic or electrical device or component thereof according to any preceding claim, wherein the crosslinking reagent has one of the following structures:



where Y_1 , Y_2 , Y_3 , Y_4 , Y_5 , Y_6 , Y_7 and Y_8 are each independently selected from hydrogen, optionally substituted cyclic, branched or straight chain C_1 - C_6 alkyl or aryl; and L is a linker moiety.

5. An electronic or electrical device or component thereof according to claim 4, wherein for compound (i) L has the following formula:

each Y_9 is independently selected from, a bond, -O-, -O-C(O)-, -C(O)-O-, -Y₁₁-O-C(O)-, -C(O)-O-Y₁₁-, -OY₁₁-, and -Y₁₁O-, where Y₁₁ is an optionally substituted cyclic, branched or straight chain C₁-C₈ alkylene; and

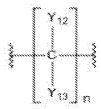
 Y_{10} is selected from optionally substituted cyclic, branched or straight chain C_1 - C_8 alkylene and a siloxane group.

6. An electronic or electrical device or component thereof according to claim 5, wherein for compound (i) L has one of the following structures:

7. An electronic or electrical device or component thereof according to claim 5, wherein for compound (i) L has one of the following structures:

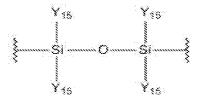


8. An electronic or electrical device or component thereof according to any of claims 5 to 7, wherein Y_{10} has the following formula:



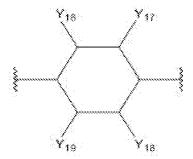
wherein each Y_{12} and Y_{13} is independently selected from H, halo, optionally substituted cyclic, branched or straight chain alkyl, or $-OY_{14}$, where Y_{14} is selected from optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl, and n is an integer from 1 to 10.

- 9. An electronic or electrical device or component thereof according to claim 8, wherein each Y₁₂ is H and each Y₁₃ is H.
- 10. An electronic or electrical device or component thereof according to claim 8, wherein each Y_{12} is fluoro and each Y_{13} is fluoro.
- 11. An electronic or electrical device or component thereof according to any of claims 8 to 10, wherein n is from 4 to 6.
- 12. An electronic or electrical device or component thereof according to any of claims 5 to 7, wherein Y_{10} has the following formula:



Wherein each Y_{15} is independently selected from optionally substituted branched or straight chain C_1 - C_6 alkyl.

- 13. An electronic or electrical device or component thereof according to claim 12, wherein each Y_{15} is methyl, and each Y_{9} is a bond.
- 14. An electronic or electrical device or component thereof according to any of claims 5 to 7, wherein Y_{10} has the following formula:



wherein Y_{16} to Y_{19} are each independently selected from H and optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl.

- 15. An electronic or electrical device or component thereof according to claim 14, wherein Y_{18} is H or vinylene, and Y_{16} , Y_{17} and Y_{19} are each H.
- 16. An electronic or electrical device or component thereof according to any preceding claim wherein the crosslinking reagent is selected from Divinyl Adipate (DVA), 1,4 Butanediol divinyl ether (BDVE), 1,4 Cyclohexanedimethanol divinyl ether (CDDE), 1,7 -Octadiene (17OD), 1,2,4 Trivinylcyclohexane (TVCH), 1,3-Divinyltetramethyldisiloxane (DVTMDS), Diallyl 1,4-cyclohexanedicarboxylate (DCHD), 1,6-Divinylperfluorohexane (DVPFH), 1H,1H,6H,6H-Perfluorohexanediol diacrylate (PFHDA) and glyoxal bis (diallyl acetal) (GBDA).
- 17. An electronic or electrical device or component thereof according to claim 4, wherein for compound (ii), L is selected from a branched or straight chain C₁-C₈ alkylene or an ether group.
- 18. An electronic or electrical device or component thereof according to any preceding claim wherein the monomer is a compound of formula I(a):

l(a)

wherein each of R₁, R₂, R₄, and R₅ to R₁₀ is independently selected from hydrogen or an optionally substituted C₁-C₆ branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1;

; or

wherein the monomer is a compound of formula I(b):

l(b)

$$R_{2} \longrightarrow \left\{ \begin{array}{c} R_{1} & C & R_{5} & R_{7} & R_{8} & X & X \\ C & C & C & C & C \\ R_{5} & R_{6} & R_{10} & X & C \\ \end{array} \right\}_{0} C \times 3_{10}$$

wherein each of R₁, R₂, R₄, and R₅ to R₁₀ is independently selected from hydrogen or an optionally substituted C₁-C₆ branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1.

- An electronic or electrical device or component thereof according to claim 18, wherein the halogen is fluorine.
- 20. An electronic or electrical device or component thereof according to claim 18 or claim 19, wherein each of R₁, R₂, R₄, and R₅ to R₁₀ is independently selected from hydrogen, methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, tert-butyl, sec-butyl, n-pentyl, neopentyl, n-hexyl, isohexyl, and 3-methylpentyl.
- 21. An electronic or electrical device or component thereof according to claim 20, wherein each of R₁, R₂, R₄, and R₅ to R₁₀ is independently selected from hydrogen or methyl.
- 22. An electronic or electrical device or component thereof according to any of claims 18 to 20, wherein a and c are each independently 0 or 1; and b is from 3 to 7.
- 23. An electronic or electrical device or component thereof according to any preceding claim wherein each X is H.
- 24. An electronic or electrical device or component thereof according to any of claims 1 to 22, wherein each X is F.
- 25. An electronic or electrical device or component thereof according to any of claims 18 to 24, wherein R_1 and R_2 are both hydrogen.
- 26. An electronic or electrical device or component thereof according to any of claims 18 to 25, wherein R₄ is hydrogen or methyl.
- 27. An electronic or electrical device or component thereof according to any of claims 18 to 26, wherein R₉ is hydrogen and R₁₀ is branched or straight chain C₁-C₆ alkyl group.
- 28. An electronic or electrical device or component thereof according to claim 27, wherein R_{10} is methyl.
- 29. An electronic or electrical device or component thereof according to any of claims 18 to 28, wherein each of R₅ to R₅ is hydrogen.
- 30. An electronic or electrical device or component thereof according to any of claims 18 to 29 wherein each of each of R₁, R₂, R₄, and R₅ to R₁₀ is hydrogen, each X is H, a=0 and c=0.
- 31. An electronic or electrical device or component thereof according to any of claims 24 to 30, wherein the compound of formula I(a) has the following formula:

where n is from 2 to 10.

32. An electronic or electrical device or component thereof according to any of claims 24 to 30, wherein the compound of formula I(a) has the following formula:

$$\begin{array}{c}
H \\
CH_3
\end{array}$$
(CF₂)_nF

where n is from 2 to 10.

- 33. An electronic or electrical device or component thereof according to claim 31, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-perfluorohexyl acrylate (PFAC4), 1H,1H,2H,2H-perfluorooctyl acrylate (PFAC6), 1H,1H,2H,2H-perfluorodecyl acrylate (PFAC8) and 1H,1H,2H,2H-perfluorododecyl acrylate (PFAC10).
- 34. An electronic or electrical device or component thereof according to claim 32, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-pefluorohexyl methacrylate (PFMAC4), 1H,1H,2H,2H-perfluorooctyl methacrylate (PFMAC6) and 1H,1H,2H,2H-perfluorodecyl methacrylate (PFMAC8).
- 35. An electronic or electrical device or component thereof according to any of claims 18 to 23, wherein the compound of formula I(a) has the following formula:

wherein a and c are each independently 0 or 1, b =3-7 and n is 4 to 10, where n=a+b+c+1.

36. An electronic or electrical device or component thereof according to any of claims 18 to 23, wherein the compound of formula I(a) has the following formula:

$$H \longrightarrow CH_3$$
 (CH₂)_nH

where n is 2 to 12.

- 37. An electronic or electrical device or component thereof according to claim 35 or claim 36, wherein the compound of formula I(a) is selected from ethyl hexyl acrylate, hexyl acrylate, decyl acrylate, lauryl dodecyl acrylate and iso decyl acrylate.
- 38. An electronic or electrical device or component thereof according to any of claims 18 to 23, wherein the monomer is a compound of formula I(b) and has the following formula:

where n is from 4 to 14, optionally where R₁, R₂ and R₃ are each H.

- 39. An electronic or electrical device or component thereof according to any of the preceding claims, wherein the electronic or electrical device or component thereof is selected from mobile phones, smartphones, pagers, radios, sound and audio systems such as loudspeakers, microphones, ringers and/or buzzers, hearing aids, personal audio equipment such as personal CD, tape cassette or MP3 players, televisions, DVD players including portable DVD players, video recorders, digi and other set-top boxes, computers and related components such as laptop, notebook, tablet, phablet or palmtop computers, personal digital assistants (PDAs), keyboards, or instrumentation, games consoles, data storage devices, outdoor lighting systems, radio antennae and other forms of communication equipment, and printed circuit boards.
- 40. A method for treating an electronic or electrical device or component thereof comprising:

exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow a protective cross-linked polymeric coating to form on a surface of the electronic or electrical device or component thereof;

wherein the monomer compound has the following formula:

where R₁, R₂ and R₄ are each independently selected from hydrogen, optionally substituted branched or straight chain C₁-C₆ alkyl or halo alkyl or aryl optionally substituted by halo, and R₃ is selected from:

$$\begin{array}{c|c} & & & & \\ & &$$

where each X is independently selected from hydrogen, optionally substituted branched or straight chain C₁-C₆ alkyl, halo alkyl or aryl optionally substituted by halo; and n₁ is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moieties and has a boiling point at standard pressure of less than 500°C.

41. A method according to claim 40, wherein the crosslinking reagent has one of the following structures:

where Y_1 , Y_2 , Y_3 , Y_4 , Y_5 , Y_6 , Y_7 and Y_8 are each independently selected from hydrogen, optionally substituted cyclic, branched or straight chain C_1 - C_6 alkyl or aryl; and L is a linker moiety.

42. A method according to claim 41, wherein for compound (i) L has the following formula:

each Y_9 is independently selected from, a bond, -O-, -O-C(O)-, -C(O)-O-, -Y₁₁-O-C(O)-, -C(O)-O-Y₁₁-, -OY₁₁-, and -Y₁₁O-, where Y₁₁ is an optionally substituted cyclic, branched or straight chain C₁-C₈ alkylene; and

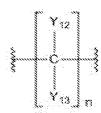
 Y_{10} is selected from optionally substituted cyclic, branched or straight chain C_1 - C_8 alkylene and a siloxane group.

43. A method according to claim 42, wherein for compound (i) L has one of the following structures:

44. A method according to claim 42, wherein for compound (i) L has one of the following structures:

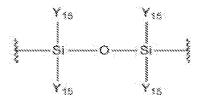


45. An method according to any of claims 42 to 44, wherein Y₁₀ has the following formula:



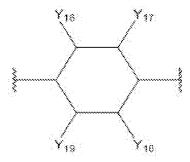
wherein each Y_{12} and Y_{13} is independently selected from H, halo, optionally substituted cyclic, branched or straight chain alkyl, or $-OY_{14}$, where Y_{14} is selected from optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl, and n is an integer from 1 to 10.

- 46. A method according to claim 45, wherein each Y₁₂ is H and each Y₁₃ is H.
- 47. A method according to claim 45, wherein each Y₁₂ is fluoro and each Y₁₃ is fluoro.
- 48. A method according to any of claims 45 to 47, wherein n is from 4 to 6.
- 49. A method according to any of claims 42 to 44, wherein Y₁₀ has the following formula:



Wherein each Y_{15} is independently selected from optionally substituted branched or straight chain C_1 - C_6 alkyl.

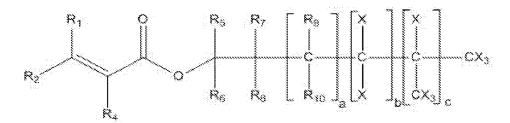
- 50. A method according to claim 49, wherein each Y₁₅ is methyl, and each Y₉ is a bond.
- 51. A method according to any of claims 42 to 44, wherein Y₁₀ has the following formula:



wherein Y_{16} to Y_{19} are each independently selected from H and optionally substituted branched or straight chain C_1 - C_8 alkyl or alkenyl.

- 52. An method according to claim 51, wherein Y_{18} is vinylene or H, and Y_{16} , Y_{17} and Y_{19} are each H.
- 53. A method according to any of claims 40 to 52, wherein the crosslinking reagent is selected from Divinyl Adipate (DVA), 1,4 Butanediol divinyl ether (BDVE), 1,4 Cyclohexanedimethanol divinyl ether (CDDE), 1,7 -Octadiene (17OD), 1,2,4 -Trivinylcyclohexane (TVCH), 1,3-Divinyltetramethyldisiloxane (DVTMDS), Diallyl 1,4-cyclohexanedicarboxylate (DCHD), 1,6-Divinylperfluorohexane (DVPFH), 1H,1H,6H,6H-Perfluorohexanediol diacrylate (PFHDA) and glyoxal bis (diallyl acetal) (GBDA).
- 54. A method according to claim 41, wherein for compound (ii), L is selected from a branched or straight chain C₁-C₈ alkylene or an ether group.
- 55. A method according to any of claims 40 to 54 wherein the monomer is a compound of formula I(a):

I(a)



wherein each of R₁, R₂, R₄, and R₅ to R₁₀ is independently selected from hydrogen or an optionally substituted C₁-C₆ branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1;

; or

wherein the monomer is a compound of formula I(b):

I(b)

$$R_{2} \longrightarrow \left\{ \begin{array}{c} R_{5} & R_{7} & \left[\begin{array}{c} R_{5} & \left[\begin{array}{c} X \\ X \end{array} \right] \left$$

wherein each of R₁, R₂, R₄, and R₅ to R₁₀ is independently selected from hydrogen or an optionally substituted C₁-C₆ branched or straight chain alkyl group; each X is independently selected from hydrogen or halogen; a is from 0-10; b is from 2 to 14; and c is 0 or 1.

- 56. A method according to claim 55, wherein the halogen is fluorine.
- 57. A method according to claim 55 or claim 56, wherein each of R₁, R₂, R₄, and R₅ to R₁₀ is independently selected from hydrogen, methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, tert-butyl, sec-butyl, n-pentyl, neopentyl, n-hexyl, isohexyl, and 3-methylpentyl.
- 58. A method according to claim 57, wherein each of R₁, R₂, R₄, and R₅ to R₁₀ is independently selected from hydrogen or methyl.
- 59. A method according to any of claims 55 to 58, wherein a and c are each independently 0 or 1; and b is from 3 to 7.
- 60. A method according to any preceding claim wherein each X is H.
- 61. A method according to any of claims 41 to 59, wherein each X is F.
- 62. A method according to any of claims 41 to 61, wherein R_1 and R_2 are both hydrogen.
- 63. A method according to any of claims 41 to 62, wherein R₄ is hydrogen or methyl.
- 64. A method according to any of claims 41 to 63, wherein R₉ is hydrogen and R₁₀ is branched or straight chain C₁-C₆ alkyl group.
- 65. A method according to claim 64, wherein R_{10} is methyl.
- 66. A method to any of claims 41 to 65, wherein each of R_5 to R_8 is hydrogen.
- 67. A method according to any of claims 41 to 66 wherein each of each of R_1 , R_2 , R_4 , and R_5 to R_{10} is hydrogen, each X is H, a=0 and c=0.
- 68. A method according to any of claims 61 to 67, wherein the compound of formula I(a) has the following formula:

$$\begin{array}{c}
H \\
\longrightarrow \\
H
\end{array}$$

$$\begin{array}{c}
O \\
\longrightarrow \\
O
\end{array}$$

$$(CF_2)_n F$$

where n is from 2 to 10.

69. A method according to any of claims 61 to 67, wherein the compound of formula I(a) has the following formula:

$$H \longrightarrow CH_3$$
 (CF₂)_nF

where n is from 2 to 10.

- 70. A method according to claim 68, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-perfluorohexyl acrylate (PFAC4), 1H,1H,2H,2H-perfluorooctyl acrylate (PFAC6), 1H,1H,2H,2H-perfluorodecyl acrylate (PFAC8) and 1H,1H,2H,2H-perfluorododecyl acrylate (PFAC10).
- 71. A method according to claim 69, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-perfluorohexyl methacrylate (PFMAC4), 1H,1H,2H,2H-perfluorooctyl methacrylate (PFMAC6) and 1H,1H,2H,2H-perfluorodecyl methacrylate (PFMAC8).
- 72. A method according to any of claims 41 to 61, wherein the compound of formula I(a) has the following formula:

wherein a and c are each independently 0 or 1, b =3-7 and n is 4 to 10, where n=a+b+c+1.

73. A method according to any of claims 41 to 61, wherein the compound of formula I(a) has the following formula:

$$\begin{array}{c}
H \\
CH_3
\end{array}$$
(CH₂)_nH

where n is 2 to 12.

- 74. A method according to claim 72 or claim 73, wherein the compound of formula I(a) is selected from ethyl hexyl acrylate, hexyl acrylate, decyl acrylate, lauryl dodecyl acrylate and iso decyl acrylate.
- 75. A method according to any of claims 41 to 60, wherein the monomer is a compound of formula I(b) and has the following formula:

where n is from 4 to 14.

76. A method according to claim 75 wherein the monomer is a compound of formula I(b) and has the following formula:

where n is from 4 to 14.

- 77. A method according to any of claims 40 to 76 wherein the electronic or electrical device or component thereof is placed in a plasma deposition chamber, a glow discharge is ignited within said chamber, and a voltage applied as a pulsed field.
- 78. A method according to claim 77 wherein applied voltage is at a power of from 40 to 500W.
- 79. A method according to claim 77 or claim 78 wherein the voltage is pulsed in a sequence in which the ratio of the time on : time off is in the range of from 1:500 to 1:1500.
- 80. A method according to any of claims 77 to 79 wherein the voltage is pulsed in a sequence where power is on for 20-50μs, and off for from 1000μs to 30000μs.
- 81. A method according to any of claims 77 to 80 wherein the voltage is applied as a pulsed field for a period of from 30 seconds to 90 minutes.
- 82. A method according to claim 81 wherein the voltage is applied as a pulsed field for from 5 to 60 minutes.
- 83. A method according to any one of claims 77 to 82, wherein in a preliminary step, a continuous power plasma is applied to the electronic or electrical device or component thereof.

- 84. A method according to claim 83, wherein the preliminary step is conducted in the presence of an inert gas.
- 85. A method according to any one of claims 77 to 84, wherein the monomer compound and/or the crosslinking reagent in gaseous form is fed into the plasma at a rate of from 80-300 mg/minute, whilst the pulsed voltage is applied.
- 86. A method according to any one of claims 77 to 85, wherein the plasma is created with a voltage at an average power of from 0.001 to 500w/m³.
- 87. A method according to any of claims 40 to 86, wherein the monomer compound is miscible with the crosslinking reagent.
- 88. A method according to any of claims 40 to 86, wherein the monomer compound is immiscible with the crosslinking reagent.
- 89. A method according to claim 87, wherein the step of exposing the electronic or electrical device or component thereof to a plasma occurs in a plasma chamber; and the monomer compound and crosslinking reagent are mixed prior to entering the chamber.
- 90. A method according to claim 87 or claim 88 wherein the step of exposing the electronic or electrical device or component thereof to a plasma occurs in a plasma chamber; and the monomer compound and crosslinking reagent are introduced separately into the chamber.
- 91. A method according to any of claims 40 to 90, wherein the crosslinking reagent is present in an amount of from 10 to 60 (v/v) % of the total volume of monomer compound and crosslinking reagent.

Abstract

The present invention provides an electronic or electrical device or component thereof comprising a cross-linked polymeric coating on a surface of the electronic or electrical device or component thereof; wherein the cross-linked polymeric coating is obtainable by exposing the electronic or electrical device or component thereof to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow formation of the cross-linked polymeric coating on a surface thereof,

wherein the monomer compound has the following formula:

where R₁, R₂ and R₄ are each independently selected from hydrogen, optionally substituted branched or straight chain C₁-C₆ alkyl or halo alkyl or aryl optionally substituted by halo, and R₃ is selected from:

$$\begin{cases} X \\ C \\ X \end{cases} = \begin{bmatrix} X \\ C \\ X \end{bmatrix}_{\Gamma_1} = \begin{bmatrix} X \\ C \\ X \end{bmatrix}_{\Gamma_1}$$

where each X is independently selected from hydrogen, a halogen, optionally substituted branched or straight chain C_1 - C_6 alkyl, halo alkyl or aryl optionally substituted by halo; and n_1 is an integer from 1 to 27; and wherein the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moieties and has a boiling point at standard pressure of less than 500°C.

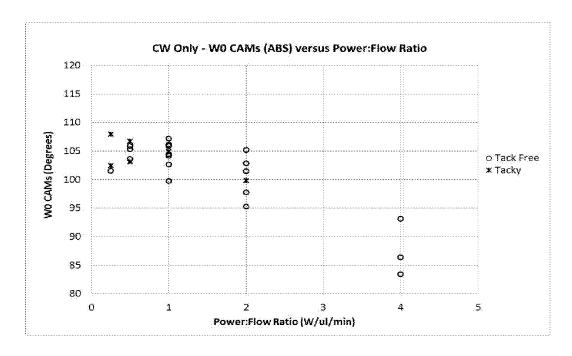


Figure 1

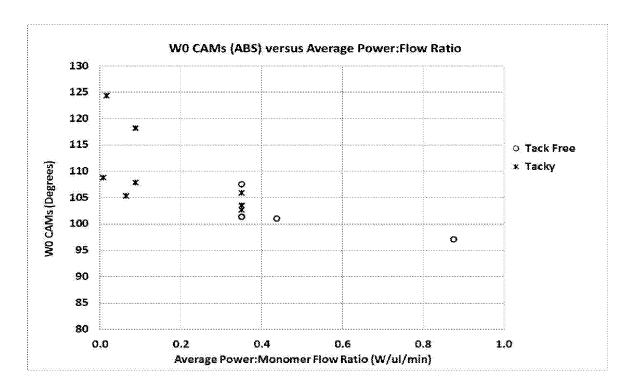


Figure 2

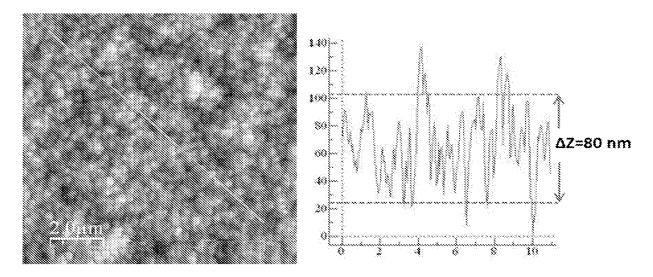


Figure 3

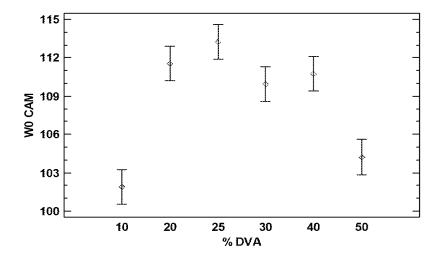


Figure 4

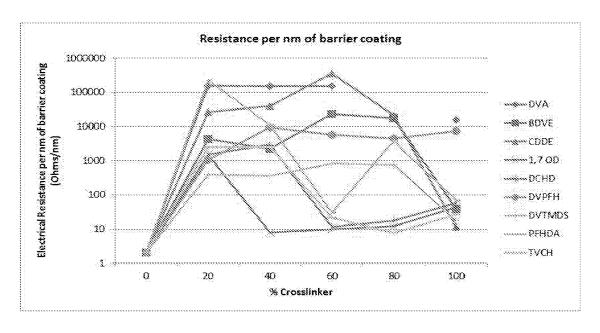


Figure 5

Figure 6

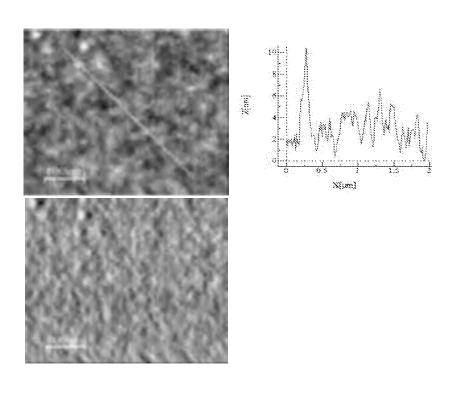
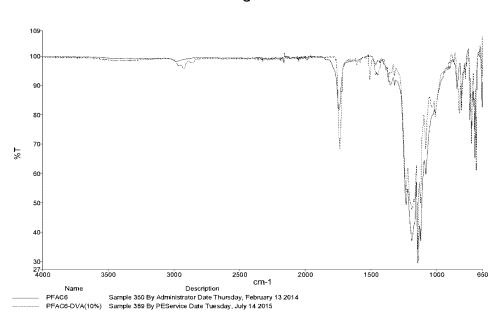


Figure 7



Electronic Patent Application Fee Transmittal					
Application Number:	157	15735130			
Filing Date:	08-	08-Dec-2017			
Title of Invention:	COATINGS				
First Named Inventor/Applicant Name:	Ste	phen Richard COUI	SON		
Filer:	Paul John Parins/Sonja Talley				
Attorney Docket Number:	GB	L0007/US			
Filed as Large Entity					
Filing Fees for U.S. National Stage under 35 USC 371					
Description		Fee Code	Quantity	Amount	Sub-Total in USD(\$)
Basic Filing:					
Pages:					
Claims:					
Miscellaneous-Filing:					
Petition:					
Patent-Appeals-and-Interference:					
Post-Allowance-and-Post-Issuance:					
Extension-of-Time:					

Description	Fee Code	Quantity	Amount	Sub-Total in USD(\$)
Extension - 1 month with \$0 paid	1251	1	200	200
Miscellaneous:				
	Tot	al in USD	(\$)	200

Electronic Acknowledgement Receipt				
EFS ID:	32913023			
Application Number:	15735130			
International Application Number:				
Confirmation Number:	7101			
Title of Invention:	COATINGS			
First Named Inventor/Applicant Name:	Stephen Richard COULSON			
Customer Number:	33072			
Filer:	Paul John Parins/Sonja Talley			
Filer Authorized By:	Paul John Parins			
Attorney Docket Number:	GBL0007/US			
Receipt Date:	15-JUN-2018			
Filing Date:	08-DEC-2017			
Time Stamp:	15:25:26			
Application Type:	U.S. National Stage under 35 USC 371			

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Payment Type	CARD
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RAM confirmation Number	061818INTEFSW15261300
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			104817		
1	Applicant Response to Pre-Exam Formalities Notice	2018-06-15- GBL0007US_Response_to_Noti ce_of_Innsufficiency.pdf	dd5a3e5142000676238de063b0dad3695a 92e9ee	no	2
Warnings:			<u>l</u>	1	
Information:					
		2018-06-15-	217014		
2		GBL0007US_Preliminary_Amen dment.pdf	8f754c0913495099afe3326be0b93545b78 83330	yes	10
	Multip	part Description/PDF files in .:	zip description	1	
	Document De	scription	Start	End	
	Preliminary Am	endment	1	1	
	Claims		2	8	
	Applicant Arguments/Remarks	Made in an Amendment	9	1	0
Warnings:					
Information:					
			1601808		
3	Other Reference-Patent/App/Search documents	PCT_GB2016_051688_As_Filed. pdf	580269b42951d2147958ac6f7f695c98d19 51127	no	42
Warnings:			I		
Information:					
			30573		
4	Fee Worksheet (SB06)	fee-info.pdf	63f6887f3ce37e3247d3a6e70d7b0e44600 7f731	no	2
Warnings:		-	l	1	
Information:					
		Total Files Size (in bytes):	10	54212	

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If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

In The United States Patent And Trademark Office

Inventor: Stephen Richard COULSON Examiner: To Be Assigned

Serial No. 15/735,130 Group Art Unit: 1762

International Filing Date: June 8, 2016 Confirmation No. 7101

International App. No. PCT/GB2016/051688 Docket No. GBL0007/US

371(c): December 8, 2017

For: COATINGS

Commissioner for Patents P.O. Box 1450 Alexandria, Virginia 22313-1450 I certify that on <u>06/15/2018</u>, this correspondence is being electronically transmitted to the United States Patent and Trademark office via the Office's EFS-Web

/Sonja A. Talley/ Sonja A. Talley

RESPONSE TO NOTIFICATION OF INSUFFICIENCY UNDER 37 CFR 1.492 and/or 1.495 (DO/EO/US)

Dear Sir or Madam:

This communication is responsive to the Notification of Insufficiency under 37 CFR 1.492 and/or 1.495 (DO/EO/US), which was mailed on March 16, 2018 in the above captioned application.

The two-month shortened-statutory period for response expired on May 16, 2018. Applicant respectfully petitions the Commissioner for Patents to extend the time for response for one month, from May 16, 2018 to June 16, 2018. It is respectfully submitted that this response is timely filed within the extended deadline. If any additional extension period is required in order for this paper to be timely filed, Applicant hereby requests an extension for such additional time period and authorizes the appropriate fees therefore to be charged to the Kagan Binder Deposit Account No. 50-1775 and to notify us of the same.

Payment in the amount of \$200.00 for a one-month extension of time is being provided by the electronic payment system of the EFS. In the event of failure of this payment by the electronic payment system of the EFS for any reason, the Commissioner is authorized to charge Kagan Binder Deposit Account No. 50-1775 and thereafter is requested to notify us of the same.

Remarks

According to the Notification, fees are due for multiple dependent claims and 70 total claims over 20.

Enclosed herewith is a Preliminary Amendment canceling claims so that the pending claims include one independent claim and 19 total claims. The claims are also amended to remove multiple dependencies.

Accordingly, no additional claims fees are due.

Respectfully Submitted,

Date: June 15, 2018 By: /Paul J. Parins/

Paul J. Parins, Reg. No. 54,358 **Customer Number 33072** Phone: 651-275-9831

Fax: 651-351-2954

#119890



United States Patent and Trademark Office

INITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS P.O. Sox 1450 Alexandria, Virginia 22313-1450 www.uspto.gov

APPLICATION NUMBER

FILING OR 371(C) DATE

FIRST NAMED APPLICANT

ATTY. DOCKET NO./TITLE

15/735,130

12/08/2017

Stephen Richard COULSON

GBL0007/US

CONFIRMATION NO. 7101 PUBLICATION NOTICE

33072 KAGAN BINDER, PLLC SUITE 200, MAPLE ISLAND BUILDING 221 MAIN STREET NORTH STILLWATER, MN 55082

Title:COATINGS

Publication No.US-2018-0171171-A1 Publication Date: 06/21/2018

NOTICE OF PUBLICATION OF APPLICATION

The above-identified application will be electronically published as a patent application publication pursuant to 37 CFR 1.211, et seg. The patent application publication number and publication date are set forth above.

The publication may be accessed through the USPTO's publically available Searchable Databases via the Internet at www.uspto.gov. The direct link to access the publication is currently http://www.uspto.gov/patft/.

The publication process established by the Office does not provide for mailing a copy of the publication to applicant. A copy of the publication may be obtained from the Office upon payment of the appropriate fee set forth in 37 CFR 1.19(a)(1). Orders for copies of patent application publications are handled by the USPTO's Public Records Division. The Public Records Division can be reached by telephone at (571) 272-3150 or (800) 972-6382, by facsimile at (571) 273-3250, by mail addressed to the United States Patent and Trademark Office, Public Records Division, Alexandria, VA 22313-1450 or via the Internet.

In addition, information on the status of the application, including the mailing date of Office actions and the dates of receipt of correspondence filed in the Office, may also be accessed via the Internet through the Patent Electronic Business Center at www.uspto.gov using the public side of the Patent Application Information and Retrieval (PAIR) system. The direct link to access this status information is currently https://portal.uspto.gov/pair/PublicPair. Prior to publication, such status information is confidential and may only be obtained by applicant using the private side of PAIR.

Further assistance in electronically accessing the publication, or about PAIR, is available by calling the Patent Electronic Business Center at 1-866-217-9197.

Office of Data Managment, Application Assistance Unit (571) 272-4000, or (571) 272-4200, or 1-888-786-0101

	PATI	ENT APPLI		N FEE DE titute for Form		TION RECOR	D		tion or Docket Num 5,130	ber
	APPI	_ICATION A			umn 2)	SMALL	ENTITY	OR	OTHER SMALL	
	FOR	NUMBE	R FILE	NUMBE	R EXTRA	RATE(\$)	FEE(\$)	1	RATE(\$)	FEE(\$)
	IC FEE FR 1.16(a), (b), or (c))	N	/A		N/A			1	N/A	300
	RCH FEE FR 1.16(k), (i), or (m))	N	N/A		N/A			1	N/A	520
	MINATION FEE FR 1.16(o), (p), or (q))	N	/A	١	√A	N/A		N/A	760	
	AL CLAIMS FR 1.16(i))	18	minus	20= *				OR	x 100 =	0.00
	PENDENT CLAIN FR 1.16(h))	^{/S} 1	minus	3 = *				1	x 460 =	0.00
FEE	PLICATION SIZE E CFR 1.16(s))	\$310 (\$15 50 sheets	paper, the 5 for sma or fraction	and drawings e e application si all entity) for ea on thereof. See CFR 1.16(s).	ze fee due is ch additional					0.00
MUL	TIPLE DEPENDE	NT CLAIM PRE	SENT (37	7 CFR 1.16(j))						0.00
* If th	ne difference in co	lumn 1 is less th	an zero,	enter "0" in colur	nn 2.	TOTAL		1	TOTAL	1580
NT A		(Column 1) CLAIMS REMAINING AFTER AMENDMENT		(Column 2) HIGHEST NUMBER PREVIOUSLY PAID FOR	(Column 3) PRESENT EXTRA	SMALL RATE(\$)	ADDITIONAL FEE(\$)	OR	SMALL RATE(\$)	ADDITIONAL FEE(\$)
₩	Total (37 CFR 1.16(i))	*	Minus	**	=	x =		OR	x =	
AMENDMENT	Independent (37 CFR 1.16(h))	*	Minus	***	=	x =		OR	x =	
₹	Application Size Fe	e (37 CFR 1.16(s))								
	FIRST PRESENTA	TION OF MULTIPI	E DEPEN	DENT CLAIM (37 C	FR 1.16(j))			OR		
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	Independent (37 CFR 1.16(h))	•	Minus	***	=	x =		OR	x =	
₹	Application Size Fe	e (37 CFR 1.16(s))						OR		
	FIRST PRESENTA	TION OF MULTIPI	E DEPEN	DENT CLAIM (37 C	CFR 1.16(j))	TOTAL		OR	TOTAL	
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UNITED STATES PATENT AND TRADEMARK OFFICE

UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS P.O. Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov

FILING or APPLICATION GRP ART FIL FEE REC'D ATTY.DOCKET.NO IND CLAIMS NUMBER 371(c) DATE UNIT TOT CLAIMS 15/735,130 12/08/2017 1762 1620 **GBL0007/US** 18

33072 KAGAN BINDER, PLLC SUITE 200, MAPLE ISLAND BUILDING 221 MAIN STREET NORTH STILLWATER, MN 55082 CONFIRMATION NO. 7101 UPDATED FILING RECEIPT



Date Mailed: 07/03/2018

Receipt is acknowledged of this non-provisional patent application. The application will be taken up for examination in due course. Applicant will be notified as to the results of the examination. Any correspondence concerning the application must include the following identification information: the U.S. APPLICATION NUMBER, FILING DATE, NAME OF APPLICANT, and TITLE OF INVENTION. Fees transmitted by check or draft are subject to collection. Please verify the accuracy of the data presented on this receipt. If an error is noted on this Filing Receipt, please submit a written request for a Filing Receipt Correction. Please provide a copy of this Filing Receipt with the changes noted thereon. If you received a "Notice to File Missing Parts" for this application, please submit any corrections to this Filing Receipt with your reply to the Notice. When the USPTO processes the reply to the Notice, the USPTO will generate another Filing Receipt incorporating the requested corrections

Inventor(s)

Stephen Richard COULSON, Abingdon, UNITED KINGDOM;

Delwyn EVANS, Abingdon, UNITED KINGDOM; Angeliki SIOKOU, Abingdon, UNITED KINGDOM; Clive TELFORD, Abingdon, UNITED KINGDOM;

Applicant(s)

P2i Ltd, Abingdon, UNITED KINGDOM

Power of Attorney: The patent practitioners associated with Customer Number 33072

Domestic Priority data as claimed by applicant

This application is a 371 of PCT/GB2016/051688 06/08/2016

Foreign Applications (You may be eligible to benefit from the **Patent Prosecution Highway** program at the USPTO. Please see http://www.uspto.gov for more information.) EUROPEAN PATENT OFFICE (EPO) 15386017.6 06/09/2015

Permission to Access Application via Priority Document Exchange: Yes

Permission to Access Search Results: Yes

Applicant may provide or rescind an authorization for access using Form PTO/SB/39 or Form PTO/SB/69 as appropriate.

If Required, Foreign Filing License Granted: 03/14/2018

page 1 of 3

The country code and number of your priority application, to be used for filing abroad under the Paris Convention, is **US 15/735,130**

Projected Publication Date: Not Applicable

Non-Publication Request: No

Early Publication Request: No

Title

COATINGS

Preliminary Class

526

Statement under 37 CFR 1.55 or 1.78 for AIA (First Inventor to File) Transition Applications: No

PROTECTING YOUR INVENTION OUTSIDE THE UNITED STATES

Since the rights granted by a U.S. patent extend only throughout the territory of the United States and have no effect in a foreign country, an inventor who wishes patent protection in another country must apply for a patent in a specific country or in regional patent offices. Applicants may wish to consider the filing of an international application under the Patent Cooperation Treaty (PCT). An international (PCT) application generally has the same effect as a regular national patent application in each PCT-member country. The PCT process **simplifies** the filing of patent applications on the same invention in member countries, but **does not result** in a grant of "an international patent" and does not eliminate the need of applicants to file additional documents and fees in countries where patent protection is desired.

Almost every country has its own patent law, and a person desiring a patent in a particular country must make an application for patent in that country in accordance with its particular laws. Since the laws of many countries differ in various respects from the patent law of the United States, applicants are advised to seek guidance from specific foreign countries to ensure that patent rights are not lost prematurely.

Applicants also are advised that in the case of inventions made in the United States, the Director of the USPTO must issue a license before applicants can apply for a patent in a foreign country. The filing of a U.S. patent application serves as a request for a foreign filing license. The application's filing receipt contains further information and guidance as to the status of applicant's license for foreign filing.

Applicants may wish to consult the USPTO booklet, "General Information Concerning Patents" (specifically, the section entitled "Treaties and Foreign Patents") for more information on timeframes and deadlines for filing foreign patent applications. The guide is available either by contacting the USPTO Contact Center at 800-786-9199, or it can be viewed on the USPTO website at http://www.uspto.gov/web/offices/pac/doc/general/index.html.

For information on preventing theft of your intellectual property (patents, trademarks and copyrights), you may wish to consult the U.S. Government website, http://www.stopfakes.gov. Part of a Department of Commerce initiative, this website includes self-help "toolkits" giving innovators guidance on how to protect intellectual property in specific countries such as China, Korea and Mexico. For questions regarding patent enforcement issues, applicants may call the U.S. Government hotline at 1-866-999-HALT (1-866-999-4258).

LICENSE FOR FOREIGN FILING UNDER

Title 35, United States Code, Section 184

Title 37, Code of Federal Regulations, 5.11 & 5.15

GRANTED

The applicant has been granted a license under 35 U.S.C. 184, if the phrase "IF REQUIRED, FOREIGN FILING LICENSE GRANTED" followed by a date appears on this form. Such licenses are issued in all applications where the conditions for issuance of a license have been met, regardless of whether or not a license may be required as set forth in 37 CFR 5.15. The scope and limitations of this license are set forth in 37 CFR 5.15(a) unless an earlier license has been issued under 37 CFR 5.15(b). The license is subject to revocation upon written notification. The date indicated is the effective date of the license, unless an earlier license of similar scope has been granted under 37 CFR 5.13 or 5.14.

This license is to be retained by the licensee and may be used at any time on or after the effective date thereof unless it is revoked. This license is automatically transferred to any related applications(s) filed under 37 CFR 1.53(d). This license is not retroactive.

The grant of a license does not in any way lessen the responsibility of a licensee for the security of the subject matter as imposed by any Government contract or the provisions of existing laws relating to espionage and the national security or the export of technical data. Licensees should apprise themselves of current regulations especially with respect to certain countries, of other agencies, particularly the Office of Defense Trade Controls, Department of State (with respect to Arms, Munitions and Implements of War (22 CFR 121-128)); the Bureau of Industry and Security, Department of Commerce (15 CFR parts 730-774); the Office of Foreign AssetsControl, Department of Treasury (31 CFR Parts 500+) and the Department of Energy.

NOT GRANTED

No license under 35 U.S.C. 184 has been granted at this time, if the phrase "IF REQUIRED, FOREIGN FILING LICENSE GRANTED" DOES NOT appear on this form. Applicant may still petition for a license under 37 CFR 5.12, if a license is desired before the expiration of 6 months from the filing date of the application. If 6 months has lapsed from the filing date of this application and the licensee has not received any indication of a secrecy order under 35 U.S.C. 181, the licensee may foreign file the application pursuant to 37 CFR 5.15(b).

SelectUSA

The United States represents the largest, most dynamic marketplace in the world and is an unparalleled location for business investment, innovation, and commercialization of new technologies. The U.S. offers tremendous resources and advantages for those who invest and manufacture goods here. Through SelectUSA, our nation works to promote and facilitate business investment. SelectUSA provides information assistance to the international investor community; serves as an ombudsman for existing and potential investors; advocates on behalf of U.S. cities, states, and regions competing for global investment; and counsels U.S. economic development organizations on investment attraction best practices. To learn more about why the United States is the best country in the world to develop technology, manufacture products, deliver services, and grow your business, visit http://www.SelectUSA.gov or call +1-202-482-6800.



United States Patent and Trademark Office

INITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS P.O. Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov

U.S. APPLICATION NUMBER NO. FIRST NAMED INVENTOR ATTY. DOCKET NO. GBL0007/US

15/735,130 Stephen Richard COULSON

INTERNATIONAL APPLICATION NO.

33072 KAGAN BINDER, PLLC SUITE 200, MAPLE ISLAND BUILDING 221 MAIN STREET NORTH STILLWATER, MN 55082

PCT/GB2016/051688 I.A. FILING DATE PRIORITY DATE 06/08/2016 06/09/2015

> CONFIRMATION NO. 7101 371 ACCEPTANCE LETTER



Date Mailed: 07/03/2018

NOTICE OF ACCEPTANCE OF APPLICATION UNDER 35 U.S.C 371 AND 37 CFR 1.495

The applicant is hereby advised that the United States Patent and Trademark Office, in its capacity as a Designated / Elected Office (37 CFR 1.495), has ACCEPTED the above identified international application for national patentability examination in the United States Patent and Trademark Office.

The United States Application Number assigned to the application is shown above. A Filing Receipt will be issued for the present application in due course. THE DATE APPEARING ON THE FILING RECEIPT AS THE "FILING DATE or 371(c) DATE" IS THE DATE ON WHICH THE LAST OF THE 35 U.S.C. 371 (c)(1) and (c)(2) REQUIREMENTS HAS BEEN RECEIVED IN THE OFFICE. THIS DATE IS SHOWN BELOW. The filing date of the above identified application is the international filing date of the international application (Article 11(3) and 35 U.S.C. 363)

> 12/08/2017 DATE OF RECEIPT OF 35 U.S.C. 371(c)(1) and (c)(2) REQUIREMENTS

The following items have been received:

- Copy of the International Application filed on 12/08/2017
- Copy of the International Search Report filed on 12/08/2017
- Preliminary Amendments filed on 06/15/2018
- Information Disclosure Statements filed on 12/08/2017
- Request for Immediate Examination filed on 12/08/2017
- U.S. Basic National Fees filed on 12/08/2017
- Authorize Access to Search Results filed on 12/08/2017
- Priority Documents filed on 12/08/2017
- Power of Attorney filed on 12/08/2017
- Authorization to Permit Access filed on 12/08/2017
- Application Data Sheet (37 CFR 1.76) filed on 12/08/2017

Applicant is notified that the above-identified application contains the deficiencies noted below. No period for reply is set forth in this notice for correction of these deficiencies. However, if a deficiency relates to the inventor's oath or declaration, the applicant must file an oath or declaration in compliance with 37 CFR 1.63, or a substitute statement in compliance with 37 CFR 1.64, executed by or with respect to each actual inventor no later than the expiration of the time period set in the "Notice of Allowability" to avoid abandonment. See 37 CFR 1.495(c).

page 1 of 2

• Properly executed inventor's oath or declaration for the following inventor(s) has not been submitted: Stephen
Richard COULSON, Delwyn EVANS, Angeliki SIOKOU, and Clive TELFORD

Applicant is reminded that any communications to the United States Patent and Trademark Office must be mailed to the address given in the heading and include the U.S. application no. shown above (37 CFR 1.5)

MAMYE I WAGSTAFF	
Telephone: (571) 272-1057	

United States Patent and Trademark Office



UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS

P.O. Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov

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	APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
	15/735,130	12/08/2017	Stephen Richard COULSON	GBL0007/US	7101
33072 7590 10/18/2018 KAGAN BINDER, PLLC SUITE 200, MAPLE ISLAND BUILDING		8	EXAMINER		
		APLE ISLAND BUILI	NG	BUIE-HATCHER, NICOLE M	
	221 MAIN STF STILLWATER			ART UNIT	PAPER NUMBER
				1767	
				NOTIFICATION DATE	DELIVERY MODE
				10/18/2018	ELECTRONIC

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Notice of the Office communication was sent electronically on above-indicated "Notification Date" to the following e-mail address(es):

docketing@kaganbinder.com

	Application No. 15/735,130	Applicant(s) COULSON et al.		
Office Action Summary	e Action Summary Examiner Art Unit AIA Status		AIA Status	
	NICOLE M BUIE-HATCHER	1767	Yes	
The MAILING DATE of this communication app	ears on the cover sheet with the c	orresnonden	ce address	
Period for Reply	ears on the cover sheet with the c	опезропаен	ce address –	
A SHORTENED STATUTORY PERIOD FOR REPLY DATE OF THIS COMMUNICATION. - Extensions of time may be available under the provisions of 37 CFR 1.13 after SIX (6) MONTHS from the mailing date of this communication. - If NO period for reply is specified above, the maximum statutory period v - Failure to reply within the set or extended period for reply will, by statute Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	36(a). In no event, however, may a reply be tin will apply and will expire SIX (6) MONTHS from , cause the application to become ABANDONE	nely filed in the mailing date of ED (35 U.S.C. § 13	of this communication. 33).	
Status				
1) Responsive to communication(s) filed on 06/15	5/2018.			
☐ A declaration(s)/affidavit(s) under 37 CFR 1.1				
	This action is non-final.			
3) An election was made by the applicant in response		set forth duri	ng the interview on	
the restriction requirement and election			Ŭ	
4) Since this application is in condition for allowan closed in accordance with the practice under E			to the merits is	
Disposition of Claims*				
5) ✓ Claim(s) 1-3,5-16 and 31-34 is/are pendir	ng in the application.			
5a) Of the above claim(s) is/are withdraw	-			
6) Claim(s) is/are allowed.	WIT HOTH CONSIGORATION.			
· · · · · · · · · · · · · · · · · · ·				
7) Claim(s) 1-3,5-16 and 31-34 is/are rejected.				
8) Claim(s) 6-7,12-13 and 16 is/are objected to				
9) Claim(s) are subject to restriction and	•			
 If any claims have been determined <u>allowable</u>, you may be eliparticipating intellectual property office for the corresponding approperty. 	=	_	iway program at a	
http://www.uspto.gov/patents/init_events/pph/index.jsp or send				
	an inquity to <u>i i i i i i coaback</u>	<u>gov.</u>		
Application Papers 10)☑ The specification is objected to by the Examine	r.			
11) The drawing(s) filed on 06/15/2018 is/are: a) □	accepted or b) ✓ objected to	by the Exam	iner.	
Applicant may not request that any objection to the di	rawing(s) be held in abeyance. See 3	37 CFR 1.85(a)		
Replacement drawing sheet(s) including the correction	on is required if the drawing(s) is obje	cted to. See 37	7 CFR 1.121(d).	
Priority under 35 U.S.C. § 119				
12) Acknowledgment is made of a claim for foreign Certified copies:	priority under 35 U.S.C. § 119(a	u)-(d) or (f).		
a)☑ All b)☐ Some** c)☐ None of th	e:			
1. Certified copies of the priority docume	ents have been received.			
2. Certified copies of the priority docume	ents have been received in Appli	cation No.		
3. Copies of the certified copies of the p application from the International Bur		eived in this	Mational Stage	
** See the attached detailed Office action for a list of the certific	, , , ,			
Attachment(s)	_			
1) Notice of References Cited (PTO-892)	3) Interview Summary			
2) ✓ Information Disclosure Statement(s) (PTO/SB/08a and/or PTO/S Paper No(s)/Mail Date 20171208.	B/08b) Paper No(s)/Mail D 4) Other:	Jate		

PTOL-326 (Rev. 11-13)

DETAILED ACTION

Notice of Pre-AIA or AIA Status

The present application, filed on or after March 16, 2013, is being examined under the first inventor to file provisions of the AIA.

Response to Amendment

The amendment filed 06/15/2018 has been entered. **Claims 1-3, 5-16, and 31-34** are pending.

Specification

The disclosure is objected to because of the following informalities: on page 1, it is unknown what is the co-pending application.

The Brief Description of Drawings section is missing from the original specification.

Appropriate correction is required.

Drawings

The drawings are objected to because view numbers must be preceded by the abbreviation "Fig.". Corrected drawing sheets in compliance with 37 CFR 1.121(d) are required in reply to the Office action to avoid abandonment of the application. Any amended replacement drawing sheet should include all of the figures appearing on the immediate prior version of the sheet, even if only one figure is being amended. The figure or figure number of an amended drawing should not be labeled as "amended." If a drawing figure is to be canceled, the

Application/Control Number: 15/735,130

Art Unit: 1767

appropriate figure must be removed from the replacement sheet, and where necessary, the

remaining figures must be renumbered and appropriate changes made to the brief description of

the several views of the drawings for consistency. Additional replacement sheets may be

necessary to show the renumbering of the remaining figures. Each drawing sheet submitted after

the filing date of an application must be labeled in the top margin as either "Replacement Sheet"

or "New Sheet" pursuant to 37 CFR 1.121(d). If the changes are not accepted by the examiner,

the applicant will be notified and informed of any required corrective action in the next Office

action. The objection to the drawings will not be held in abeyance.

Claim Objections

Claims 6 and 7 are objected to because of the following informalities: both claims 6 and

7 do not end with periods. Appropriate correction is required.

Claims 12 and 13 are objected to because of the following informalities: the word

"Wherein" should be lowercase in claim 12. Appropriate correction is required.

Claim 16 objected to because of the following informalities: the capital names used for

the names of the compounds should be lowercase. The abbreviations are appropriate.

Appropriate correction is required.

Claim Rejections - 35 USC § 112

The following is a quotation of 35 U.S.C. 112(b):

(b) CONCLUSION.—The specification shall conclude with one or more claims particularly pointing out and distinctly claiming the subject matter which the inventor or a joint inventor regards as the

invention.

The following is a quotation of 35 U.S.C. 112 (pre-AIA), second paragraph:

The specification shall conclude with one or more claims particularly pointing out and distinctly claiming the subject matter which the applicant regards as his invention.

Ex. 1028 - Favored - p. 246

Page 3

Application/Control Number: 15/735,130

Art Unit: 1767

Claims 1-3, 5-16, and 31-34 are rejected under 35 U.S.C. 112(b) or 35 U.S.C. 112 (pre-AIA), second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which the inventor or a joint inventor, or for pre-AIA the applicant regards as the invention.

Regarding claim 1, the instant claim recites, "wherein the protective cross-linked polymeric coating is obtainable by ..." on line 5. It is unknown by what other methods may produce the protective cross-linked polymeric coating.

Regarding claim 1, it is unknown how the following structure

relates to the instant claimed limitations.

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103 which forms the basis for all obviousness rejections set forth in this Office action:

A patent for a claimed invention may not be obtained, notwithstanding that the claimed invention is not identically disclosed as set forth in section 102, if the differences between the claimed invention and the prior art are such that the claimed invention as a whole would have been obvious before the effective filing date of the claimed invention to a person having ordinary skill in the art to which the claimed invention pertains. Patentability shall not be negated by the manner in which the invention was made.

The factual inquiries set forth in *Graham v. John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103 are summarized as follows:

- 1. Determining the scope and contents of the prior art.
- 2. Ascertaining the differences between the prior art and the claims at issue.

Art Unit: 1767

- 3. Resolving the level of ordinary skill in the pertinent art.
- 4. Considering objective evidence present in the application indicating obviousness or nonobviousness.

This application currently names joint inventors. In considering patentability of the claims, the examiner presumes that the subject matter of the various claims was commonly owned as of the effective filing date of the claimed invention(s) absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and effective filing dates of each claim that was not commonly owned as of the effective filing date of the later invention in order for the examiner to consider the applicability of 35 U.S.C. 102(b)(2)(C) for any potential 35 U.S.C. 102(a)(2) prior art against the later invention.

Claims 1-3, 5-6, 8-11, 16, and 31-34 are rejected under 35 U.S.C. 103 as being unpatentable over Coulson (WO 2007/083122 A1, referred to hereinafter as WO '122) in view of Coulson (US 2010/0203347 A1, referred to hereinafter as US '347).

Regarding claims 1, 5-6, 8-10, 16, and 31-34, Coulson (WO '122) discloses an electronic or electrical device with a polymeric coating formed by exposing the device to pulsed plasma (P3/L20-CP4/3). The monomer used to formed the polymer includes a compound is of formula (IV):

where R^7 is includes H or C_1 alkyl, and in particular H and x is an integer of from 1 to 9 (P12/L14-22).

However, Coulson (WO '122) does not disclose the crosslinking reagent of the instant claims. Coulson (US '347) teaches a polymeric coating formed by exposing an item to plasma

comprising one or more organic monomeric compounds, at least one of which comprises two carbon-carbon double bonds (crosslinking reagents) [0029]. The crosslinking reagents include where Z is perfluoroalkyl group or optionally substituted alkyl groups, wherein the ester group is not required [0030-0031]. The compounds also include the following compound:

wherein each R¹⁴ and R¹⁵ is independently selected from hydrogen, alkyl or haloalkyl [0032, 0035]. One of the crosslinking reagents is of the following formula:

Since the crosslinking reagent meets the claimed limitations of the formula as recited in claim 5, it would be expected that the crosslinking reagent has a boiling point of less than 500°C, absent objective evidence to the contrary. The comonomers may include fluoro (meth)acrylates [0028]. Coulson (WO '122) and Coulson (US '347) are analogous art concerned with the same field of endeavor, namely plasma formed coatings. It would have been obvious to one of ordinary skill in the art before the effective filing date of the claimed invention to add the compound have two carbon-carbon double bonds per the teachings of Coulson (US '347), and the motivation to do so would have been as Coulson (US '347 suggests such polymeric layers are useful on the surface [0029]. The coatings have a high degree of liquid-proofing protection [0011]. Since the coating has a high degree of liquid-proofing protection, it would be expected that the protective cross-linked polymer coating forms a liquid repellent surface defined by a

static water contact angle (WCA) of at 90° or may be adjusted to achieve this static water contact angle. Due to two carbon-carbon double bounds, crosslinking would result, and the resulting coating is tougher and less flexible. In regards to the length of the carbon chain in the alkyl group for Y¹⁰, an obviousness rejection based on similarity in chemical structure and function entails the motivation of one skilled in the art to make a claimed compound in the expectation that compounds similar in structure will have similar properties. *In re Payne*, 606 F. 2d 303, 313, 203 USPQ 245, 254 (CCPA 979). See *In re Papesch*, 315.2d 381, 137 USPQ 43 (CCPA

1963) and In re Dillon, 919 F.2d 688, 16 USPQ2d 1897 (Fed. Cir. 1991). See MPEP 2144.09.

Regarding claim 2, Coulson (WO '122) does not disclose the protective cross-linked polymeric coating is a physical barrier to mass and electron transport. The Office realizes that all of the claimed effects or physical properties are not positively stated by the reference(s). However, the reference(s) teaches all of the claimed ingredients, claimed amounts, and substantially similar process of making. Therefore, the claimed effects and physical properties, i.e. the protective cross-linked polymeric coating is a physical barrier to mass and electron transport would implicitly be achieved by a composition with all the claimed ingredients, claimed amounts, and substantially similar process of making. See MPEP § 2112.01. If it is the applicant's position that this would not be the case: (1) evidence would need to be provided to support the applicant's position; and (2) it would be the Office's position that the application contains inadequate disclosure that there is no teaching as to how to obtain the claimed properties with only the claimed ingredients, claimed amounts, and substantially similar process of making.

Regarding claim 11, Coulson (US '347) teaches the formula of (V) wherein n is 1 to 10 [0032]. Therefore, similar results would be expected when n is 4 to 6. An obviousness rejection

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Art Unit: 1767

based on similarity in chemical structure and function entails the motivation of one skilled in the

art to make a claimed compound in the expectation that compounds similar in structure will have

similar properties. In re Payne, 606 F. 2d 303, 313, 203 USPQ 245, 254 (CCPA 979). See In re

Papesch, 315.2d 381, 137 USPQ 43 (CCPA 1963) and In re Dillon, 919 F.2d 688, 16 USPQ2d

1897 (Fed. Cir. 1991). See MPEP 2144.09.

Allowable Subject Matter

Claims 7 and 12-15 would be allowable if rewritten to overcome the rejection(s) under

35 U.S.C. 112(b) or 35 U.S.C. 112 (pre-AIA), 2nd paragraph, set forth in this Office action and

to include all of the limitations of the base claim and any intervening claims.

The following is a statement of reasons for the indication of allowable subject matter: the

closest prior art of record, Coulson (WO '122) does not teach or suggest each L has one of the

structures as recited in claim 7, wherein Y^{10} meets the formula as recited in claim 12, wherein

each Y^{15} is methyl and each Y^{9} is a bond, or Y^{10} has the formula as recited in claim 14.

Prior Art

The prior art made of record and not relied upon is considered pertinent to applicant's

disclosure. Badyal et al. (US RE43,651 E) teaches a method of coating a surface with a polymer

layer.

York et al. (WO 2004/065684 A2) teaches fibrous products.

Legein et al. (US 2016/0284518 A1) teaches low pressure plasma processes.

Petersen et al. (US 2014/0342103 A1) teaches atmospheric plasma polymerization.

Badyal et al. (EP 1557489 A1) teaches surface coatings.

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Sanyal et al. (US 2010/0021998 A1) teaches plasma polymerization.

Legein (US 8,852,693 B2) teaches a coated electronic device.

Rauschnabel et al. (US 6,613,393 B1) teaches a method for applying wear protection layers.

Bongiovanni et al. "High performance UV-cured coatings for wood protection", Progress in Organic Coatings, 2002, 45, 359-363.

Morita et al. (WO 2006/129800 A1) teaches a finishing agent for photo lithography.

Correspondence

Any inquiry concerning this communication or earlier communications from the examiner should be directed to NICOLE M. BUIE-HATCHER whose telephone number is (571)270-3879. The examiner can normally be reached on M-F, 8 am - 4:30 pm.

Examiner interviews are available via telephone, in-person, and video conferencing using a USPTO supplied web-based collaboration tool. To schedule an interview, applicant is encouraged to use the USPTO Automated Interview Request (AIR) at http://www.uspto.gov/interviewpractice.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Mark Eashoo can be reached on (571)272-1197. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/NICOLE M. BUIE-HATCHER/ Primary Examiner, Art Unit 1767

Notice of References Cited Application/Control No. 15/735,130 Applicant(s)/Patent Under Reexamination COULSON et al. Examiner NICOLE M BUIE-HATCHER Art Unit 1767 Page 1 of 1

U.S. PATENT DOCUMENTS

*		Document Number Country Code-Number-Kind Code	Date MM-YYYY	Name	CPC Classification	US Classification
*	Α	US-6613393-B1	09-2003	Rauschnabel; Johannes	B05D1/62	204/192.12
*	В	US-8852693-B2	10-2014	Legein; Filip	B05D1/62	118/728
*	С	US-RE43651-E	09-2012	Badyal; Jas P. S.	D06M10/02	427/487
*	D	US-20100021998-A1	01-2010	Sanyal; Suparna	C12N5/0068	435/366
*	Е	US-20100203347-A1	08-2010	Coulson; Stephen	A43B7/12	428/500
*	F	US-20140342103-A1	11-2014	Petersen; Julien	B05D5/083	427/569
*	G	US-20160284518-A1	09-2016	Legein; Filip	H01J37/32146	1/1
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	K					
	L					
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*		Document Number Country Code-Number-Kind Code	Date MM-YYYY	Country	Name	CPC Classification
	N	WO-2004065684-A2	08-2004	wo	York et al.	D06M10/025
	0	EP-1557489-A1	07-2005	EP	Badyal et al.	D06M10/02
	Р	WO-2006129800-A1	12-2006	wo	Morita et al.	G03F7/0005
	Q					
	R					
	S					
	Т					

NON-PATENT DOCUMENTS

*		Include as applicable: Author, Title Date, Publisher, Edition or Volume, Pertinent Pages)
	U	Bongiovanni et al. "High performance UV-cured coatings for wood protection", Progress in Organic Coatings, 2002, 45, 359-363. (Year: 2002)
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	w	
	х	

^{*}A copy of this reference is not being furnished with this Office action. (See MPEP § 707.05(a).) Dates in MM-YYYY format are publication dates. Classifications may be US or foreign.

U.S. Patent and Trademark Office PTO-892 (Rev. 01-2001)

Notice of References Cited

Part of Paper No. 20181012

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Searc	h Notes	Application/Control No.		Applicant(s)/Patent Under Reexamination		
		15/735,130	COULSON et al			
		Examiner	Art Unit			
		NICOLE M BUIE-HATCHER	1767			
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Search Notes			Date	Examiner		
Interference S	earch					
US Class/CPC Symbol	US Subclass	/CPC Group	Date	Examiner		

Session Began October 15, 2018 at 09:38 AM

Task Began October 15, 2018 09:38 AM

Explore substances by ID: dimethylpolysiloxane diacrylate (ID 1)

No answers

Task Began October 15, 2018 09:39 AM

Explore substances by ID: diacrylate dimethylpolysiloxane (ID 2)

No answers

Task Began October 15, 2018 09:39 AM

Explore references by patent: (ID 3)

Patent Number: us20180171171

Answer Type: References

Result Count:

Retrieve substance information in 1 reference (ID 4)

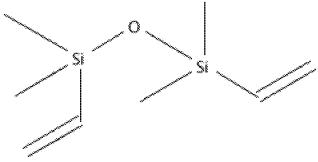
From ID: 3

Answer Type: Substances

Result Count: 17

Task Began October 15, 2018 09:41 AM

Explore substances by SIMILARITY_2D_EXACT



Candidates: 9

Candidates Selected (ID 6)

99 (most similar)

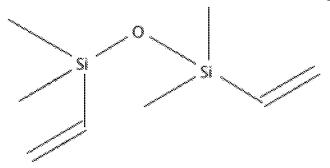
95-98 90-94 85-89 80-84 75-79

Answer Type: Substances

Result Count: 1536

Refine Answer set by structure (ID 7)

Structure: exact From ID: 6



Limiters:

Polymers

Answer Type: Substances

Result Count: 394

Retrieve reference information in 394 substances (ID 8)

From ID: 7

Answer Type: References

Result Count: 2

Task Began October 15, 2018 09:54 AM

Explore substances by property: (ID 9)

Property: Experimental - Boiling Point, >500

Answer Type: Substances

Result Count: 852

Task Began October 15, 2018 10:08 AM

Explore references by patent: (ID 10)

Patent Number: us20180171171

Answer Type: References

Result Count: 1

Retrieve substance information in 1 reference (ID 11)

From ID: 10

Answer Type: Substances

Result Count: 17

Retrieve reference information in 17 substances (ID 12)

From ID: 11

Answer Type: References

Result Count: 668

Refine by research topic (ID 13)

Research Topic: crosslinked polymer

From ID: 12

Answer Type: References

Result Count: 139

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Bibliographic Data

15/735,130 Application No: • Yes ONo Foreign Priority claimed: \square No Yes ☐ Met After Allowance 35 USC 119 (a-d) conditions met: /NICOLE M. BUIE-Verified and Acknowledged: HATCHER/ Examiner's Signature Initials COATINGS Title:

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12/08/2017	526	1767	GBL0007/US
RULE			

APPLICANTS

P2i Ltd, Abingdon, UNITED KINGDOM

INVENTORS

Stephen Richard COULSON Abingdon, UNITED KINGDOM

Delwyn EVANS Abingdon, UNITED KINGDOM

Angeliki SIOKOU Abingdon, UNITED KINGDOM

Clive TELFORD Abingdon, UNITED KINGDOM

CONTINUING DATA

This application is a 371 of PCT/GB2016/051688 06/08/2016

FOREIGN APPLICATIONS

EPO 15386017.6 06/09/2015

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\$1,620

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U.S. Patent and Trademark Office; U.S. DEPARTMENT OF COMMERCE

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First Named Inventor Steph		hen Richard COULSON	
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Attorney Docket Number		GBL0007/US	
	Filing Date First Named Inventor Art Unit Examiner Name	Filing Date First Named Inventor Steph Art Unit Examiner Name	Filing Date First Named Inventor Stephen Richard COULSON Art Unit Examiner Name

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Receipt date: 12/08/2017 15/735,130 - GAU: 1767 **Application Number** Filing Date INFORMATION DISCLOSURE First Named Inventor Stephen Richard COULSON STATEMENT BY APPLICANT Art Unit (Not for submission under 37 CFR 1.99) **Examiner Name** Attorney Docket Number GBL0007/US 2007/083122 WO A1 2007-07-26 P2I LTD /n.м.в./⁴ /и.м.в. /⁵ 0834352 ΕP **A**1 1998-04-08 CIBA-GEIGY AG /n.m.B./6 2014/155099 WO A1 2014-10-02 SEMBLANT LTD

/N.M.B./7 WO Α1 2015/128627 2015-09-03 P2I LTD WO 2016/198855 Α1 2016-12-15 P2I LTD /N.M.B./8/N.M.B./9 2016/198856 WO A1 2016-12-15 P2I LTD If you wish to add additional Foreign Patent Document citation information please click the Add button **NON-PATENT LITERATURE DOCUMENTS** Include name of the author (in CAPITAL LETTERS), title of the article (when appropriate), title of the item Cite Examiner Ţ5 (book, magazine, journal, serial, symposium, catalog, etc), date, pages(s), volume-issue number(s), Initials* publisher, city and/or country where published. International Search Report and Written Opinion for International Application No. PCT/GB2016/051688, mailed $/N.M.B./|_1$ November 8, 2016 (9 pages) If you wish to add additional non-patent literature document citation information please click the Add button **EXAMINER SIGNATURE** Examiner Signature Date Considered /NICOLE M. BUIE-HATCHER/ 10/12/2018 *EXAMINER: Initial if reference considered, whether or not citation is in conformance with MPEP 609. Draw line through a citation if not in conformance and not considered. Include copy of this form with next communication to applicant.

Receipt date: 12/08/2017 15/735,130 - GAU: 1767

	Application Number		
INFORMATION BIOOLOGICE	Filing Date		
INFORMATION DISCLOSURE	First Named Inventor	Steph	en Richard COULSON
STATEMENT BY APPLICANT (Not for submission under 37 CFR 1.99)	Art Unit		
(Notice Sasmission and or or a not)	Examiner Name		
	Attorney Docket Number	er	GBL0007/US

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Task Began October 12, 2018 05:13 PM

Explore references by patent: (ID 1)

Patent Number: us20180171171

Answer Type: References

Result Count: 1

Detailed display

From ID: 1

Type: hydrophobic coating for protection of an electronic or electric device

Retrieve substance information in 1 reference (ID 2)

From ID: 0

Answer Type: Substances

Result Count: 17

Retrieve reference information in 17 substances (ID 3)

From ID: 2

Answer Type: References

Result Count: 755

Refine by research topic (ID 4)

Research Topic: plasma From ID: 3

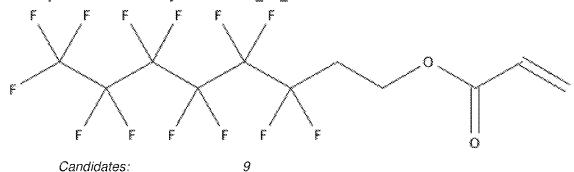
Answer Type: References

Result Count: 42

Prepared 6 reference answers from Answer set 4 for printing in PDF form

Task Began October 12, 2018 05:21 PM

Explore substances by SIMILARITY_2D_EXACT



Co. A. Carrott

Candidates Selected (ID 6)

99 (most similar)

Answer Type: Substances

Result Count: 1802

Retrieve reference information in 1,802 substances (ID 7)

From ID: 6

Answer Type: References

Result Count: 755

Refine by research topic (ID 8)

Research Topic: plasma polymerization

From ID: 7

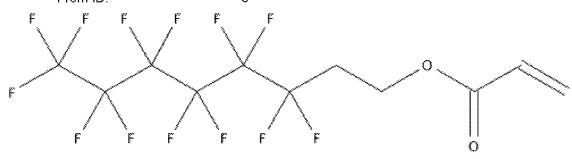
SciFinder® Page 2

Answer Type: References

Result Count: 17

Refine Answer set by structure (ID 9)

Structure: exact From ID: 6



Limiters:

Polymers

Answer Type: Substances

Result Count: 1800

Retrieve reference information in 1,800 substances (ID 10)

From ID: 9

Answer Type: References

Result Count: 37

Prepared 9 reference answers from Answer set 10 for printing in PDF form

Task Began October 12, 2018 06:02 PM

Explore references by research topic: plasma polymerization of fluoroacrylates

Research Topic: plasma polymerization of fluoroacrylates

Result Count: 3

Candidates Selected (ID 11)

2 references were found containing the two concepts "plasma polymerization"

and "fluoroacrylates".

Answer Type: References

Result Count: 2

Task Began October 12, 2018 06:05 PM

Explore references by research topic: plasma fluoroacrylates crosslinked coating

Research Topic: plasma fluoroacrylates crosslinked coating

Result Count: 4

Candidates Selected (ID 12)

1150 references were found containing the concept "crosslinked coating", and

either the concept "plasma" or the concept "fluoroacrylates".

Automatically removed 25 duplicate MEDLINE answer(s)

2 references were found containing the concept "plasma fluoroacrylates".

Automatically removed 25 duplicate MEDLINE answer(s)

Answer Type: References

Result Count: 1127

Prepared 1 reference answer from Answer set 12 for printing in PDF form

 $\mathsf{SciFinder}$

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Doc code: IDS PTO/SB/08a (03-15)
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Doc description: Information Disclosure Statement (IDS) Filed

Approved for use through 07/31/2016. OMB 0851-0031 U.S. Patent and Trademark Office; U.S. DEPARTMENT OF COMMERCE

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	Application Number		15735130		
6 D 3 000 475 500 N 66 G 500 N 675 D 3 N 500 N 675 675 675 675 675 6 000 1000	Filing Date	2017-12-08			
INFORMATION DISCLOSURE	First Named Inventor	Steph	phen Richard COULSON		
STATEMENT BY APPLICANT (Not for submission under 37 CFR 1.99)	Art Unit		1767		
(Notice Submission ander or or it is of	Examiner Name	BUIE	E-HATCHER, NICOLE M		
	Attorney Docket Numb	er	GBL0007/US		
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(Not for submission under 37 CFR 1.99)

Application Number		15735130		
Filing Date		2017-12-08		
First Named Inventor Steph		en Richard COULSON		
Art Unit		1767		
Examiner Name BUII		E-HATCHER, NICOLE M		
Attorney Docket Number	er	GBL0007/US		

	"Plasma treatment can apply liquid repellency to any material", Technical Textiles International, May 2005, pages 23-25 (3 pages).									
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INFORMATION DISCLOSURE STATEMENT BY APPLICANT

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Application Number		15735130		
Filing Date		2017-12-08		
First Named Inventor Steph		en Richard COULSON		
Art Unit		1767		
Examiner Name BUI		E-HATCHER, NICOLE M		
Attorney Docket Number		GBL0007/US		

		CERTIFICATION	STATEMENT		
Plea	ase see 37 CFR 1	.97 and 1.98 to make the appropriate selecti	ion(s):		
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OR	!				
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	See attached ce	rtification statement.			
\boxtimes	The fee set forth	in 37 CFR 1.17 (p) has been submitted here	ewith.		
\boxtimes	A certification sta	atement is not submitted herewith.			
SIGNATURE A signature of the applicant or representative is required in accordance with CFR 1.33, 10.18. Please see CFR 1.4(d) for the form of the signature.					
Sigr	nature	/Paul J. Parins/	Date (YYYY-MM-DD)	2018-11-21	
Nan	Name/Print Paul J. Parins Registration Number 54,358				
pub 1.14 app	lic which is to file I. This collection lication form to the	rmation is required by 37 CFR 1.97 and 1.98 (and by the USPTO to process) an application is estimated to take 1 hour to complete, included USPTO. Time will vary depending upon the his form and/or suggestions for reducing this	on. Confidentiality is gove uding gathering, preparing e individual case. Any co	rned by 35 U.S.C. 122 and 37 CFR and submitting the completed mments on the amount of time you	

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Electronic Patent Application Fee Transmittal					
Application Number:	15735130				
Filing Date:	08-	Dec-2017			
Title of Invention:	COATINGS				
First Named Inventor/Applicant Name:	Stephen Richard COULSON				
Filer:	Paul John Parins/Bethany Lichtscheidl				
Attorney Docket Number:	GBL0007/US				
Filed as Large Entity					
Filing Fees for U.S. National Stage under 35 USC 371					
Description		Fee Code	Quantity	Amount	Sub-Total in USD(\$)
Basic Filing:					
Pages:					
Claims:					
Miscellaneous-Filing:					
Petition:					
Patent-Appeals-and-Interference:					
Post-Allowance-and-Post-Issuance:					
Extension-of-Time:					

Description	Fee Code	Quantity	Amount	Sub-Total in USD(\$)
Miscellaneous:				
SUBMISSION- INFORMATION DISCLOSURE STMT	1806	1	240	240
	Tot	al in USD	(\$)	240

EFS ID: Application Number: International Application Number:	34374605 15735130 7101
International Application Number:	7101
	7101
Confirmation Number:	
Title of Invention:	COATINGS
First Named Inventor/Applicant Name:	Stephen Richard COULSON
Customer Number:	33072
Filer:	Paul John Parins/Bethany Lichtscheidl
Filer Authorized By:	Paul John Parins
Attorney Docket Number:	GBL0007/US
Receipt Date:	21-NOV-2018
Filing Date:	08-DEC-2017
Time Stamp:	13:19:45
Application Type:	U.S. National Stage under 35 USC 371

Payment information:

Submitted with Payment	yes
Payment Type	CARD
Payment was successfully received in RAM	\$240
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			595496				
2	Non Patent Literature	Plasma_treatment.pdf	5eaec8c580a621d9715f7a8b6ad643d1d6ff eb70	no	3		
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Buie-Hatcher, Nicole

GBL0007/US

In The United States Patent And Trademark Office

Examiner:

Docket No.

Group Art Unit: 1767

Confirmation No. 7101

Stephen Richard COULSON Inventor:

Serial No. 15/735,130

International Filing Date: June 8, 2016

International App. No. PCT/GB2016/051688

371(c):

December 8, 2017

For:

COATINGS

I CERTIFY THAT ON MARCH 14, 2019, THIS CORRESPONDENCE IS BEING ELECTRONICALLY TRANSMITTED TO THE UNITED STATES PATENT AND TRADEMARK OFFICE VIA THE OFFICE'S EFS-WEB.

Mail Stop Amendment Commissioner for Patents P.O. Box 1450 Alexandria, Virginia 22313-1450

TRANSMITTAL

Dear Sir or Madam:

Enclosed herewith are the following in connection with the above-identified patent application:

- A. Declarations (37 CFR 1.63) for Utility Application Using an Application Data Sheet, executed by:
 - · Stephen Richard Coulson,
 - Delwyn Evans,
 - Angeliki Siokou
- B. A Substitute Statement in Lieu of an Oath or Declaration for Utility Design Patent Application or Design Patent Application (35 U.S.C. 115(d) And 37 CFR 1.64), executed by:
 - assignee (Dr. Stephen Coulson) on behalf of Clive Telford.
- C. It is believed that no surcharge is required in filing these declarations as the \$140.00 Surcharge for filing the Declaration late was paid on the same day that the application was filed (December 8, 2017).

Date: MARCH 14,2019

Respectfully Submitted

By:

Paul J. Parins, Reg. No. 54,358

Customer Number 33072

Phone: 651-275-9831 Fax: 651-351-2954

DECLARATION (37 CFR 1.63) FOR UTILITY OR DESIGN APPLICATION USING AN APPLICATION DATA SHEET (37 CFR 1.76)

Title of Invention	COATINGS					
Attorney Docket No.	GBL0007/US (P2494US00-CD7QC)					
As the below named inventor, I hereby declare that:						
This declaration						
	United States application or PCT international application no. PCT/GB2016/051688 filed on June 8, 2016; and/or					
	The application having the above-identified attorney docket number on the specification as filed, or the above-identified title on the specification as filed.					
The above-ide	ntified application was made or authorized to be made by me.					
I believe that I	am the original inventor or an original joint inventor of a claimed invention in the application.					
I hereby state t including the cl	hat I have reviewed and understand the contents of the above-identified application, aims.					
I am aware of t	he duty to disclose information which is material to patentability as defined in 37 CFR 1.56.					
I hereby acknowledge that any willful false statement made in this declaration is punishable under 18 U.S.C. 1001 by fine or imprisonment of not more than five (5) years, or both.						
LEGAL NAME OF INVENTOR						
Inventor: Stephen Richard COULSON Date (optional):						
Signature:	SKGL					

DECLARATION (37 CFR 1.63) FOR UTILITY OR DESIGN APPLICATION USING AN APPLICATION DATA SHEET (37 CFR 1.76)

Title of Invention	COATINGS				
Attorney Docket No.	GBL0007/US (P2494US00-CD7QC)				
As the below	named inventor, I hereby declare that:				
This declaration	green, and an amount and a second a second and a second a				
	United States application or PCT international application no. PCT/GB2016/051688 filed on June 8, 2016; and/or				
	The application having the above-identified attorney docket number on the specification as filed, or the above-identified fitle on the specification as filed.				
The above-ide	entified application was made or authorized to be made by me.				
I believe that I	arn the original inventor or an original joint inventor of a claimed invention in the application.				
I hereby state including the o	that I have reviewed and understand the contents of the above-identified application, claims.				
I am aware of	the duty to disclose information which is material to patentability as defined in 37 CFR 1.56.				
I hereby acknowledge that any willful false statement made in this declaration is punishable under 18 U.S.C. 1001 by fine or imprisonment of not more than five (5) years, or both.					
LEGAL NAME OF INVENTOR					
Inventor: Delw	ryn EVANS Date (optional): 27 FEIS 2019				
Signature:	DUL.				

DECLARATION (37 CFR 1.63) FOR UTILITY OR DESIGN APPLICATION USING AN APPLICATION DATA SHEET (37 CFR 1.76)

Title of Invention	COATINGS				
Attorney Docket No.	GBL0007/US (P2494US00-CD7QC)				
As the below r	named inventor, I hereby declare that:				
This declaration	n				
,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,	United States application or PCT international application no. PCT/GB2016/051688 filed on June 8, 2016; and/or				
	The application having the above-identified attorney docket number on the specification as filed, or the above-identified title on the specification as filed.				
The above-ide	ntified application was made or authorized to be made by me.				
I believe that I	am the original inventor or an original joint inventor of a claimed invention in the application.				
I hereby state including the c	that I have reviewed and understand the contents of the above-identified application, laims.				
I am aware of	the duty to disclose information which is material to patentability as defined in 37 CFR 1.56.				
I hereby acknowledge that any willful false statement made in this declaration is punishable under 18 U.S.C. 1001 by fine or imprisonment of not more than five (5) years, or both.					
LEGAL NAME OF INVENTOR					
Inventor: Angeliki SIOKOU Date (optional): 27 02.19					
Signature:					

Document Description: Oath or declaration filed

PT(VAIA/02 (07-13)

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Title of Invention	COATINGS			
This statement is directed to: The attached application. OR United States application or PCT international application number LEGAL NAME of inventor to whom this substitute statement applies: (E.g., Given Name (first and middle (if any)) and Family Name or Surname) Clive Telford Residence (except for a deceased or legally incapacitated inventor): City State Country Mailing Address (except for a deceased or legally incapacitated inventor):				
City.		State	Žir.	Country
City State Zip Country I believe the above-named inventor or joint inventor to be the original inventor or an original joint inventor of a claimed invention in the application. The above-identified application was made or authorized to be made by me. I hereby acknowledge that any willful false statement made in this statement is punishable under 18 U.S.C. 1001 by line or imprisonment of not more than five (5) years, or both.				
Relationship to the inventor to whom this substitute statement applies: Legal Representative (for deceased or legally incapacitated inventor only). Assignee, Person to whom the inventor is under an obligation to assign, Person who otherwise shows a sufficient proprietary interest in the matter (petition under 37 CFR 1.46 is required), or Joint Inventor.				

[Page 1 of 2]

This collection of information is required by 35 U.S.C. 115 and 37 CFR 1.63. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.11 and 1.14. This collection is estimated to take 1 minute to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form ancier suggestions for reducing this burden, should be sent to the Charl Micromation Officer, U.S. Patent and Tradement Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450, DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

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SUBSTITUTE STATEMENT

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Circumstances permitting execution of th	is substitute statemer	νŧ			
Inventor is deceased,					
Inventor is under legal incapacit	ty,				
Inventor cannot be found or rea	ched after diligent effo	ont or			
Inventor has refused to execute	ra i di sar e rator sed e ed		FR 1.63.		
If there are joint inventors, please check					
An application data sheet under	and the second second second second		valent) naming the enti	re inventive entity has been	
or is currently submitted.	NATAN TERMEN PROTESTANDA MARI	William Series et	Minde is a series in the series of the seri	PRESIDENT CONTRACTOR OF THE SECOND CONTRACTOR	
<u>or</u>					
An application data sheet under Statement Supplemental Sheet information is attached. See 37	(PTO/AIA/11 or equiv				
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PERSON EXECUTING THIS SUBSTITUTE	E STATEMENT:				
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Signature: SLA					
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P2i Ltd Applicant Name:					
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Note: Use an additional PTO/AIA/02 form for after diligent effort, or has refused to execu				annot be found or reached	

[Page 2 of 2]

#### Privacy Act Statement

The **Privacy Act of 1974 (P.L. 93-579)** requires that you be given certain information in connection with your submission of the attached form related to a patent application or patent. Accordingly, pursuant to the requirements of the Act, please be advised that: (1) the general authority for the collection of this information is 36 U.S.C. 2(b)(2); (2) furnishing of the information solicited is voluntary; and (3) the principal purpose for which the information is used by the U.S. Patent and Trademark. Office is to process and/or examine your submission related to a patent application or patent. If you do not furnish the requested information, the U.S. Patent and Trademark Office may not be able to process and/or examine your submission, which may result in termination of proceedings or abandonment of the application or expiration of the patent.

The information provided by you in this form will be subject to the following routine uses:

- The information on this form will be treated confidentially to the extent allowed under the
  Freedom of Information Act (5 U.S.C. 552) and the Privacy Act (5 U.S.C. 552a). Records from
  this system of records may be disclosed to the Department of Justice to determine whether
  disclosure of these records is required by the Freedom of Information Act.
- A record from this system of records may be disclosed, as a routine use, in the course of presenting evidence to a court, magistrate, or administrative tribunal, including disclosures to opposing coursel in the course of settlement negotiations.
- 3. A record in this system of records may be disclosed, as a routine use, to a Member of Congress submitting a request involving an individual, to whom the record pertains, when the individual has requested assistance from the Member with respect to the subject matter of the record.
- 4. A record in this system of records may be disclosed, as a routine use, to a contractor of the Agency having need for the information in order to perform a contract. Recipients of information shall be required to comply with the requirements of the Privacy Act of 1974, as amended, pursuant to 5 U.S.C. 552a(m).
- 5. A record related to an International Application filed under the Patent Cooperation Treaty in this system of records may be disclosed, as a routine use, to the International Bureau of the World Intellectual Property Organization, pursuant to the Patent Cooperation Treaty.
- A record in this system of records may be disclosed, as a routine use, to another federal agency for purposes of National Security review (35 U.S.C. 181) and for review pursuant to the Atomic Energy Act (42 U.S.C. 218(c)).
- 7. A record from this system of records may be disclosed, as a routine use, to the Administrator, General Services, or his/her designee, during an inspection of records conducted by GSA as part of that agency's responsibility to recommend improvements in records management practices and programs, under authority of 44 U.S.C. 2904 and 2906. Such disclosure shall be made in accordance with the GSA regulations governing inspection of records for this purpose, and any other relevant (i.e., GSA or Commerce) directive. Such disclosure shall not be used to make determinations about individuals.
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- A record from this system of records may be disclosed, as a routine use, to a Federal, State, or local law enforcement agency, if the USPTO becomes aware of a violation or potential violation of law or regulation.

Electronic Acknowledgement Receipt				
EFS ID:	35420272			
Application Number:	15735130			
International Application Number:				
Confirmation Number:	7101			
Title of Invention:	COATINGS			
First Named Inventor/Applicant Name:	Stephen Richard COULSON			
Customer Number:	33072			
Filer:	Paul John Parins/Bethany Lichtscheidl			
Filer Authorized By:	Paul John Parins			
Attorney Docket Number:	GBL0007/US			
Receipt Date:	14-MAR-2019			
Filing Date:	08-DEC-2017			
Time Stamp:	11:11:11			
Application Type:	U.S. National Stage under 35 USC 371			

### **Payment information:**

Submitted with Payment	no
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### File Listing:

Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)	
		CDI 0007HC Tunnersitted for D	212108			
1 Transmittal Letter	GBL0007US_Transmittal_for_D eclarations.pdf	470f712ee11cbccf3c5990c6345f95e9e317 29d6	no	2		
<b>Warnings:</b> Ex. 1028 - Favored - p. 282						

Information:							
2	Oath or Declaration filed	GBL0007US_Declaration.pdf	398658	no	3		
Warnings:		4c68a					
Information:							
	CDI 0007US S. L. VII. L. SL. VII.	634125					
3 Oath or Declaration filed	GBL0007US_Substitute_Statem ent.pdf	3eb370671cae47e410a2009cc6dfb7f38bed 56b7	no	3			
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	Total Files Size (in bytes): 1244891						

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If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

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If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

#### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

o Tack Free ☀ Tacky

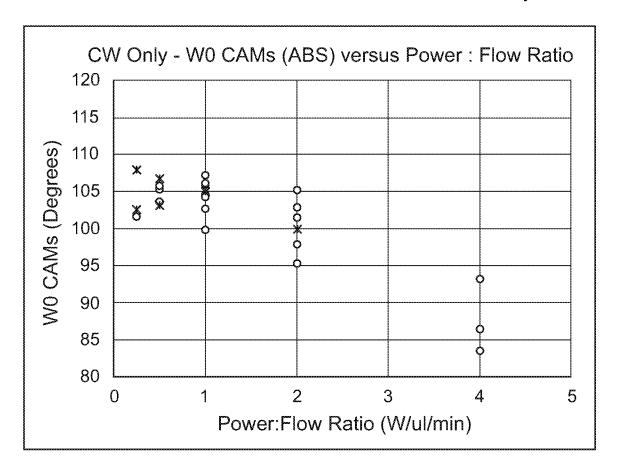


Fig. 1

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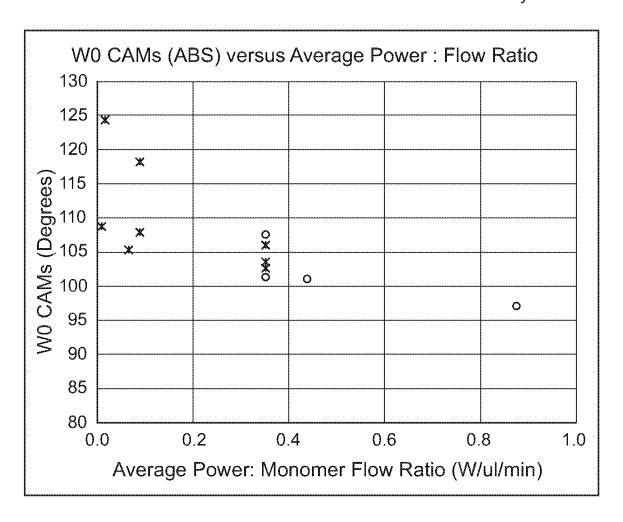
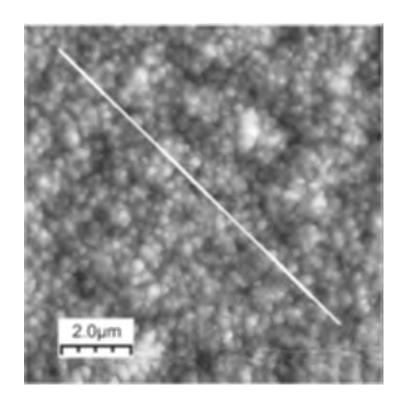
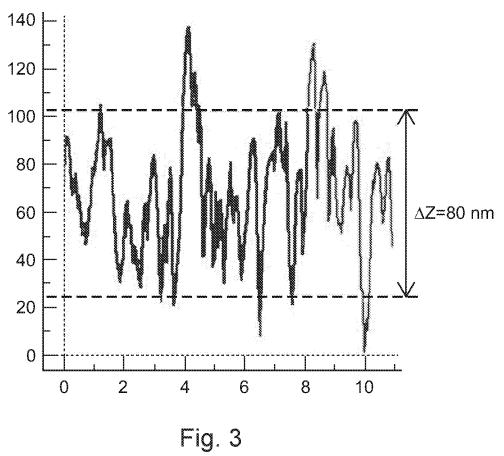


Fig. 2





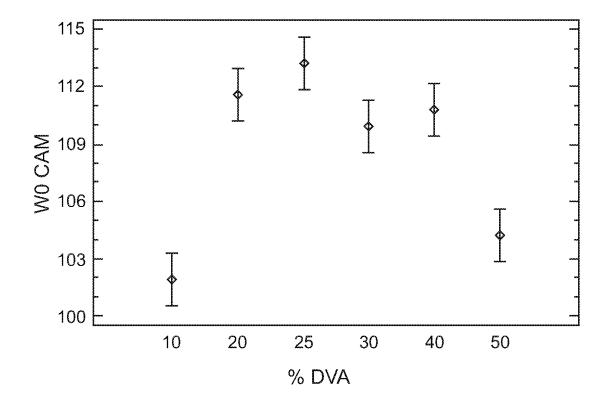
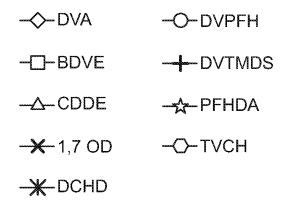


Fig. 4



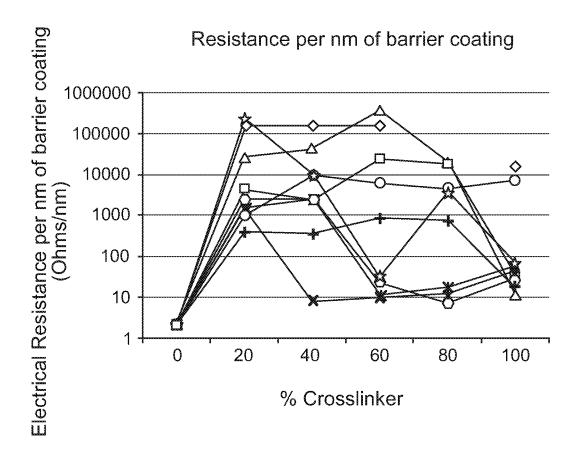
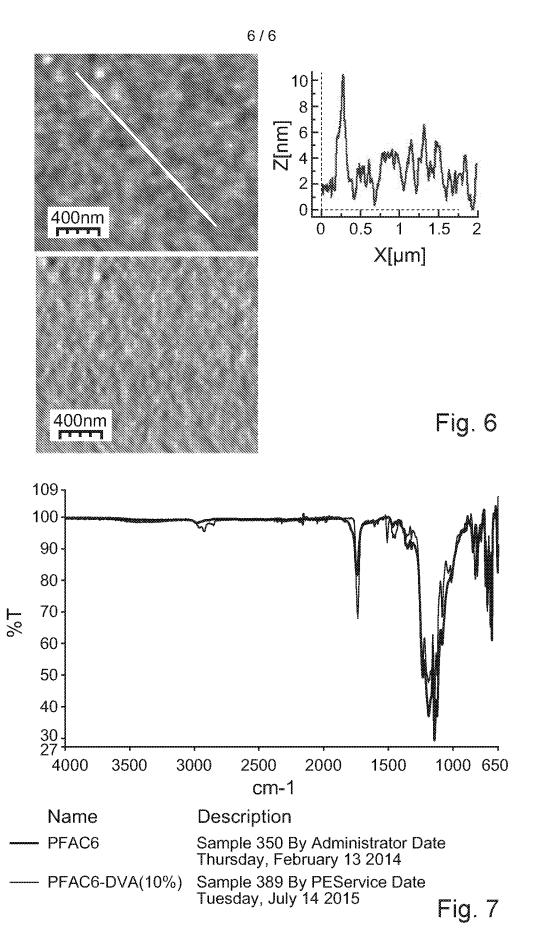


Fig. 5



Electronic Patent Application Fee Transmittal						
Application Number:	157	735130				
Filing Date:	08-	Dec-2017				
Title of Invention:	COATINGS					
First Named Inventor/Applicant Name:	Stephen Richard COULSON					
Filer:	Paul John Parins/Bethany Lichtscheidl					
Attorney Docket Number:	GB	L0007/US				
Filed as Large Entity						
Filing Fees for U.S. National Stage under 35 USC 371						
Description		Fee Code	Quantity	Amount	Sub-Total in USD(\$)	
Basic Filing:						
Pages:						
Claims:						
Miscellaneous-Filing:						
Petition:						
Patent-Appeals-and-Interference:						
Post-Allowance-and-Post-Issuance:						
Extension-of-Time:						

Description	Fee Code	Quantity	Amount	Sub-Total in USD(\$)		
Extension - 2 months with \$0 paid	1252	1	600	600		
Miscellaneous:						
	Tot	al in USD	(\$)	600		

Electronic Acknowledgement Receipt				
EFS ID:	35447276			
Application Number:	15735130			
International Application Number:				
Confirmation Number:	7101			
Title of Invention:	COATINGS			
First Named Inventor/Applicant Name:	Stephen Richard COULSON			
Customer Number:	33072			
Filer:	Paul John Parins/Bethany Lichtscheidl			
Filer Authorized By:	Paul John Parins			
Attorney Docket Number:	GBL0007/US			
Receipt Date:	18-MAR-2019			
Filing Date:	08-DEC-2017			
Time Stamp:	11:00:25			
Application Type:	U.S. National Stage under 35 USC 371			

## **Payment information:**

Submitted with Payment	yes
Payment Type	CARD
Payment was successfully received in RAM	\$600
RAM confirmation Number	031819INTEFSW11013800
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The Director of the USPTO is hereby authorized to charge indicated fees and credit any overpayment as follows:

File Listing	<b>1:</b>				
Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.
			182316		
1		GBL0007US_Response_to_Non -Final_OA.pdf	d1a8000f854fd5dc38c714c95a6dbbe3917 3fd87	yes	13
	Multip	part Description/PDF files in .	zip description		
	Document De	scription	Start	Eı	nd
	Applicant Arguments/Remarks	10	13		
	Claims	3	9		
	Amendment/Req. Reconsiderat	ion-After Non-Final Reject	1	2	
Warnings:					
Information:					
			178887		
2	Drawings-only black and white line drawings	GBL0007US_replacement_dra wings.pdf	2ac523b93f2e1ef0Seb5f5646d4401cfec3a2 eba	no	6
Warnings:					
Information:					
			30658		
3	Fee Worksheet (SB06)	fee-info.pdf	1e28cad9fa35fafad8b5e7122f8b7073acbf2 419	no	2
Warnings:					
Information:					
		Total Files Size (in bytes)	39	91861	

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### New Applications Under 35 U.S.C. 111

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New International Application Filed with the USPTO as a Receiving Office

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#### Remarks

Claims 1-3, 5-16, and 31-34 are pending. With this Response, claims 1, 6, 8, 13, 15, and 16 are amended; and claims 5, 7, 12, and 14 are cancelled. Upon entry of the current amendments, claims 1-3, 6, 8-11, 13, 15, 16, and 31-34 are pending for consideration. The present amendments are fully supported by the application as originally filed and do not introduce new matter.

Applicant respectfully requests reconsideration and further examination of the application in view of the amendments above and remarks below.

## **Allowable Subject Matter**

Applicant gratefully acknowledges the indication in the Office Action that claims 7 and 12-15 would be allowable if rewritten to overcome the rejection(s) under 35 U.S.C. 112(b) or 35 U.S.C. 112 (pre-AIA), 2nd paragraph, set forth in this Office action and to include all of the limitations of the base claim and any intervening claims. As explained below, claim 1 is amended to include subject matter of claims 7, 12, and 14 (using alternative "or" language), therefore claim 1 is considered allowable.

Claim 1 is amended to limit L for compound (i) to having the features of claim 5 in combination with 7 (i.e. "Formula A"), or to having the features of claim 5 in combination with claim 12 or claim 14 (i.e. "Formula B").

The terms "Formula A" and "Formula B" have been added to refer to the optional features included in claim 1, in order to avoid unnecessary repetition of these features in the dependent claims. Hence the inclusion of these terms in claim 1 does not add matter.

Claim 1 has been amended to broaden the definitions of Y1 to Y8 and R1, R2, R4, and R5 to R10 to be independently selected from hydrogen or an optionally substituted C1-C6 branched or straight chain alkyl group. Basis for this amendment is found on page 5, lines 22 and 23, and on page 5, lines 5 and 6 of the PCT application as published.

## **Specification**

According to the Office Action, the disclosure is objected to because of the following informalities:

- 1) On page 1, "it is unknown what is the co-pending application." With this reply, the paragraph on page 1 that begins with "Monosaturated monomers" and ends with "(see co-pending application)" is deleted thereby rendering this objection moot.
- 2) A Brief Description of Drawings section is missing. With this reply, said section is added to the specification, thereby rendering this objection moot.

Accordingly, it is respectfully requested that the specification objections be withdrawn.

## **Drawing Objections**

According to the Office Action, the drawings are objected to because each of the view numbers must be preceded by the abbreviation "Fig."

With this reply, each of figures 1-7 are amended to replace "Figure" with "Fig." as requested in the Office Action.

Accordingly, it is respectfully requested the outstanding drawing objections be withdrawn.

## **Claim Objections**

Claims 6 and 7 are objected to because of the following informalities: both claims 6 and 7 do not end with periods. Claim 6 is amended to include a period and claim 7 is cancelled.

Accordingly, it is respectfully requested that the objection of claims 7 and 6 be withdrawn.

Claims 12 and 13 are objected to because of the following informalities: the word "Wherein" should be lowercase in claim 12. Claim 12 is canceled thereby rendering this objection moot.

Accordingly, it is respectfully requested that the objection of claims 12 and 13 be withdrawn.

Claim 16 objected to because of the following informalities: the capital names used for the names of the compounds should be lowercase. The abbreviations are appropriate. Claim 16 is amended as suggested by the Office Action, thereby rendering this objection moot.

Accordingly, it is respectfully requested that the objection of claim 16 be withdrawn.

## Rejection Under 35 U.S.C. §112(b)

Claims 1-3, 5-16, and 31-34 are rejected under 35 U.S.C. 112(b) or 35 U.S.C. 112 (pre-AIA), second paragraph, as being indefinite.

According to the Office Action, the phrase "wherein the protective cross-linked polymeric coating is obtainable by" in claim 1 is unclear. With this reply, the word "obtainable" in said phrase is replaced with "obtained," thereby rendering this rejection moot.

According to the Office Action, the structure is in claim 1 is unclear. With this reply, said structure is delected thereby rendering this rejection moot. Applicant notes that said structure was intended to be deleted with the Preliminary Amendent previously filed.

Accordingly, it is respectfully requested that the rejection of claims 1-3, 5-16, and 31-34 under 35 U.S.C. 112(b) or 35 U.S.C. 112(pre-AIA), second paragraph, as being indefinite, be withdrawn.

## Rejection Under 35 U.S.C. §103

Claims 1-3, 5-6, 8-11, 16, and 31-34 are rejected under 35 U.S.C. 103 as being unpatentable over Coulson (WO 2007/083122 Al, referred to hereinafter as WO '122) in view of Coulson (US 2010/0203347 Al, referred to hereinafter as US '347).

Independent claim 1 is amended in manner to include the subject matter of claims 7, 12, or 14 (in an alternative manner), thereby rendering this rejection moot.

Accordingly, it is respectfully requested that the rejection of claims 1-3, 5-6, 8-11, 16, and 31-34, under 35 U.S.C. 103 as being unpatentable over Coulson (WO 2007/083122 Al, referred to hereinafter as WO '122) in view of Coulson (US 2010/0203347 Al, referred to hereinafter as US '347) be withdrawn.

## **Conclusion**

In view of the above amendments and remarks, it is respectfully submitted that all of the pending claims are in condition for allowance. Approval of the application and allowance of the claims are earnestly solicited. In the event that a phone conference between the Examiner and the Applicants' undersigned attorney would help resolve any issues in the application, the Examiner is invited to contact said attorney at (651) 275-9831.

Respectfully Submitted,

Dated: March 18, 2019

By:/Paul J. Parins/

Paul J. Parins, Reg. No. 54,358

Customer No. 33072 Phone: 651-275-9831 Facsimile: 651-351-2954

#124050

Doc code: IDS Doc description: Information Disclosure Statement (IDS) Filed

PTO/SB/08a (03-15)

Approved for use through 07/31/2016. OMB 0651-0031

Mation Disclosure Statement (IDS) Filed

U.S. Patent and Trademark Office; U.S. DEPARTMENT OF COMMERCE

Under the Paperwork Reduction Act of 1995, no persons are required to respond to a collection of information unless it contains a valid OMB control number.

	Application Number		15735130	
INFORMATION DISCLOSURE	Filing Date		2017-12-08	
	First Named Inventor Stephe		en Richard COULSON	
( Not for submission under 37 CFR 1.99)	Art Unit		1767	
(Not for Submission ander of STR 1.55)	Examiner Name BUIE		IIE-HATCHER, NICOLE M	
	Attorney Docket Number		GBL0007/US	

				Remove				
Examiner Initial*	Cite No	Patent Number	Kind Code ¹	Issue Date	Name of Patentee or Applicant of cited Document		Lines where es or Relevant	
	1	4822632	A	1989-04-18	Williams et al.			
	2	6764758	B1	2004-07-20	Grunze et al.			
If you wis	h to add	additional U.S. Pater	nt citatio	n information pl	ease click the Add button.		Add	
	T	1	U.S.P.	ATENT APPLI	CATION PUBLICATIONS	1	Remove	
Examiner Initial*	Cite No	Publication Number	Kind Code ¹	Publication Date	Name of Patentee or Applicant of cited Document	Releva		Lines where es or Relevant
	1	20040028931	A1	2004-02-12	Bletsos et al.			
	2	20120051018	A1	2012-03-01	Ollgaard			
	3	20120171386	A1	2012-07-05	Bicker et al.			
	4	20120321775	A1	2012-12-20	Parker			

( Not for submission under 37 CFR 1.99)

Application Number		15735130			
Filing Date		2017-12-08			
First Named Inventor	Steph	en Richard COULSON			
Art Unit		1767			
Examiner Name	BUIE	E-HATCHER, NICOLE M			
Attorney Docket Number		GBL0007/US			

	5	20140010969	A1	2014-01	-09	Bicker et al.		
	6	20090267489	A1	2009-10-29		Arita et al.		
	7	20100189914	A1	2010-07	'-29	Coulson		
	8	20120009231	A1	2012-01-12		Herbert et al.		
	9	20140322525	A1	2014-10-30		Legein et al.		
	10	20150291830	A1	2015-10	)-15	Galbreath et al	I.	
	11	20040046165	A1	2004-03	i <b>-</b> 11	Hunze et al.		
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	1	2434368	GB	GB		2007-07-25	P2i Limited	
	2	2493264	GB	GB		2013-01-30	P2i Ltd	

( Not for submission under 37 CFR 1.99)

Application Number		15735130			
Filing Date		2017-12-08			
First Named Inventor	Steph	en Richard COULSON			
Art Unit		1767			
Examiner Name	BUIE	E-HATCHER, NICOLE M			
Attorney Docket Number		GBL0007/US			

3	3	03063938	wo	A1	2003-08-07	ASTRAZENECA AB	
2	4	2005021833	wo	A2	2005-03-10	SURFACE INNOVATIONS LIMITED	
Ę	5	2009046093	WO	A2	2009-04-09	EXOGENESIS CORPORATION	
6	6	2009030435	WO	A1	2009-03-12	UNIV AUGSBURG	
7	7	2011029628	WO	A1	2011-03-17	SCHOTT AG	
3	8	2711153	EP	A1	2014-03-26	The Goodyear Tire and Rubber Company	
ę	9	2005/089961	wo	A1		THE SECRETARY OF STATE OF DEFENCE	
1	10	02/28548	WO	A2	2002-04-11	DOW CORNING CORPORATION	
1	11	2013/132250	wo	A1	2013-09-12	SEMBLANT LTD	
1	12	2012/158953	wo	A2	2012-11-22	GADGET IP LLC	
1	13	2014026967	wo	A2	2014-02-20	EUROPLASMA NV	

( Not for submission under 37 CFR 1.99)

Application Number		15735130		
Filing Date		2017-12-08		
First Named Inventor Steph		en Richard COULSON		
Art Unit		1767		
Examiner Name	BUIE	E-HATCHER, NICOLE M		
Attorney Docket Number		GBL0007/US		

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	1	Search Report under Section 17(5) from UK Patent Application No. GB1403558.8, mailed on August 13, 2014 (4 pages)							
	2	nternational Search Report and Written Opinion from International Application No. PCT/GB2015/050521, mailed June 5, 2015 (16 pages)							
	3	Search Report under Section 17(5) from UK Patent Application No. GB1503054.7, mailed on November 30, 2015 (4 pages)							
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Standard ST ⁴ Kind of doo	Γ.3). ³ F cument l	JSPTO Patent Documents at <a href="www.USPTO.GOV">www.USPTO.GOV</a> or MPEP 901.04. ² Enter office that issued the document, by the two-letter code (WIPO Japanese patent documents, the indication of the year of the reign of the Emperor must precede the serial number of the patent document the appropriate symbols as indicated on the document under WIPO Standard ST.16 if possible. ⁵ Applicant is to place a check mark here slation is attached.							

( Not for submission under 37 CFR 1.99)

Application Number		15735130
Filing Date		2017-12-08
First Named Inventor	Steph	en Richard COULSON
Art Unit		1767
Examiner Name BUIL		E-HATCHER, NICOLE M
Attorney Docket Number		GBL0007/US

#### **CERTIFICATION STATEMENT**

Please see 37 CFR 1.97 and 1.98 to make the appropriate selection(s):

That each item of information contained in the information disclosure statement was first cited in any communication from a foreign patent office in a counterpart foreign application not more than three months prior to the filing of the information disclosure statement. See 37 CFR 1.97(e)(1).

OR

That no item of information contained in the information disclosure statement was cited in a communication from a foreign patent office in a counterpart foreign application, and, to the knowledge of the person signing the certification after making reasonable inquiry, no item of information contained in the information disclosure statement was known to any individual designated in 37 CFR 1.56(c) more than three months prior to the filing of the information disclosure statement. See 37 CFR 1.97(e)(2).

See attached certification statement.

- The fee set forth in 37 CFR 1.17 (p) has been submitted herewith.
- × A certification statement is not submitted herewith.

#### **SIGNATURE**

A signature of the applicant or representative is required in accordance with CFR 1.33, 10.18. Please see CFR 1.4(d) for the form of the signature.

Signature	/Paul J. Parins/	Date (YYYY-MM-DD)	2019-03-18
Name/Print	Paul J. Parins	Registration Number	54,358

This collection of information is required by 37 CFR 1.97 and 1.98. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.14. This collection is estimated to take 1 hour to complete, including gathering, preparing and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. **SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.** 

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## **EUROPEAN PATENT APPLICATION**

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(71) Applicant: The Goodyear Tire & Rubber Company Akron, Ohio 44316 (US) (72) Inventors:

- Siffer, Frederic Gerard Auguste Akron, Ohio 44313 (US)
- Richards, Jimmy Lee Rootstown, Ohio 44272 (US)
- (74) Representative: Kutsch, Bernd Goodyear S.A. Patent Department Avenue Gordon Smith 7750 Colmar-Berg (LU)
- (54) Method of coating a metal mold surface with a polymer coating, mold for rubber products and method of molding rubber products
- (57) A method of freating a metal surface (1) of a mold for molding and curing rubber products is disclosed. The method comprises applying an almospheric plasma comprising an ionized gas containing a polymerizable

precursor to a metal molding surface of the mold metal surface to form and bond a polymer layer of plasma polymerized precursor on the mold metal surface (1).

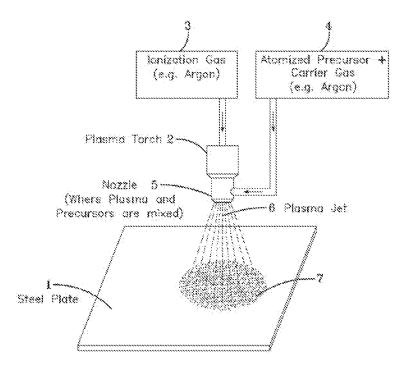


FIG-1

#### Description

#### Field of the Invention

[0001] This invention relates to preparation of a metal mold surface for molding rubber products by plasma jet directed polymer coating, a coated mold for rubber products and production of rubber products. Such rubber products are rubber tires and industrial products, particularly rubber tires.

#### Background

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[0802] Rubber products, such as for example pneumatic rubber tires, are typically produced by molding and curing sulfur curable green, or uncured, rubber tires in a tire mold in which the uncured pneumatic rubber tire is pressed outwardly against a metal mold surface by means of an internal fluid expanding rubber bladder (e.g. by application of internal steam pressure to the bladder envelop). By such method the uncured rubber tire is pressed against and shaped against the metal mold surface which defines the tire's tread pattern and configuration of its sidewalls and during which a portion of the rubber in proximity to the mold surface flows to permit the rubber to assume the contours of the mold surface. By application of heat to the mold, the tire is cured within the mold. The mold is then opened and the cured tire removed from the mold. Such pneumatic rubber tire molding and curing is well known to those having skill in such art. [0803] It is important that such molded and sulfur cured rubber products such as tires be readily removable from their tire molds. A release coating is sometimes applied to the surface of the uncured tire to allow the cured tire to be suitably removed from the mold. It is both economically and physically desirable to provide a durable release coating on a tire mold surface itself to provide a releasability of the molded tire from the tire mold over repeated tire molding and curing cycles.

[0004] Further, by repeated tire molding and curing cycles, the mold surface of a metal tire mold may be subject to a buildup of materials, or debris, on its surface. Buildup of materials on the mold surface may sometimes be referred to mold fouling which may, for example, affect cosmetic appearance of the surface of the molded tire. Such fouled metal tire mold surface may be periodically cleaned to remove such mold fouling when appropriate. Such tendency for mold fouling is well known to those having skill in such art.

[0805] Therefore, in practice, it may be desired to promote releasability of sulfur cured rubber products such as tires, as well as various industrial product such as, for example belts, hoses, track pads and shoes, from their mold surfaces and, further, to retard, diminish or substantially prevent or release mold fouling of the surface of the rubber (e.g., tire) mold. [0806] For this invention, it desired to evaluate a plasma induced polymer coating of a metal mold surface for molding rubber products (e.g., tires) to promote releasability of a cured rubber product.

#### 35 Summary and Practice of the Invention

[0007] The invention relates to a method in accordance with claim 1, to a mold in accordance with claim 13 and to methods in accordance with claims 14 and 15 respectively.

[0008] Dependent claims refer to pireferred embodiments of the invention:

[0009] In one aspect, the present invention is directed to a method of treating a metal (e.g. steel or aluminum) mold surface for molding and curing rubber products (e.g. tires) which comprises applying a thin polymer layer to the surface of the mold to facilitate removal of a rubber product having been cured with in such mold.

100101 A clurality of such coatings on the metal mold surface may also be provided.

[0011] Further, a mold for rubber products (e.g. tires) is provided with a coaled metal surface (e.g. steel or aluminum surface) prepared by said method.

[0012] In addition, a method of molding and curing a rubber product (e.g. pneumatic rubber tire) is provided.

[0013] In accordance with this invention, a method of treating a metal surface of a mold for molding and curing rubber products is provided which comprises directing an almospheric plasma jet comprising an ionized gas containing a polymerizable precursor (e.g. atomized liquid polymerizable precursor) to a metal molding surface of the mold to form and bond a thin polymer layer of plasma polymerized precursor on said metal mold surface to promote release of a cured rubber product from said treated mold surface.

[0014] In one embodiment, said ionized gas for said plasma is electrically ionized by, for example, allowing said gas to pass between a set of positive and negative electrodes (e.g. an electric arc) to cause the gas to be in a high state of high energy environment to thereby promote polymerization of the introduced polymerizable precursor.

[0615] Such deposited polymer on said metal mold surface from said plasma is in a form of a polymer layer having a thickness, for example, in a range of from 10 nm to 500 nm, preferably 20 nm to 100 nm, and is provided from polymerizable precursors so long the resulting polymer has an ability to promote releasability of a cured rubber product from the deposited polymer layer on the metal mold surface.

[0015] Representative of such rubber products are pneumatic rubber tires and engineered (e.g. industrial) rubber products comprising for example belts (e.g. power and transmission belts), hoses, vehicular track pads and shoes.

[0017] It is to be appreciated that the atmospheric plasma jets may be in a form of pulse or continuous plasmas. The plasmas may be generated, for example, from dietectric barrier jets, microwave torches, radio-frequency torches, low frequency torches, current-carrying are plasma jets, or by any other means of producing a suitable atmospheric plasma able to polymerize organic precursors of polymers known to one skilled in such art.

[0018] While there are additional methods of creating an almospheric plasma jet, a current-carrying arc plasma jet, appears to be an acceptable means of applying the thin polymer layer to the metal moid surface.

[0619] In practice, to the ionized gas may be introduced a small particle sized (e.g. atomized) liquid polymerizable precursor (polymer precursor, namely a precursor to form a polymer) to form a plasma jet, expelled through and from a torch, comprised of the ionized gas and atomized precursor for which polymerization of the precursor is promoted by the high energy environment provided by the plasma.

[0920] The plasma jet is directed (e.g. directed from its torch) to a molding surface of the steel or aluminum mold to form and deposit a thin polymer layer (e.g. coaling) from the polymerized precursor on said steel surface of the mold by polymerization of said polymerizable precursor.

[0021] In one embodiment, it is desired that the polymer layer is bonded to the sleet or aluminum mold surface in a sense of chemical bonding of the polymerized coating to the surface of the steet or aluminum is created as a result of the high plasma energy environment.

[0022] In one embodiment, it is desired that the polymer layer resists significant adhesion of the rubber to thereby permit the molded and cured rubber to be releasable from the mold surface, (e.g. to permit removal of the molded rubber product from the mold surface).

[0023] In one embodiment, it is desired that the polymer comprising a crosslinked polymer in a sense of it has been observed that the plasma polymerized coalings usually exhibit some degree of crosslinking. In this manner, it is believed that a degree of crosslink density within the polymer coating promotes mechanical strength to withstand abrasion when taking cured tires out of the mold.

[0024] By atmospheric pressure plasma, it is meant that the pressure of the plasma is equal to or slightly above the ambient pressure of the surroundings. The pressure of the plasma may be desirably somewhat higher than ambient, such that the plasma pressure is sufficient to induce the desired flow rate through a plasma generator which contains an ionized gas promoting electric arc.

[0025] An ionizable gas (e.g. carrier gas for the polymerizable polymer precursor) may be, for example a noble gas such as, for example helium, argon, xenon and neon as well as, for example, oxygen, nitrogen, carbon dioxide, nitrous oxide and air. In one embodiment, such ionizable gas is argon.

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[0026] A carrier gas to carry the atomized liquid polymerizable precursor to the plasma plume may be for example also a noble gas such as, for example, helium, argon, xenon and neon as well as, for example, oxygen, nitrogen, carbon dioxide, nitrous oxide and air. In one embodiment such carrier gas is argon.

[0027] Representative of such polymerizable precursors may be comprised of, for example, and not intended to be limiting, one or more of disiloxanes, disilazanes, perfluoroacrylates (e.g. perfluoroacrylate and perfluoroacrylate), fluorinated silanes, vinylidene fluoride.

[0028] For example, such polymerizable precursors may be a dislipxane such as, for example, hexamethyldisiloxene; a dislipzane such as for example hexamethyldisilozane; fluorinated silane comprising, for example, 1 H, 1 H, 2H-per-fluorocityltriethoxysilane and other fluorinated compounds comprising at least one of 1 H, 1 H, 2H-per-fluoro-1-alkylene where said alkylene group comprises of decene, octene, hexane or buttene groups or 1 H, 1 H, 2H-per-fluoro-1-alkanol where said alkanol group comprises of decanol, octanol, hexanol or buttened groups and fluorinated compounds such as, for example per-fluoromethylcyclohexane, per-fluorodimethylcyclohexane, hexafluorobenzene, per-fluoromethyl vinyl ether, per-fluorocityl vinyl ether, per-fluorocityl vinyl ether, per-fluorocityl ethylene.

[0029] Such dislovane, disitazane, fluorinated silane, vinytidene fluoride and fluorinated compounds may be combined, if desired and appropriate with said perfluoroacrylates to form a precursor for said plasma coating method and to thereby form a copolymer thereof.

(0030) The practice of the method of this invention is considered as being significant in a sense of being able to refresh the coating on the metal mold surface at a rubber product (e.g. pneumatic tire) manufacturing facility, particularly where a thin coating is desired to be applied by an energy and polymer precursor efficient method.

[0031] For a batch process wherein an object is exposed to an atmospheric plasma, the object may be exposed, for example, to the plasma for a period of from 1 to 100 seconds, in a continuous process, the exposure time may be characterized by a residence time expressed as the length or circumference of the object to be treated (e.g. in centimeters) with the plasma torch divided by the torch velocity (e.g. in cm/sec) multiplied by the number of plasma treatments applied to the object. The residence time in such a continuous process would then range, for example, from 0.5 to 100 seconds. [0032] The flow rate of atomized precursor into the plasma generator necessary to obtain an effective amount of

polymerized or partially polymerized precursor onto the moid surface will depend on the desired face velocity in the plasma generator, i.e. the gas velocity (e.g. in cm/sec) exiting the plasma torch. Necessary flow rate may be determined by one skilled in the art without undue experimentation.

[0033] The following drawings are provided for further understanding of the invention and are not intended to be limiting.

#### The Drawings

[0034] A method is depicted for providing a thin polymer layer for enhancing releasability of suifur cured rubber from a steel or aluminum surface, such as for example a metal surface of a mold for rubber products, such as for example tires and industrial products such as for example, belts, hoses, tracks and shoes, particularly a tire mold.

#### In the Drawings

[0035] In Figure 1 a steel plate 1, is sandblasted.

[0036] To a cylindrical torch 2 device is fed a stream 3 of argon gas in which the argon gas is caused to pass through between positive and negative electrodes to ionize the argon gas to a high energy level at a flow rate through the torch 2 device of about 800 liters/hour (l/hr).

[0037] To the torch 2 device is iniroduced a stream of atomized precursor 4 carried by a carrier gas 5, namely argon gas where the atomized precursor 4 stream, namely a polymer precursor, is introduced into the stream 3 of argon gas, at substantially ambient atmospheric pressure, in the nozzle 6 of the torch 2 device to create a plasma jet 7 comprising the stream 3 of ionized gas which contains the atomized precursor 4.

[0038] The plasma jet 7 is passed over the steel plate 1 to deposit a thin polymer coating 8 onto the steel plate 1 created by the atomized precursor 4 contained in the stream 3 of ionized argon gas.

[0039] The invention is further described with reference to the following examples.

#### EXAMPLE

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[0040] In this example, treatments of sandblasted steel plates by an atmospheric plasma jet by which a polymerizable precursor is introduced are evaluated for providing a coating of polymer on the steel surface and compared to a steel plate without such treatment.

[0041] The plasma is provided as an electrical arc-formed ionized argon gas to which an atomized liquid polymerizable precursor is introduced. A polymerization of the precursor is promoted by the highly ionized argon gas in a sense being in a nature of what is believed to be a free radical polymerization process.

[0042] The liquid polymerizable precursor used was comprising 70 weight percent hexamethyldisiloxane (HMDSO) and 30 weight percent 1H,1H,2H,2H-periluoroociyl acrylate (PFOA).

[0043] For this evaluation, 1 cm by 3 cm steel plates were positioned on a table and held stationary while the plasma jet was applied by the plasma torch device where the plasma jet was comprising the high energy ionized argon gas containing the atomized liquid polymerizable precursor for which one or more passes of the torch device, and thereby the jet, over the surface of the steel plates.

[0044] An electrical arc power of 70 Watts was provided through which the argon gas was passed within the torch device to cause its ionization.

[0045] The argon gas flow rate through the torch was about 800 liters/hour to which an atomized liquid polymerizable precursor is injected to form the plasma torch.

[0046] The plasma torch comprised of the ionized argon gas and polymerizable precursor is directed to and thereby applied to the surface of the steel plate in one and three passes. The period of exposure for each pass was about 15 seconds.

[0047] Three samples of the steel plate were evaluated and identified herein as Samples A, B and C.

[0048] The surface of all of the steel Samples had been pre-sandblasted with a fine grain sand.

[0049] Control Sample A was a steel plate without exposure by the plasma torch.

50 [0050] Experimental Sample B was a steel plate to which one pass of the plasma torch was applied to deposit the precursor in a form of a polymer layer on the steel plate.

[0051] Experimental Sample C was a steel plate to which three individual passes of the plasma torch was applied to deposit the precursor in a form of a polymer layer on the steel plate in a more homogeneous form therefore reducing the occurrence of pin holes.

[0052] To the steel plate Control Sample A and Experimental polymer coated Samples B and C were applied suffor curable rubber blocks and the rubber blocks cured under conditions of elevated temperature and pressure against the Control and Experimental steel plates (for example at a temperature of about 155°C for about 35 minutes).

[0053] After cooling to room temperature (about 23°C), the energy needed to pull the cured rubber blocks from each

of the steel plate Control Sample A and polymer coated Experimental steel plates B and C was measured and reported in the following Table 1.

#### Table 1

Sample	Coating Passes	Average Separation Energy, milliJoules (mJ)
A (Centrol)	Ö	1494
8 (Experimental)	4.	438
C (Experimental)	3	112

[0054] From Table 1 it can be seen that the energy needed to pull the cured rubber off of the steel plate drastically decreases when a plasma coating is applied as compared to the reference, or control, steel plate. Higher number of passes leads to a more homogeneous coating and therefore better release performances.

[0055] This is considered as being significant in a sense of that a thin plasma coated polymerized layer plays the role of a release coating.

[0056] Therefore, it is concluded that the level of performance of aforementioned plasma polymers as a release coating is similar to the one obtained with other more traditional release coatings. In addition, the plasma coating process can be done onsite using minimal amounts of energy, chemicals and handling while providing flexibility, on-demand mold coating as well as the possibility to automate the plasma coating process.

#### Claims

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- A method of treating a metal surface of a mold for molding and curing rubber products, the method comprising applying an atmospheric plasma comprising an ionized gas containing a polymerizable precursor to a metal molding surface of the mold metal surface to form and bond a polymer layer of plasma polymerized precursor on the mold metal surface.
- The method of claim 1 wherein the metal mold surface comprises steel or aluminum.
  - The method of claim 1 or 2 wherein the mold is a mold for molding and curing pneumatic rubber tires or rubber products comprising belts, hoses, vehicular track pads or shoes.
- 4. The method of at least one of the previous claims wherein the polymerizable precursor is an atomized liquid polymerizable precursor contained in a partier gas for the precursor and introduced into said ionized gas.
  - 5. The method of claim 1 or 4, wherein the carrier gas for said polymerizable precursor and/or the ionized plasma gas comprises at least one of argon, helium, neon, xenon, oxygen, nitrogen, nitrous oxide and carbon dioxide.
- 46 8. The method of at least one of the previous claims wherein the precursor is a polymerizable monomer comprising at least one of disiloxenes, disiloxenes, perfluoroacryletes, fluorinated silones, vinylidene fluoride.
- 7. The method of at least one of the previous claims wherein the precursor is a polymerizable monomer comprising at least one of alkyldisiloxane comprising hexamethyldisiloxane; disilazane comprising hexamethyldisilazane; fluorinated silane comprising 1 H, 1 H, 2H-perfluorocctyltriethoxysilane, 1 H, 1 H, 2H-perfluoro-1-alkylene wherein the alkylene group comprises decene, octene, hexane or butene groups, or 1 H, 1 H, 2H-perfluoro-1-alkanol wherein the alkanol group comprises of decanol, octanol, hexanol or butanol groups; vinylidene fluoride and fluorinated compounds comprising at least one of perfluoromethylcyclohexane, perfluorodimethylcyclohexane, hexafluorobenzene, perfluoromethyl vinyl ether, perfluoromethylcyclobutane, perfluoromethylcyclopentane, perfluoromethyldecalin, tetrafluoroethylene, hexafluoropropylene and perfluorobulyl ethylene.
  - 8. The method of at least one of the previous claims wherein the precursor is a perfluoroacrylate comprising at least one of perfluoroactylate and perfluoroactylate.
  - 9. The method of claim 1 or 7 wherein the precursor is combined with a perfluorcacrylate.

- 10. The method of claim 9 wherein the perfluoroacrylate comprises at least one of perfluoroacrylate and perfluoroacrylate.
- 11. The method of at least one of the previous claims wherein the precursor is a dislloxane comprising hexamethyldisilazane.
  - 12. The method of at least one of the previous claims wherein the precursor is a fluorinated compound comprising at least one of 1 H, 1 H, 2H-perfluorocctyltriethoxysilane, 1 H, 1 H, 2H-perfluoro-1-alkylene, 1 H, 1 H, 2H-perfluoroctyltriethoxysilane, perfluorocethylcyclohexane, perfluoroctyltriethoxysilane, perfluoroctyltriethoxysilane, perfluorocethylcyclopentane, perfluoro
  - 13. A mold for molding and curing rubber products prepared by the method of at least one of the previous claims.
- 15 14. A method of preparing a rubber product comprising molding and curing a rubber product in the mold of claim 43.

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15. A method of preparing a pneumatic rubber tire comprising molding and curing a pneumatic rubber tire in the mold of claim 13.

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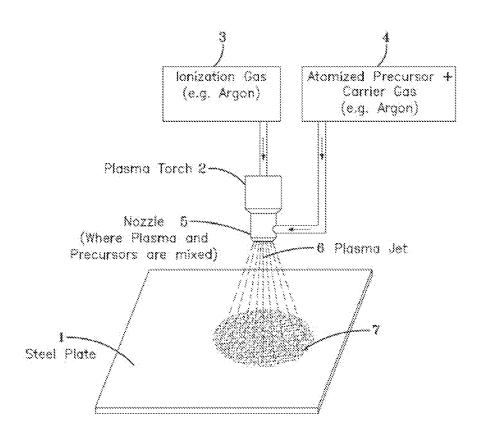


FIG-1



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## UK Patent Application (B) GB (B) 2 434 368

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(71) Applicant(s): P2i Limited (incorporated in the United Kingdom) Unit 14, Central 127, Milton Park, ASINGDON, Oxfordshire, OX14 4SA, United Kingdom

(72) Inventor(s): Stephen Coulson

(74) Agent and/or Address for Service: Greaves Brewster Indigo House, Cheddar Business Park, Wedmore Road, CHEDDAR, Somerset, BS27 3E8, United Kingdom

(51) INT CL

C08F 2/52 (2006.01) COSJ 7/16 (2006.01) D06M 14/18 (2006.01) B29C 59/14 (2006.01) DO6M 10/02 (2006.01)

(52) UK CL (Edition X ): COP PKN C3W W227

(56) Documents Cited: GB 2341864 A WO 2005/089961 A1 WO 2003/063938 A1

(SS) Field of Search: Other: ONLINE: CAS ONLINE, EPODOC, WPI.

#### (54) Abstract Title: Polymeric coating of a laboratory consumable

(57) A laboratory consumable having a polymeric coating, formed by exposing said consumable to pulsed plasma comprising a compound of formula (I)

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R^1 \\
R^2
\end{array}$$

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R^3 \\
R^4
\end{array}$$
(1)

where R1, R2 and R3 are independently selected from hydrogen, alkyl, haloalkyl or aryl optionally substituted by halo; andR⁴ is a group X-R⁵ where R⁵ is an alkyl or helicalkyl group and X is a bond; a group of formula -C(O)O(CH_e)_nY- where n is an integer of from 1 to 10 and Y is a bond or a sulphonamide group; or a group  $-(O)_{c}R^{6}(O)_{o}(CH_{2})_{r}$  where  $R^{6}$  is any optionally substituted by halo, p is 0 or 1, q is 0 or 1 and t is 0 or an integer of from 1 to 10, provided that where g is 1, t is other than 0, for a sufficient period of time to allow a polymeric layer to form on the surface of the laboratory consumable.

Consumables of this type show significantly reduced liquid sample retention, maximum sample recovery and ultra non-binding properties, as well as oil and water-repellency.

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#### Novel Products

The present invention relates to novel products in the form of laboratory consumables, which are treated to prevent retention of reagents on the surface thereof, as well as to processes for their production.

There is a need for laboratory equipment and in particular laboratory consumables, to be as non-binding to reagents or 10 samples as possible so as to avoid retention of these on the equipment. In particular, the biochemical and chemical reactions conducted for example for research or analysis, and in particular medical or forensic diagnostics, often use very small volumes of samples or reagents. Retention of these in 15 any significant proportion on the surface of laboratory equipment, which comes into direct contact with these, used to handle the samples or reagents used in these types of reaction can affect the results significantly. Furthermore, since many of the reagents are extremely costly, there are economic 20 implications as a result of reagent loss in this way.

Various approaches have been made to address this problem.

These include processes aimed at creating ultra smooth surfaces or solid materials such as those used to produce pipette tips in order to reduce retention. Other approaches have included processes aimed at adding additives to migrate through the polymer dip coating the tips in a liquid repellent media.

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Plasma deposition techniques have been quite widely used for the deposition of polymeric coatings onto a range of surfaces, and in particular onto fabric surfaces. This technique is recognised as being a clean, dry technique that generates little waste compared to conventional wet chemical methods. Using this method, plasmas are generated from organic molecules, which are subjected to an electrical field. When this is done in the presence of a substrate, the radicals of

the compound in the plasma polymerise on the substrate.

Conventional polymer synthesis tends to produce structures containing repeat units that bear a strong resemblance to the monomer species, whereas a polymer network generated using a plasma can be extremely complex. The properties of the resultant coating can depend upon the nature of the substrate as well as the nature of the monomer used and conditions under which it is deposited.

- The applicants have found that by utilising a specific type of monomer under particular deposition conditions, laboratory consumables having ultra-non-binding nano-coatings thereon can be produced.
- 15 According to the present invention there is provided a laboratory consumable having a polymeric coating, formed by exposing said consumable to pulsed plasma comprising a compound of formula (I)

$$R^{1}$$
 $R^{2}$ 
 $R^{3}$ 
 $R^{4}$ 
 $R^{3}$ 

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where R¹, R² and R³ are independently selected from hydrogen, alkyl, haloslkyl or aryl optionally substituted by halo; and R⁶ is a group X-R⁵ where R⁵ is an alkyl or haloslkyl group and X is a bond; a group of formula -C(O)O(CH₂),Y- where n is an integer of from 1 to 10 and Y is a bond or a sulphonamide group; or a group -(O)_PR⁶(O)_q(CH₂)₁- where R⁶ is aryl optionally substituted by halo, p is 0 or 1, q is 0 or 1 and t is 0 or an integer of from 1 to 10, provided that where q is 1, t is other than 0, for a sufficient period of time to allow a polymeric layer to form on the surface of the laboratory consumable.

As used herein, the expression "in a gaseous state" refers to gases or vapours, either alone or in mixture, as well as aerosols.

- As used herein, the term "laboratory consumable" refers to any piece of equipment, which comes into contact with liquid reagents, polar and apolar solvents or samples in the course of normal laboratory use, including use for example in medical environments such as hospitals and research laboratories.
- Generally these will be single use disposable devices, although some may be used repeatedly, for example after washing or cleaning of some type.
- Particular examples of laboratory consumables include pipette tips, filtration membranes, microplates (including 96 well plates), immunoassay products (such as lateral flow devices), centrifuge tubes (including microcentrifuge tubes), microtubes, specimen tubes, test tubes, blood collection tubes, flat based tubes, aseptically produced containers, general labware,
- 20 burettes, curvettes, needles, hypodermic syringes, sample vials/bottles, screw cap containers, or weighing bottles. These can be made from a wide variety of materials such as, thermosetting resins, thermoplastic resins polyolefins, acetals, polyamidic resins, acrylic resins (PMMA), hydrocarbons
- or fluorocarbons such as polyethylene (PE), polypropylene (PP), polymethylpentene (PMP or TPX®), polystyrene (PS), polyvinyl chloride (PVC), polyoxymethylene (POM), nylon (PA6), polycarbonates (PC), polytetrafluoroethylene (PTFE), tetrafluoroethylene-perfluoropropylene (FEP), perfluoroalkoxy
- 30 (PFA), polyvinylidene fluoride (PVDF), ethylenetetrafluorethylene (ETFE), ethylene-cholortrifiuoroethylene (E-CTFE) and glass.

In a particular embodiment, the laboratory consumable is a pipette tip. By utilising the method defined above, highly advantageous pipette tips have been produced. In particular, the main features and benefits of using this approach are that both the inside and the outside of the tip are enhanced, allowing much less binding, reducing loss of liquid and maximizing sample recovery.

Laboratory consumables treated in this way provide are water and oil repellent, and also resistant to clogging. They may have useful "shake dry" properties also, reducing risk of contamination after washing.

Furthermore, the enhancement material is molecularly bound to
the surface and so there are no leachables; the modification
becomes part of the device. The coated tip's surface displays
much lower surface energies then polypropylene and so will bind
less material. As a result, liquid retention by the pipette
tip in use is virtually eliminated (as can be illustrated

20 photographically) meaning that the results obtained using these
tips are more accurate and reliable, and reagent costs can be
reduced, in some cases by more than 10%.

Similar advantages accrue in relation to other laboratory consumables which are used to handle in particular small volumes of sample or reagent, such as microtitre plates. microplates (including 96 well plates), centrifuge tubes (including microcentrifuge tubes) and microtubes.

- 30 The fact that needles and syringes can be made so that virtually no liquid is retained therein is a significant advantage since it means that accurate dosing of materials, in particular medicines and the like, can be ensured.
- 35 Another type of laboratory consumable to benefit significantly from the modification of the invention are membranes such as

those used as filtration membranes or those used in immunoassay devices. These membranes are made of porous material. It is necessary for liquid to pass through the pores of the material. In the case of filtration membranes, particles above a particular size are retained on the surface. In the case of membranes used immunoassays, liquid and small particles such as gold or latex particles are required also to pass through the pores of the membrane. Although these membranes are able to fulfil these functions, they are generally made of a liquid absorbent material such as cellulose. This means that significant quantities of liquid will be retained within the membrane, as they will become saturated in use. This can be problematic, in particular where only small volumes of liquid are being used.

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Membranes treated in accordance with the invention retain their porosity, as the coating layer deposited thereon is only molecules thick. Therefore, the liquid or even small particles can continue to pass through them, in particular when a positive pressure is applied to the liquid, or a negative pressure is applied to the side of the membrane to draw the liquid through. However, larger particles will not. However, substantially all the liquid applied will pass through the membrane, since it will not adhere or be absorbed by the material of the membrane.

Precise conditions under which the plasma polymerization takes place in an effective manner will vary depending upon factors such as the nature of the polymer, the laboratory consumable etc. and will be determined using routine methods and/or the techniques.

Suitable plasmas for use in the method of the invention include non-equilibrium plasmas such as those generated by radiofrequencies (Rf), microwaves or direct current (DC). They may operate at atmospheric or sub-atmospheric pressures as are

known in the art. In particular however, they are generated by radiofrequencies (Rf).

Various forms of equipment may be used to generate gaseous plasmas. Generally these comprise containers or plasma chambers in which plasmas may be generated. Particular examples of such equipment are described for instance in WO2005/089961 and WO02/28548, but many other conventional plasma generating apparatus are available.

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In general, the item to be treated is placed within a plasma chamber together with the material to be deposited in gaseous state, a glow discharge is ignited within the chamber and a suitable voltage is applied, which may be pulsed.

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The gas used within the plasma may comprise a vapour of the monomeric compound alone, but it may be combined with a carrier gas, in particular, an inert gas such as helium or argon. In particular helium is a preferred carrier gas as this can minimises fragmentation of the monomer.

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When used as a mixture, the relative amounts of the monomer vapour to carrier gas is suitably determined in accordance with procedures which are conventional in the art. The amount of monomer added will depend to some extent on the nature of the particular monomer being used, the nature of the laboratory disposable being treated, the size of the plasma chamber etc. Generally, in the case of conventional chambers, monomer is delivered in an amount of from 50-250mg/min, for example at a rate of from 100-150mg/min. Carrier gas such as helium is suitably administered at a constant rate for example at a rate of from 5-90, for example from 15-30sccm. In some instances, the ratio of monomer to carrier gas will be in the range of from 100:1 to 1:100, for instance in the range of from 10:1 to 1:100, and in particular about 1:1 to 1:10. The precise ratio

selected will be so as to ensure that the flow rate required by the process is achieved.

In some cases, a preliminary continuous power plasma may be struck for example for from 2-10 minutes for instance for about 4 minutes, within the chamber. This may act as a surface pretreatment step, ensuring that the monomer attaches itself readily to the surface, so that as polymerisation occurs, the coating "grows" on the surface. The pre-treatment step may be conducted before monomer is introduced into the chamber, in the presence of only the inert qas.

The plasma is then suitably switched to a pulsed plasma to allow polymerisation to proceed, at least when the monomer is present.

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In all cases, a glow discharge is suitably ignited by applying a high frequency voltage, for example at 13.56MHz. This is suitably applied using electrodes, which may be internal or external to the chamber, but in the case of the larger chambers are internal.

Suitably the gas, vapour or gas mixture is supplied at a rate of at least 1 standard cubic centimetre per minute (sccm) and preferably in the range of from 1 to 100sccm.

In the case of the monomer vapour, this is suitably supplied at a rate of from 80-300mg/minute, for example at about 120mg per minute depending upon the nature of the monomer, whilst the pulsed voltage is applied.

Gases or vapours may be drawn or pumped into the plasma region. In particular, where a plasma chamber is used, gases or vapours may be drawn into the chamber as a result of a reduction in the pressure within the chamber, caused by use of an evacuating

pump, or they may be pumped or injected into the chamber as is common in liquid handling.

Polymerisation is suitably effected using vapours of compounds of formula (I), which are maintained at pressures of from 0.1 to 200mtorr, suitably at about 80-100mtorr.

The applied fields are suitably of power of from 40 to 500W, suitably at about 100W peak power, applied as a pulsed field.

The pulses are applied in a sequence which yields very low average powers, for example in a sequence in which the ratio of the time on: time off is in the range of from 1:500 to 1:1500. Particular examples of such sequence are sequences where power is on for 20-50µs, for example about 30µs, and off for from 1000µs to 30000µs, in particular about 20000µs. Typical average powers obtained in this way are 0.01W.

The fields are suitably applied from 30 seconds to 90 minutes, preferably from 5 to 60 minutes, depending upon the nature of the compound of formula (I) and the laboratory consumable etc.

Soitably a plasma chamber used is of sufficient volume to accommodate multiple laboratory consumables, for example up to 500,000 pipette tips at the same time.

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A particularly suitable apparatus and method for producing laboratory consumables in accordance with the invention is described in WO2005/089961, the content of which is hereby incorporated by reference.

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In particular, when using high volume chambers of this type, the plasma is created with a voltage as a pulsed field, at an average power of from 0.001 to  $500\text{w/m}^3$ , for example at from 0.001 to  $100\text{w/m}^3$  and suitably at from 0.005 to  $0.5\text{w/m}^3$ .

These conditions are particularly suitable for depositing good quality uniform coatings, in large chambers, for example in chambers where the plasma zone has a volume of greater than  $500\,\mathrm{cm}^3$ , for instance  $0.5\mathrm{m}^3$  or more, such as from  $0.5\mathrm{m}^3-10\mathrm{m}^4$  and suitably at about  $\mathrm{Im}^3$ . The layers formed in this way have good mechanical strength.

The dimensions of the chamber will be selected so as to accommodate the particular laboratory consumable being treated.

10 For instance, generally cuboid chambers may be suitable for a wide range of applications, but if necessary, elongate or rectangular chambers may be constructed or indeed cylindrical, or of any other suitable shape.

The chamber may be a sealable container, to allow for batch processes, or it may comprise inlets and outlets for laboratory consumables, to allow it to be utilised in a continuous process. In particular in the latter case, the pressure conditions necessary for creating a plasma discharge within the chamber are maintained using high volume pumps, as is conventional for example in a device with a "whistling leak". However it will also be possible to process certain items at atmospheric pressure, or close to, negating the need for "whistling leaks"

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The monomers used are selected from monomers of formula (I) as defined above. Suitable haloalkyl groups for  $R^1$ ,  $R^2$ ,  $R^3$  and  $R^3$  are fluoroalkyl groups. The alkyl chains may be straight or branched and may include cyclic moleties.

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For R⁵, the alkyl chains suitably comprise 2 or more carbon atoms, suitably from 2-20 carbon atoms and preferably from 6 to 12 carbon atoms.

35 For  $R^{2}$ ,  $R^{2}$  and  $R^{3}$ , alkyl chains are generally preferred to have from 1 to 6 carbon atoms.

Preferably  $R^5$  is a haloalkyl, and more preferably a perhaloalkyl group, particularly a perfluoroalkyl group of formula  $C_nF_{2m+1}$  where m is an integer of 1 or more, suitably

5 from 1-20, and preferably from 4-12 such as 4, 6 or 8.

Suitable alkyl groups for  $R^3$ ,  $R^4$  and  $R^3$  have from 1 to 6 carbon atoms.

- In one embodiment, at least one of  $R^1$ ,  $R^2$  and  $R^3$  is hydrogen. In a particular embodiment  $R^1$ ,  $R^2$ ,  $R^3$  are all hydrogen. In yet a further embodiment however  $R^3$  is an alkyl group such as methyl or propyl.
- Where X is a group -C(O)O(CH₂)_nY-, n is an integer which provides a suitable spacer group. In particular, n is from 1 to 5, preferably about 2.
- Suitable sulphonamide groups for Y include those of formula  $20 \text{ N(R}^3)\,\text{SO}_2$  where R is hydrogen or alkyl such as  $C_{1-4}$  alkyl, in particular methyl or ethyl.

In one embodiment, the compound of formula (I) is a compound of formula (II)

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$$CH_2=CH-R^5$$
 (II)

where R' is as defined above in relation to formula (I).

30 In compounds of formula (II), X in formula (I) is a bond.

However in a preferred embodiment, the compound of formula (I) is an acrylate of formula (III)

35  $CH_2 = CR^7C(O)O(CH_2)_nR^5$  (III)

where n and  $R^5$  as defined above in relation to formula (I) and  $R^7$  is hydrogen,  $C_{1-10}$  alkyl, or  $C_{1-10}$ haloalkyl. In particular  $R^7$  is hydrogen or  $C_{1-6}$ alkyl such as methyl. A particular example of a compound of formula (III) is a compound of formula (IV)

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$$\stackrel{\mathsf{H}}{\longrightarrow} \stackrel{\mathsf{O}}{\longrightarrow} 0 \stackrel{(\mathsf{CF}_2)_{\mathsf{X}}}{\longrightarrow} \mathsf{CF}_3 \qquad (\mathsf{IV})$$

where R⁷ is as defined above, and in particular is hydrogen and x is an integer of from 1 to 9, for instance from 4 to 9, and 10 preferably 7. In that case, the compound of formula (IV) is 1H,1H,2H,2H-heptadecafluorodecylacylate.

In a further aspect, the invention provides a method for reducing liquid sample retention by a laboratory consumable, said method comprising exposing said consumable to a pulsed plasma comprising a compound of formula (I)

$$R^{1}$$
 $R^{2}$ 
 $R^{3}$ 
 $R^{4}$ 

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where  $R^1$ ,  $R^2$  and  $R^3$  are independently selected from hydrogen, alkyl, haloalkyl or aryl optionally substituted by halo; and  $R^4$  is a group X-R⁵ where  $R^5$  is an alkyl or haloalkyl group and X is a bond; a group of formula  $-C(O)O(CH_2)_nY$ — where n is an integer of from 1 to 10 and Y is a bond or a sulphonamide group; or a group  $-(O)_nR^6(O)_q(CH_2)_z$ — where  $R^6$  is aryl optionally substituted by halo, p is 0 or 1, q is 0 or 1 and t is 0 or an integer of from 1 to 10, provided that where q is 1, t is other than 0,

in a gaseous state for a sufficient period of time to allow a polymeric layer to form on the surface of the laboratory consumable.

5 Suitably, the laboratory consumable is placed in a plasma deposition chamber, a glow discharge is ignited within said chamber, and a voltage applied as a pulsed field.

Suitable monomers and reaction conditions for use in this 10 method are as described above.

The invention will now be particularly described by way of example.

# 15 Example 1

### Pipette Tips

A set of polypropylene pipette tips were placed into a plasma chamber with a processing volume of ~ 300 litres. The chamber was connected to supplies of the required gases or vapours, via 20 a mass flow controller and/or liquid mass flow meter and a mixing injector as appropriate.

The chamber was evacuated to between 3 - 10 mtorr base pressure 25 before allowing helium into the chamber at 20 sccm until a pressure of 80 mtorr was reached. A continuous power plasma was then struck for 4 minutes using RF at 13.56 MHz at 300 W.

After this period, 1H,1H,2H,2H-heptadecafluorodecylacylate (CAS 30 # 27905-45-9) of formula

$$\stackrel{\text{H}}{\longrightarrow} \stackrel{\text{O}}{\longrightarrow} 0 \stackrel{\text{(CF}_2)_7}{\longrightarrow} \text{CF}_3$$

was brought into the chamber at a rate of 120 milli grams per minute and the plasma switched to a pulsed plasma at 30 micro seconds on-time and 20 milli seconds off-time at a peak power of 100 W for 40 minutes. On completion of the 40 minutes the plasma power was turned off along with the processing gases and vapours and the chamber evacuated back down to base pressure. The chamber was then vented to atmospheric pressure and the pipette tips removed.

10 It was found that the pipette tips was covered with an ultranon-binding, pin-hole free modification that had a surface
energy much lower that a Teflon® (PTFE) coated surface. The
coating was distributed both inside and outside the pipette tip
and was molecularly attached.

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The performance of the pipette tips was significantly enhanced, as liquid retention was virtually eliminated and maximum sample recovery was possible.

### 20 Example 2

#### Filter paper

filter papers (Whatman number 1) were placed into the chamber and processed under the same conditions. It was found that despite the fact the plasma enhanced paper appears the same, had the same handle and could not be differentiated either by appearance or feel; repelled water to an extent that the water formed beads on the surface which roll around effortlessly. Despite this, the droplets could be forced through the microporous paper if sufficient force was applied as the process does not block up the pores. The paper remained un-wet but can still carry out its filtering capabilities.

By using the same method as outlined in Example 1, cellulose

### Claims

 A laboratory consumable having a polymeric coating, formed by exposing said consumable to pulsed plasma comprising a compound of formula (I)

$$\begin{array}{c}
R^{1} \\
R^{2}
\end{array}$$

$$\begin{array}{c}
R^{3} \\
R^{4}
\end{array}$$

$$\begin{array}{c}
(0) \\
\end{array}$$

where R¹, R² and R³ are independently selected from hydrogen,

alkyl, haloalkyl or aryl optionally substituted by halo; andR²
is a group K-R³ where R³ is an alkyl or haloalkyl group and X
is a bond; a group of formula -C(O)O(CH₂)_nY- where n is an
integer of from 1 to 10 and Y is a bond or a sulphonamide
group; or a group -(O)₂R⁶(O)_q(CH₂)_c- where R⁶ is aryl optionally

substituted by halo, p is 0 or 1, q is 0 or 1 and t is 0 or an
integer of from 1 to 10, provided that where q is 1, t is other
than 0, for a sufficient period of time to allow a polymeric
layer to form on the surface of the laboratory consumable.

- 20 2. A laboratory consumable according to claim 1 which is selected from pipette tips, filtration membranes, microplates immunoassay products, centrifuge tubes, microtubes, speciman tubes, test tubes, blood collection tubes, flat based tubes, aseptically produced containers, general labware, burettes, curvettes, needles, hypodermic syringes, sample vials/bottles, screw cap containers, or weighing bottles.
  - 3. A laboratory consumable according to claim 2 which is a pipette tip.
  - 4. A laboratory consumable according to any one of the preceding claims wherein the laboratory consumable is exposed to the pulsed plasma within a plasma deposition chamber.

5. A laboratory consumable according to any one of the preceding claims wherein the compound of formula (I) is a compound of formula (II)

where  $R^b$  is as defined in claim 1, or a compound of formula (III)

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$$CH_0 = CR^3C(O)O(CH_2)_{11}R^5$$
 (III)

where n and  $R^5$  as defined in claim 1 and  $R^7$  is hydrogen,  $C_{1-10}$  alkyl, or  $C_{1-10}$ haloalkyl.

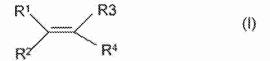
- 15 6. A laboratory consumable according to claim 5 wherein the compound of formula (1) is a compound of formula (III).
  - 7. A laboratory consumable according to claim 6 wherein the compound of formula (III) is a compound of formula (IV)

$$\begin{array}{c} H \\ \downarrow \\ R^{7} \end{array} O \\ O \\ CF_{2})_{x} \\ CF_{3} \end{array} \qquad (IV)$$

where  $\mathbb{R}^7$  is as defined in claim 4, and x is an integer of from 1 to 9.

- 8. A laboratory consumable according to claim 7 wherein the compound of formula (IV) is is 1H,1H,2H,2Hheptadecafluorodecylacylate.
- 30 9. A method for reducing liquid sample retention by a laboratory consumable, said method comprising exposing said consumable to a pulsed plasma comprising a compound of formula (I)

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- 5 where R¹, R² and R³ are independently selected from hydrogen, alkyl, haloalkyl or aryl optionally substituted by halo; and R⁴ is a group X-R⁵ where R⁵ is an alkyl or haloalkyl group and X is a bond; a group of formula -C(O)O(CH₂)_nY- where n is an integer of from 1 to 10 and Y is a bond or a sulphonamide
  10 group; or a group -(O)_pR⁶(O)_q(CH₂)_v- where R⁶ is aryl optionally substituted by halo, p is 0 or 1, q is 0 or 1 and t is 0 or an integer of from 1 to 10, provided that where q is 1, t is other
- in a gaseous state for a sufficient period of time to allow a 15 polymeric layer to form on the surface of the laboratory consumable.

than 0,

- 10. A method according to claim 9 wherein the laboratory consumable is placed in a plasma deposition chamber, a glow discharge is ignited within said chamber, and a voltage applied as a pulsed field.
  - 11. A method according to claim 9 or claim 10 wherein applied voltage is at a power of from 40 to 500W.
  - 12. A method according to any one of claims 9 to 11 wherein the voltage is pulsed in a sequence in which the ratio of the time on : time off is in the range of from 1:500 to 1:1500.
- 30 13. A method according to claim 12 wherein the voltage is pulsed in a sequence where power is on for  $20-50\mu s$ , and off for from  $1000\mu s$  to  $30000\mu s$ .

- 14. A method according to any one of claims 9 to 13 wherein the voltage is applied as a pulsed field at for a period of from 30 seconds to 90 minutes.
- 5 15. A method according to claim 14 wherein the voltage is applied as a pulsed field for from 5 to 60 minutes.
  - 16. A method according to any one of claims 9 to 15, wherein in a preliminary step, a continuous power plasma is applied to the laboratory consumable.
    - 17. A method according to claim 16 wherein the preliminary step is conducted in the presence of an inert gas.
- 15 18. A method according to any one of claims 9 to 17 wherein the compound of formula (I) in gaseous form is fed into the plasma at a rate of from 80-300 mg/minute, whilst the pulsed voltage is applied.
- 20 19. A method according to any one of claims 9 to 18 wherein the plasma is created with a voltage at an average power of from 0.001 to  $500\text{w/m}^3$ .
- 20. A method according to claim 19 wherein the plasma is created with a voltage at an average power of from 0.001 to  $100 \text{w/m}^3$ .
- 21. A method according to claim 20 wherein the plasma is created with a voltage at an average power of from 0.005 to  $30-0.5 \mathrm{w/m}^3$ .
  - 22. A method according to any one of claims 9 to 21 wherein wherein the compound of formula (I) is a compound of formula (II)

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where  $R^5$  is as defined in claim 1, or a compound of formula (III)

5 
$$CH_2=CR^{\gamma}C(O)O(CH_2)_{\alpha}R^{\beta}$$
 (III)

where n and  $R^b$  as defined in claim I and  $R^7$  is hydrogen,  $C_{1\sim 10}$  alkyl, or  $C_{1\sim 10}$ haloalkyl.

- 10 24. A method according to claim 23 wherein the compound of formula (I) is a compound of formula (III).
  - 25. A method according to claim 24 wherein the compound of formula (III) is a compound of formula (IV)

where R' is as defined in claim 4, and x is an integer of from 1 to 9.

- 26. A method according to claim 25 wherein the compound of formula (IV) is 1H, 1H, 2H, 2H-heptadecafluorodecyl acrylate.
- 27. A laboratory consumable substantially as hereinbefore described with reference to the examples.
  - 28. A method for reducing liquid sample retention by a laboratory consumable, substantially as hereinbefore described with reference to the Examples.

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Examiner:

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Claims searched:

1-28

Date of search:

21 April 2006

# Patents Act 1977: Search Report under Section 17

Documents considered to be relevant:

Category	Relevant to claims	Identity of document and passage or figure of particular relevance
Х		GB2341864 A (SECRETARY OF STATE FOR DEFENCE), see whole document, note particularly page 1 lines 3-12.
X	2 6	WO2005/089961 A1 (SECRETARY OF STATE OF DEFENCE), see whole document, note particularly page 1 lines 3-11.
Х		WO03/063938 A1 (ASTRAZENECA), see whole document, note particularly page 1 lines 7-11, page 2 lines 30-33, page 8 lines 10-18, Claims 1-6.

Categories:

X	Document indicating lack of novelty or inventive	Ą	Document indicating technological background and/or state
}	step		of the art.
¥	Document indicating lack of inventive step if combined with one or more other documents of	þ	Document published on or after the declared priority date but before the filling date of this invention.
&	same category. Member of the same patent family	E	Patent document published on or after, but with priority date
}.			earlier than, the filing date of this application

### Field of Search:

Search of GB, EP, WO & US patent documents classified in the following areas of the UKCX:

Worldwide search of patent documents classified in the following areas of the IPC

The following online and other databases have been used in the preparation of this search report

ONLINE: CAS ONLINE, EPODOC, WPL

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(71) Applicant(s):

P2i Ltd

(Incorporated in the United Kingdom) 127 (9-12 North) Millon Park, ABINGDON, Oxfordshire, OX14 4SA, United Kingdom

(72) Inventor(s):

Stephen Coulson Paul De Blaquiere Delwyn Evans Warren Lee

(74) Agent and/or Address for Service:

Greaves Brewster LLP Copa House, Station Road, CHEDDAR, Somerset, BS27 3AH, United Kingdom

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(56) Documents Cited: GS 2434379 A EP 1087034 A1 WO 2009/010753 A2

GB 1495480 A EP 1074588 AZ WO 2007/015779 A2

(58) Field of Search:

INTICL BOSD, COSF, COSJ, DOSM

Other: EPODOC; WPI

WO 2000/020130 A1

### (54) Title of the Invention: Surface coatings Abstract Title: Formation of a liquid repellent coating, using plasma polymerisation

(57) The plasma deposition process involves exposing a substrate to a monomer in a plasma polymerisation process. This allows a polymeric layer to form on the surface of the substrate. The process is subjected to at least one cycle of varying pressure. The preferred monomer is a mono-clefin, especially a compound of formula (I) or (II).

CH2=CHR'(I)

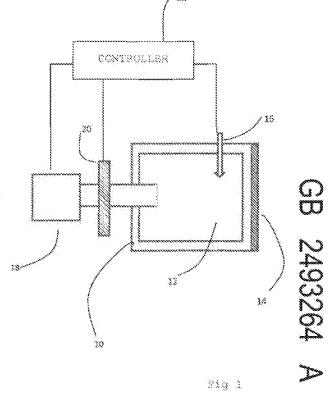
CH2=CR3C(O)O(CH2),R3 (II)

R1 = alkyl or haloalkyl;

R2 = H, 1-10C alkyl or 1-10C haloalkyl;

n = 1-10.

An especially preferred monomer is 1H,1H,2H,2Hheptadecafluorodecylacrylate. Preferably, pulsed plasma is used. The substrate is preferably fabric, metal, glass, ceramic, paper or polymer, especially an item of clothing



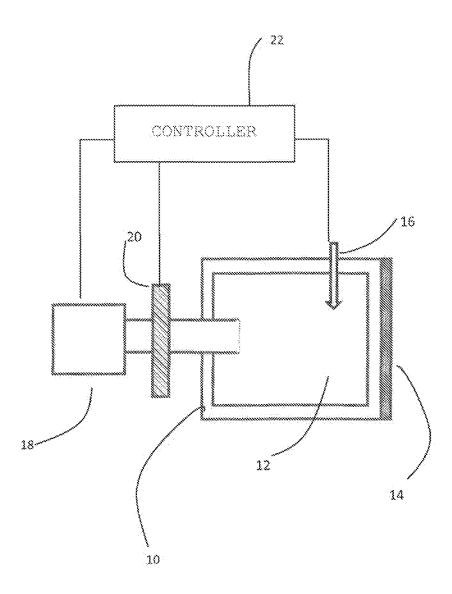


Fig 1

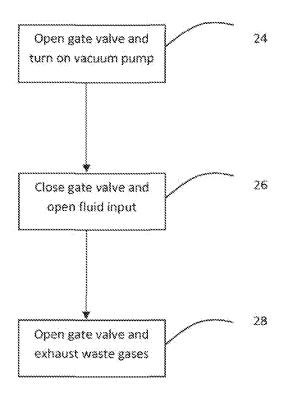


Fig 2

## Surface Coatings

The present invention relates to the coating of surfaces, in particular to the production of oil and water repellent surfaces, as well as to coated articles obtained thereby.

It is known to apply water and oil repellent coatings by a wide variety of methods. Such coatings typically include fluorocarbon chains, with the degree of oil and water repellency being a function of the number and length of fluorocarbon groups or moieties that can be fitted into the available space.

Plasma deposition techniques have been quite widely used for the deposition of polymeric coatings onto a range of surfaces. This 15 technique is recognised as being a clean, dry technique that generates little waste compared to conventional wet chemical methods. Using this method, plasmas are generated from small organic molecules, which are subjected to an ionising electrical field under low pressure conditions. When this is done in the 20 presence of a substrate, the ions, radicals and excited molecules of the monomer in the plasma polymerise in the gas phase and react with a growing polymer film on the substrate. Conventional polymer synthesis tends to produce structures containing repeat units which bear a strong resemblance to the 25 monomer species, whereas a polymer network, generated using a plasma can be extremely complex.

US patent 6,551,950 discloses use of plasma polymerisation to form an oil or water repellent surface using long chain hydrocarbons and fluorocarbons.

The applicants have discovered that by increasing the residency time of the monomer in the processing chamber, the quality of the coating and efficiency of its deposition can be increased.

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Residency time is the average amount of time that a particle spends in a particular system. Residency time can be defined by the equation

$$t = C/Q \tag{1}$$

5 where  $\tau$  is residence time, C is the capacity of the processing chamber and Q is the flow rate of the gas through the system at the pressure in the chamber.

From the above equation, it can be seen that for any given gas flow rate, an increase chamber size results in a longer residency time. Likewise, the slower the gas flow rate, the greater the residency time. Therefore, slow flow rates and large chambers will result in long residency times.

The longer a molecule remains in the processing chamber, the great the likelihood that it will undergo a deposition process, such as polymerisation, and attach to the surface.

A first aspect of the present invention provides a method of forming a liquid repellent coating on a surface of a substrate, said method comprising exposing said surface to a monomer in a plasma deposition process under conditions that maintain the monomer in situ for a period of time to allow a polymeric layer to form on the surface, wherein the conditions comprise at least one cycle of varying pressure.

The use of variable pressure cycling increases the residency time during which molecules of the monomer are prevalent in the processing chamber. This is due to the cycling keeping the pressure within the optimal range. It is desirable that the chamber remains within the optimal pressure range to maximise polymerisation.

30

In comparison to coating at static pressure, the use of variable pressure cycling improves the evenness of the coating, reduces

the processing time and uses less monomer to produce a given thickness of coating.

The processing chamber may be provided with valve such as a gas exhaust gate, which is closed during introduction of the monomer. The exhaust gate may be adjusted to open, closed or partially open.

The at least one cycle may comprise continually introducing the nonomer into the processing chamber (for example by injection) and allowing the pressure to rise. Exhaust gas may be expelled from the processing chamber at the end of each cycle.

The processing chamber may be at least partly evacuated at the 15 beginning of each cycle; this enables it to reach the low pressure optimal for the polymer coating to form.

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As gases are expelled at the end of each cycle, waste gases such as water vapour can be vented without affecting the low pressure achieved at the beginning of the cycle.

Whilst leaving the exhaust gate of the chamber closed throughout the process ensures that optimum pressure can easily be reached, a combination of monomer input and outgassing from the treated product causes the pressure within the chamber to build up until it exceeds optimum levels. By contrast, if the gate remains open for the duration of the process, outgassing products can be effectively removed; this allows the pressure to be brought down to optimum levels. However, a significant amount of the input monomer will simply flow through the chamber without undergoing polymerisation. The use of variable pressure cycling enables undesired outgassing products to be vented from the chamber but following venting the gate is closed to bring the chamber back to the bottom of the optimal pressure range. The closing of the gate at a particular point in the cycle is important to maintain

the monomer in situ (increase its residency time) so polymerisation can occur.

The at least one cycle of varying pressure may comprise a time based cycle. For example the processing chamber may be at least partly evacuated after a predetermined amount of time in each cycle. Time cycling is particularly beneficial in that it allows for the monomer to be retained in situ so increasing residency time which assists in improving the coating process.

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The at least one cycle of varying pressure may comprise a pressure based cycle. For example the processing chamber may be at least partly evacuated if the pressure falls outside the optimum range. The processing chamber may comprise pressure sensors to determine the pressure within the chamber. Feedback from the pressure sensor may be used to adjust an exhaust gate of the processing chamber. For example, the exhaust gate may be closed if the pressure drops below the optimum range and the exhaust gate may be opened if the pressure rises above the optimum range.

A pressure based system is suitable for low outgassing products, such as hearing aids and mobile telephones, as it produces a long cycle within the desired pressure range. For higher outgassing products, such as shoes, pressure based cycles can be too short. Bowever, a time based cycle can be designed which allows the cycle to be of good length, gives the monomer a good residency time, whilst still staying near the optimal pressure.

- 30 The variable pressure may be maintained below a maximum pressure. For example, the maximum pressure may be at or below 150 mTorr. Suitably, the maximum pressure may be at or below 125 mTorr.
- 35 Each cycle may be between 45 and 75 seconds. Each cycle may be approximately 60 seconds.

The method may comprise exposing the surface to two or more cycles of varying pressure and in particular up to four cycles. Said two or more cycles may comprise between 5 and 12 cycles. Alternatively, said two or more cycles may comprise 8 or 9 cycles.

In one embodiment, the deposition process is a gas process. The deposition process may be a plasma process, for example a plasma polymerisation process.

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Where the deposition process is a plasma polymerisation process, the coating may be applied using a pulsed plasma.

The liquid repellent coating may comprise an oil or water 15 repellent coating.

The polymeric layer may be uniform. However, it may also be advantageous to form a non uniform polymeric layer, for example where the coating is used in a bio array.

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The liquid repellent coating is suitable for surfaces on a wide range of substrates, for example fabric, metal, glass, ceramics, paper or polymer substrates. Items such as clothing (including footwear), laboratory consumables (including pipette tips).

filtration membranes, electronic devices (including mobile phones, audio equipments, laptop computers and hearing aids), microfluidic devices and photovoltaic modules (such as solar panels) can all suitably be treated using the method of this invention.

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Plasma polymers are typically generated by subjecting a coating forming precursor to an ionising electric field under low pressure conditions. Deposition occurs when excited species generated by the action of the electric field upon the precursor (radicals, ions, excited molecules etc) polymerise in the gas

phase and react with the substrate surface to form a growing polymer film.

Suitable plasmas for use in the method described herein include non-equilibrium plasmas such as those generated by radiofrequencies (RF), microwaves or direct current (DC). They may operate at atmospheric or sub-atmospheric pressures as are known in the art. In particular however, they are generated by radiofrequencies (RF).

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Various forms of equipment may be used to generate gaseous plasmas. Generally these comprise containers or plasma chambers in which plasmas may be generated. Particular examples of such equipment are described for instance in WO2005/089961 and WO02/28548, but many other conventional plasma generating apparatus are available.

In general, the item to be treated is placed within a plasma chamber together with the material to be deposited in gaseous state, a glow discharge is ignited within the chamber and a suitable voltage is applied, which may be pulsed.

The gas used within the plasma may comprise a vapour of the monomeric compound alone, but it may be combined with a carrier gas, in particular, an inert gas such as helium or argon. In particular belium is a preferred carrier gas as this can minimise fragmentation of the monomer.

When used as a mixture, the relative amount of the monomer vapour to carrier gas is suitably determined in accordance with procedures which are conventional in the art. The amount of monomer added will depend to some extent on the nature of the particular monomer being used, the nature of the substrate being treated, the size of the plasma chamber etc. Generally, in the case of conventional chambers, monomer is delivered in an amount of from 50-600mg/min, for example at a rate of from 100-

150mg/min. Carrier gas such as helium is suitably administered at a constant rate for example at a rate of from 5-90, for example from 15-30sccm. In some instances, the ratio of monomer to carrier gas will be in the range of from 100:1 to 1:100, for instance in the range of from 10:1 to 1:100, and in particular about 1:1 to 1:10. The precise ratio selected will be so as to ensure that the flow rate required by the process is achieved.

In some cases, a preliminary continuous power plasma may be struck for example for from 0.5-10 minutes for instance for about 4 minutes, within the chamber. This may act as a surface pre-treatment step, ensuring that the monomer attaches itself readily to the surface, so that as polymerisation occurs, the coating "grows" on the surface. The pre-treatment step may be conducted before monomer is introduced into the chamber, in the presence of only the inert gas.

The plasma is then suitably switched to a pulsed plasma to allow polymerisation to proceed, at least when the monomer is present.

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In all cases, a glow discharge is suitably ignited by applying a high frequency voltage, for example at 13.56MHz. This is suitably applied using electrodes, which may be internal or external to the chamber, but in the case of the larger chambers are internal.

Suitably the gas, vapour or gas mixture is supplied at a rate of at least 1 standard cubic centimetre per minute (sccm) and preferably in the range of from 1 to 100sccm.

30

In the case of the monomer vapour, this is suitably supplied at a rate of from 80-300mg/minute, for example at about 120mg per minute depending upon the nature of the monomer, whilst the pulsed voltage is applied.

Gases or vapours may be drawn or pumped into the plasma region. In particular, where a plasma chamber is used, gases or vapours may be drawn into the chamber as a result of a reduction in the pressure within the chamber, caused by use of an evacuating pump, or they may be pumped or injected into the chamber as is common in liquid handling.

Polymerisation is suitably effected using vapours of compounds of formula (I), which are kept within a pressure range of from 40 to 150mtorr, suitably at about 80-120mtorr.

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The applied fields are suitably of power of from 0.2W to 20W, more suitably about 2W, applied as a pulsed field. These powers are suitable for use in a chamber having a volume of 50cm. For larger or smaller chambers, a suitable power giving the same power density can be used. For deposition of the polymeric layer, the pulses are applied in a sequence which yields very low average powers, suitably at a duty cycle of up to 10% (i.e. an on:off ratio of up to 10% on). More suitable, the duty cycle is from 0.1% to 1%. The on pulses can be long or short.

The fields are suitably applied from 30 seconds to 90 minutes, preferably from 5 to 60 minutes, depending upon the nature of the compound of formula (I) and the item being treated etc.

Suitably a plasma chamber used is of sufficient volume to accommodate multiple items.

A particularly suitable apparatus and method for producing items in accordance with the invention is described in WO2005/089961, the content of which is hereby incorporated by reference.

In particular, when using high volume chambers of this type, the plasma is created with a voltage as a pulsed field, at an average power of from 0.001 to 500w/m³, for example at from 0.001

to  $100w/m^3$  and suitably for the functional layer, at from 9.005 to  $0.5w/m^3$ .

These conditions are particularly suitable for depositing good quality uniform coatings, in large chambers, for example in chambers where the plasma zone has a volume of greater than 500cm³, for instance 0.5m³ or more, such as from 0.5m³-10m³ and suitably at about 1m³. The layers formed in this way have good mechanical strength.

10

The dimensions of the chamber will be selected so as to accommodate the particular item being treated. For instance, generally cuboid chambers may be suitable for a wide range of applications, but if necessary, elongate or rectangular chambers may be constructed or indeed cylindrical, or of any other suitable shape.

The chamber may be a sealable container, to allow for batch processes, or it may comprise inlets and outlets for the items, for example yarn, to allow it to be utilised in a continuous process. In particular in the latter case, the pressure conditions necessary for creating a plasma discharge within the chamber are controlled using high volume pumps, as is conventional for example in a device with a "whistling leak".

25 However it will also be possible to process certain items at atmospheric pressure, or close to, negating the need for "whistling leaks"

The monomer may comprise a compound of formula (1)

30

where  $R^1$ ,  $R^2$  and  $R^3$  are independently selected from hydrogen, alkyl, haloalkyl or aryl optionally substituted by halo; and  $R^4$  is

a group K-R° where R° is an alkyl or haloalkyl group and X is a bond; a group of formula -C(0)O(CH₂),Y- where n is an integer of from 1 to 10 and Y is a bond or a sulphonamide group; or a group -(0),R°(0),(CH₂),- where R° is aryl optionally substituted by halo, 5 p is 0 or 1, q is 0 or 1 and t is 0 or an integer of from 1 to 10, provided that where q is 1, t is other than 0.

The monomers used are selected from monomers of formula (I) as defined above. Suitable haloalkyl groups for R¹, R², R² and R° are fluoroalkyl groups. The alkyl chains may be straight or branched and may include cyclic moieties.

For  $\mathbb{R}^5$ , the alkyl chains suitably comprise 2 or more carbon atoms, suitably from 2-20 carbon atoms and preferably from 5 to 12 carbon atoms.

15

For  $R^1$ ,  $R^2$  and  $R^3$ , alkyl chains are generally preferred to have from 1 to 6 carbon atoms.

Freferably  $R^{0}$  is a haloalkyl, and more preferably a perhaloalkyl group, particularly a perfluoroalkyl group of formula  $C_{\alpha}F_{2\alpha+1}$  where m is an integer of 1 or more, suitably from 1-20, and preferably from 4-12 such as 4, 6 or 8.

Suitable alkyl groups for  $\mathbb{R}^3$ ,  $\mathbb{R}^3$  and  $\mathbb{R}^3$  have from 1 to 6 carbon 25 atoms.

In one embodiment, at least one of  $R^3$ ,  $R^2$  and  $R^3$  is hydrogen. In a particular embodiment  $R^3$ ,  $R^2$ ,  $R^3$  are all hydrogen. In yet a further embodiment however  $R^3$  is an alkyl group such as methyl or propyl.

Where X is a group  $-C(O)O(CR_1) \circ Y$ , n is an integer which provides a suitable spacer group. In particular, n is from 1 to 5, preferably about 2.

35

Suitable sulphonamide groups for Y include those of formula –  $N(R^7)SO_8$  where  $R^7$  is hydrogen or alkyl such as  $C_{1-8}alkyl$ , in particular methyl or ethyl.

5 In one embodiment, the compound of formula (I) is a compound of formula (II)

$$CH_0=CH-R^3$$
 (II)

10 where  $R^5$  is as defined above in relation to formula (1).

In compounds of formula (II), X in formula (I) is a bond.

However in a preferred embodiment, the compound of formula (I) is an acrylate of formula (III)

$$CH_0 = CR^2C(O) O(CH_0) R^3$$
 (III)

where n and  $R^5$  as defined above in relation to formula (I) and  $R^7$  is hydrogen,  $C_{1-10}$  alkyl, or  $C_{1-10}$  haloalkyl. In particular  $R^7$  is hydrogen or  $C_{1-10}$  alkyl such as methyl. A particular example of a compound of formula (III) is a compound of formula (IV)

$$H \longrightarrow C CF_2)_{\kappa} \setminus CF_3 \qquad (IV)$$

25

where  $R^7$  is as defined above, and in particular is hydrogen and x is an integer of from 1 to 9, for instance from 4 to 9, and preferably 7. In that case, the compound of formula (IV) is 1H, 1H, 2H, 2H-heptadecafluorodecylacrylate.

30

In one embodiment, the compound of formula (IV) is a compound of formula (V)  $\ensuremath{\text{(V)}}$ 

$$\begin{array}{c} H \\ \\ R^8 \end{array} O(CH_2)_0(CF_2)_5CF_3 \end{array} \qquad (V)$$

where R8 is hydrogen or methyl, ethyl or propyl group; and n=1 to 5.

A second aspect of the present invention provides a hydrophobic and/or eleophobic substrate which comprises a coating of a polymer which has been applied by the method.

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The substrate may comprise, for example, a fabric or an item of clothing (including footwear) comprising said fabric.

Additionally, the substrate may comprise an electronic device, microfluidic device, laboratory consumable or photovoltaic module.

Preferred features of the second aspect of the invention may be as described above in connection with the first aspect.

- Throughout the description and claims of this specification, the words "comprise" and "contain" and variations of the words, for example "comprising" and "comprises", mean "including but not limited to", and do not exclude other moieties, additives, components, integers or steps.
- Throughout the description and claims of this specification, the singular encompasses the plural unless the context otherwise requires. In particular, where the indefinite article is used, the specification is to be understood as contemplating plurality as well as singularity, unless the context requires otherwise.

Other features of the present invention will become apparent from the following example. Generally speaking the invention extends to any novel one, or any novel combination, of the features disclosed in this specification (including any accompanying claims and drawings). Thus features, integers, characteristics, compounds, chemical moleties or groups described in conjunction with a particular aspect, embodiment or example of the invention are to be understood to be applicable to any other aspect, embodiment or example described herein unless incompatible therewith.

Moreover unless stated otherwise, any feature disclosed herein may be replaced by an alternative feature serving the same or a similar purpose.

15 The invention will now be particularly described by way of example, with reference to the accompanying diagrammatic drawings in which:

Figure 1 illustrates the apparatus for carrying out the 20 invention; and Figure 2 is a flow diagram showing the events within each variable pressure cycle.

The apparatus used for forming the polymeric coating is shown in simplified from in Figure 1. A processing chamber 10 has a processing area 12 within it and an access door 14. Equipment for generating a plasma within the processing chamber are not shown.

30 A fluid input 16 is provided to allow monomer to be introduced into the chamber. Other fluids may also be introduced via this fluid input, for example a carrier gas or a pre-treatment gas.

The processing chamber 10 is connected to a vacuum pump 18 via a gate valve 20. The gate valve 20 can be closed or opened (either partially or fully). When opened, the gate valve allows waste

gases to be exhausted and when used in conjunction with the vacuum pump enables the pressure within the processing area reduced.

5 The apparatus also has a controller 22 which controls the fluid input, gate valve and vacuum pump. The controller may be a microprocessor, PC or any other suitable device.

Figure 2 shows the sequence of events during a cycle. In a 10 first step 24, the gate valve is opened and the vacuum pump partially evacuates the processing chamber. In particular the pressure in the processing chamber is brought down to a desired low pressure. The low pressure allows for increased residency time of monomer(s) in the processing chamber as it/they cannot escape from the valve and flow downstream with the evacuated 15 gases. In a second step 26, the gate valve is closed (or at least partially closed) and the fluid input is opened, this allows the monomer to be introduced into the chamber. The pressure in the chamber gradually increases and in particular 20 this is as a result of vaporisation of monomers introduced into the chamber. In a third step 28, the gate valve is opened so that the waste gases can be exhausted. The controller controls the actions of the vacuum pump, gate valve and fluid input during these steps in each cycle. The cycle is repeated to 25 maintain the desired residency time of the monomer. The cycling rate can be selected either by the time taken to reach a given pressure or by the duration that is needed to provide monomer conversion to polymer. A typical number of cycles that is used is four. Further the use of cycling allows for the deposition 30 rate to be moved within the chamber so that there is more uniformity of coating. In particular the deposition is moved from the edges of the chamber to the centre to improve coating.

The cycles may be time based or pressure based. In the case of time based cycles, the controller further includes a clock or

counter which is used to control the length of each cycle and the timing of the events in each cycle.

If a pressure based cycle is used, one or more pressure sensors are provided in the processing area, with pressure data being output to the controller. In this case, each step occurs when the pressure within the processing area reaches a pre-determined value.

### 10 Example 1

In a first example, a hearing aid was coated using a monomer of formula VI below.

Formula VI

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The plasma polymerisation coating was applied in an inductively coupled glow discharge reactor with a leak rate of better than 6 × 10 mol s⁻¹ and a monomer flow rate of 4mg/min or 3.2 mol s⁻¹. This was connected to a two stage Edwards rotary pump via a liquid nitrogen cold trap, a thermocouple pressure gauge, and a monomer tube containing the monomer. A 13.56 MHz radio frequency (BF) generator was used to power the electrical discharge.

A hearing aid having a textured ABS plastic exterior surface was placed into the centre of the chamber, which was then evacuated down to 20mTcrr.

The glow discharge was ignited the chamber subjected to continuous wave (CW) of 150W for 30s, during which the monomer was injected into the chamber in two doses 3 seconds apart.

A pulsed wave (PW) was then applied at 450W at a duty cycle of 35 microseconds on and 10 milliseconds off (0.35%), whilst the

monomer was injected into the chamber; 140 shots with 3 seconds between each shot.

During the PW phase, the pressure within the chamber was cycled 5 as follows: Once the chamber pressure reached SOmTorr, the gate valve was closed for 60 seconds. After 60 seconds has elapsed, the gate was opened by a fixed percentage (between 50 to 190%). When pressure fell to 50mTorr, the gate was again closed for 60seconds and the cycle repeated. This was repeated for the duration of the PW cycle (i.e. approximately 420 seconds).

This method was repeated without pressure cycling. The conditions were exactly the same as above, except that the gate valve is left open for the duration of the continuous wave phase and PW phase.

The effect of the plasma deposition on the surface energy of the treated substrate was then determined. Water was applied to the surface of the treated hearing aid and the contact angle measured. The tests were carried out on the coated hearing aid surface and also after the coating had been subjected to 1,000 abrasions (using the same test as example 2). The results of coatings applied using pressure cycling and without pressure cycling are listed in the table below.

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Conditions	Initial contact	Contact angle after
	angle (average)	1,000 abrasions
		(average)
With pressure	105.98	110.28
cycling		
Without pressure	103.63	99,03
cycling		

Table 1

As seen from table 1, higher contact angles are achieved with pressure cycling (both initially and after abrasions). This

shows that the desired performance levels of the coating has been achieved more quickly with pressure cycling than without.

### Example 2

5 In a second example, a mobile telephone was coated, which included various substrates of plastic, glass and metal. The same apparatus and monomer was used as described in example 1.

The chamber was evacuated to 20mTorr before igniting the glow discharge and subjecting the chamber to continuous wave (CW) of 150W for 30s, during which the monomer was injected into the chamber in two doses 3 seconds apart.

A pulsed wave (FW) was then applied at 300W at a duty cycle of 35 microseconds on and 10 milliseconds off (0.35%), whilst monomer was injected into the chamber; 140 shots with 3 seconds between each shot.

During the PW phase, the pressure within the chamber was cycled as follows: Once the chamber pressure reached 50mTorr, the gate was closed for 60 seconds. After 60 seconds has elapsed, the gate was opened by a fixed percentage (between 50 to 100%). When pressure fell to 50mTorr, the gate was gain closed for 60seconds and the cycle repeated. This was repeated for the duration of the PW cycle (i.e. approximately 420 seconds).

This method was repeated without the pressure cycling.

The effect of the plasma deposition on the surface energy of the treated substrate was then determined. Water was applied to the surface of the treated hearing aid and the contact angle measured. The tests were carried out on the coated hearing aid surface and also after the coating had been subjected to 1,000 abrasions using the Taber 5750 linear abrader with a standard Martindate abrasion fabric, 500g weight with a 1 inch abrasion

path. The results of coatings applied using pressure cycling and without pressure cycling are listed in the table below.

Conditions	Initial contact	Contact angle after	
	angle (average)	1,000 abrasions	
		(average)	
Untreated	89,04	77.4125	
Without pressure cycling	91.78	113.3875	
With pressure cycling	115,15	114,2375	

Table 2

5

As with the previous example, the coating applied with pressure cycling results in a higher contact angle. The contact angle changes very little following the abrasions. As before, this is believed to be due to the coating forming more quickly.

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## Example 3

In a third example, a mobile telephone was coated, which included various substrates of plastic, glass and metal. The same apparatus and monomer was used as described in example 1.

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The chamber was evacuated to 20mTorr before igniting the glow discharge and subjecting the chamber to continuous wave (CW) of 150W for 30s, during which the monomer was injected into the chamber in one dose.

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A pulsed wave (PW) was then applied at 450W at a duty cycle of 35 microseconds on and 10 milliseconds off (0.35%), whilst monomer was injected into the chamber; 130 shots with 3 seconds between each shot.

During the PW phase, the pressure within the chamber was cycled by alternatively opening and closing the gate valve for periods of 60 seconds.

5 This method was repeated without the pressure cycling, during which the gate was left open for the duration of the CW and PW phases.

The effect of the plasma deposition on the surface energy of the treated substrate was then determined using the same methods outlined in example 2.

Conditions	Average contact	Average contact
	angle (interior)	angle (exterior)
Without pressure	88.52	713.47
cycling		
With pressure	113.62	121.42
cycling		

Table 3

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15 Table 3 shows greatly increased deposition on the inside of the telephone where pressure cycling is used, as compared to no cycling.

As shown by the examples, the use of variable pressure cycling speeds up the process of achieving the performance levels desired, i.e. it speeds up the processing time.

As shown in example 3, not only is an improvement seen on the outside of products but also on the inside of complex products. To achieve the same results with a non-cycling process would require a longer processing time and more monomer (although the same performance level may still not be reached).

By speeding up the processing time, pressure cycling also has the effect of using less monomer. Other advantages include improved evenness of coating. In addition, the desired performance levels occur across the whole processing chamber.

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The increase in residency time has additional advantages, for example improved penetration of the coatings into nooks and crannies, for example in electronic devices. In addition, the increased residency time enables coatings to be formed from monomers which do not polymerise well in conventional conditions, for example shorter chain monomers such as 1H,1H,2H,2H-tridecafluoro-octyl acrylate.

### Claims

1. A method of forming a liquid repellent coating on a surface of a substrate, said method comprising exposing said surface to a monomer in a plasma deposition process under conditions that maintain the monomer in situ for a period of time to allow a polymeric layer to form on the surface, wherein the conditions comprise at least one cycle of varying pressure.

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2. A method according to claim 1, wherein the substrate is exposed to a plasma in a processing chamber and the processing chamber is at least partly evacuated at the beginning of each cycle.

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- 3. A method according to claim 1 or claim 2 wherein each cycle comprises continually introducing the monomer into a processing chamber and allowing the pressure to rise.
- 20 4. A method according to claim 2 or claim 3, wherein exhaust gas is expelled from the processing chamber at the end of each cycle.
- 5. A method according to any preceding claim wherein a valve is used to control pressure in and out of the processing chamber, with the valve being used to increase residency time of the monomer in the chamber to allow the polymeric layer to form.
- 6. A method according to any preceding claim, wherein the 30 maximum pressure is at or below 150 mTorr and more particularly is at or below 125 mTorr.
  - 7. A method according to any preceding claim, wherein the at least one cycle of varying pressure may comprise a time based cycle and/or a pressure based cycle.

- 8. A method according to any preceding claim, wherein each cycle is between 45 and 75 seconds.
- 9. A method according to any preceding claim, wherein exposing the surface to at least one cycle of varying pressure comprises exposing the surface to two or more cycles of varying pressure and in particular said two or more cycles comprises between 5 and 12 cycles and more particularly 8 or 9 cycles.
- 10 10. A method according to any preceding claim wherein the monomer is a compound of formula (I)

$$\begin{array}{c} R^{1} \\ R^{2} \end{array} \qquad \begin{array}{c} R3 \\ R^{4} \end{array} \qquad (0)$$

- where R', R' and R' are independently selected from hydrogen, alkyl, haloalkyl or aryl optionally substituted by halo; andR' is a group X-R' where R' is an alkyl or haloalkyl group and X is a bond; a group of formula -C(O)O(CH₂)₂Y- where n is an integer of from 1 to 10 and Y is a bond or a sulphonamide group; or a group -(O)₂R'(O)₃(CH₂)₂- where R' is aryl optionally substituted by halo, pis 0 or 1, q is 0 or 1 and t is 0 or an integer of from 1 to 10, provided that where q is 1, t is other than 0.
- A method according to claim 10 wherein the compound of formula (I) is a compound of formula (II)

where  $\mathbb{R}^5$  is as defined in claim 10, or a compound of formula 30 (III)

$$CH_0 = CR^2C(O)O(CH_0)_0R^3$$
 (III)

where n and R° as defined in claim 10 and R' is hydrogen,  $C_{1-10}$  alkyl, or  $C_{1-10}$  halcalkyl.

- 12. A method according to claim 11 wherein the compound of formula (I) is a compound of formula (III).
  - 13. A method according to claim 12 wherein the compound of formula (III) is a compound of formula (IV)

$$\begin{array}{c} H \\ \downarrow \\ F^{7} \\ \end{array} \\ O \\ \begin{array}{c} (CF_{2})_{x} \\ CF_{3} \\ \end{array} \\ (IV)$$

10

where  $\mathbb{R}^2$  is as defined in claim II, and x is an integer of from I to 9.

- 15 14. A method according to claim 13 wherein the compound of formula (IV) is 18,18,28,28-heptadecafluorodecylacrylate.
  - 15. A method according to claim 13 wherein the compound of formula (IV) is a compound of formula (V)

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$$\begin{array}{c} H \\ \downarrow \\ H \end{array} \begin{array}{c} O \\ R^6 \\ O(CH_2)_n(CF_2)_5 CF_3 \end{array} \hspace{1cm} (V)$$

where  $\mathbb{R}^s$  is hydrogen or methyl, ethyl or propyl group; and n=1 to 5.

16. A method according to any preceding claim, wherein the plasma deposition process is a plasma polymerisation process and the coating is applied using a pulsed plasma.

- 17. A method according to claim 16, wherein the pulsing sequence comprises a ratio of on:off of 1:200 to 1:1500.
- 5 18. A method according to any one of claims 16 or 17, wherein the pulsing sequence comprises power on for 20-50 µs and off for from 500 µs to 30000 µs.
- 19. A method according to any preceding claim wherein the average power levels are between 1 W and 1 kW and more particularly between 300 W and 500 W.

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- 20. A method according to any preceding claim, wherein in a preliminary step, a continuous power plasma is applied to the surface.
  - 21. A method according to claim 20 wherein the preliminary step is conducted in the presence of an inert gas.
- 20 22. A method according to any preceding claim wherein the liquid repellent coating comprises an oil or water repellent coating.
- 23. A method according to any preceding claim, wherein the surface is a surface of a fabric, metal, glass, ceramics, paper or polymer substrate.
  - 24. A method of forming a liquid repellent coating on a surface of a substrate, which method is substantially as herein
- 30 described with reference to the accompanying illustrative drawings.
- 25. A liquid repellent substrate which comprises a coating of a polymer which has been applied by a method according to any one of the preceding claims.

- 26. A substrate according to claim 25 which comprises a fabric and in particular an item of clothing.
- 27. A liquid repellent substrate, which substrate is 5 substantially as herein described with reference to the accompanying illustrative drawings.



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Application No: GB1212808.8 Examiner: Mr Robert Black
Claims searched: 1-24 Date of search: 23 November 2012

## Patents Act 1977: Search Report under Section 17

Documents considered to be relevant:

Category	Relevant to claims	Identity of document and passage or figure of particular relevance
Y	<b>∤</b> ++15-	EP 1087034 A1 (METALLE) see especially WPI abstract 2001-345753 and the EPODOC abstract
. <b>A</b> .		GB 2434379 A (P2i) see especially the abstract, page 11 lines 10-19 and page 16 lines 1-20
Y	1-5, 7 and 23	WO 2007/015779 A2 (3M) see especially the abstract and claim 1
Y	1-7 and 23	EP 1074588 A2 (BECTON) see especially claim 8, and paragraphs 0047 and 0048
Y	1-5, 7, 22 and 23	WO 00/20130 A1 (MoD) see especially the abstract and page 13 lines 12-34
Y	1-7 and 23	GB 1495480 A (AIRCO) see especially page 1 lines 10-14 and 45-56, page 1 line 85 to page 2 line 3, page 3 line 118 to page 4 line 10, page 4 lines 84-97, page 5 lines 16-31, and page 7 lines 8-24
Y	1-5, 7, 22 and 23	WO 2009/010753 A2 (P2i) see especially the abstract, figure 1, page 4 lines 20 and 21, page 6 line 19 to page 7 line 11, page 9 lines 5-16, and page 12 lines 1-5

#### Categories:

*******			
X	Document indicating lack of novelty or inventive	A	Document indicating technological background and/or state
	step		of the art.
X	Document indicating lack of inventive step if combined with one or more other documents of	p.	Document published on or after the declared priority date but before the filing date of this invention.
	same category.		
&	Member of the same patent family	$\mathbf{E}$	Patent document published on or after, but with priority date
-			earlier than, the filing date of this application.

#### Field of Search:

Search of GB, EP, WO & US patent documents classified in the following areas of the UKC X:

Worldwide search of patent documents classified in the following areas of the IPC

B05D; C08F; C08I; D06M



27

The following online and other databases have been used in the preparation of this search report

EPODOC; WPI

## International Classification:

Subclass	Subgroup	Valid From
B05D	0007/24	01/01/2006
C08F	0002/52	01/01/2006
C08J	0007/04	01/01/2006
D06M	0010/10	01/01/2006
B05D	0005/00	01/01/2006

#### PATENT COOPERATION TREATY

#### From the INTERNATIONAL SEARCHING AUTHORITY NOTIFICATION OF TRANSMITTAL OF Leland, Emma THE INTERNATIONAL SEARCH REPORT AND Greaves Brewster LLP THE WRITTEN OPINION OF THE INTERNATIONAL European Patent Attorneys SEARCHING AUTHORITY, OR THE DECLARATION Copa House Station Road GB-Cheddar, Somerset BS27 3AH ROYAUME UNI (PCT Rule 44.1) Date of mailing (day/month/year) 5 June 2015 (05-06-2015) Applicant's or agent's file reference EL/P2310PC00 FOR FURTHER ACTION See paragraphs 1 and 4 below International filing date International application No. (day/month/year) PCT/GB2015/050521 24 February 2015 (24-02-2015) Applicant P2I LTD The applicant is hereby notified that the international search report and the written opinion of the International Searching X Authority have been established and are transmitted herewith. Filing of amendments and statement under Article 19: The applicant is entitled, if he so wishes, to amend the claims of the International Application (see Rule 46): The time limit for filing such amendments is normally two months from the date of transmittal of the International Search Report. Directly to the International Bureau of WIPO, 34 chemin des Colombettes How? 1211 Geneva 20, Switzerland, Fascimile No.: (41-22) 338.82.70 For more detailed instructions, see PCT Applicant's Guide, International Phase, paragraphs 9.004 - 9.011. The applicant is hereby notified that no international search report will be established and that the declaration under Article 17(2)(a) to that effect and the written opinion of the International Searching Authority are transmitted herewith. With regard to any protest against payment of (an) additional fee(s) under Rule 40.2, the applicant is notified that: 3. the protest together with the decision thereon has been transmitted to the International Bureau together with any applicant's request to forward the texts of both the protest and the decision thereon to the designated Offices. no decision has been made yet on the protest; the applicant will be notified as soon as a decision is made. 4. Reminders The applicant may submit comments on an informal basis on the written opinion of the International Searching Authority to the International Bureau. These comments will be made available to the public after international publication. The International Bureau will send a copy of such comments to all designated Offices unless an international preliminary examination report has been or is to be established. Shortly after the expiration of 18 months from the priority date, the international application will be published by the International Bureau. If the applicant wishes to avoid or postpone publication, a notice of withdrawal of the international application, or of the priority claim, must reach the International Bureau before the completion of the technical preparations for international publication (Rules 90*bls*.1 and 90*bls*.3). Within 19 months from the priority date, but only in respect of some designated Offices, a demand for international preliminary examination must be filed if the applicant wishes to postpone the entry into the national phase until 30 months from the priority date (in some Offices even later); otherwise, the applicant must, within 20 months from the priority date, perform the prescribed acts for entry into the national phase before those designated Offices. In respect of other designated Offices, the time limit of 30 months (or later) will apply even if no demand is filed within 19 months. For details about the applicable time

Name and mailing address of the International Searching Authority

European Patent Office, P.B. 5818 Patentlaan 2 NL-2280 HV Rijswijk Tel. (+31-70) 340-2040 Fax: (+31-70) 340-3016 Authorized officer

limits, Office by Office, see www.wipo.int/pct/en/texts/time_limits.html and the *PCT Applicant's Guide*, National Chapters.

Within 19 months from the priority date, the applicant may request that a supplementary international search be carried out by a different International Searching Authority that offers this service (Rule 45*bis.*1). The procedure for requesting supplementary international search is described in the *PCT Applicant's Guide*, International Phase, paragraphs 8.006-8.032.

VAN DER HOEVEN, Maria Tel: +31 (0)70 340-3342

### PATENT COOPERATION TREATY

# PCT

## INTERNATIONAL SEARCH REPORT

(PCT Article 18 and Rules 43 and 44)

Applicant's or agent's file reference  EL/P2310PC00	FOR FURTHER ACTION as well	see Form PCT/ISA/220 as, where applicable, item 5 below.					
International application No.	International filing date (day/month/year)	(Earliest) Priority Date (day/month/year)					
PCT/GB2015/050521	24 February 2015 (24-02-2015)	28 February 2014 (28-02-2014)					
Applicant							
201170							
P2! LTD							
This international search report has been according to Article 18. A copy is being tra	prepared by this International Searching Autho Insmitted to the International Bureau.	rity and is transmitted to the applicant					
This international search report consists o	f a total of 6 sheets.						
X It is also accompanied by	a copy of each prior art document cited in this	report.					
Basis of the report		:6.					
	international search was carried out on the bas application in the language in which it was filed						
a translation of the	e international application into rnished for the purposes of international searcl	, which is the language					
b. This international search rauthorized by or notified to	report has been established taking into accoun o this Authority under Rule 91 (Rule 43.6 <i>bis</i> (a)	t the rectification of an obvious mistake ).					
c. With regard to any nucleo	otide and/or amino acid sequence disclosed	in the international application, see Box No. !.					
2. X Certain claims were fou	nd unsearchable (See Box No. II)						
3. Unity of invention is lac	king (see Box No III)						
4. With regard to the <b>title,</b>							
X the text is approved as su	bmitted by the applicant						
the text has been established by this Authority to read as follows:							
5. With regard to the abstract,	husitha d bu tha amplia ant						
the text is approved as submitted by the applicant the text has been established, according to Rule 38.2, by this Authority as it appears in Box No. IV. The applicant							
may, within one month fro	m the date of mailing of this international searc	ch report, submit comments to this Authority					
6. With regard to the <b>drawings</b> ,							
a. the figure of the <b>drawings</b> to be p	oublished with the abstract is Figure No						
as suggested by t	• •						
	s Authority, because the applicant failed to sug						
	s Authority, because this figure better characte e published with the abstract	nizes the invention					
b. X none of the figures is to be	, published with the abstract						

International application No. PCT/GB2015/050521

This international search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:  1. Claims Nos.:	Box No. II Observations where certain claims were found unsearchable (Continuation of item 2 of first sheet)
because they relate to subject matter not required to be searched by this Authority, namely:  2.	This international search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:
because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:  SEE FURTHER INFORMATION sheet PCT/ISA/210  3. Claims Nos.:  because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).  Box No. III Observations where unity of invention is lacking (Continuation of item 3 of first sheet)  This International Searching Authority found multiple inventions in this international application, as follows:  1. As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.  2. As all searchable claims could be searched without effort justifying an additional fees, this Authority did not invite payment of additional fees.  3. As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:	
Box No. III Observations where unity of invention is lacking (Continuation of item 3 of first sheet)  This International Searching Authority found multiple inventions in this international application, as follows:  1. As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.  2. As all searchable claims could be searched without effort justifying an additional fees, this Authority did not invite payment of additional fees.  3. As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:	because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:
This International Searching Authority found multiple inventions in this international application, as follows:  1. As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.  2. As all searchable claims could be searched without effort justifying an additional fees, this Authority did not invite payment of additional fees.  3. As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:	
1. As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.  2. As all searchable claims could be searched without effort justifying an additional fees, this Authority did not invite payment of additional fees.  3. As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:	Box No. III Observations where unity of invention is lacking (Continuation of item 3 of first sheet)
2. As all searchable claims could be searched without effort justifying an additional fees, this Authority did not invite payment of additional fees.  3. As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:  4. No required additional search fees were timely paid by the applicant. Consequently, this international search report is	This International Searching Authority found multiple inventions in this international application, as follows:
additional fees.  3. As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:  4. No required additional search fees were timely paid by the applicant. Consequently, this international search report is	
only those claims for which fees were paid, specifically claims Nos.:  4. No required additional search fees were timely paid by the applicant. Consequently, this international search report is	
4. No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:	
	4. No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:
Remark on Protest  The additional search fees were accompanied by the applicant's protest and, where applicable, the payment of a protest fee.  The additional search fees were accompanied by the applicant's protest but the applicable protest	payment of a protest fee.  The additional search fees were accompanied by the applicant's protest but the applicable protest
fee was not paid within the time limit specified in the invitation.  No protest accompanied the payment of additional search fees.	

International application No PCT/GB2015/050521

A. CLASSIFICATION OF SUBJECT MATTER INV. B05D3/04 B05D3 B05D7/22 B05D3/14 ADD. According to International Patent Classification (IPC) or to both national classification and IPC B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) B05D Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) EPO-Internal, WPI Data C. DOCUMENTS CONSIDERED TO BE RELEVANT Citation of document, with indication, where appropriate, of the relevant passages Relevant to claim No. Category* US 4 822 632 A (WILLIAMS JOEL L [US] ET AL) 18 April 1989 (1989-04-18) Х 1-35 examples I, II US 2014/010969 A1 (BICKER MATTHIAS [DE] ET 1 - 35Х AL) 9 January 2014 (2014-01-09) example 1 WO 2011/029628 A1 (SCHOTT AG [DE]; BICKER 1-35 Χ MATTHIAS [DE]; BAUCH HARTMUT [DE]; LOHMEYER MAN) 17 March 2011 (2011-03-17) page 21, line 20 - line 32 example 1 claims 1,8 χ US 2012/321776 A1 (VETRECIN ROBERT [US]) 1-35 20 December 2012 (2012-12-20) examples 2,3 Χ Further documents are listed in the continuation of Box C. See patent family annex. Special categories of cited documents: "I" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier application or patent but published on or after the international "X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other step when the document is taken alone "Y" document of particular relevance; the claimed invention cannot be special reason (as specified) considered to involve an inventive step when the document is combined with one or more other such documents, such combination "O" document referring to an oral disclosure, use, exhibition or other being obvious to a person skilled in the art document published prior to the international filing date but later than the priority date claimed "&" document member of the same patent family Date of the actual completion of the international search Date of mailing of the international search report 28 May 2015 05/06/2015 Authorized officer Name and mailing address of the ISA/ European Patent Office, P.B. 5818 Patentlaan 2 NL - 2280 HV Rijswijk Tel. (+31-70) 340-2040, Fax: (+31-70) 340-3016 Slembrouck, Igor

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International application No PCT/GB2015/050521

C(Continua	tion). DOCUMENTS CONSIDERED TO BE RELEVANT	
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	WO 2005/021833 A2 (SURFACE INNOVATIONS LTD [GB]; WARD LUKE [GB]; BADYAL JAS PAL [GB]) 10 March 2005 (2005-03-10) cited in the application	

1

Information on patent family members

International application No
PCT/GB2015/050521

Patent document Publication cited in search report date		Patent family member(s)	Publication date
US 4822632 A	18-04-1989	NONE	
US 2014010969 A	1 09-01-2014	CA 2824074 DE 102011009057 EP 2666181 KR 20140007842 US 2014010969 WO 2012097972	A1 26-07-2012 A1 27-11-2013 A 20-01-2014 A1 09-01-2014
WO 2011029628 A	1 17-03-2011	DE 102009041132 DK 2477755 EP 2477755 US 2012171386 WO 2011029628	T3 16-02-2015 A1 25-07-2012 A1 05-07-2012
US 2012321776 A	1 20-12-2012	AU 2012271807 CA 2839402 CN 103619493 EP 2720807 US 2012321776 WO 2012174054	A1 20-12-2012 A 05-03-2014 A1 23-04-2014 A1 20-12-2012
WO 2005021833 A	2 10-03-2005	GB 2419358 WO 2005021833	

## FURTHER INFORMATION CONTINUED FROM PCT/ISA/ 210

Continuation of Box II.2

Claims Nos.: 36, 37

(1) claim 36 refers to a method of applying a coating whereby the characterising technical feature(s) should be disclosed by the illustrative drawings; this authority found no clear and unambiguous indication in the drawings that could possibly constitute a characterising technical feature for a method of applying a coating; furthermore, the passages of the description relating to the illustrative figures are disclosing so many technical details that it is impossible for the skilled reader to determine which one is essential to the definition of the invention; as a consequence, the subject matter of claim 36 is considered as lacking clarity (Article 6 PCT) and

non-compliance with the substantive provisions is to such an extent that no meaningful search of claim 36 could be carried out at all (Article 17(2) PCT).

(2) claim 37 refers to a device whereby the characterising technical feature(s) should be disclosed by the illustrative drawings; this authority found no clear and unambiguous indication in the drawings that could possibly constitute a characterising technical feature for a device according to the invention; furthermore, the passages of the description relating to the illustrative figures are disclosing so many technical details that it is impossible for the skilled reader to determine which one is essential to the definition of the invention; as a consequence, the subject matter of claim 37 is considered as lacking clarity (Article 6 PCT) and

The non-compliance with the substantive provisions is to such an extent that no meaningful search of claim 37 could be carried out at all (Article 17(2) PCT).

The applicant's attention is drawn to the fact that claims relating to inventions in respect of which no international search report has been established need not be the subject of an international preliminary examination (Rule 66.1(e) PCT). The applicant is advised that the EPO policy when acting as an International Preliminary Examining Authority is normally not to carry out a preliminary examination on matter which has not been searched. This is the case irrespective of whether or not the claims are amended following receipt of the search report or during any Chapter II procedure. If the application proceeds into the regional phase before the EPO, the applicant is reminded that a search may be carried out during examination before the EPO (see EPO Guidelines C-IV, 7.2), should the problems which led to the Article 17(2) declaration be overcome.

#### PATENT COOPERATION TREATY

From the INTERNATIONAL SEARCHING AUTHORITY To: WRITTEN OPINION OF THE see form PCT/ISA/220 INTERNATIONAL SEARCHING AUTHORITY (PCT Rule 43bis.1) Date of mailing (day/month/year) see form PCT/ISA/210 (second sheet) Applicant's or agent's file reference FOR FURTHER ACTION see form PCT/ISA/220 See paragraph 2 below International application No. International filing date (day/month/year) Priority date (day/month/year) PCT/GB2015/050521 24.02.2015 28.02.2014 International Patent Classification (IPC) or both national classification and IPC INV. B05D3/04 B05D3/14 B05D7/22 Applicant P2I LTD This opinion contains indications relating to the following items: Box No. I Basis of the opinion ☐ Box No. II Priority Box No. III Non-establishment of opinion with regard to novelty, inventive step and industrial applicability Box No. IV Lack of unity of invention Reasoned statement under Rule 43bis.1(a)(i) with regard to novelty, inventive step and industrial Box No. V applicability; citations and explanations supporting such statement ☐ Box No. VI Certain documents cited ☐ Box No. VII Certain defects in the international application Box No. VIII Certain observations on the international application **FURTHER ACTION** 2. If a demand for international preliminary examination is made, this opinion will usually be considered to be a written opinion of the International Preliminary Examining Authority ("IPEA") except that this does not apply where the applicant chooses an Authority other than this one to be the IPEA and the chosen IPEA has notifed the International Bureau under Rule 66.1 bis(b) that written opinions of this International Searching Authority will not be so considered. If this opinion is, as provided above, considered to be a written opinion of the IPEA, the applicant is invited to submit to the IPEA a written reply together, where appropriate, with amendments, before the expiration of 3 months from the date of mailing of Form PCT/ISA/220 or before the expiration of 22 months from the priority date, whichever expires later. For further options, see Form PCT/ISA/220. Name and mailing address of the ISA: Authorized Officer Date of completion of this opinion Telephone No. +31 70 340-0 European Patent Office see form P.B. 5818 Patentlaan 2

PCT/ISA/210

NL-2280 HV Rijswijk - Pays Bas Tel. +31 70 340 - 2040 Fax: +31 70 340 - 3016

# WRITTEN OPINION OF THE INTERNATIONAL SEARCHING AUTHORITY

International application No. PCT/GB2015/050521

В	ox No. I	Basis of the opinion
1. W	/ith rega	rd to the language, this opinion has been established on the basis of:
$\boxtimes$	the in	ternational application in the language in which it was filed.
		islation of the international application into , which is the language of a translation furnished for the uses of international search (Rules 12.3(a) and 23.1 (b)).
2. 🗆		opinion has been established taking into account the rectification of an obvious mistake authorized notified to this Authority under Rule 91 (Rule 43bis.1(a))
3. □		regard to any <b>nucleotide and/or amino acid sequence</b> disclosed in the international application, this on has been established on the basis of a sequence listing:
	а. 🗆	forming part of the international application as filed:
		☐ in the form of an Annex C/ST.25 text file.
		$\square$ on paper or in the form of an image file.
	b. □	furnished together with the international application under PCT Rule 13ter.1(a) for the purposes of international search only in the form of an Annex C/ST.25 text file.
	c. 🗆	furnished subsequent to the international filing date for the purposes of international search only:
		☐ in the form of an Annex C/ST.25 text file (Rule 13ter.1(a)).
		☐ on paper or in the form of an image file (Rule 13 <i>ter</i> .1(b) and Administrative Instructions, Section 713).
4. 🏻	the re	dition, in the case that more than one version or copy of a sequence listing has been filed or furnished, quired statements that the information in the subsequent or additional copies is identical to that appart of the application as filed or does not go beyond the application as filed, as appropriate, were hed.

5. Additional comments:

# WRITTEN OPINION OF THE INTERNATIONAL SEARCHING AUTHORITY

International application No. PCT/GB2015/050521

	x No. III Non-establishment of opinion with regard to novelty, inventive step and industrial plicability
	e questions whether the claimed invention appears to be novel, to involve an inventive step (to be non vious), or to be industrially applicable have not been examined in respect of
	the entire international application
$\boxtimes$	claims Nos. <u>36, 37</u>
bec	cause:
	the said international application, or the said claims Nos. relate to the following subject matter which does not require an international search (specify):
	the description, claims or drawings (indicate particular elements below) or said claims Nos. are so unclear that no meaningful opinion could be formed (specify):
	the claims, or said claims Nos. are so inadequately supported by the description that no meaningful opinion could be formed (specify):
$\boxtimes$	no international search report has been established for the whole application or for said claims Nos. 36, 37
	a meaningful opinion could not be formed without the sequence listing; the applicant did not, within the prescribed time limit:
	☐ furnish a sequence listing in the form of an Annex C∕ST.25 text file, and such listing was not available to the International Searching Authority in the form and manner acceptable to it; or the sequence listing furnished did not comply with the standard provided for in Annex C of the Administrative Instructions.
	☐ furnish a sequence listing on paper or in the form of an image file complying with the standard provided for in Annex C of the Administrative Instructions, and such listing was not available to the International Searching Authority in the form and manner acceptable to it; or the sequence listing furnished did not comply with the standard provided for in Annex C of the Administrative Instructions.
	□ pay the required late furnishing fee for the furnishing of a sequence listing in response to an invitation under Rule 13ter.1(a) or (b).
$\boxtimes$	See Supplemental Box for further details

# WRITTEN OPINION OF THE INTERNATIONAL SEARCHING AUTHORITY

International application No. PCT/GB2015/050521

Box No. V Reasoned statement under Rule 43bis.1(a)(i) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

1. Statement

Novelty (N)

Yes: Claims

<u>2-35</u>

No:

Claims

1

Inventive step (IS)

Yes: Claims

No: Claims

<u>1-35</u>

Industrial applicability (IA)

Yes: Claims

<u>1-35</u>

No: Claims

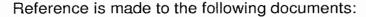
2. Citations and explanations

see separate sheet

Box No. VIII Certain observations on the international application

The following observations on the clarity of the claims, description, and drawings or on the question whether the claims are fully supported by the description, are made:

see separate sheet



- D1 US 4 822 632 A (WILLIAMS JOEL L [US] ET AL) 18 April 1989 (1989-04-18)
- D2 US 2014/010969 A1 (BICKER MATTHIAS [DE] ET AL) 9 January 2014 (2014-01-09)
- D3 WO 2011/029628 A1 (SCHOTT AG [DE]; BICKER MATTHIAS [DE]; BAUCH HARTMUT [DE]; LOHMEYER MAN) 17 March 2011 (2011-03-17)
- D4 US 2012/321776 A1 (VETRECIN ROBERT [US]) 20 December 2012 (2012-12-20)
- D5 WO 2005/021833 A2 (SURFACE INNOVATIONS LTD [GB]; WARD LUKE [GB]; BADYAL JAS PAL [GB]) 10 March 2005 (2005-03-10)cited in the application

#### Re Item III

# Non-establishment of opinion with regard to novelty, inventive step and industrial applicability

(1) Claim 36 refers to a method of applying a coating whereby the characterising technical feature(s) should be disclosed by the illustrative drawings; this authority found no clear and unambiguous indication in the drawings that could possibly constitute a characterising technical feature for a method of applying a coating; furthermore, the passages of the description relating to the illustrative figures are disclosing so many technical details that it is impossible for the skilled reader to determine which one is essential to the definition of the invention; as a consequence, the subject matter of claim 36 is considered as lacking clarity (Article 6 PCT) and

The non-compliance with the substantive provisions is to such an extent that no meaningful search of claim 36 could be carried out at all (Article 17(2) PCT).

(2) Claim 37 refers to a device whereby the characterising technical feature(s) should be disclosed by the illustrative drawings; this authority found no clear and unambiguous indication in the drawings that could possibly constitute a characterising technical feature for a device according to the invention; furthermore, the passages of the description relating to the illustrative figures are disclosing so many technical

details that it is impossible for the skilled reader to determine which one is essential to the definition of the invention; as a consequence, the subject matter of claim 37 is considered as lacking clarity (Article 6 PCT) and

The non-compliance with the substantive provisions is to such an extent that no meaningful search of claim 37 could be carried out at all (Article 17(2) PCT).

## Re Item VIII

## Certain observations on the international application

### Clarity (Article 6 PCT)

- (3) The scope of claim 1 is not clear and lacks support according to Article 6 PCT for the following reasons:
- (3.1) the term "housing" used in claim 1 is vague and unclear (Article 6 PCT);

actually, a plasma treatment chamber containing substrates could be regarded as a device according to claim 1, where the plasma treatment chamber would be regarded as a "housing" and the substrates would be regarded as the "components" according to the wording of claim 1;

this interpretation is also supported by the statement in the application's description on p. 6, l. 7-10, where the "device" is stipulated as comprising a "laboratory equipment"; said passage further stipulates that "articles of laboratory equipment (...) may be housed in a container"; this passage of the description support the vague interpretation of the terms "device", "housing" and "elements" used in claim 1;

## (3.2) step (i) of claim 1 reads:

"applying liquid coating precursor to at least part of the internal surface of the housing and/or at least part of the one or more components",

it appears that the method of claim 1 may be directed to coat only "at least part of the one or more components";

#### (3.3) the term

"the pressure being sufficient to cause the liquid coating precursor to evaporate"

used in claim 1 constitutes an attempt to define the pressure in terms of the result to be achieved, which merely amounts to a statement of the underlying problem, without providing the technical features necessary for achieving this result; this attempt is also considered lacking clarity in that the liquid coating precursor may be applied:

- by spraying (i.e. in an atomised state; cf. claim 15),
- as multiple droplets (i.e. as liquid aggregates; cf. claim 16),
- within a porous material (i.e. the liquid coating precursor soaks a material; cf. claim 17),

those differences in applications result in differences of vacuum level needed to evaporate the liquid coating precursor; furthermore the term does not specify if the pressure should be sufficient to evaporate partly or completely the liquid coating precursor;

it is furthermore known that a reduction of the total pressure will necessarily lead to an increase partial pressure of any liquid subjected to said reduced total pressure, resulting in a partial evaporation of said liquid.

as a consequence, the term

"the pressure being sufficient to cause the liquid coating precursor to evaporate"

is considered as lacking clarity (Article 6 PCT) and can not be considered as possibly restricting the scope of claim 1;

- (3.4) claim 1 therefore covers embodiments where :
  - (i) components are coated with a liquid coating precursor,
  - (ii) coated components of step (i) are subjected to reduced pressure in a plasma treatment chamber
  - (iii) liquid coating precursor is subjected to plasma treatment;

those embodiments do not correspond to the scope of the invention as can be derived from the application; claim 1 is therefore considered as lacking support (Article 6 PCT).

(4) It is clear from the description's examples that

the fact that in step (ii) evaporation of the liquid coating precursor occurs is essential to the definition of the invention.

Since independent claim 1 does not contain this feature it does not meet the requirement following from Article 6 PCT taken in combination with Rule 6.3(b) PCT that any independent claim must contain all the technical features essential to the definition of the invention.

(5) In claim 1, step (ii) refers to a "closed housing";

in claim 20, it is stipulated that a vapour is introduced through an aperture in the housing;

in claim 21, it is stipulated that the liquid coating precursor is applied in step (i) whilst the housing is closed;

it appears from the examples, that the mobile phone enclosures (which are believed to correspond to the term "housing" used in the claims) has apertures;

the topological definition of a closed structure is that it has no part left open and that its internal surface does not contact its external surface;

it appears however that

the "closed housing" referred to in claim 1 may have apertures through which diffusion of vapours may occur (cf. claim 22, examples),

the apertures are not sealed at any moment of the method,

there is therefore a confusion around the term "closed" used in the claims, as it appears clearly that the housing may well be an open structure with apertures;

the subject matter of the claims is therefore considered as lacking clarity for this reason as well (Article 6 PCT).

#### Re Item V

Reasoned statement with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

### Novelty (Article 33(3) PCT)

(6) In view of the clarity objections raised under Item VIII, this authority found that document D1 discloses (cf. document D1, passages cited in the search report):

a method of applying a coating to a device, the device comprising one or more components housed in a housing, the method comprising the steps of:

- (i) applying liquid coating precursor to least part of the one or more components;
- (ii) applying below atmosphere pressure conditions to the closed housing, with the one or more components inside the housing; and
- (iii) initiation of the liquid coating precursor with a plasma, to thereby cause the coating to form from the liquid precursor on at least some of the internal surfaces of the device.

The subject matter of at least claim 1 is therefore not novel (Article 33(2) PCT).

(7) The attention of the applicant is brought to the fact that a similar objection against at least the subject matter of claim 1 could be raised on the basis of documents D2, D3 and D4 each considered independently.

## Inventive Step (Article 33(3) PCT)

(8) At this stage of the procedure, this authority is of the opinion that if the subject matter of any of claims 2-35 should appear to be novel over the disclosure of any of documents D1-D4, it is not clear how the subject matter of claims 2-35 could eventually contribute to the inventive step of the present application: the technical features of claim 2-35 seem to represent nothing more than some out of many alternatives the person skilled in the art would select without the need of inventive skills, especially as no particular advantage can be derived from said technical features. Consequently, and without any acknowledgement concerning the novelty of said claims, it is considered at this stage of the procedure that the subject-matter of claims 2-35, if ever novel, lacks an inventive step (Article 33(3) PCT)).

## Industrial Applicability (Article 33(4) PCT)

(9) The industrial applicability of the subject matter of claims 1-35 (method of applying a coating to a device) is obvious (Article 33(4) PCT).

Electronic Patent Application Fee Transmittal						
Application Number:		735130				
Filing Date:	08-	Dec-2017				
Title of Invention:		COATINGS				
First Named Inventor/Applicant Name:	Ste	phen Richard COU	LSON			
Filer:	Pai	Paul John Parins/Bethany Lichtscheidl				
Attorney Docket Number:	GBL0007/US					
Filed as Large Entity						
Filing Fees for U.S. National Stage under 35 USC 371						
Description		Fee Code	Quantity	Amount	Sub-Total in USD(\$)	
Basic Filing:						
Pages:						
Claims:						
Miscellaneous-Filing:						
Petition:						
Patent-Appeals-and-Interference:						
Post-Allowance-and-Post-Issuance:						
Extension-of-Time:						

Description	Fee Code	Quantity	Amount	Sub-Total in USD(\$)
Miscellaneous:				
SUBMISSION- INFORMATION DISCLOSURE STMT	1806	1	240	240
	Tot	al in USD	(\$)	240

Electronic Acknowledgement Receipt				
EFS ID:	35453134			
Application Number:	15735130			
International Application Number:				
Confirmation Number:	7101			
Title of Invention:	COATINGS			
First Named Inventor/Applicant Name:	Stephen Richard COULSON			
Customer Number:	33072			
Filer:	Paul John Parins/Bethany Lichtscheidl			
Filer Authorized By:	Paul John Parins			
Attorney Docket Number:	GBL0007/US			
Receipt Date:	18-MAR-2019			
Filing Date:	08-DEC-2017			
Time Stamp:	15:21:19			
Application Type:	U.S. National Stage under 35 USC 371			
Filing Date: Time Stamp:	08-DEC-2017 15:21:19			

## **Payment information:**

Submitted with Payment	yes
Payment Type	CARD
Payment was successfully received in RAM	\$240
RAM confirmation Number	031919INTEFSW15220300
Deposit Account	
Authorized User	

The Director of the USPTO is hereby authorized to charge indicated fees and credit any overpayment as follows:

File Listin	g:				
Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)
			1044070	no	6
1	1 Information Disclosure Statement (IDS) Form (SB08)	GBL0007US_SIDS.pdf	8baa48276c651d906bc79b07cf8410e92d2 abc63		
Warnings:					
Information:					
		EP2711153A1.pdf	1839209		
2	Foreign Reference		273f4666e3147230e78d058f4086ca67f021 e5be	no	11
Warnings:					
Information:					
			2735053	no	20
3	Foreign Reference	GB2434368A.pdf	702b3464b3a0e3421e053cd93f13359fb3d e23fd		
Warnings:			-		
Information:					
		GB2493264A.pdf	3598172	no	30
4	Foreign Reference		a4ee32791598c1fac8a32ef19f63ddb9e4ec 066f		
Warnings:				'	
Information:					
	Non Patent Literature	NPL_1_GB_Search_Report.pdf	1398326	no	4
5			c413532c08977af64faea9ca161d96d73982 2f79		
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Warnings:					
	n the PDF is too large. The pages should be pper and may affect subsequent processing		tted, the pages will be re	sized upon er	ntry into the
Information					
8	Foreign Reference	WO03063938A1.pdf	2895400 378ea27fee4daf174f3631dce6ac971e93f57 f50	no	21
Warnings:					
Information:					
			895493	no	23
9 Foreig	Foreign Reference	WO2002028548.pdf	2adf3502c0ae0c5b771be85abbbf71ca02e 5ad95		
Warnings:		•			
Information					
10	Fareira Deference	WO2005021833A2.pdf	4660650	no	30
	Foreign Reference		e045f75c5014592f618f9d5d50b566d30486 fa87		
Warnings:					
Information					
11 Foreign Referen		WO2009030435A1.pdf	6162869	no	38
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Warnings:					
Information:					
12	Foreign Reference	WO2009046093A2.pdf	5660317	no	31
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Warnings:					
Information					

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13	Foreign Reference	WO2011029628A1.pdf	3b81efc196e17bbe58bb7f3fe79461a93fcf4 c70	no	62
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Information:					
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14	Foreign Reference	WO2012158953.pdf	643d47c6b265353102c52f09d0788b8c674 b7a92	no	36
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Information:					
			1448452		
15	Foreign Reference	WO2013132250.pdf	504c26eaa5100f6931b000001c12088ecf36 bdba	no	35
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Information:					
			6009473		
16	Foreign Reference	WO2014026967A2.pdf	fdf42a9ec122303b1ac3efac89f699adee578 7ab	no	46
Warnings:		-	1		
Information:					
			1227145		
17	Foreign Reference	WO2005089961.pdf	1472f0b0c75ae062271df78bfc9f88e8267df 909	no	29
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Information:					
			30379		
18	Fee Worksheet (SB06)	fee-info.pdf	09b294e9188deda138c308ea0e3b111a922 83867	no	2
Warnings:		-	1		
Information:					
		Total Files Size (in bytes	514	20345	

This Acknowledgement Receipt evidences receipt on the noted date by the USPTO of the indicated documents, characterized by the applicant, and including page counts, where applicable. It serves as evidence of receipt similar to a Post Card, as described in MPEP 503.

#### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

#### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

#### PATENT APPLICATION

### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

Inventor: Stephen Richard COULSON | Examiner: Buie-Hatcher, Nicole

Serial No. 15/735,130 | Group Art Unit: 1767

International Filing Date: June 8, 2016 Confirmation No. 7101

International App. No. PCT/GB2016/051688 | Docket No. GBL0007/US

371(c): December 8, 2017

Mail Stop Amendment Commissioner for Patents P.O. Box 1450 Alexandria, Virginia 22313-1450

For:

**COATINGS** 

I CERTIFY THAT ON MARCH 18, 2019 THIS CORRESPONDENCE IS BEING ELECTRONICALLY TRANSMITTED TO THE UNITED STATES PATENT AND TRADEMARK OFFICE VIA THE OFFICE'S EFS-Web.

/Bethany Lichtscheidl/
BETHANY L. LICHTSCHEIDL

#### AMENDMENT AND RESPONSE

Dear Examiner Buie-Hatcher:

This communication is responsive to the non-final Office Action issued on October 18, 2018, in connection with the above-identified patent application.

The three-month shortened-statutory period for response expired on January 18, 2019. Applicant respectfully petitions the Commissioner for Patents to extend the time for response for two months, from January 18, 2019 to March 18, 2019. It is respectfully submitted that this response is timely filed within the extended deadline. If any additional extension period is required in order for this paper to be timely filed, Applicant hereby requests an extension for such additional time period and authorizes the appropriate fees therefore to be charged to the Kagan Binder Deposit Account No. 50-1775 and to notify us of the same.

Payment in the amount of \$600.00 for a two-month extension of time is being provided by the electronic payment system of the EFS. In the event of failure of this payment by the electronic payment system of the EFS for any reason, the Commissioner is authorized to charge Kagan Binder Deposit Account No. 50-1775 and thereafter is requested to notify us of the same.

## **Specification Amendments**

Please delete the paragraph on page 1 that begins with "Monosaturated monomers" and ends with "(see co-pending application)."

Please add the following paragraph (section heading) on page 17 immediately prior to the paragraph beginning with "The present invention will now ....":

Brief Description of Drawings

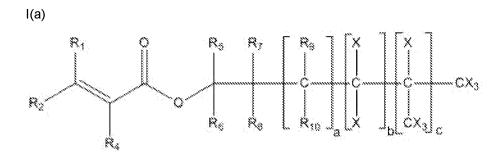
#### **Claim Amendments**

Please amend claims 1, 6, 8, 13, 15, and 16, and cancel claims 5, 7, 12, and 14 as follows:

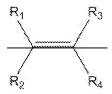
1. (Currently Amended) An electronic or electrical device or electronic or electrical component thereof comprising a protective cross-linked polymeric coating on a surface of said device or component;

wherein the protective cross-linked polymeric coating is obtained by exposing said device or component to a plasma comprising a monomer compound and a crosslinking reagent for a period of time sufficient to allow formation of the protective cross-linked polymeric coating on a surface thereof,

wherein the monomer compound is a compound of formula I(a):



wherein each of  $R_1$ ,  $R_2$ ,  $\underline{R_4}$  and  $R_5$  to  $\underline{R_8}$   $\underline{R_{10}}$  is hydrogen; each of  $R_4$ ,  $R_9$  and  $R_{10}$  is independently hydrogen or methyl is independently selected from hydrogen or an optionally substituted  $\underline{C_1}$ - $\underline{C_6}$  branched or straight chain alkyl group; each X is independently hydrogen or halogen; a is from 0-10; b is from 3 to 7; and c is 0 or 1; [[:]]



and wherein the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moieties and has a boiling point of less than 500 °C at standard pressure, the crosslinking reagent having one of the following structures:

where Y₁, Y₂, Y₃, Y₄, Y₅, Y₆, Y₇ and Y₈ are each hydrogen are each independently selected from hydrogen, optionally substituted cyclic, branched or straight chain C₁-C₆ alkyl or aryl; and L is a linker moiety;

wherein for compound (i) L is of Formula A having one of the following structures:

and  $Y_{10}$  is selected from optionally substituted cyclic, branched or straight chain  $C_1$ - $C_8$  alkylene and a siloxane group;

or wherein for compound (i) L is of Formula B having the following formula:

where each  $Y_9$  is independently selected from, a bond, -O-, -O-C(O)-, -C(O)-O-, -Y₁₁-O-C(O)-, -C(O)-O-Y₁₁-, -OY₁₁-, and -Y₁₁O-, where Y₁₁ is an optionally substituted cyclic, branched or straight chain  $C_1$ - $C_8$  alkylene; and wherein Y₁₀ has the following formula:

$$\xi -$$
 $\xi -$ 
 $\xi -$ 

and each  $Y_{15}$  is independently selected from optionally substituted branched or straight chain  $C_{\underline{1}}$ - $\underline{C_6}$  alkyl;

or wherein Y₁₀ has the following formula:

and  $Y_{16}$  to  $Y_{19}$  are each independently selected from H and optionally substituted branched or straight chain  $C_1$ - $C_8$  alkyl or alkenyl.

- 2.( Previously Presented) The device or component according to claim 1, wherein the protective cross-linked polymeric coating is a physical barrier to mass and electron transport.
- 3.( Previously Presented) The device or component according to claim 1, wherein the protective cross-linked polymeric coating forms a liquid repellent surface defined by a static water contact angle (WCA) of at least 90°.
- 4. (Cancelled)
- 5. (Cancelled)
- 6. (Currently Amended) The device or component according to claim 5 claim 1, wherein when for compound (i) L is of the formula B, L has one of the following structures:

## 7. (Cancelled)

8. (Currently Amended) The device or component according to elaim 5 claim 1, whereinwhen for compound (i) L is of Formula B, Y₁₀ has the following formula:

wherein each  $Y_{12}$  and  $Y_{13}$  is independently selected from H, halo, optionally substituted cyclic, branched or straight chain alkyl, or  $-OY_{14}$ , where  $Y_{14}$  is selected from optionally substituted branched or straight chain  $C_1$ - $C_8$  alkyl or alkenyl, and n is an integer from 1 to 10.

- 9. (Previously Presented) The device or component according to claim 8, wherein each  $Y_{12}$  is H and each  $Y_{13}$  is H.
- 10. (Previously Presented) The device or component according to claim 8, wherein each  $Y_{12}$  is fluoro and each  $Y_{13}$  is fluoro.
- 11. (Previously Presented) The device or component claim 8, wherein n is from 4 to 6.
- 12. (Cancelled)

13. (Currently Amended) The device or component according to elaim 12 claim 1, wherein when for compound (i) L is of Formula B and  $Y_{10}$  has the following formula:

each Y₁₅ is methyl, and each Y₉ is a bond.

14. (Cancelled)

15. (Currently Amended) The device or component according to elaim 14 claim 1, wherein when for compound (i) L is of Formula B and  $Y_{10}$  has the following formula:

 $Y_{18}$  is H or vinylene, and  $Y_{16}$ ,  $Y_{17}$  and  $Y_{19}$  are each H.

16. (Currently Amended) The device or component according to claim 1, wherein the crosslinking reagent is selected from Divinyl Adipate divinyl adipate (DVA), 1,4 Butanediol 1,4-butanediol divinyl ether (BDVE), 1,4 Cyclohexanedimethanol 1,4-cyclohexanedimethanol divinyl ether (CDDE), 1,7 Octadiene 1,7-octadiene (17OD), 1,2,4-trivinylcyclohexane—Trivinylcyclohexane (TVCH), 1,3-Divinyltetramethyldisiloxane1,3-divinyltetramethyldisiloxane (DVTMDS), Diallyl diallyl 1,4-cyclohexanedicarboxylate (DCHD), 1,6-Divinylperfluorohexane (DVPFH), 1H,1H,6H,6H-Perfluorohexanediol perfluorohexanediol diacrylate (PFHDA) and glyoxal bis (diallyl acetal) (GBDA).

17-30. (Cancelled)

U.S. Serial No. 15/735,130 Page 8

31. (Previously Presented) The device or component according to claim 1, wherein the compound of formula I(a) has the following formula:

where n is from 2 to 10.

32. (Previously Presented) The device or component thereof according to claim 1, wherein the compound of formula I(a) has the following formula:

$$\begin{array}{c} H \\ \\ \\ H \end{array} \begin{array}{c} O \\ \\ CH_3 \end{array} (CF_2)_n F$$

where n is from 2 to 10.

- 33. (Previously Presented) The device or component according to claim 31, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-perfluorohexyl acrylate (PFAC4), 1H,1H,2H,2H-perfluoroctyl acrylate (PFAC6), 1H,1H,2H,2H-perfluorodecyl acrylate (PFAC8) and 1H,1H,2H,2H-perfluorododecyl acrylate (PFAC10).
- 34. (Previously Presented) The device or component thereof according to claim 32, wherein the compound of formula I(a) is selected from 1H,1H,2H,2H-pefluorohexyl methacrylate (PFMAC4), 1H,1H,2H,2H-perfluorooctyl methacrylate (PFMAC6) and 1H,1H,2H,2H-perfluorodecyl methacrylate (PFMAC8).

35-91. (Cancelled)

# **Drawing Amendments**

Please replace each of original figures 1-7 with each of the respective, enclosed replacement sheets, which shall replace each of original figures 1-7.

Each of figures 1-7 are amended to replace "Figure," which precedes each of the view numbers, with "Fig." as requested in the Office Action.

U.S. Patent and Trademark Unice; U.S. DEFAN INENT OF COMMENCE.

PÆ	PATENT APPLICATION FEE DETERMINATION RECORD Substitute for Form PTO-875				Application	n or Docket Number 5/735,130	Filing Date 12/08/2017	To be Mailed	
								LARGE SM	IALL MICRO
					ATION AS FII	LED - PAR	RT I		
	FOR		(Column		(Column 2)		DATE (4)		/A\
	FOR BASIC FEE	IN	UMBER FI	TED	NUMBER EXTRA		RATE (\$)	+	FEE (\$)
	(37 CFR 1.16(a), (b), o	or (c))	N/A		N/A		N/A		
	SEARCH FEE (37 CFR 1.16(k), (i), or		N/A		N/A		N/A		
1	EXAMINATION FEE (37 GFR 1.16(o), (p), o		N/A		N/A		N/A		
	TAL CLAIMS DFR 1.16(i))		miı	nus 20 = *			x \$80 =		
IND	EPENDENT CLAIM	s	m	ninus 3 = *			x \$420 =	1	
	APPLICATION SIZE CFR 1.16(s))	FEE (37 of p for s frac	aper, the small entit	ation and drawin application size f y) for each addit of. See 35 U.S.C	fee due is \$310 i ional 50 sheets i	(\$155 or			
	MULTIPLE DEPENI	DENT CLAIM PR	ESENT (37	' CFR 1.16(j))					
* If th	ne difference in co	olumn 1 is less	han zero	enter "0" in colu	ımn 2.		TOTAL		
				APPLICA1	TION AS AME	NDED - PA	ART II		
		(Column 1)		(Column 2)	(Column 3	3)			
ΞΝΤ	03/18/2019	CLAIMS REMAINING AFTER AMENDMENT		HIGHEST NUMBER PREVIOUSLY PAID FOR	PRESENT EX	(TRA	RATE (\$)	ADDIT	IONAL FEE (\$)
Ĭ	Total (37 CFR 1.16(i))	* 15	Minus	** 20	= 0		x \$100 =		0
AMENDMENT	Independent (37 CFR 1.16(h))	* 1	Minus	*** 3	= 0		x \$460 =		0
₹	Application S	Size Fee (37 Cl	R 1.16(s	))					
	☐ FIRST PRES	SENTATION O	F MULTIF	PLE DEPENDEN	IT CLAIM (37 CF	-R			
	377					•	TOTAL ADD'L FE	E	0
		(Column 1)		(Column 2)	(Column 3	3)			
F		CLAIMS REMAINING AFTER AMENDMENT		HIGHEST NUMBER PREVIOUSLY PAID FOR	PRESENT EX	(TRA	RATE (\$)	ADDIT	IONAL FEE (\$)
MEN	Total (37 CFR 1.16(i))	*	Minus	**	=		x \$0 =		
AMENDM	Independent (37 CFR 1.16(h))	*	Minus	***	=		x \$0 =		
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٩	☐ FIRST PRES	SENTATION O	F MULTIF	PLE DEPENDEN	IT CLAIM (37 CF	-R			
							TOTAL ADD'L FE	E	
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***	f the "Highest Numb	per Previously Pa	d For" IN T	HIS SPACE is less	s than 3, enter "3".				
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This collection of information is required by 37 CFR 1.16. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.14. This collection is estimated to take 12 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, VA 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS

ADDRESS. SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, VA 22313-1450.

If you need assistance in completing the form, call 1-800-PTO-9199 and select option 2.

# **SCORE Placeholder Sheet for IFW Content**

Application Number: 15735130 Document Date: 03/18/2019

The presence of this form in the IFW record indicates that the following document type was received in electronic format on the date identified above. This content is stored in the SCORE database.

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# Drawing

At the time of document entry (noted above):

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- External customers may access SCORE content via PAIR using the Supplemental Content tab

Form Revision Date: August 26, 2013

### United States Patent and Trademark Office



UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office Address: COMMISSIONER FOR PATENTS

P.O. Box 1450 Alexandria, Virginia 22313-1450 www.uspto.gov

### NOTICE OF ALLOWANCE AND FEE(S) DUE

33072 7590 05/15/2019
KAGAN BINDER, PLLC
SUITE 200, MAPLE ISLAND BUILDING
221 MAIN STREET NORTH
STILLWATER, MN 55082

EXAMINER

BUIE-HATCHER, NICOLE M

ART UNIT PAPER NUMBER

1767

DATE MAILED: 05/15/2019

APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
15/735.130	12/08/2017	Stephen Richard COULSON	GBL0007/US	7101

TITLE OF INVENTION: COATINGS

APPLN. TYPE	ENTITY STATUS	ISSUE FEE DUE	PUBLICATION FEE DUE	PREV. PAID ISSUE FEE	TOTAL FEE(S) DUE	DATE DUE
nonprovisional	UNDISCOUNTED	\$1000	\$0.00	\$0.00	\$1000	08/15/2019

THE APPLICATION IDENTIFIED ABOVE HAS BEEN EXAMINED AND IS ALLOWED FOR ISSUANCE AS A PATENT. <u>PROSECUTION ON THE MERITS IS CLOSED</u>. THIS NOTICE OF ALLOWANCE IS NOT A GRANT OF PATENT RIGHTS. THIS APPLICATION IS SUBJECT TO WITHDRAWAL FROM ISSUE AT THE INITIATIVE OF THE OFFICE OR UPON PETITION BY THE APPLICANT. SEE 37 CFR 1.313 AND MPEP 1308.

THE ISSUE FEE AND PUBLICATION FEE (IF REQUIRED) MUST BE PAID WITHIN THREE MONTHS FROM THE MAILING DATE OF THIS NOTICE OR THIS APPLICATION SHALL BE REGARDED AS ABANDONED. THIS STATUTORY PERIOD CANNOT BE EXTENDED. SEE 35 U.S.C. 151. THE ISSUE FEE DUE INDICATED ABOVE DOES NOT REFLECT A CREDIT FOR ANY PREVIOUSLY PAID ISSUE FEE IN THIS APPLICATION. IF AN ISSUE FEE HAS PREVIOUSLY BEEN PAID IN THIS APPLICATION (AS SHOWN ABOVE), THE RETURN OF PART B OF THIS FORM WILL BE CONSIDERED A REQUEST TO REAPPLY THE PREVIOUSLY PAID ISSUE FEE TOWARD THE ISSUE FEE NOW DUE.

#### HOW TO REPLY TO THIS NOTICE:

I. Review the ENTITY STATUS shown above. If the ENTITY STATUS is shown as SMALL or MICRO, verify whether entitlement to that entity status still applies.

If the ENTITY STATUS is the same as shown above, pay the TOTAL FEE(S) DUE shown above.

If the ENTITY STATUS is changed from that shown above, on PART B - FEE(S) TRANSMITTAL, complete section number 5 titled "Change in Entity Status (from status indicated above)".

For purposes of this notice, small entity fees are 1/2 the amount of undiscounted fees, and micro entity fees are 1/2 the amount of small entity fees.

II. PART B - FEE(S) TRANSMITTAL, or its equivalent, must be completed and returned to the United States Patent and Trademark Office (USPTO) with your ISSUE FEE and PUBLICATION FEE (if required). If you are charging the fee(s) to your deposit account, section "4b" of Part B - Fee(s) Transmittal should be completed and an extra copy of the form should be submitted. If an equivalent of Part B is filed, a request to reapply a previously paid issue fee must be clearly made, and delays in processing may occur due to the difficulty in recognizing the paper as an equivalent of Part B.

III. All communications regarding this application must give the application number. Please direct all communications prior to issuance to Mail Stop ISSUE FEE unless advised to the contrary.

IMPORTANT REMINDER: Maintenance fees are due in utility patents issuing on applications filed on or after Dec. 12, 1980. It is patentee's responsibility to ensure timely payment of maintenance fees when due. More information is available at www.uspto.gov/PatentMaintenanceFees.

PART B - FEE(S) TRANSMITTAL Complete and send this form, together with applicable fee(s), by mail or fax, or via EFS-Web. Mail Stop ISSUE FEE By mail, send to: By fax, send to: (571)-273-2885 Commissioner for Patents P.O. Box 1450 Alexandria, Virginia 22313-1450 INSTRUCTIONS: This form should be used for transmitting the ISSUE FEE and PUBLICATION FEE (if required). Blocks 1 through 5 should be completed where appropriate. All further correspondence including the Patent, advance orders and notification of maintenance fees will be mailed to the current correspondence address as indicated unless corrected below or directed otherwise in Block 1, by (a) specifying a new correspondence address; and/or (b) indicating a separate "FEE ADDRESS" for maintenance fee notifications. Note: A certificate of mailing can only be used for domestic mailings of the Fee(s) Transmittal. This certificate cannot be used for any other accompanying CURRENT CORRESPONDENCE ADDRESS (Note: Use Block 1 for any change of address) papers. Each additional paper, such as an assignment or formal drawing, must have its own certificate of mailing or transmission. Certificate of Mailing or Transmission 33072 7590 05/15/2019 I hereby certify that this Fee(s) Transmittal is being deposited with the United KAGAN BINDER, PLLC States Postal Service with sufficient postage for first class mail in an envelope SUITE 200, MAPLE ISLAND BUILDING addressed to the Mail Stop ISSUE FEE address above, or being transmitted to the USPTO via EFS-Web or by facsimile to (571) 273-2885, on the date below. 221 MAIN STREET NORTH (Typed or printed name STILLWATER, MN 55082 (Signature (Date APPLICATION NO. FILING DATE FIRST NAMED INVENTOR ATTORNEY DOCKET NO. CONFIRMATION NO. 15/735,130 12/08/2017 Stephen Richard COULSON GBL0007/US 7101 TITLE OF INVENTION: COATINGS APPLN. TYPE ENTITY STATUS ISSUE FEE DUE PUBLICATION FEE DUE PREV. PAID ISSUE FEE TOTAL FEE(S) DUE DATE DUE nonprovisional UNDISCOUNTED \$1000 \$0.00 \$0.00 \$1000 08/15/2019 EXAMINER ART UNIT CLASS-SUBCLASS BUIE-HATCHER, NICOLE M 1767 526-245000 1. Change of correspondence address or indication of "Fee Address" (37 For printing on the patent front page, list CFR 1.363). (1) The names of up to 3 registered patent attorneys or agents OR, alternatively, ☐ Change of correspondence address (or Change of Correspondence Address form PTO/SB/122) attached. (2) The name of a single firm (having as a member a registered attorney or agent) and the names of up to 2 registered patent attorneys or agents. If no name is "Fee Address" indication (or "Fee Address" Indication form PTO/ listed, no name will be printed. SB/47; Rev 03-09 or more recent) attached. Use of a Customer Number is required. 3. ASSIGNEE NAME AND RESIDENCE DATA TO BE PRINTED ON THE PATENT (print or type) PLEASE NOTE: Unless an assignee is identified below, no assignee data will appear on the patent. If an assignee is identified below, the document must have been previously recorded, or filed for recordation, as set forth in 37 CFR 3.11 and 37 CFR 3.81(a). Completion of this form is NOT a substitute for filing an assignment. (A) NAME OF ASSIGNEE (B) RESIDENCE: (CITY and STATE OR COUNTRY) Please check the appropriate assignee category or categories (will not be printed on the patent) : 🔲 Individual 🖵 Corporation or other private group entity 🖵 Government ☐Issue Fee Publication Fee (if required) Advance Order - # of Copies 4a. Fees submitted: 4b. Method of Payment: (Please first reapply any previously paid fee shown above) Lectronic Payment via EFS-Web ■ Enclosed check Non-electronic payment by credit card (Attach form PTO-2038) 🖵 The Director is hereby authorized to charge the required fee(s), any deficiency, or credit any overpayment to Deposit Account No. 5. Change in Entity Status (from status indicated above) NOTE: Absent a valid certification of Micro Entity Status (see forms PTO/SB/15A and 15B), issue Applicant certifying micro entity status. See 37 CFR 1.29 fee payment in the micro entity amount will not be accepted at the risk of application abandonment. NOTE: If the application was previously under micro entity status, checking this box will be taken Applicant asserting small entity status. See 37 CFR 1.27 to be a notification of loss of entitlement to micro entity status. NOTE: Checking this box will be taken to be a notification of loss of entitlement to small or micro Applicant changing to regular undiscounted fee status.

entity status, as applicable.

NOTE: This form must be signed in accordance with 37 CFR 1.31 and 1.33. See 37 CFR 1.4 for signature requirements and certifications.

Authorized Signature _

Typed or printed name

Date _

Registration No. _

### United States Patent and Trademark Office



UNITED STATES DEPARTMENT OF COMMERCE United States Patent and Trademark Office

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Alexandria, Virginia 22313-1450 www.uspto.gov

APPLICATION NO. FILING DATE FIRST NAMED INVENTOR ATTORNEY DOCKET NO. CONFIRMATION NO. 15/735,130 12/08/2017 Stephen Richard COULSON GBL0007/US 7101 **EXAMINER** 33072 7590 05/15/2019 KAGAN BINDER, PLLC BUIE-HATCHER, NICOLE M SUITE 200, MAPLE ISLAND BUILDING ART UNIT PAPER NUMBER 221 MAIN STREET NORTH STILLWATER, MN 55082 1767 DATE MAILED: 05/15/2019

### Determination of Patent Term Adjustment under 35 U.S.C. 154 (b)

(Applications filed on or after May 29, 2000)

The Office has discontinued providing a Patent Term Adjustment (PTA) calculation with the Notice of Allowance.

Section 1(h)(2) of the AIA Technical Corrections Act amended 35 U.S.C. 154(b)(3)(B)(i) to eliminate the requirement that the Office provide a patent term adjustment determination with the notice of allowance. See Revisions to Patent Term Adjustment, 78 Fed. Reg. 19416, 19417 (Apr. 1, 2013). Therefore, the Office is no longer providing an initial patent term adjustment determination with the notice of allowance. The Office will continue to provide a patent term adjustment determination with the Issue Notification Letter that is mailed to applicant approximately three weeks prior to the issue date of the patent, and will include the patent term adjustment on the patent. Any request for reconsideration of the patent term adjustment determination (or reinstatement of patent term adjustment) should follow the process outlined in 37 CFR 1.705.

Any questions regarding the Patent Term Extension or Adjustment determination should be directed to the Office of Patent Legal Administration at (571)-272-7702. Questions relating to issue and publication fee payments should be directed to the Customer Service Center of the Office of Patent Publication at 1-(888)-786-0101 or (571)-272-4200.

#### OMB Clearance and PRA Burden Statement for PTOL-85 Part B

The Paperwork Reduction Act (PRA) of 1995 requires Federal agencies to obtain Office of Management and Budget approval before requesting most types of information from the public. When OMB approves an agency request to collect information from the public, OMB (i) provides a valid OMB Control Number and expiration date for the agency to display on the instrument that will be used to collect the information and (ii) requires the agency to inform the public about the OMB Control Number's legal significance in accordance with 5 CFR 1320.5(b).

The information collected by PTOL-85 Part B is required by 37 CFR 1.311. The information is required to obtain or retain a benefit by the public which is to file (and by the USPTO to process) an application. Confidentiality is governed by 35 U.S.C. 122 and 37 CFR 1.14. This collection is estimated to take 30 minutes to complete, including gathering, preparing, and submitting the completed application form to the USPTO. Time will vary depending upon the individual case. Any comments on the amount of time you require to complete this form and/or suggestions for reducing this burden, should be sent to the Chief Information Officer, U.S. Patent and Trademark Office, U.S. Department of Commerce, P.O. Box 1450, Alexandria, Virginia 22313-1450. DO NOT SEND FEES OR COMPLETED FORMS TO THIS ADDRESS. SEND TO: Commissioner for Patents, P.O. Box 1450, Alexandria, Virginia 22313-1450. Under the Paperwork Reduction Act of 1995, no persons are required to respond to a collection of information unless it displays a valid OMB control number.

### **Privacy Act Statement**

The Privacy Act of 1974 (P.L. 93-579) requires that you be given certain information in connection with your submission of the attached form related to a patent application or patent. Accordingly, pursuant to the requirements of the Act, please be advised that: (1) the general authority for the collection of this information is 35 U.S.C. 2(b) (2); (2) furnishing of the information solicited is voluntary; and (3) the principal purpose for which the information is used by the U.S. Patent and Trademark Office is to process and/or examine your submission related to a patent application or patent. If you do not furnish the requested information, the U.S. Patent and Trademark Office may not be able to process and/or examine your submission, which may result in termination of proceedings or abandonment of the application or expiration of the patent.

The information provided by you in this form will be subject to the following routine uses:

- The information on this form will be treated confidentially to the extent allowed under the Freedom of Information Act (5 U.S.C. 552) and the Privacy Act (5 U.S.C 552a). Records from this system of records may be disclosed to the Department of Justice to determine whether disclosure of these records is required by the Freedom of Information Act.
- 2. A record from this system of records may be disclosed, as a routine use, in the course of presenting evidence to a court, magistrate, or administrative tribunal, including disclosures to opposing counsel in the course of settlement negotiations.
- 3. A record in this system of records may be disclosed, as a routine use, to a Member of Congress submitting a request involving an individual, to whom the record pertains, when the individual has requested assistance from the Member with respect to the subject matter of the record.
- 4. A record in this system of records may be disclosed, as a routine use, to a contractor of the Agency having need for the information in order to perform a contract. Recipients of information shall be required to comply with the requirements of the Privacy Act of 1974, as amended, pursuant to 5 U.S.C. 552a(m).
- 5. A record related to an International Application filed under the Patent Cooperation Treaty in this system of records may be disclosed, as a routine use, to the International Bureau of the World Intellectual Property Organization, pursuant to the Patent Cooperation Treaty.
- 6. A record in this system of records may be disclosed, as a routine use, to another federal agency for purposes of National Security review (35 U.S.C. 181) and for review pursuant to the Atomic Energy Act (42 U.S.C. 218(c)).
- 7. A record from this system of records may be disclosed, as a routine use, to the Administrator, General Services, or his/her designee, during an inspection of records conducted by GSA as part of that agency's responsibility to recommend improvements in records management practices and programs, under authority of 44 U.S.C. 2904 and 2906. Such disclosure shall be made in accordance with the GSA regulations governing inspection of records for this purpose, and any other relevant (i.e., GSA or Commerce) directive. Such disclosure shall not be used to make determinations about individuals.
- 8. A record from this system of records may be disclosed, as a routine use, to the public after either publication of the application pursuant to 35 U.S.C. 122(b) or issuance of a patent pursuant to 35 U.S.C. 151. Further, a record may be disclosed, subject to the limitations of 37 CFR 1.14, as a routine use, to the public if the record was filed in an application which became abandoned or in which the proceedings were terminated and which application is referenced by either a published application, an application open to public inspection or an issued patents.
- is referenced by either a published application, an application open to public inspection or an issued patents.

  9. A record from this system of records may be disclosed, as a routine use, to a Federal, State, or local law enforcement agency, if the USPTO becomes aware of a violation or potential violation of law or regulation.

	<b>Application No.</b> 15/735,130	Applicant(s	
Notice of Allowability	Examiner NICOLE M BUIE-HATCHER	Art Unit 1767	AIA (FITF) Status Yes
The MAILING DATE of this communication appear All claims being allowable, PROSECUTION ON THE MERITS IS (herewith (or previously mailed), a Notice of Allowance (PTOL-85) of NOTICE OF ALLOWABILITY IS NOT A GRANT OF PATENT RICE of the Office or upon petition by the applicant. See 37 CFR 1.313	OR REMAINS) CLOSED in thi or other appropriate communic GHTS. This application is subje	s application. If not ation will be mailed	t included d in due course. <b>THIS</b>
1. ☐ This communication is responsive to response filed 03/18/20 ☐ A declaration(s)/affidavit(s) under 37 CFR 1.130(b) was/			
2. An election was made by the applicant in response to a rest restriction requirement and election have been incorporated		ring the interview of	on; the
3. The allowed claim(s) is/are See Continuation Sheet. As a repart Prosecution Highway program at a participating into information, please see <a href="http://www.uspto.gov/patents/inityPPHfeedback@uspto.gov">http://www.uspto.gov/patents/inityPPHfeedback@uspto.gov</a> .	ellectual property office for the	corresponding app	
4. Acknowledgment is made of a claim for foreign priority unde	er 35 U.S.C. § 119(a)-(d) or (f).		
Certified copies:			
a) □AII b) □ Some *c) □ None of the:			
<ol> <li>Certified copies of the priority documents have</li> <li>Certified copies of the priority documents have</li> </ol>		No	
Copies of the certified copies of the priority do	, ,		e application from the
International Bureau (PCT Rule 17.2(a)).		· ····o ···aiio···ai· otag	
* Certified copies not received:			
Applicant has THREE MONTHS FROM THE "MAILING DATE" noted below. Failure to timely comply will result in ABANDONM THIS THREE-MONTH PERIOD IS NOT EXTENDABLE.		reply complying wit	th the requirements
5. CORRECTED DRAWINGS (as "replacement sheets") must	be submitted.		
including changes required by the attached Examiner's Paper No./Mail Date		:he Office action of	
Identifying indicia such as the application number (see 37 CFR 1. sheet. Replacement sheet(s) should be labeled as such in the hea		_	t (not the back) of each
6. DEPOSIT OF and/or INFORMATION about the deposit of B attached Examiner's comment regarding REQUIREMENT F	IOLOGICAL MATERIAL must OR THE DEPOSIT OF BIOLO	be submitted. Note GICAL MATERIAL	e the 
Attachment(s)			
1. ✓ Notice of References Cited (PTO-892)	5. 🗹 Examiner's A		
2. ✓ Information Disclosure Statements (PTO/SB/08), Paper No./Mail Date 20181121 / 20190318.	6. <b>☑</b> Examiner's S	tatement of Reasor	ns for Allowance
Examiner's Comment Regarding Requirement for Deposit of Biological Material	7. 🗌 Other		
4. ✓ Interview Summary (PTO-413), Paper No./Mail Date. 20190430.			
/NICOLE M. BUIE-HATCHER/			
Primary Examiner, Art Unit 1767			

U.S. Patent and Trademark Office PTOL-37 (Rev. 08-13)

Notice of Allowability

Part of Paper No./Mail Date 20190430

Continuation of 3. The allowed claim(s) is/are: 1-3,6,8-11,13,15-16 and 31-34

Application/Control Number: 15/735,130

Art Unit: 1767

# Notice of Pre-AIA or AIA Status

The present application, filed on or after March 16, 2013, is being examined under the first inventor to file provisions of the AIA.

#### **EXAMINER'S AMENDMENT**

An examiner's amendment to the record appears below. Should the changes and/or additions be unacceptable to applicant, an amendment may be filed as provided by 37 CFR 1.312. To ensure consideration of such an amendment, it MUST be submitted no later than the payment of the issue fee.

Authorization for this examiner's amendment was given in an interview with Paul Parins (Reg. No. 54,358) on 05/09/2019.

The application has been amended as follows:

In claim 1, line 23, replace "Formula A" with "formula A".

In claim 1, line 27, replace "Formula B" with "formula B".

In claim 8, line 1, replace "whereinwhen" with "wherein when".

In claim 8, line 2, replace "Formula B" with "formula A".

In claim 13, line 2, replace "Formula B" with "formula B".

In claim 15, line 2, replace "Formula B" with "formula B".

#### Reasons for Allowance

The following is an examiner's statement of reasons for allowance:

Page 2

Application/Control Number: 15/735,130

Art Unit: 1767

A) The closest prior art of record, Coulson (WO 2007/082122 A1) does not teach or suggest the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moieties and has a boiling point of less than 500°C at standard pressure, the crosslinking reagent having one of the structures (i) or (ii).

B) Coulson (US 2010/0203347 A1) teaches a compound with more than one double bond [0030] and alkyne compounds comprise chains of carbon atoms, including one or more carbon-carbon triple bonds [0046]. However, Coulson (US '347) does not teach or suggest the crosslinking reagent comprises two or more unsaturated bonds attached by means of one or more linker moieties and has a boiling point of less than 500°C at standard pressure, the crosslinking reagent having one of the structures (i) or (ii).

Any comments considered necessary by applicant must be submitted no later than the payment of the issue fee and, to avoid processing delays, should preferably accompany the issue fee. Such submissions should be clearly labeled "Comments on Statement of Reasons for Allowance."

#### Prior Art

The prior art made of record and not relied upon is considered pertinent to applicant's disclosure. Chen et al. (CN 103881013 A, Machine translation attached) teaches a fluorine containing hydrogel.

Page 3

Application/Control Number: 15/735,130 Page 4

Art Unit: 1767

Correspondence

Any inquiry concerning this communication or earlier communications from the

examiner should be directed to NICOLE M. BUIE-HATCHER whose telephone number is

(571)270-3879. The examiner can normally be reached on M-F, 8 am - 4:30 pm.

Examiner interviews are available via telephone, in-person, and video conferencing using

a USPTO supplied web-based collaboration tool. To schedule an interview, applicant is

encouraged to use the USPTO Automated Interview Request (AIR) at

http://www.uspto.gov/interviewpractice.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's

supervisor, Mark Eashoo can be reached on (571)272-1197. The fax phone number for the

organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent

Application Information Retrieval (PAIR) system. Status information for published applications

may be obtained from either Private PAIR or Public PAIR. Status information for unpublished

applications is available through Private PAIR only. For more information about the PAIR

system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR

system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would

like assistance from a USPTO Customer Service Representative or access to the automated

information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/NICOLE M. BUIE-HATCHER/

Primary Examiner, Art Unit 1767

Application/Control Number: 15/735,130 Art Unit: 1767

Page 5

Examiner-Initiated Interview Summary	15/735,130	COULSON et al.	
Examiner-initiated interview Summary	Examiner	Art Unit	AIA (FITF) Status
	NICOLE M BUIE-HATCHER	1767	Yes
All participants (applicant, applicant's representative, PTO per	rsonnel):		
(1) NICOLE M BUIE-HATCHER.	(3)		
(2) <u>Paul Parins (Reg. No. 54,358)</u> .	(4)		
Date of Interview: 09 May 2019.			
Type: ☑ Telephonic ☐ Video Conference ☐ Personal [copy given to: ☐ applicant ☐ app	olicant's representative]		
Exhibit shown or demonstration conducted:	0.		
Issues Discussed 101 112 102 103 (For each of the checked box(es) above, please describe below the issue and detailed description	Others of the discussion)		
Claim(s) discussed: <u>1-3,6,8-11,13,15-16 and 31-34</u> .			
Identification of prior art discussed: N/A.			
Substance of Interview (For each issue discussed, provide a detailed description and indicate if agreement wa or a portion thereof, claim interpretation, proposed amendments, arguments of any approximately approximately according to the control of the control o	• •	ntification or clari	fication of a reference
On 5/1/19, the Examiner stated that claim 8 did not further lime. Formula should be lowercase in each instance in claims 1, 8, should be separated in claim 8. On 5/9/19, Mr. Parrins called	13, and 15. Furthermore the	words "where	
Applicant recordation instructions: It is not necessary for applicant to provi	de a separate record of the substance	e of interview.	
<b>Examiner recordation instructions</b> : Examiners must summarize the substance of an interview should include the items listed in MPEP 713.04 for thrust of each argument or issue discussed, a general indication of any other outcome of the interview, to include an indication as to whether or not agreement.	complete and proper recordation inclu pertinent matters discussed regarding	ıding the identifi ı patentability an	cation of the general
☐ Attachment			
/NICOLE M. BUIE-HATCHER/ Primary Examiner, Art Unit 1767			

Application No.

U.S. Patent and Trademark Office PTOL-413B (Rev. 8/11/2010) Applicant(s)

		Mating of Deference	- Cit-d		Application/ 15/735,130	Control No.		Applicant(s)/Par Reexamination COULSON et a	
Notice of References Cited				Examiner NICOLE M	BUIE-HATCHER		Art Unit 1767	Page 1 of 1	
				U.S. F	PATENT DOCU	MENTS			
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*		Inclu	ıde as applicat	ole: Author	, Title Date, Pub	lisher, Edition or Volur	me, F	ertinent Pages)	
	U	Machine translation of CN 103	881013 A, re	trived Apr	il 2019. (Year:	2019)			
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	w								

*A copy of this reference is not being furnished with this Office action. (See MPEP § 707.05(a).) Dates in MM-YYYY format are publication dates. Classifications may be US or foreign.

U.S. Patent and Trademark Office PTO-892 (Rev. 01-2001)

Χ

Notice of References Cited

Part of Paper No. 20190430

Search Notes

Application/Control No.	Applicant(s)/Patent Under Reexamination
15/735,130	COULSON et al.
Examiner	Art Unit
NICOLE M BUIE-HATCHER	1767

CPC - Searched*				
Symbol	Date	Examiner		
C09D 133/16	05/09/2019	NMB		
B05D 1/62, 5/083	05/09/2019	NMB		
H01L 21/0274	05/09/2019	NMB		

CPC Combination Sets - Searched*		
Symbol	Date	Examiner

US Classification - Searched*				
Class	Subclass	Date	Examiner	

^{*} See search history printout included with this form or the SEARCH NOTES box below to determine the scope of the search.

Search Notes				
Search Notes	Date	Examiner		
Inventor name search, PALM and EAST	03/26/2019	NMB		
Global Dossier	04/30/2019	NMB		
Scifinder	04/30/2019	NMB		
EAST text search, see search history	04/30/2019	NMB		

Interference Search			
US Class/CPC Symbol	US Subclass/CPC Group Date Examiner		
C09D	133/16	05/09/2019	NMB
B05D	1/62 , 5/083	05/09/2019	NMB
H01L	21/0274	05/09/2019	NMB

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	Application/Control No.	Applicant(s)/Patent Under Reexamination
Search Notes	15/735,130	COULSON et al.
	Examiner	Art Unit
	NICOLE M BUIE-HATCHER	1767

E 4000 E 1 440

Issue Classif	ication

Application/Control No.	Applicant(s)/Patent Under Reexamination
15/735,130	COULSON et al.
Examiner	Art Unit
NICOLE M BUIE-HATCHER	1767

CPC					
Symbol				Туре	Version
C09D	/ 133	1	16	F	2013-01-01
C09D	/ 5	1	00	l I	2013-01-01
B05D	/ 5	J	08	I	2013-01-01
B05D	/ 1	1	62	I	2013-01-01
B05D	/ 5	1	083	I	2013-01-01
H01J	/ 37	1	32082	I	2013-01-01
H01J	/ 37	J	32467	I	2013-01-01
H01J	/ 2237	f	332	A	2013-01-01
H01J	/ 2237	7	3382	A	2013-01-01
H01L	/ 21	1	02118	A	2013-01-01
H01L	/ 21	1	02274	A	2013-01-01
H01L	/ 21	7	56	A	2013-01-01
H01L	/ 23		293	A	2013-01-01
H01L	/ 23	J	564	A	2013-01-01
H05K	/ 3	J	285	Α	2013-01-01

CPC Combination Sets					
Symbol	Туре	Set	Ranking	Version	

NONE		Total Claims	s Allowed:
(Assistant Examiner)	(Date)	15	5
/NICOLE M. BUIE-HATCHER/ Primary Examiner, Art Unit 1767	09 May 2019	O.G. Print Claim(s)	O.G. Print Figure
(Primary Examiner)	(Date)	1	none

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Issue	e Classification

Application/Control No.	Applicant(s)/Patent Under Reexamination
15/735,130	COULSON et al.
Examiner	Art Unit
NICOLE M BUIE-HATCHER	1767

INTERNATIONAL CLASSIFICATION		
CLAIMED		
C09D133/16	/ 133	/ 16
C09D5/00	/ 5	00
B05D5/08	<i>f</i> 5	08
B05D1/00	1	<i>t</i> 00
H01L21/02	/ 21	<i>t</i> 02
H01L21/56	/ 21	56
H01L23/29	/ 23	/ 29
H01L23/00	/ 23	<i>∦</i> 00
H05K3/28	/ 3	28
NON-CLAIMED		
H01J37/32	/ 37	32

US ORIGINAL CLASSIFICATION						
CLASS	SUBCLASS					

CROSS REFERENCES(S)						
CLASS		SUBCLASS (ONE SUBCLASS PER BLOCK)				

NONE	Total Claims Allowed:			
(Assistant Examiner)	(Date)	15		
/NICOLE M. BUIE-HATCHER/ Primary Examiner, Art Unit 1767	09 May 2019	O.G. Print Claim(s)	O.G. Print Figure	
(Primary Examiner)	(Date)	1	none	

Issue Classif	ication

Application/Control No.	Applicant(s)/Patent Under Reexamination
15/735,130	COULSON et al.
Examiner	Art Unit
NICOLE M BUIE-HATCHER	1767

	Claims renumbered in the same order as presented by applicant CPA T.D. R.1.47														
CLAIM	CLAIMS														
Final	Original	Final	Original	Final	Original	Final	Original	Final	Original	Final	Original	Final	Original	Final	Original
1	1	9	13	-	25	-	37	-	49	-	61	-	73	-	85
2	2	-	14	-	26	-	38	-	50	-	62	-	74	-	86
3	3	10	15	-	27	=	39	-	51	-	63	-	75	-	87
-	4	11	16	-	28	-	40	-	52	-	64	-	76	-	88
-	5	-	17	-	29	-	41	-	53	-	65	-	77	-	89
4	6	-	18	-	30	-	42	-	54	-	66	-	78	-	90
-	7	-	19	12	31	-	43	-	55	-	67	-	79	-	91
5	8	-	20	14	32	-	44	-	56	-	68	-	80		
6	9	-	21	13	33	-	45	-	57	-	69	-	81		
7	10	-	22	15	34	-	46	-	58	-	70	-	82		
8	11	-	23	-	35	-	47	-	59	-	71	-	83		
-	12	-	24	-	36	-	48	-	60	-	72	-	84		

NONE	Total Claims Allowed:			
(Assistant Examiner)	(Date)	15		
/NICOLE M. BUIE-HATCHER/ Primary Examiner, Art Unit 1767	09 May 2019	O.G. Print Claim(s)	O.G. Print Figure	
(Primary Examiner)	(Date)	1	none	

U.S. Patent and Trademark Office

## **EAST Search History**

## **EAST Search History (Prior Art)**

Ref #	Hits	Search Query	DBs	Default Operator	Plurals	Time Stamp
S32	5	("9617353").PN.	US- PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT	OR	OFF	2019/03/26 17:33
S33	13	("20040028931"   "20040046165"   "20090267489"   "20100189914"   "20120009231"   "20120051018"   "20120171386"   "20120321775"   "20140010969"   "20140322525"   "20150291830"   "4822632"   "6764758").PN.	US- PGPUB; USPAT	OR	ON	2019/03/26 17:38
S34	2	("6764758").PN.	US- PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT	OR	OFF	2019/03/26 17:56
S35	14	("4822632" "20140010969" "20120321776" "20120051018" "20140010969" "20120321776" "20120171386" "4822632").pn.	US- PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT	OR	ON	2019/04/29 17:01
S41	832	(c09d133/16 b05d5/083).cpc. AND b05d1/62.cpc.	US- PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT	OR	ON	2019/04/30 10:23
S40	0	(c09d133/16 b05d5/083).cpc. SAME b05d1/62.cpc.	US- PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT	OR	ON	2019/04/30 10:23
S42	12	(c09d133/16 b05d5/083).cpc. AND h01l21/0274.cpc.	US- PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT	OR	ON	2019/04/30 10:25
S44	75	S43 AND (coulson-s\$ evans-d\$ siokou-a\$	US-	OR	ON	2019/04/30

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		telford-c\$).in.	PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT			16:37
S43	365816	((\$3fluor\$4 fluorin\$4) near5 (monomer\$3 acry \$4 methacry \$4)) or (\$3fluor\$4 fluorin\$4) near5 olefin or (\$3fluor\$4 fluorin\$4) near5 (alken\$2 alkenylene vinyl) \$3fluoroolefin\$2 \$3fluoroalkene\$2 \$3fluoroalkeny \$2 \$3fluoromonomer\$2 \$3fluoroalkylene\$2 polyfluoroalkyl polyfluoroalkeny  \$3fluorocarbon\$2 \$3fluoro\$viny \$2 \$3fluoromethacry \$4 \$3fluoroacry \$4	US- PGPUB; USPAT; FPRS; EPO; JPO; DERWENT	OR	ON	2019/04/30 16:37
S45	17	S43 AND "p2i".as.	US- PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT	OR	ON	2019/04/30 16:45
S46	2	cn-103881013-\$.did.	US- PGPUB; USPAT; USOCR; FPRS; EPO; JPO; DERWENT	OR	ON	2019/04/30 18:27

# **EAST Search History (Interference)**

Ref #	Hits	Search Query	DBs	Default Operator	Plurals	Time Stamp
L3	232749	((\$3fluor\$4 fluorin\$4) near5 (monomer\$3 acry \$4 methacry \$4)) or (\$3fluor\$4 fluorin\$4) near5 olefin or (\$3fluor\$4 fluorin\$4) near5 (alken\$2 alkenylene vinyl) \$3fluoroolefin\$2 \$3fluoroalkene\$2 \$3fluoroalkeny \$2 \$3fluoromonomer\$2 \$3fluoroalkylene\$2 polyfluoroalkyl polyfluoroalkeny  \$3fluorocarbon\$2 \$3fluoro\$viny \$2 \$3fluoromethacry \$4 \$3fluoroacry \$4	US- PGPUB; USPAT	OR	ON	2019/05/09 16:55
L2	1250650	crosslink\$6 (cross ADJ link\$6) curable cured curing vulcaniz\$6	US- PGPUB; USPAT	OR	ON	2019/05/09 16:55
L1	29561	(divinyl WITH adipate) (butanediol WITH divinyl WITH ether) (cyclohexanedimethanol WITH divinyl WITH ether) (octadiene) trivinylcyclohexane divinyltetramethyldisiloxane (diallyl WITH cyclohexanedicarboxylate) divinylperfluorohexane (perfluorohexanediol WITH diacrylate) (glyoxal WITH bis wITH diallyl WITH acetal) dialkyn\$2 ((di bis) NEAR2 alkyn\$2)	US- PGPUB; USPAT	OR	ON	2019/05/09 16:55
L4	129	L1 SAME L2 SAME L3	US- PGPUB; USPAT	OR	ON	2019/05/09 17:02

Ex. 1028 - Favored - p. 418

L6	29	L1 SAME3 L2 SAME3 L3 SAME3 plasma	US-	OR	ON	2019/05/09
			PGPUB;			17:03
<b>L</b>			USPAT			
L5	3	L1 SAME L2 SAME L3 SAME plasma	US-	OR	ON	2019/05/09
			PGPUB;			17:03
			USPAT			

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Receipt date: 03/18/2019

15/735,130 - GAU: 1767

Doc code: IDS Doc description: Information Disclosure Statement (IDS) Filed PTO/SB/08a (03-15)
Approved for use through 07/31/2016. OMB 0651-0031
U.S. Patent and Trademark Office; U.S. DEPARTMENT OF COMMERCE

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INFORMATION DISCLOSURE	Application Number	15	5735130	
	Filing Date		017-12-08	
	First Named Inventor Stephen		en Richard COULSON	
STATEMENT BY APPLICANT (Not for submission under 37 CFR 1.99)	Art Unit	17	1767	
(Not lot submission under or of N 1.33)	Examiner Name	BUIE-HATCHER, NICOLE M		
	Attorney Docket Number		GBL0007/US	

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	Examiner Initial*	Cite No	Patent Number	Kind Code ¹	Issue Date	Name of Patentee or Applicant of cited Document	Pages,Columns,Lines where Relevant Passages or Relevant Figures Appear
/1	N.M.B./	1	4822632	А	1989-04-18	Williams et al.	
/r	I.M.B./	2	6764758	B1	2004-07-20	Grunze et al.	
	If you wisl	n to add	additional U.S. Pate	nt citatio	n information pl	ease click the Add button.	Add
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	Examiner Initial*	Cite No	Publication Number	Kind Code ¹	Publication Date	Name of Patentee or Applicant of cited Document	Pages,Columns,Lines where Relevant Passages or Relevant Figures Appear
/	N.M.B./	1	20040028931	A1	2004-02-12	Bletsos et al.	
/	N.M.B./	2	20120051018	A1	2012-03-01	Oligaard	
/1	N.M.B./	3	20120171386	A1	2012-07-05	Bicker et al.	
/	N.M.B./	4	20120321775	A1	2012-12-20	Parker	

Receipt date: 03/18/2019 15/735,130 - GAU: 1767 Application Number 15735130 Filing Date 2017-12-08 INFORMATION DISCLOSURE First Named Inventor Stephen Richard COULSON STATEMENT BY APPLICANT Art Unit 1767 ( Not for submission under 37 CFR 1.99) **Examiner Name** BUIE-HATCHER, NICOLE M Attorney Docket Number GBL0007/US 2014-01-09 Α1 20140010969 Bicker et al. /M.M.B./ /N.M.B./ Α1 2009-10-29 20090267489 Arita et al. Α1 2010-07-29 Coulson 20100189914 /m.m.b./ 20120009231 Α1 2012-01-12 Herbert et al. /w.m.b./ Α1 2014-10-30 20140322525 egein et al. /N.M.B./ 10 20150291830 Α1 2015-10-15 Galbreath et al. /N.M.B./ Α1 2004-03-11 20040046165 Hunze et al. /N.M.B./ 11 Add If you wish to add additional U.S. Published Application citation information please click the Add button. Remove FOREIGN PATENT DOCUMENTS Pages, Columns, Lines Name of Patentee or Examiner Cite Foreign Document Country Kind **Publication** where Relevant **T**5 Applicant of cited Initial* No Number³ Code2i Code⁴ Date Passages or Relevant Document Figures Appear 2007-07-25 2434368 GB Α P2i Limited 1 /N.M.B./

P2i Ltd

/N.M.B./

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2013-01-30

Receipt date: 03/18/2019

# **INFORMATION DISCLOSURE STATEMENT BY APPLICANT**

( Not for submission under 37 CFR 1.99)

		15/735,130 - GAU: 1767					
Application Number		15735130					
Filing Date		2017-12-08					
First Named Inventor Stephe		en Richard COULSON					
Art Unit		1767					
Examiner Name BUIE		E-HATCHER, NICOLE M					
Attorney Docket Number	er	GBL0007/US					

/N.M.B./ 3	03063938	wo	A1	2003-08-07	ASTRAZENECA AB
/N.M.B./ 4	2005021833	wo	A2	2005-03-10	SURFACE INNOVATIONS LIMITED
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/N.M.B./ 10	02/28548	wo	A2	2002-04-11	DOW CORNING CORPORATION
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/N.M.B./ 13	201402 <del>696</del> 7	wo	A2	2014-02-20	EUROPLASMA NV

Receipt date: 03/18/2019 15/735,130 - GAU: 1767 Application Number 15735130 Filing Date 2017-12-08 INFORMATION DISCLOSURE First Named Inventor Stephen Richard COULSON STATEMENT BY APPLICANT Art Unit 1767 ( Not for submission under 37 CFR 1.99) **Examiner Name** BUIE-HATCHER, NICOLE M Attorney Docket Number GBL0007/US Add If you wish to add additional Foreign Patent Document citation information please click the Add button Remove **NON-PATENT LITERATURE DOCUMENTS** Include name of the author (in CAPITAL LETTERS), title of the article (when appropriate), title of the item Examiner Cite **T**5 (book, magazine, journal, serial, symposium, catalog, etc), date, pages(s), volume-issue number(s), Initials* publisher, city and/or country where published. Search Report under Section 17(5) from UK Patent Application No. GB1403558.8, mailed on August 13, 2014 (4 /N.M.B./ 1

5, 2015 (16 pages)

nternational Search Report and Written Opinion from International Application No. PCT/GB2015/050521, mailed June

Search Report under Section 17(5) from UK Patent Application No. GB1503054.7, mailed on November 30, 2015 (4 /N.M.B./ pages)

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### **EXAMINER SIGNATURE**

Date Considered **Examiner Signature** 04/29/2019 /NICOLE M. BUIE-HATCHER/

*EXAMINER: Initial if reference considered, whether or not citation is in conformance with MPEP 609. Draw line through a citation if not in conformance and not considered. Include copy of this form with next communication to applicant.

pages)

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Receipt date: 11/21/2018 15/735,130 - GAU: 1767

Doc code: IDS Doc description: Information Disclosure Statement (IDS) Filed PTO/SB/08a (03-15)
Approved for use through 07/31/2016. OMB 0651-0031
U.S. Patent and Trademark Office; U.S. DEPARTMENT OF COMMERCE

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	Application Number		15735130	
00.3 pm /m	Filing Date		2017-12-08	
INFORMATION DISCLOSURE	First Named Inventor Steph		hen Richard COULSON	
STATEMENT BY APPLICANT (Not for submission under 37 CFR 1.99)	Art Unit		1767	
(Notion Submission ander or or it iso)	Examiner Name But		BUIE-HATCHER, NICOLE M	
	Attorney Docket Numb	er	GBL0007/US	

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				FOREIG	N PAT	ENT DOCUM	ENTS			
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	date:			1 2 1 1 21 1		· ·	735,130 - GAU:	1.70
				Application Number	************	15735130		
			\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	Filing Date		2017-12-08		
			DISCLOSURE	First Named Inventor	First Named Inventor Stephen Richard COULSON			
STATEMENT BY APPLICANT (Not for submission under 37 CFR 1.99)				Art Unit	nntnnnnnnn	1767		
				Examiner Name	Examiner Name BUIE-HATCHER, NICOLE M			
				Attorney Docket Number		GBL0007/US		
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## (19) 中华人民共和国国家知识产权局



# (12) 发明专利申请



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(21) 申请号 201410100313.4

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(71) 申请人 南京工业大学 地址 211816 江苏省南京市浦口区浦珠南路 30 号

(72)发明人 陈苏 汤文岐 张强 周振方

(74) 专利代理机构 南京天华专利代理有限责任 公司 32218

代理人 徐冬涛 袁正英

(51) Int. Cl.

COSF 220/24 (2006, 01)

COSF 220/28 (2006, 01)

COSF 222/38 (2006, 01)

008F 220/20(2005,01)

COSF 220/22 (2006.01)

COSF 226/10 (2006, 01)

008F 220/06 (2006, 01)

GOSF 212/36 (2006, 01)

COBF 220/54 (2006, 01)

COSF 2/52 (2006, 01)

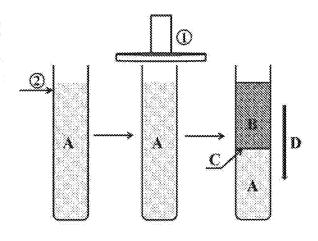
权利要求书1页 说明书4页 附图2页

### (54) 发明名称

采用等离子前端聚合快速制备含氟水凝胶的 方法

### (57) 摘要

本发明涉及一种采用等离子前端聚合快速制备含额水凝胶的方法。其具体步骤如下:将活性单体、含氮单体,引发剂和交联剂按比例溶于有机溶剂中,配成均匀溶液;将配成均匀溶液混合移入反应器中,对反应器上端利用低温等离子体处理,使引发剂受热分解,引发聚合反应;关闭等离子体反应器,依靠聚合反应放出的热向未反应区域扩散,直到整个反应器内所有原料全部转化为聚合物;用去离子水源洗,得到含氟水凝胶。该方法具有速度快、效率高、成本低、成型快等优点,避~免了传统制备方法中工艺流程复杂。时间长等缺~



CX 103881013 3

- 1. 采用等离子前端聚合快速制备含氟水凝胶的方法,其具体步骤如下:
- (1) 将活性单体、含氟单体、引发剂和交联剂按比例将于有机溶剂中,配成均匀溶液;
- (2) 将配成均匀溶液混合移入反应器中,对反应器上端利用低温等离子体处理,使引发 剂受热分解,引发聚合反应:
- (3) 关闭等离子体反应器,依靠聚合反应放出的热向未反应区域扩散,直到整个反应器 内所有原料全部转化为聚合物;
  - (4) 取出所得的聚合物,用去离子水漂洗,得到含氰水凝胶。
- 2. 根据权利要求 1 所述的方法, 其特征在于步骤 (1) 配成均匀溶液中, 活性单体占溶液总质量的 20 ~ 40%; 含氟单体占溶液总质量的 20 ~ 40%; 交联剂占溶液总质量的 0.5 ~ 5%; 引发剂所占质量比为 0.25 ~ 4%; 有机溶剂占溶液总质量的 15 ~ 45%。
- 3. 根据权利要求 1 所述的方法, 其特征在于 ; 所述活性单体为丙烯酸羟乙酯、丙烯酸羟丙酯、甲基丙烯酸羟乙酯、甲基丙烯酸羟丙酯、N, N-二甲基丙烯酰胺, 丙烯酰胺, 双丙酮丙烯酰胺、N-羟甲基丙烯酰胺、N-异丙基丙烯酰胺、N-乙烯基吡咯烷酮、N-乙烯基己内酰胺、丙烯酸辛、丙烯酸或甲基丙烯酸中的至少一种。
- 4. 根据权利要求1所述的方法,其特征在于;含氯单体为甲基丙烯酸氯烷烃酯、甲基丙烯酸全氯烷基乙基酯,甲基丙烯酸三氟乙酯、N-乙基全氟辛基磺酰胺基丙烯酸酸乙酯、全氟烷烃丙烯酸酯、甲基丙烯酸全氟辛基乙酯、丙烯酸六氟丁酯、甲基丙烯酸六氟丁酯、甲基丙烯酸十二氟庚酯、三氟甲基丙烯酸或丙烯酸全氟烷基乙酯中的一种。
- 5. 根据权利要求1所述的方法,其特征在于;所述有机容剂为;甘油、二甲亚砜、N,N'-二甲基甲酰胺或1-甲基-2-吡咯烷酮中的一种。
- 6. 根据权利要求 1 所述的方法, 其特征在于: 所述交联剂为 N, N'- 亚甲基双丙烯酰胺、二乙烯基苯或二甲基丙烯酸乙二醇酯中的一种。
- 7. 根据权利要求1所述的方法,其特征在于;所述引发剂为偶氮二异丁腈,偶氮二异庚腈,过氧化苯甲酰、过硫酸铵、过硫酸钾、叔丁基过氧化氢中的其中一种。
- 8. 根据权利要求 1 所述的方法, 其特征在于; 低温等离子体处理电压为  $30\sim80$ V; 电流为  $1\sim3$ A; 处理时间为  $5\sim60$ s。

## 采用等离子前端聚合快速制备含氟水凝胶的方法

### 技术领域

[0001] 本发明属于功能高分子材料制备技术领域,尤其涉及一种采用等离子前端聚合快速制备含氟水凝胶的方法。

### 背景技术

[0002] 水凝胶材料是一种能够在水中溶胀并保持大量水分且不被溶解的三维网络交联结构的新型高分子材料。根据不同的对环境刺激响应能力水凝胶材料可分为普通水凝胶和智能水凝胶。智能水凝胶能够感知外界环境的细微变化(如温度、pH值、电、光、磁场、特定生物分子等的变化),以体积显著变化的形式反馈给人们,从而实现与外界环境的信息交流。

[0003] 含氟聚合物具有高度化学稳定性、极强的耐高低温性能、突出的不粘性、异常的润滑性、优异的电绝缘性、耐老化性、抗辐射性和极小的吸水性等优良的综合性能,广泛应用于航空航天、石油化工、机械、电子电器、建筑、纺织等诸多领域。

[0004] 前端聚合又被称为前沿聚合或者波聚合,是一种通过局部反应区域在聚合物单体中的移动而将聚合物单体转变为聚合物的一种新型自由基反应模式。前端聚合是对含有单体和引发剂的混合物进行局部加热,引发剂受热分解,引发单体发生聚合反应。而后撤离热源,聚合反应本身产生的热量向未反应区域扩散,继续引发聚合,从而产生反应热扩散偶合循环的狭窄反应区域,此区域称为聚合前端。聚合前端不断自发向前推进,直到容器内的单体聚合完全。

[0005] 由于前端聚合体系中所需的热量均来自于其本身,热扩散来自于对流传热,反应时间很短,因此它具有很好的节能环保等特性。大部分的前端聚合反应都是在无溶剂或少溶剂的条件下进行,这就意味着最终的产物无需从溶剂中分离,制备过程没有大量有机溶剂的使用和后处理,这就意味着保护环境,节省成本。前端聚合可以在高浓度单体和引发剂下反应而不会发生暴聚现象,反应快、效率高;相对于常规制备工艺,聚合前端的高温和快速移动能力能够有效地抑制相分离,得到具有特殊结构的聚合物。

[0006] 根据前端聚合反应的不同引发方式和反应机理,可以将前端聚合反应分为两类,即:热差前端聚合和等温前端聚合。随着研究的广泛深入,前端聚合家族又引进了光引发前端聚合。

[0007] 等离子体是由部分电子被剥夺后的原子及原子被电离后产生的正负电子组成的离子化气体状物质,广泛存在于宇宙中,被认为是除去固、液、气外物质存在的第四态。等离子技术在薄膜沉积、刻蚀、等离子引发聚合以及表面处理等方面具有广阔的应用前景。目前尚未有发现采用等离子体前端聚合制备智能水凝胶材料的报道。

#### 发明内容

[0008] 本发明的目的在于提供一种采用等离子体前端聚合快速制备含氟水凝胶的方法。该方法具有速度快,效率高,成本低,成型快等优点,避免了传统制备方法中工艺流程复杂、

时间长等缺点,在工业生产中具有较高的应用价值。

[0009] 本发明的技术方案为:采用等离子前端聚合快速制备含氟水凝胶的方法,其具体步骤如下:

[0010] (1) 将活性单体、含氟单体、引发剂和交联剂按比例溶于有机溶剂中,配成均匀溶液;

[0011] (2) 将配成均匀溶液混合移入反应器中,对反应器上端利用低温等离子体处理,使引发剂受热分解,引发聚合反应;

[0012] (3) 关闭等离子体反应器,依靠聚合反应放出的热向未反应区域扩散,直到整个反应器内所有原料全部转化为聚合物;

[0013] (4) 取出所得的聚合物,用去离子水漂洗,得到含氟水凝胶。

[0014] 上述步骤(1)配成均匀溶液中,优选活性单体占溶液总质量的20~40%;含氟单体占溶液总质量的20~40%;交联剂占溶液总质量的0.5~5%;引发剂所占质量比为0.25~4%;有机容剂占溶液总质量的15~45%。

[0015] 优选上述活性单体为丙烯酸羟乙酯、丙烯酸羟丙酯、甲基丙烯酸羟乙酯、甲基丙烯酸羟乙酯、甲基丙烯酸羟丙酯、N,N-二甲基丙烯酰胺、丙烯酰胺、双丙酮丙烯酰胺、N-羟甲基丙烯酰胺、N-异丙基丙烯酰胺、N-乙烯基吡咯烷酮、N-乙烯基己内酰胺、丙烯酸辛、丙烯酸或甲基丙烯酸中的至少一种。

[0016] 优选上述含氟单体为甲基丙烯酸氟烷烃酯、甲基丙烯酸全氟烷基乙基酯、甲基丙烯酸三氟乙酯、N-乙基全氟辛基磺酰胺基丙烯酸酸乙酯、全氟烷烃丙烯酸酯、甲基丙烯酸全氟辛基乙酯、丙烯酸六氟丁酯、甲基丙烯酸六氟丁酯、甲基丙烯酸十二氟庚酯、三氟甲基丙烯酸或丙烯酸全氟烷基乙酯中的一种。

[0017] 优选上述有机溶剂为:甘油、二甲亚砜、N, N'-二甲基甲酰胺或1-甲基-2-吡咯烷 顧中的一种。

[0018] 优选上述交联剂为 N, N' - 亚甲基双丙烯酰胺、二乙烯基苯或二甲基丙烯酸乙二醇 酯中的一种。

[0019] 优选上述引发剂为偶氮二异丁腈、偶氮二异庚腈、过氧化苯甲酰、过硫酸铵、过硫酸钾、叔丁基过氧化氢中的其中一种。

[0020] 优选低温等离子体处理电压为 30 ~ 80V;电流为 1 ~ 3A;处理时间为 5 ~ 60s。

[0021] 有着效果:

[0022] 1. 等离子前端聚合法与传统的聚合方法相比,所需的外供能量是短暂一次性的,聚合启动后即可停止供能,在聚合反应过程中不需要外加热源来维持整个体系的聚合,具有可控性,且聚合反应时间较短,节能,无污染。

[0023] 2. 等离子前端聚合体系中的热量产生于反应本身, 热扩散源于对流传热, 需要较少反应时间, 因此是一种极佳的节能反应模式, 在工业化生产中具有很高的应用价值。

#### 附图说明

[0024] 图 1 为等离子体前端聚合制备水凝胶示意图;其中①为等离子体装置,②为反应器, A 为单体混合物, B 为聚合物, C 为前端端面, D 为前端传播方向;

[0025] 图 2 为实施例 1 等离子体前端聚合制备水凝胶的前端位置与时间变化关系图:

[0026] 图 3 为为实施例 1 等离子体前端聚合制备智能水凝胶的温度曲线图。

### 具体实施方式

[0027] 以下通过具体实施例说明本发明,但本发明并不仅限定于这些实施例。

[0028] 实施例 [

[0029] 称取 60.0g 丙烯酸羟乙酯, 4.25g N, N'-亚甲基双丙烯酰胺, 100.0g 甲基丙烯酸氟烷烃酯溶于 135.0gN, N'-二甲基甲酰胺中配成均匀溶液, 然后加入 0.75g 过硫酸钾, 混合均匀后倒入玻璃反应器内, 调节等离子体装置电压为 30V 电流为 3A, 引发聚合反应, 其中等离子体处理时间为 5s。关闭等离子体反应装置, 借助放热反应的热自催化作用而完成单体聚合。用去离子水冲洗聚合物, 得到智能水凝胶。如图 2 和 3 所示, 从图中可以发现只存在纯粹的前端聚合, 而没有自发本体聚合的发生。对水凝胶红外分析, 1200cm "处为 C-F 的特征峰, 2450cm"处为 0-H 的红外特征峰, 说明得到的水凝胶为(丙烯酸羟乙酯-甲基丙烯酸氟烷烃酯) 共聚物。

[0030] 实施例 2

[0031] 称取 40.0g 丙烯酸羟丙酯,0.5g 二甲基丙烯酸乙二醇酯,20g 甲基丙烯酸全氟烷基乙基酯溶于 38.0g N,N'-二甲基甲酰胺中配成均匀溶液,然后加入 1.5g 过硫酸铵,混合均匀后倒入玻璃反应器内,调节等离子体装置电压为 80V 电流为 1A,引发聚合反应,其中等离子体处理时间为 50s。关闭等离子体反应装置,借助放热反应的热自催化作用而完成单体聚合。用去离子水冲洗聚合物,得到智能水凝胶。对水凝胶红外分析,1190cm¹处为 C+F 的特征峰,2445cm¹处为 0-H 的红外特征峰,说明得到的水凝胶为(丙烯酸羟丙酯-甲基丙烯酸全氟烷基乙基酯) 共聚物。

[0032] 实施例 3

[0033] 称取 20.0g N-乙烯基吡咯烷酮,1.5g二甲基丙烯酸乙二醇酯,19.0g 甲基丙烯酸 六氟丁酯溶于 7.5g 二甲亚砜中配成均匀溶液,然后加入 2.0g 过氧化苯甲酰,混合均匀后倒入玻璃反应器内,调节等离子体装置电压为 50V 电流为 3A,引发聚合反应,其中等离子体处理时间为 30s。关闭等离子体反应装置,借助放热反应的热自催化作用而完成单体聚合。用去离子水冲洗聚合物,得到智能水凝胶。对水凝胶红外分析,1185cm、处为 C-F 的特征峰,1440cm、处为内酰胺吸收峰,说明得到的水凝胶为 (N-乙,烯基吡咯烷酮-甲基丙烯酸六氟丁酯) 共聚物。

[0034] 实施例 4

[0035] 称取 25.0g 丙烯酸,4.0g 二乙烯基苯,24.0g 甲基丙烯酸全氟辛基乙酯溶于 26,0g 甘油中配成均匀溶液,然后加入 1.0g 偶氮二异庚腈,混合均匀后倒入玻璃反应器内,调节 等离子体装置电压为 30V 电流为 1A,引发聚合反应,其中等离子体处理时间为 60s。关闭 等离子体反应装置,借助放热反应的热自催化作用而完成单体聚合。用去离子水冲洗聚合物,得到智能水凝胶。对水凝胶热重分析,物质质量和温度的关系可以说明得到的水凝胶为 (丙烯酸-甲基丙烯酸全氟辛基乙酯) 共聚物。

[0036] 实施例 5

[0037] 称取 100, 0g N-异丙基丙烯酰胺, 3, 0g N, N'-亚甲基双丙烯酰胺, 75, 0g 甲基丙烯酸十二氟庚酯溶于 120, 0g1-甲基-2-吡咯烷酮中配成均匀溶液, 然后加入 2, 0g 偶氮二异

丁腈,混合均匀后倒入玻璃反应器内,调节等离子体装置电压为 40V 电流为 2.5A,引发聚合 反应,其中等离子体处理时间为 40s。关闭等离子体反应装置,借助放热反应的热自催化作用而完成单体聚合。用去离子水冲洗聚合物,得到智能水凝胶。对水凝胶热重分析,物质质量和温度的关系可以说明得到的水凝胶为 (N-异丙基丙烯酰胺-甲基丙烯酸十二氟庚酯) 共聚物。

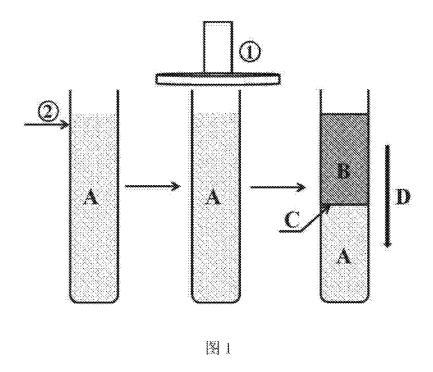
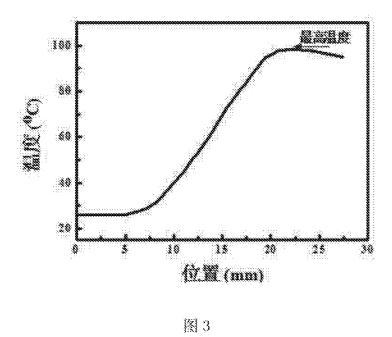


图 2



SciFinder® Page 1

#### Session Began April 30, 2019 at 06:06 PM

# Task Began April 30, 2019 06:06 PM

Explore references by research topic: plasma polymerization of fluoroacrylate and crosslinking monomer

Research Topic: plasma polymerization of fluoroacrylate and crosslinking monomer

Result Count: 9

Candidates Selected (ID 1)

4 references were found containing the two concepts "plasma polymerization"

and "fluoroacrylate".

Automatically removed 4 duplicate MEDLINE answer(s)

173 references were found containing the two concepts "plasma polymerization"

and "crosslinking monomer".

Automatically removed 4 duplicate MEDLINE answer(s)

50 references were found containing both of the concepts "fluoroacrylate" and

"crosslinking monomer".

Automatically removed 4 duplicate MEDLINE answer(s)

Answer Type: References

Result Count: 223

#### Prepared 4 reference answers from Answer set 1 for printing in PDF form

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NUMBER	371(c) DATE	UNIT	FIL FEE REC'D	ATTY.DOCKET.NO	TOT CLAIMS	IND CLAIMS
15/735 130	12/08/2017	1767	1620	GBI 0007/LIS	18	1

33072 KAGAN BINDER, PLLC SUITE 200, MAPLE ISLAND BUILDING 221 MAIN STREET NORTH STILLWATER, MN 55082 UNITED STATES OF AMERICA CONFIRMATION NO. 7101 CORRECTED FILING RECEIPT



Date Mailed: 07/02/2019

Receipt is acknowledged of this non-provisional utility patent application. The application will be taken up for examination in due course. Applicant will be notified as to the results of the examination. Any correspondence concerning the application must include the following identification information: the U.S. APPLICATION NUMBER, FILING DATE, NAME OF FIRST INVENTOR, and TITLE OF INVENTION. Fees transmitted by check or draft are subject to collection.

Please verify the accuracy of the data presented on this receipt. If an error is noted on this Filing Receipt, please submit a written request for a corrected Filing Receipt, including a properly marked-up ADS showing the changes with strike-through for deletions and underlining for additions. If you received a "Notice to File Missing Parts" or other Notice requiring a response for this application, please submit any request for correction to this Filing Receipt with your reply to the Notice. When the USPTO processes the reply to the Notice, the USPTO will generate another Filing Receipt incorporating the requested corrections provided that the request is grantable.

Inventor(s)

Stephen Richard COULSON, Abingdon, UNITED KINGDOM;

Delwyn EVANS, Abingdon, UNITED KINGDOM; Angeliki SIOKOU, Abingdon, UNITED KINGDOM;

Clive TELFORD, Abingdon, UNITED KINGDOM, Deceased;

Applicant(s)

P2i Ltd, Abingdon, UNITED KINGDOM

Power of Attorney: The patent practitioners associated with Customer Number 33072

Domestic Priority data as claimed by applicant

This application is a 371 of PCT/GB2016/051688 06/08/2016

**Foreign Applications** (You may be eligible to benefit from the **Patent Prosecution Highway** program at the USPTO. Please see <a href="http://www.uspto.gov">http://www.uspto.gov</a> for more information.) EUROPEAN PATENT OFFICE (EPO) 15386017.6 06/09/2015

Permission to Access Application via Priority Document Exchange: Yes

Permission to Access Search Results: Yes

Applicant may provide or rescind an authorization for access using Form PTO/SB/39 or Form PTO/SB/69 as appropriate.

If Required, Foreign Filing License Granted: 03/14/2018

The country code and number of your priority application, to be used for filing abroad under the Paris Convention,

is **US 15/735,130** 

Projected Publication Date: Not Applicable

Non-Publication Request: No Early Publication Request: No

Title

**COATINGS** 

**Preliminary Class** 

526

Statement under 37 CFR 1.55 or 1.78 for AIA (First Inventor to File) Transition Applications: No

#### PROTECTING YOUR INVENTION OUTSIDE THE UNITED STATES

Since the rights granted by a U.S. patent extend only throughout the territory of the United States and have no effect in a foreign country, an inventor who wishes patent protection in another country must apply for a patent in a specific country or in regional patent offices. Applicants may wish to consider the filing of an international application under the Patent Cooperation Treaty (PCT). An international (PCT) application generally has the same effect as a regular national patent application in each PCT-member country. The PCT process **simplifies** the filing of patent applications on the same invention in member countries, but **does not result** in a grant of "an international patent" and does not eliminate the need of applicants to file additional documents and fees in countries where patent protection is desired.

Almost every country has its own patent law, and a person desiring a patent in a particular country must make an application for patent in that country in accordance with its particular laws. Since the laws of many countries differ in various respects from the patent law of the United States, applicants are advised to seek guidance from specific foreign countries to ensure that patent rights are not lost prematurely.

Applicants also are advised that in the case of inventions made in the United States, the Director of the USPTO must issue a license before applicants can apply for a patent in a foreign country. The filing of a U.S. patent application serves as a request for a foreign filing license. The application's filing receipt contains further information and guidance as to the status of applicant's license for foreign filing.

Applicants may wish to consult the USPTO booklet, "General Information Concerning Patents" (specifically, the section entitled "Treaties and Foreign Patents") for more information on timeframes and deadlines for filing foreign patent applications. The guide is available either by contacting the USPTO Contact Center at 800-786-9199, or it can be viewed on the USPTO website at http://www.uspto.gov/web/offices/pac/doc/general/index.html.

For information on preventing theft of your intellectual property (patents, trademarks and copyrights), you may wish to consult the U.S. Government website, http://www.stopfakes.gov. Part of a Department of Commerce initiative, this website includes self-help "toolkits" giving innovators guidance on how to protect intellectual property in specific countries such as China, Korea and Mexico. For questions regarding patent enforcement issues, applicants may call the U.S. Government hotline at 1-866-999-HALT (1-866-999-4258).

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# Title 35, United States Code, Section 184

### Title 37, Code of Federal Regulations, 5.11 & 5.15

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The applicant has been granted a license under 35 U.S.C. 184, if the phrase "IF REQUIRED, FOREIGN FILING LICENSE GRANTED" followed by a date appears on this form. Such licenses are issued in all applications where the conditions for issuance of a license have been met, regardless of whether or not a license may be required as set forth in 37 CFR 5.15. The scope and limitations of this license are set forth in 37 CFR 5.15(a) unless an earlier license has been issued under 37 CFR 5.15(b). The license is subject to revocation upon written notification. The date indicated is the effective date of the license, unless an earlier license of similar scope has been granted under 37 CFR 5.13 or 5.14.

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#### **NOT GRANTED**

No license under 35 U.S.C. 184 has been granted at this time, if the phrase "IF REQUIRED, FOREIGN FILING LICENSE GRANTED" DOES NOT appear on this form. Applicant may still petition for a license under 37 CFR 5.12, if a license is desired before the expiration of 6 months from the filing date of the application. If 6 months has lapsed from the filing date of this application and the licensee has not received any indication of a secrecy order under 35 U.S.C. 181, the licensee may foreign file the application pursuant to 37 CFR 5.15(b).

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further correspondence	ncluding the Patent, adv	asce orders and notificatio	E and PUBLICATION FEI a of maintenance less will dence address; and/or (b) i	be mailed to the cur adjecting a separate	rent corresp : "FEE ADI	oondence address as i DRESS" for mainten	ndicated unless corrected ance fee notifications.
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APPLICATION NO.	FILING DATE		FIRST NAMED INVENTOR	[	ATTORNE	Y DOCKET NO.	CONFIRMATION NO.
15/735,130 TITLE OF INVENTION	12/08/2017 E COATINGS	: : : : : : : : : : : : : : : : : : : :	Stephen Richard COULSO	N.	GBI	L00077US	7101
APPLN: TYPE	ENTITY STATUS	ISSUE FEE DUE	PUBLICATION FEE DUE	PREV. PAID ISSU	e fee t	OTAL FEE(S) DUE	DATEDUE
Isnoisiverpen	UNDISCOUNTED	\$1000	\$0.00	\$0.00		\$1000	08/15/2019
EXA	ENER	ART UNIT	CLASS-SUBCLASS				
BUIE-HATCH	ER, NICOLE M	1767	526-245000				
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		A TO BE PRINTED ON	THE PATENT (print or ty)	); (*)	***************************************	4:4:::::::::::::::::::::::::::::::::::	
PLEASE NOTE: Unl recorded, or filed for	ess an assignee is identif recordation, as set forth	ied below, no assignee dat in 37 CFR 3.11 and 37 CR	a will appear on the patent R 3.81(a). Completion of	. If an assignee is it this form is NOT a	tentified be substitute	low, the document m for filing an assignm	ast have been previously ent.
(A) NAME OF ASSI	GNEE		(B) RESIDENCE: (CITY	and STATE OR C	OUNTRY	)	
P2i Ltd			Abingdon, U	nited Kingdo	m		
Please check the appropr	iate assigner category o	r categories (will not be p	rinted on the patent): 🛄 J	idividual 🏖 Corpo	ration or ot	her private group ent	ity 🔾 Government
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4b. Method of Payment:		previously paid fee show					
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Authorized Signature	The state of the s			Date Augus	t 12, 201	9	

Typed or primed name Paul I. Parins

Registration No. 54,358

Electronic Patent Application Fee Transmittal							
Application Number:	15735130						
Filing Date:	08-Dec-2017						
Title of Invention:	co	ATINGS					
First Named Inventor/Applicant Name:	Stephen Richard COULSON						
Filer:	Paul John Parins/Bethany Lichtscheidl						
Attorney Docket Number:	GBL0007/US						
Filed as Large Entity							
Filing Fees for U.S. National Stage under 35 USC 371							
Description		Fee Code	Quantity	Amount	Sub-Total in USD(\$)		
Basic Filing:							
Pages:							
Claims:							
Miscellaneous-Filing:							
Petition:							
Patent-Appeals-and-Interference:							
Post-Allowance-and-Post-Issuance:							
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Description	Fee Code	Quantity	Amount	Sub-Total in USD(\$)
Extension-of-Time:				
Miscellaneous:				
	Tot	al in USD	(\$)	1000

Electronic Acknowledgement Receipt					
EFS ID:	36847528				
Application Number:	15735130				
International Application Number:					
Confirmation Number:	7101				
Title of Invention:	COATINGS				
First Named Inventor/Applicant Name:	Stephen Richard COULSON				
Customer Number:	33072				
Filer:	Paul John Parins/Bethany Lichtscheidl				
Filer Authorized By:	Paul John Parins				
Attorney Docket Number:	GBL0007/US				
Receipt Date:	12-AUG-2019				
Filing Date:	08-DEC-2017				
Time Stamp:	15:05:07				
Application Type:	U.S. National Stage under 35 USC 371				
Payment information:					

# **Payment information:**

Submitted with Payment	yes
Payment Type	CARD
Payment was successfully received in RAM	\$1000
RAM confirmation Number	E20198BF05367563
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Document Number	Document Description	File Name	File Size(Bytes)/ Message Digest	Multi Part /.zip	Pages (if appl.)
			297603		
1	Issue Fee Payment (PTO-85B)	GBL0007US_Issue_Fee.pdf	80f2dbd1b1fb615e227e0e21f7a15084e49e c76c	no	1
Warnings:					
Information:		_			
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2	Fee Worksheet (SB06)	fee-info.pdf	4e40642dec0dcc86f1d6acaa6b3f005b4f13f 1fa	no	2
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Total Files Size (in bytes):

#### New Applications Under 35 U.S.C. 111

If a new application is being filed and the application includes the necessary components for a filing date (see 37 CFR 1.53(b)-(d) and MPEP 506), a Filing Receipt (37 CFR 1.54) will be issued in due course and the date shown on this Acknowledgement Receipt will establish the filing date of the application.

#### National Stage of an International Application under 35 U.S.C. 371

If a timely submission to enter the national stage of an international application is compliant with the conditions of 35 U.S.C. 371 and other applicable requirements a Form PCT/DO/EO/903 indicating acceptance of the application as a national stage submission under 35 U.S.C. 371 will be issued in addition to the Filing Receipt, in due course.

### New International Application Filed with the USPTO as a Receiving Office

If a new international application is being filed and the international application includes the necessary components for an international filing date (see PCT Article 11 and MPEP 1810), a Notification of the International Application Number and of the International Filing Date (Form PCT/RO/105) will be issued in due course, subject to prescriptions concerning national security, and the date shown on this Acknowledgement Receipt will establish the international filing date of the application.

327617

# United States Patent and Trademark Office



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APPLICATION NO.	ISSUE DATE	PATENT NO.	ATTORNEY DOCKET NO.	CONFIRMATION NO.
15/735,130	09/24/2019	10421876	GBL0007/US	7101

33072

7590

09/04/2019

KAGAN BINDER, PLLC SUITE 200, MAPLE ISLAND BUILDING 221 MAIN STREET NORTH STILLWATER, MN 55082

## ISSUE NOTIFICATION

The projected patent number and issue date are specified above.

# **Determination of Patent Term Adjustment under 35 U.S.C. 154 (b)**

(application filed on or after May 29, 2000)

The Patent Term Adjustment is 0 day(s). Any patent to issue from the above-identified application will include an indication of the adjustment on the front page.

If a Continued Prosecution Application (CPA) was filed in the above-identified application, the filing date that determines Patent Term Adjustment is the filing date of the most recent CPA.

Applicant will be able to obtain more detailed information by accessing the Patent Application Information Retrieval (PAIR) WEB site (http://pair.uspto.gov).

Any questions regarding the Patent Term Extension or Adjustment determination should be directed to the Office of Patent Legal Administration at (571)-272-7702. Questions relating to issue and publication fee payments should be directed to the Application Assistance Unit (AAU) of the Office of Data Management (ODM) at (571)-272-4200.

APPLICANT(s) (Please see PAIR WEB site http://pair.uspto.gov for additional applicants):

Stephen Richard COULSON, Abingdon, UNITED KINGDOM; P2i Ltd, Abingdon, UNITED KINGDOM Delwyn EVANS, Abingdon, UNITED KINGDOM; Angeliki SIOKOU, Abingdon, UNITED KINGDOM; Clive TELFORD, Abingdon, UNITED KINGDOM, Deceased;

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